



“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” – शिक्षणमहर्षी डॉ. बापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

PADMABHUSHAN DR. VASANTRAO DADA PATIL MAHAVIDYALAYA

TASGAON, Dist. Sangli, Pin- 416 312 ☎ - STD : 02346-250665, 250575 FAX : 250575

● **Affiliated to Shivaji University, Kolhapur** ●

E-mail : san.pdvpm.tas@gmail.com Website : www.pdvpmtasgaon.edu.in

Established Year : June 1962 ▶ P. B. No. : 14 ▶ Jr. College No. : J22-10-001 ▶ Sr. College Code No. : $\frac{SUAC4}{X}$ Jr.: C-8



NAAC Reaccredited 'B' (2.76)

ISO Certified : 9001:2015

**Shikshanmaharshi
Dr. Bapuji Salunkhe**
B.A., B.T., D.Litt.
FOUNDER

Hon. Chandrakant (Dada) Patil
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B.Com.
Ex- Minister of Revenue, Public Works
Govt. of Maharashtra

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M.Sc. B.Ed.
SECRETARY

Dr. Milind S. Hujare
M.Sc., Ph.D.
PRINCIPAL

Ref.No. : PDVPMT /

Date :

Criterion II

Teaching Learning and Evaluation

2.2.1

Learning levels of the students

POLICY ON ADVANCED AND SLOW LEARNERS



Accredited with Grade B++ By NAAC

**“Dissemination of Education through Knowledge, Science and Culture”-
Shikshanmaharshi Dr. Bapuji Salunkhe**

Shri Swami Vivekanand Shikshan Santha's Kolhapur

**PADMABHUSHAN DR. VASANTRAODADA PATIL
MAHAVIDYALAYA TASGAON DIST- SANGLI**

416 312 (Maharashtra) Phone No: (02346)250665

(Affiliated to Shivaji University, Kolhapur)

NAAC Reaccredited B⁺⁺ with CGPA 2.76 AISHE:C11096

College email Id :pdvptas.san@gmail.com

College Website :pdvpmtasgaon.edu.in

Internal Quality Assurance Cell (IQAC)

**POLICY ON ADVANCED AND
SLOW LEARNERS**

POLICY ON ADVANCED AND SLOW LEARNERS

This policy paper is recognized as the “Policy on Advanced and Slow Learners” of the Padmabhusan Dr Vasantodada Patil Mahavidyalaya, Tasgaon. Through this policy the Institute pronounces its assurance to the vital facilitation and prop up to the advanced learners to be excellent achievers and slow learners to be better performing and achieving students in the academic and personal life. The policy also persuades the teaching Departments to develop significant strategies and scientific implementations to benefit both the advanced learners and slow learners in the Institute education system without forgetting the average performers.

A) Introduction

The Padmabhusan Dr Vasantodada Patil Mahavidyalaya, Tasgaon is an academic institution for higher level of knowledge generation and dissemination through effective teaching programs and research. It is established in the 1962. The Institute is offering graduation, Post-Graduation and Research programs to the students from the rural as well as urban backgrounds. Majority of the students are from the rural background and also represents the middle or lower middle class social environment. The Institute is an aided co education Institute as per State Government rules and regulations and therefore caters to the needs of the students from the minority communities as well. The admissions to all the programs are through Shivaji University rules and regulations. The ultimate aim of the programs in the Institute is to make all students admitted in the Institute equipped and competent to face the challenges of the life and contribute to the social development through the critical reflection and academic achievements. This is possible through the process of

making the academics more exigent and competent for the advanced learners and also by supporting the slow learners to achieve more in their academics and personal life. In every academic program there will be some students who can do really well and learn more with the comprehension capacity, retention ability and hard working practices. On the other side some students may find disadvantage in their learning process due to various personal or systemic reasons. In both these situations the students need special attention and interventions to make their learning activity more enriching and effective.

B) The Concepts

B1.) Advanced Learners:

The term advanced learner in this policy refers to the students who can engage learning activities faster than the other students in the class and achieve high scores and make significant achievements in their life. They are more potential with their comprehension, retention, memory, critical thinking, creativity and contextualization practices. They also may have hard working behaviors and usually achieve more than the majority of the classmates. These students are in a great extend gifted and talented than the others in the class. These students can take up higher level learning and academic responsibilities. They can bring some new concepts, strategies, and also can take the leadership in the teaching learning activities.

B.2) Slow Learners:

The slow learners are always the poor achievers and lag behind with the academic life. They may fail in exams or will score only poor grades. They will find it difficult to understand the lessons and may have difficulties in their

comprehension, retention, reproduction and integration. They may fail in articulations and critical reflections. Their motivation levels also may be poor and may find it difficult to adjust with the teaching learning process and can drop out or fail in the program. The poor performance may not be a sign of the poor capacity or talent but may be due to inappropriate teaching methods, poor family situations, inadequate motivations and supports, unscientific learning practices or even the inability to converse in an unfamiliar language.

C) Methods of assessment:

The students can be assessed to identify their learning levels by different mechanisms.

1. The merit in the qualified examinations. Competitive entrance examination, and
2. The performance in the initial stage of the program including the bridge program and orientations assessed through the qualitative methods can be taken as indicators of their achievement as they get admitted to the program in the Institute **<45 marks**.

This level can be set as the benchmark of the student. Subsequently they can be assessed continuously through the examination results at various levels and stages, their performances in the extra - curricular and co - curricular activities throughout the program. The assessment procedure can be a statistical process of making the three levels like High, Average and Low on the basis of the central tendencies (mean plus and minus Standard deviations). The change of the students and the levels achievements can also be compared with the previous semester so that the advancements can be assessed and appropriate interventions can be made.

C.1) Policy Guidelines For Advanced Learners

1. Advanced learners are motivated to strive for higher goals. They are provided with additional inputs for better career planning and growth through offering special coaching for higher level competitive examinations
2. Motivating them to involve in research projects to inculcate research orientation and higher studies aspirations
3. Helping them to participate in group discussions, technical quizzes to develop analytical and problem solving abilities in them and thereby, to improve their presentation skills. Encouraging them to participate in National International Conferences and also to make presentations
4. Stirring the advanced learners to make quality publications and creative contributions to the academic as well as to the practical world
5. They are given representation in student council and various working committees.
6. Advanced learners are encouraged to undertake research projects under Institutional Research Promotion Scheme in which seed money of Rs. 5000/-nper project is provided under the guidance of teacher.
7. The mentors give higher goals and also make the advanced learners get higher levels of personality development and stress management trainings.
8. They are made the supporters to the average and the slow learners.

C.2) Policy Guidelines for Slow learners


1. The slow learners are not labeled as poor achievers or problem students in the class or department so that their motivation and their interest are not negatively affected. This can also make them more stressed, and introvert in the class.
2. They should be treated as any other student in the class but they can be provided extra classes for improvement and achievement.
3. The Department and individual teachers help the slow learners by giving proper guidance and support to them.
4. Organize bridge classes and remedial programs for them.
5. Conduct extra classes for the difficult subjects (based on the previous Institute results) in the curriculum.
6. Special attention is given to the students in the tutorial classes, who are identified as the slow learners.
7. Slow learners are specially advised and counseled by a teacher guardian and the subject expert.
8. Corrective classes are conducted for the weaker students based on the results of class tests.
9. The students are given with training on communication skills, personality development, time management and motivational sessions.
10. Design special coaching sessions or tutorial sessions to bridge the gap between the slow learners and advanced learners are provided.
11. Academic and personal counseling are given to the slow learners by the tutor, mentor and the counseling cell.
12. Bilingual explanation and discussions are imparted to the slow learners after the class hours for better understanding.
13. Provision of simple and standard lecture notes/course materials and special preparation for the exams will be good.
14. Getting the support of the advanced learners to the slow learners in making their

learning process more participatory and interesting.

15. Peer education strategies are effectively used
16. Encouraging the group learning activities and practical will be useful to the slow learners.
17. The support of the alumni is also effectively used to motivation and mentoring to the slow learners.

The process of giving additional inputs to the advanced learners and the slow learners is not an effort to make the students average achievers but to make very one better achievers. Even if the students are identified as slow learners or advanced learners in the class they should not be labeled as the extra ordinary or poor but they need to be treated equally in the class with supportive care and appropriate pedagogical systems so that the talented can make more achievement and the less talented also make their goals in life achieved. The mentoring and facilitating efforts of the teachers and the Institute administrative system will be a great investment in achieving the best result and performance of the students.




Prin. Dr. Milind S. Hujare
Principal
Padmabhushan Dr. Vasantroodada Patil
Mahavidyalaya, Targaon (Sangli).



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Dr. Bapuji Salunkhe**
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Criterion II

Teaching Learning and Evaluation

2.2.1

Learning levels of the students



Bridge Course



Seminars



Class Tests



Remedial Coaching



Reports of Activities Conducted for


○ Slow and

○ Advanced Learners


Shri Swami Vivekanand Shiksha Samiti's
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon, Dist-Sangli,
Time-Table (Senior Science) 2020-2021

B. Sc. I
Time-Table for Online Bridge course
(With effect from 28th October 2020)

Period	Time	Div.	Inauguration	Day 1	Day 2	Day 3	Day 4	Day 5	Day 6	Day 7	Day 8	Day 9	Day 10
			28 th Oct 20	29 th Oct 20	31 st Oct 20	2 nd Nov 20	3 rd Nov 20	4 th Nov 20	5 th Nov 20	6 th Nov 20	7 th Nov 20	9 th Nov 20	10 th Nov 20
			WED	THURS	SAT	MON	TUES	WED	THURS	FRI	SAT	MON	TUES
1	11.00 to 11.40	A	Address by Principal	Phy	Stats	Eng	Eng	Phy	Stats	Phy	Phy	Eng	Eng
		B		Chem	Bot	Eng	Eng	Zoo	Bot	Bot	Chem	Eng	Eng
2	12.00 to 12.40	A	Maths	Chem/ C. Sc.	Maths	Maths	Stats	Chem/ C. Sc.	Maths	Chem/ C. Sc.	Chem/ C. Sc.	Maths	Stats
		B	Chem	Phy	Zoo	Chem	Bot	Phy	Zoo	Phy	Phy	Chem	Zoo


(Dr. S. D. Jadhav)
Incharge,
Time-Table Committee.




(Dr. Milind S. Hujare)
Principal
Padmabhushan Dr. Vasantrodada Patil
Mahavidyalaya, Tasgaon, (Sangli)



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**Shri Swami Vivekanand Shikshan
Sanstha, Kolhapur's**



**PADMABHUSHAN DR. VASANTRAODADA PATIL
MAHAVIDYALAYA, TASGAON**

Tal. Tasgaon, Dist.: Sangli.

DEPARTMENT OF MATHEMATICS

A Report On

Bridge course

Submitted To

Internal Quality Assurance Cell

Year 2020-21


Department of Mathematics

B.Sc-I (2019-20)

Notice

Date:01/08/2019

All the students of B.Sc part I are here by inform that our Department of Mathematics is going to start a 'Bridge course' for incoming students of B.Sc. Part-I.As far as syllabus is concerned there is academic gap between scope of syllabus of XII Std and B.Sc. Course.

Head 
HEAD
Department of Mathematics
P. D. V. P. Mahavidyalaya
TASGAON, (Sangli)

Department of Mathematics

B.Sc-I (2019-20)

Bridge Course

Syllabus

- Intersection of sets- Union of sets - Complement of Sets
- Difference of two sets - Theorems and solved problems
- Trigonometric formulae of $\sin \theta$, $\cos \theta$, $\tan \theta$, $\sin 2\theta$, $\cos 2\theta$, $\tan 2\theta$, $\sin 3\theta$, $\cos 3\theta$, $\tan 3\theta$.
- Matrix-Types of matrix, determinant of matrix, order of matrix.
- Probability-Introduction to probability, Algebra set, Formulae, Random variable, Probability distribution.
- Derivative & Integration-Derivative Formulae, Integration Formulae, Examples.
- Limit & Continuity-Introduction, Definition, Examples.



HEAD
Department of MATHS
P. D. V. P. Mahavidyalaya,
TASGAON. (Sangli)

Department of Mathematics
B.Sc-I (2019-20)
Bridge Course
Time Table

Date 01/08/2019

I hereby informed that we are starting bridge course in Mathematics for the students of B.Sc.Part-I. Come regularly as schedule below.

Sr.No	Name of Faculty	Topic	Time	Day	Date
1	Mr.P.V.Patil	Intersection of sets	12:50pm	Monday	05/08/2019
		Difference of two sets	12:00pm	Tuesday	06/08/2019
		Limit & Continuity	12:00pm	Wednesday	07/08/2019
2	Miss.P.D.Jirage	Matrix	12:50pm	Friday	09/08/2019
		Matrix	12:50pm	Saturday	10/08/2019
3	Mr.B.H.Tarange	Probability	12:50pm	Monday	12/08/2019
4	Miss.S.A.Mane	Derivative & Integration	12:00pm	Wednesday	13/08/2019
5	Miss.B.B.Patil	Trigonometry	12:50pm	Friday	14/08/2019


HEAD
Department OF MATHS
Department of Mathematics
P.D. V.P. Patil
TASGAON. (Sangli)

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Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon
Student Attendance Sheet

Class -

Teacher Name -

Subject -

Sr. No	Name of the Student	Roll No	Signature of the Student and Date					
			5/8/2019	6/8/2019	7/8/2019	9/8/2019	10/8/2019	12/8/2019
27	Shinde Sakshi .S	5084	S.S.Shinde	S.S.Shinde	S.S.Shinde	S.S.Shinde	S.S.Shinde	S.S.Shinde
28	Shinde Surjit .S	5086	S.S.S.	S.S.S.	S.S.S.	S.S.S.	S.S.S.	S.S.S.
29	Wagh Varshali .A	5093	V.Wagh	V.Wagh	V.Wagh	V.Wagh	V.Wagh	V.Wagh
30	Chavan Gauri .K	5096	G.Chavan	G.Chavan	G.Chavan	G.Chavan	G.Chavan	G.Chavan
31	Desai Nikhil .S	5097	N.S.Desai	N.S.Desai	N.S.Desai	N.S.Desai	N.S.Desai	N.S.Desai
32	Gujar Kranta .A	5098	K.Gujar	K.Gujar	K.Gujar	K.Gujar	K.Gujar	K.Gujar
33	Thorat Parth .S	5090	P.S.T	P.S.T	P.S.T	P.S.T	P.S.T	P.S.T
34	Tadrey Pallavi .S	5101	Jadray	Jadray	Jadray	Jadray	Jadray	Jadray
35	Mali Sanket .V	5106	S.V.Mali	S.V.Mali	S.V.Mali	S.V.Mali	S.V.Mali	S.V.Mali
36	Mate Priti .A	5109	P.A.M	P.A.M	P.A.M	P.A.M	P.A.M	P.A.M
37	Patil Akash .S	5112	A.Patil	A.Patil	A.Patil	A.Patil	A.Patil	A.Patil
38	Patil Kamal .A	5113	K.P.Patil	K.P.Patil	K.P.Patil	K.P.Patil	K.P.Patil	K.P.Patil
39	Patil Pooja .T	5114	P.Patil	P.Patil	P.Patil	P.Patil	P.Patil	P.Patil
40	Patil Rohan .N	5116	R.Patil	R.Patil	R.Patil	R.Patil	R.Patil	R.Patil
41	Patil Prajanta .R	5100	P.Patil	P.Patil	P.Patil	P.Patil	P.Patil	P.Patil
42	Patil Shreya .S	5118	S.Patil	S.Patil	S.Patil	S.Patil	S.Patil	S.Patil
43	Salunkhe Artika .S	5122	A.S	A.S	A.S	A.S	A.S	A.S
44	Sawant Rutuja .S	5124	R.S	R.S	R.S	R.S	R.S	R.S
45	Mane Shubham .N	5107	S.M	S.M	S.M	S.M	S.M	S.M
46	Hankate Vikram .S	5151	V.H	V.H	V.H	V.H	V.H	V.H
47	Rajmane Suyog .S	5121	S.M	S.M	S.M	S.M	S.M	S.M
48	Mithare Rishabh .S	5108	M.M	M.M	M.M	M.M	M.M	M.M
49	Kharode Saumik .A	5104	S.A.K	S.A.K	S.A.K	S.A.K	S.A.K	S.A.K
50	Kochti Kaitiki .U	5105	V.H	V.H	V.H	V.H	V.H	V.H

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Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon

Student Attendance Sheet

Class -

Teacher Name -

Subject -

Sr. No	Name of the Student	Roll No	Signature of the Student and Date					
			13/8/2019	14/8/2019				
1	Bhosale Sonali S.	5002	Bhosale	Bhosale	-	-	-	-
2	Chavan Manoj R.	5004	Chavan	Chavan	-	-	-	-
3	Chavan Vaidhavi V.	5006	V.V.Chavan	V.V.Chavan	-	-	-	-
4	Desai Vijay V.	5008	Desai	Desai	-	-	-	-
5	Gajkwad Rohan D.	5010	R.D.G.	R.D.G.	-	-	-	-
6	God Subhangi M.	5013	God	God	-	-	-	-
7	Hajare Smita M.	5015	Hajare S.	Hajare S.	-	-	-	-
8	Ingle PRAKASH B.	5016	P.B.Ingle	P.B.Ingle	-	-	-	-
9	Jadhav Rutuja D.	5020	R.D.J.	R.D.J.	-	-	-	-
10	Jadhav Vishal J.	5023	V.Jadhav	V.Jadhav	-	-	-	-
11	Kadam Snehal M.	5025	S.M.K	S.M.K	-	-	-	-
12	Kamble Sakshi M.	5027	Skamble	Skamble	-	-	-	-
13	Kumbhar Rajyoti P.	5030	P.Kumbhar	P.Kumbhar	-	-	-	-
14	Lokhande Rohit M.	5031	R.M.L	R.M.L	-	-	-	-
15	More Arjun R.	5038	A.R.More	A.R.More	-	-	-	-
16	Nalavde Nita M.	5041	N.M.N	N.M.N	-	-	-	-
17	Patil Apurva P.	5044	A.Patil	A.Patil	-	-	-	-
18	Patil Jyoti R.	5047	J.P.Patil	J.P.Patil	-	-	-	-
19	Patil Nikhil V.	5048	N.Patil	N.Patil	-	-	-	-
20	Pawar Kajal V.	5063	K.Pawar	K.Pawar	-	-	-	-
21	Pisad Nivedita S.	5070	N.S.P	N.S.P	-	-	-	-
22	Pol Shubham S.	5071	S.S.Pol	S.S.Pol	-	-	-	-
23	Salunkhe Divya S.	5072	D.S.S	D.S.S.	-	-	-	-
24	Shendage Mital V.	5076	M.Shendage	M.Shendage	-	-	-	-
25	Shinde Gautam V.	5080	G.V.S	G.V.S	-	-	-	-
26	Shinde Pritam G.	5083	P.Shinde	P.Shinde	-	-	-	-

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Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tazgaon

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Class -

Teacher Name -

Subject -

Sr. No	Name of the Student	Roll No	Signature of the Student and Date					
			13/12/2019	14/12/2019				
27	Shinde Sakshi.S	5084	S.S.Shinde	S.S.Shinde	-	-	-	-
28	Shinde Sumit.S	5086	S.S.S.	S.S.S.	-	-	-	-
29	Wagh Varshali.A	5093	V.Wagh	V.Wagh	-	-	-	-
30	Chavhan Gauri.K	5096	G.Chavhan	G.Chavhan	-	-	-	-
31	Desai Nikhil.S	5097	N.S.Desai	N.S.Desai	-	-	-	-
32	Gujar Karan.A	5098	K.Gujar	K.Gujar	-	-	-	-
33	Thorat Prith.S	5090	P.S.T.	P.S.T.	-	-	-	-
34	Jadhav Pallavi.S	5101	Jadhav	Jadhav	-	-	-	-
35	Mali Sanket.V	5106	S.V.Mali	S.V.Mali	-	-	-	-
36	More Priti.A	5109	P.A.M	P.A.M	-	-	-	-
37	Patil Akash.S	5112	A.Patil	A.Patil	-	-	-	-
38	Patil Karan.R	5113	K.P.Patil	K.P.Patil	-	-	-	-
39	Patil Pooja.T	5114	P.Patil	P.Patil	-	-	-	-
40	Patil Rohan.N	5116	R.Patil	R.Patil	-	-	-	-
41	Hake Prajanta.R	5100	P.Hake	P.Hake	-	-	-	-
42	Patil Shreya.S	5118	S.Patil	S.Patil	-	-	-	-
43	Salunkhe Rutik.V	5122	R.V.S	R.V.S	-	-	-	-
44	Sawant Rutuja.S	5124	SRS	SRS	-	-	-	-
45	Mane Subhram.N	5107	S.M.N	S.M.N	-	-	-	-
46	Hankare Vikram.G	5151	V.G.H	V.G.H	-	-	-	-
47	Rajmane Suyog.M	5121	S.M.R	S.M.R	-	-	-	-
48	Mithare Rinesh.A	5108	M.Mithare	M.Mithare	-	-	-	-
49	Khatode Santosh.A	5104	S.A.K	S.A.K	-	-	-	-
50	Kashti Kartiki.V	5105	V.K.A.K	V.K.A.K	-	-	-	-

Name of the Student

20

Roll number

DEPARTMENT OF MATHEMATICS

BASIC MATHEMATICS MCQ TEST

Signature of
Teacher

B.Sc.Part - I

Day and Date Friday 23/08/2019

Time 12.00

Marks 20

Instruction: Use other papers for rough work.

Question Select correct alternative and tick mark for proper alternative.

20

1. If $z = 1+i$ then using De Moivre's theorem $z^4 =$ -----

- a) -4 b) 4 c) 4i d) none of the above

2. An integer ----- is the example of prime number.

- a) 2 b) 3 c) 5 d) All of them

3. Find the real numbers between complex number $3 + 5i$.

- a) 3, 5 b) -3, -5 c) i d) 3

4. If $\sin\theta = \frac{\sqrt{3}}{2}$ then $\theta =$ -----.

- a) $\frac{\pi}{3}$ b) $\frac{3\pi}{2}$ c) $\frac{\pi}{2}$ d) $\frac{\pi}{4}$

5. The expansion of $\cos x$ is -----

a) $1 + x + \frac{x^2}{2!} + \frac{x^3}{3!} + \dots$ b) $x - \frac{x^3}{3!} + \frac{x^5}{5!} - \dots$

c) $1 - x + \frac{x^2}{2!} - \frac{x^3}{3!} + \dots$ d) $1 - \frac{x^2}{2!} + \frac{x^4}{4!} - \dots$

6. The standard equation of ellipse is -----.

a) $x^2 + y^2 = a^2$ b) $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$

c) $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ d) $y^2 = 4ax$

7. If two dice are thrown then the number of sample spaces are

- a) 6 b) 12 c) 36 d) 4

8. If $A = \{1, 2, 3, 4, 5\}$ and $B = \{7, 6, 8, 5, 4, 6, 3\}$ then $A \cap B =$ -----

- a) $\{1, 2, 3, 4\}$ b) $\{3, 4, 5\}$ c) $\{6, 8, 5\}$ d) $\{1, 2, 3, 4, 5, 6, 7, 8\}$

9. If $x^2 + 5x + 6 = 0$ then solution of this equation is
 a) 3,2 b) -3,-2, c) 5,6 d) -5,-6
10. If A is any $m \times n$ matrix, AB and BA are both defined then B is an ----- matrix.
 a) $m \times n$ b) $n \times m$ c) $m \times m$ d) $n \times n$
11. $\sum_1^3 n^3$ is equal to -----
 a) 30 b) 12 c) 48 d) 36
12. $\int \cos x =$ -----
 a) $\sin x + C$ b) $-\sin x + C$ c) $\tan x + C$ d) $\cot x + C$
13. $1 + 2x + 3x^2 + \dots$ is equal to -
 a) $(1-x)^{-1}$ b) $(1+x)^{-1}$ c) $(1+x)^{-2}$ d) $(1-x)^{-2}$
14. The value of 8P is -----.
 a) 76 b) 56 c) 86 d) 66
15. The value of mC is --- a) m b) c c) 1 d) $m - 1$
16. $(\cos \theta + i \sin \theta)^n =$ -----
 a) $\cos n \theta + \sin n \theta$ b) $\sin n \theta + i \cos n \theta$
 c) $\cos n \theta + i \sin n \theta$ d) $\cos n \theta - i \sin n \theta$
17. $\lim_{x \rightarrow 0} \frac{\sin cx}{x} = ?$ a) 1 b) $\frac{1}{c}$ c) 0 d) c
18. Evaluate the determinant $\begin{vmatrix} 5 & -3 \\ 6 & 1 \end{vmatrix}$
 a) 23 b) -13 c) -23 d) -8
19. Solve the equations $x + y = 5$ and $x - y = 1$ then values of x and y are
 a) 3,2 b) 4,1 c) 9,8 d) unsolvable equations
20. Find the derivative of x^5 with respect to x.
 a) x^5 b) $4x^5$ c) $5x^4$ d) $\frac{x^6}{6}$

No.	Roll no.		Name of the Student	Marks
1	5001	SHRI	BHASME SHREYAS SANJAY	AB
2	5002	MISS	BHOSALE SAYALI SHANKAR	16
3	5003	SHRI	CHAVAN ANURAG RAJESH	AB
4	5004	MISS	CHAVAN MONAL RAMESH	11
5	5005	MISS	CHAVAN NIKITA SAMHAJI	14
6	5006	SHRI	CHAVAN VAIBHAV VIJAY	AB
7	5007	MISS	DESAI ANJALI DINKAR	11
8	5008	SHRI	DESAI VIJAY VASANT	AB
9	5009	SHRI	GAIKWAD PRAMOD TANAJI	12
10	5010	SHRI	GAIKWAD ROHAN DATTATRAY	AB
11	5011	MISS	GEJAGE KOMAL KRISHNADEV	14
12	5012	MISS	GHADAGE PRIYANKA CHANDRAKANT	13
13	5013	MISS	GOD SHUBHANGI MAHADEV	AB
	5014		GURAV ANIKET DEVIDAS	AB
14	5015	MISS	HAJARE SMITA MAHALING	14
15	5016	MISS	INGOLE PRAJAKTA BAPU	15
16	5017	SHRI	JADHAV ANIRUDDHA LALASO	10
17	5018	SHRI	JADHAV DASHARATH GANPATI	17
18	5019	MISS	JADHAV PRANALI ANIL	12
19	5020	MISS	JADHAV RUTUJA DHANAJI	14
20	5021	MISS	JADHAV SNEHA SHASHIKANT	10
21	5022	MISS	JADHAV SNEHAL SHAHAJI	AB
22	5023	SHRI	JADHAV VISHAL JAGANNATH	AB
23	5024	SHRI	JAMDADE SANGRAM GUNDERAO	AB
24	5025	MISS	KADAM SNEHAL MOHAN	14
25	5026	MISS	KALBAGE ASHWETA JAYWANT	AB
26	5027	MISS	KAMBLE SAKSHI MILIND	18
27	5028	SHRI	KAMBLE SHIMON BHARAT	AB
28	5029	SHRI	KONDAKE KUNDAN MAHADEV	9
29	5030	MISS	KUMBHAR PRAJYOTI PRALHAD	12
30	5031	SHRI	LOKHANDE ROHIT MAHADEV	6
31	5032	MISS	MALI BHAGYASHRI MOHAN	12
32	5033	SHRI	MANE AVISHKAR LALASAHEB	

No.	Roll no.		Name of the Student	Marks
33	5034	MISS	MANE GEETANJALI PIRAJI	11
34	5035	MISS	MANE MAYURI MAHADEV	14
35	5036	SHRI	MANE SWAPNIL CHANDRASHEKHAR	16
36	5037	MISS	MOHITE MAYURI ABASO	10
37	5038	SHRI	MOHITE SHUBHAM PANDURANG	AB
38	5039	SHRI	MORE ARJUN RANGRAO	AB
39	5040	MISS	MORE PRATIKSHA RAVASAHEB	16
40	5041	MISS	NALAVADE NITA MAHADEV	11
41	5042	SHRI	NIMBALKAR PRITHVIRAJ ASHOK	AB
42	5043	SHRI	PATIL ADITYA HANAMANT	AB
43	5044	MISS	PATIL APURVA PRAKASH	10
44	5045	SHRI	PATIL HANMANT DNYANESHWAR	13
45	5046	SHRI	PATIL JAYSHANKAR HANMANT	11
46	5047	MISS	PATIL JYOTI RAVASAHEB	14
47	5048	SHRI	PATIL NIKHIL VIJAY	13
48	5049	MISS	PATIL NISHA BABASO	11
49	5050	SHRI	PATIL PANDURANG UTTAM	AB
50	5051	SHRI	PATIL PRADIP PANDURANG	AB
51	5052	SHRI	PATIL PRATIK CHANKRAKANT	AB
52	5053	SHRI	PATIL ROHAN MANIK	11
53	5054	SHRI	PATIL ROHAN TUKARAM	AB
54	5055	SHRI	PATIL ROHIT RAJENDRA	AB
55	5056	SHRI	PATIL RUSHIKESH BALASO	AB
56	5057	MISS	PATIL RUTUJA RAMESH	14
57	5058	SHRI	PATIL RUTVIK RAMESH	AB
58	5059	MISS	PATIL SAKSHI PANDIT	14
59	5060	SHRI	PATIL SHUBHAM PRAKASH	AB
60	5061	MISS	PATIL SHUBHANGI SHIVAJI	14
61	5062	SHRI	PATIL SUYASH UTTAMRAO	11
62	5063	MISS	PAWAR KAJAL VISHNU	11
63	5064	MISS	PAWAR MEGHA PARASHRAM	9
64	5065	SHRI	PAWAR PRATIK NANDKUMAR	AB
65	5066	SHRI	PAWAR RAJESH SHIVAJI	17

No.	Roll no.		Name of the Student	Marks
66	5067	SHRI	PAWAR SUDIP ARJUN	AB
67	5068	SHRI	PAWAR VAIBHAV PRAFULL	AB
68	5069	MISS	PHADTARE GAYATRI MAHADEV	15
69	5070	MISS	PISAL NIVEDITA SHANKAR	14
70	5071	SHRI	POL SHUBHAM SUNIL	7
71	5072	MISS	SALUNKHE DIVYA SHAHAJI	14
72	5073	SHRI	SALUNKHE VAIBHAV TUKARAM	AB
73	5074	SHRI	SARAVADE ROSHAN CHANDRAKANT	14
74	5075	MISS	SHENDAGE AKSHADA JAGANANTH	13
75	5076	MISS	SHENDAGE MINAL UTTAM	13
76	5077	MISS	SHENDAGE SARIKA TANAJI	15
77	5078	MISS	SHIKALGAR MOHABBAT MOULA	14
78	5079	MISS	SHINDE DHANASHRI VILAS	14
79	5080	SHRI	SHINDE GAUTAM VIJAY	AB
80	5081	SHRI	SHINDE JOTIRAM BABURAO	AB
81	5082	MISS	SHINDE PRANALI JALINDAR	11
82	5083	SHRI	SHINDE PRITAM GORAKHNATH	AB
83	5084	MISS	SHINDE SAKSHI SANTOSH	AB
84	5085	SHRI	SHINDE SOURABH GANESH	AB
85	5086	SHRI	SHINDE SUMIT SAMBHAJI	AB
86	5087	MISS	SHIPURE SARIKA SHARANAPPA	AB
87	5088	SHRI	SURYAWANSHI SOURABH MANIK	AB
88	5089	SHRI	THOMBARE PRASHANT RAMESH	13
89	5090	SHRI	THORAT PARTH SARJERAO	16
90	5091	MISS	THORAVAT SHWETA RAJESH	12
91	5092	SHRI	THORWAT SHUBHAM SADASHIV	10
92	5093	MISS	WAGH VARSHALI ARUN	13
93	5094	MISS	BHANDURGE PRADNYA PRATAP	10
94	5095	MISS	BHATIYA SUKHMANKAUR JARNAILSINGH	9
95	5096	MISS	CHAVAN GOURI KIRAN	13
96	5097	SHRI	DESAI NIKHIL SANJAY	AB
97	5098	MISS	GUJAR KANTA ANANT	18

No.	Roll no.		Name of the Student	Marks
98	5099	SHRI	HAJARE ABHISHEK MAHADEV	AB
99	5100	MISS	HAKA PRAJAKTA BABURAO	AB
100	5101	MISS	JADHAV PALLAVI SADASHIV	14
101	5102	MISS	JADHAV SHRUTIKA MAHADEV	14
102	5103	MISS	KADAM HARSHADA PANDHARINATH	9
103	5104	SHRI	KHARADE SAURABH AJIT	AB
104	5105	MISS	KOSHTI KARTIKI UDAY	AB
105	5106	SHRI	MALI SANKET VILAS	AB
106	5107	SHRI	MANE SHUBHAM NANASAHEB	AB
107	5108	SHRI	MITHARE RINESH RAJENDRA	11
108	5109	MISS	MORE PRITI ARUN	AB
109	5110	SHRI	PARDESHI GAUTAM AVINASH	AB
110	5111	SHRI	PATIL ABHISHEK ANKUSH	12
111	5112	SHRI	PATIL AKASH SAMPAT	12
112	5113	MISS	PATIL KOMAL RAMESH	10
113	5114	MISS	PATIL POOJA TANAJI	AB
114	5115	MISS	PATIL PRAJAKTA BHARAT	11
115	5116	SHRI	PATIL ROHAN NAMDEV	9
116	5117	SHRI	PATIL SAURABH TANAJI	AB
117	5118	MISS	PATIL SHREYA SARJERAO	10
118	5119	SHRI	PATIL SHRIDHAR GAJANAN	AB
119	5120	MISS	PATIL VAISHNAVI NANDKUMAR	11
120	5121	SHRI	RAJMANE SUYOG MOHAN	12
121	5122	SHRI	SALUNKHE RUTIK VINAYAK	8
122	5123	MISS	SALUNKHE SHIVANI SHIVAJI	AB
123	5124	MISS	SAWANT RUTUJA SANDESH	AB
124	5125	MISS	VADGAVE SAMRUDDHI SANDEEP	18
125	5151	SHRI	HANKARE VIKRAM GUNDA	AB


HEAD
 Department OF MATHS
 P. D. V. P. Mahavidyalaya
 TASGAON. (Sangli)



“Dissemination of Education through Knowledge, Science and Culture” -Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha's Kolhapur

PADMABHUSHAN DR. VASANTRAODADA PATIL
MAHAVIDYALAYA TASGAON DIST- SANGLI
416 312 (Maharashtra) Phone No: (02346)250665
(Affiliated to Shivaji University, Kolhapur)

DEPARTMENT OF PHYSICS

REPORT ON

BRIDGE COURSE

2020-2021

“Dissemination of Education through Knowledge, Science and Culture”

-Shikshanmaharshi Dr. BapujiSalunkhe

Shri Swami Vivekanand Shikshan Sanstha's

Padmabhushan Dr. Vasandraodada Patil Mahavidyalaya, Tasgaon (Dist.Sangli)

DEPARTMENT OF PHYSICS

Bridge course

2020-2021

Duration of the Bridge course:

1 week-

From 28//10/2020

Syllabus of the Bridge course:

Unit I

Introduction of bachelor degree course in Physics-CBCS program, Structure of course, Course and program outcome, Explanation of syllabus related practicals.

Unit II

CBCS and Examination pattern, Newton's law of gravitation, Keplers law, Periodic motion, Simple harmonic motion, Defination and explanation of energy, Defination of surface tension and its unit, dimensions, applications.

Unit III

Inroduction about various branches of physics, opportunities after B. Sc., M. Sc. in physics, Employment areas and career opportunities in physics graduates, Inroduction about various apparatus in dark room and general laboratory, Meaning of mechanics, Rotational and angular velocity

Unit IV

Nature of practical examination, Marks distribution, differential equations and its types, linear differential equations, homogenous equations and nonhomogenous differential equations, solutions of differential equations

Unit V

Introduction about scope of syllabus, Explanation of nature of question paper, passing scheme, General Introduction related to elasticity

Unit VI

General discussion on mechanics, Basics of motion-types of motion, Types of force, Newtons law of motion, Basics of frame of reference and its concept.

Dr. Ghodake J. S

"Dissemination of Education for Knowledge, Science and Culture"
Shikshamharashi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha Kolhapur

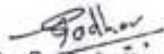
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon

Notice

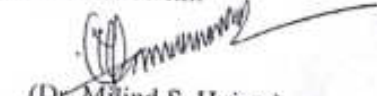
B. Sc. I Bridge Course Schedule

All science faculty members are hereby informed that B. Sc. I online classes will be started from 28th October 2020 as a bridge course.

- ❖ On 28th October 2020 at 11.00 am the bridge course will be inaugurated by the address of Prin. Dr. M. S. Hujare.
- Whatsapp groups are created for Division A and Division B. Teachers should communicate their links to the students on these groups, before scheduled time.
- Faculty members should use their mobile phones or laptops and engage lectures using Zoom app / Google meet.
- The faculty members should maintain the record of online bridge course.
- Collect the students' attendance using Google form.
- All faculty members should engage their lectures in the allotted classrooms as following:
 - > B. Sc. I Division A : Mathematics Department
 - > B. Sc. I Division B : Physics Lecture Hall
- The detailed schedule of bridge course is attached herewith.


(Dr. S. D. Jadhav)

Incharge,
Time-Table Committee


(Dr. Milind S. Hujare)
Principal

Shri Swami Vivekanand Shikshan Sanstha's
 Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon. Dist-Sangli.
Time-Table (Senior Science) 2020-2021

B. Sc. I
Time-Table for Online Bridge course
 (With effect from 28th October 2020)

Period	Time	Div.	Inauguration										
			28 th Oct 20	29 th Oct 20	Day 1	Day 2	Day 3	Day 4	Day 5	Day 6	Day 7	Day 8	Day 9
1	11.00 to 11.40	A	WED	THURS	MON	TUES	WED	THURS	FRI	SAT	MON	TUES	Eng
		B	Address by Principal	Phy & Math	Eng	Eng	Phy & Zoo	Stats	Bot	Stats	Phy & Chem	Eng	
2	12.00 to 12.40	A		Chem	Chem	Maths	Chem / C. Sc.	Chem / C. Sc.	Chem / C. Sc.	Phy & Bot	Chem / C. Sc.	Phy & Chem	Eng
		B	Chem	Chem / C. Sc.	Phy & Bot	Zoo	Maths	Stats	Bot	Maths	Chem / C. Sc.	Phy & Chem	

Sachin
 (Dr. S. D. Jadhav)
 Incharge,
 Time-Table Committee

Milind S. Hujare
 (Dr. Milind S. Hujare)

“Dissemination of Education through Knowledge, Science and Culture”

-Shikshanmaharshi Dr. BapujiSalunkhe

Shri Swami Vivekanand Shikshan Sanstha's

Padmabhushan Dr. Vasanturadada Patil Mahavidyalaya, Tasgaon (Dist.Sangli)

DEPARTMENT OF PHYSICS

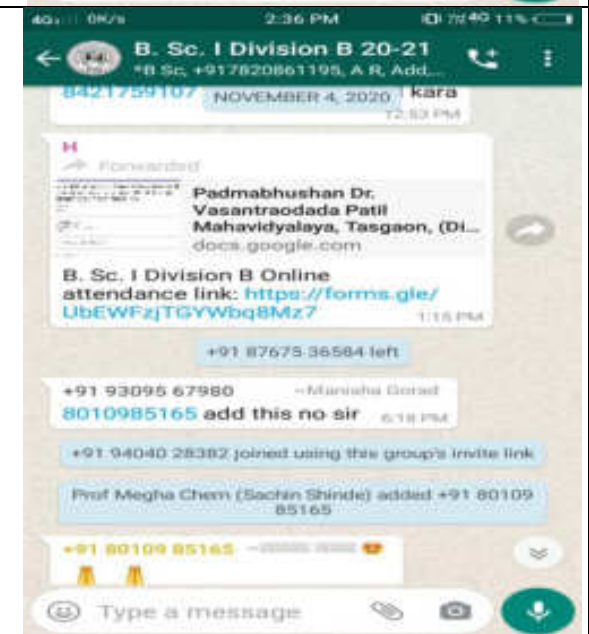
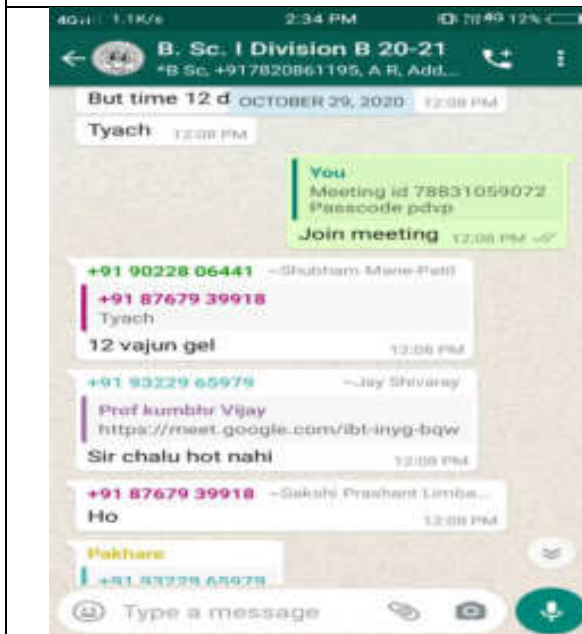
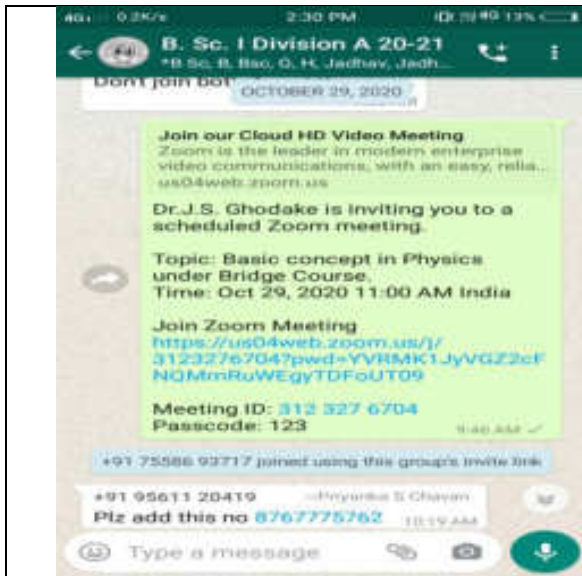
Bridge course (2020-2021)

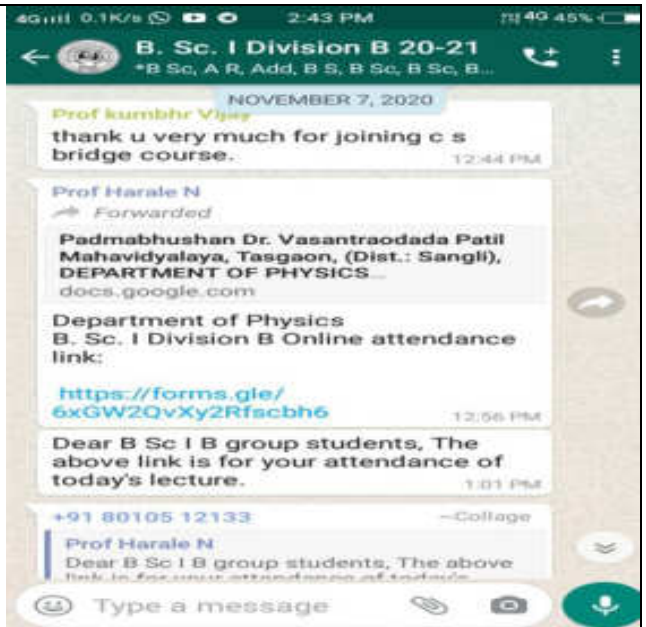
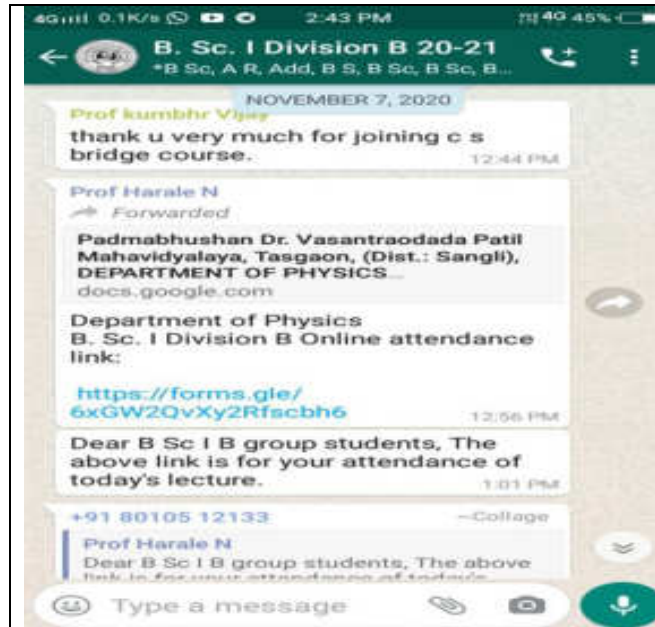
Time table

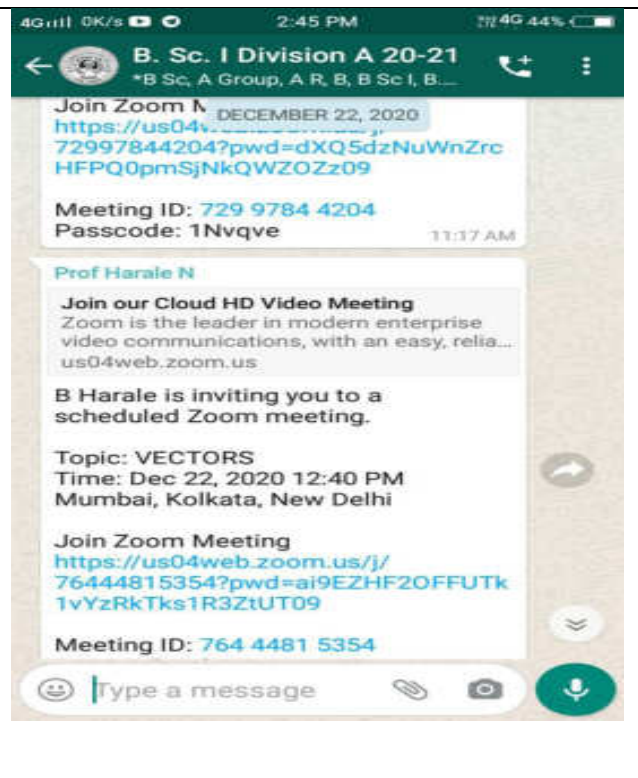
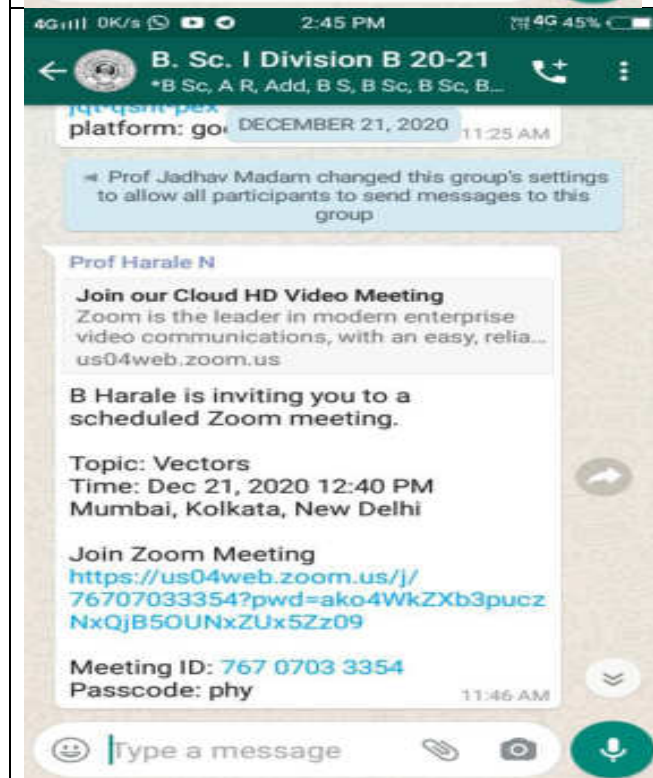
Time	Thursday 29/10/2020	Wednesday 04/11/2020	Friday 06/11/2020	Saturday 07/11/2020
11.00am- 11.40am	I(BSH)	I(JSG)	I(BSH)	I(JSG)
12.00pm- 12.40pm	I(JSG)	I(BSH)	I(JSG)	I(BSH)

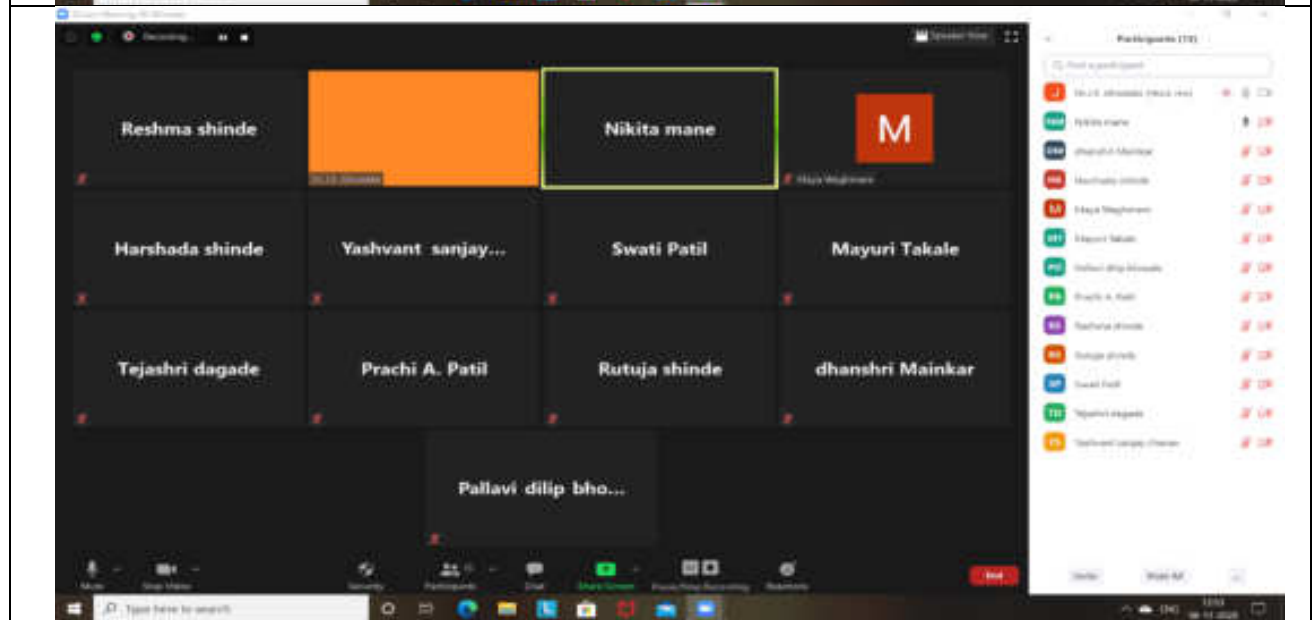
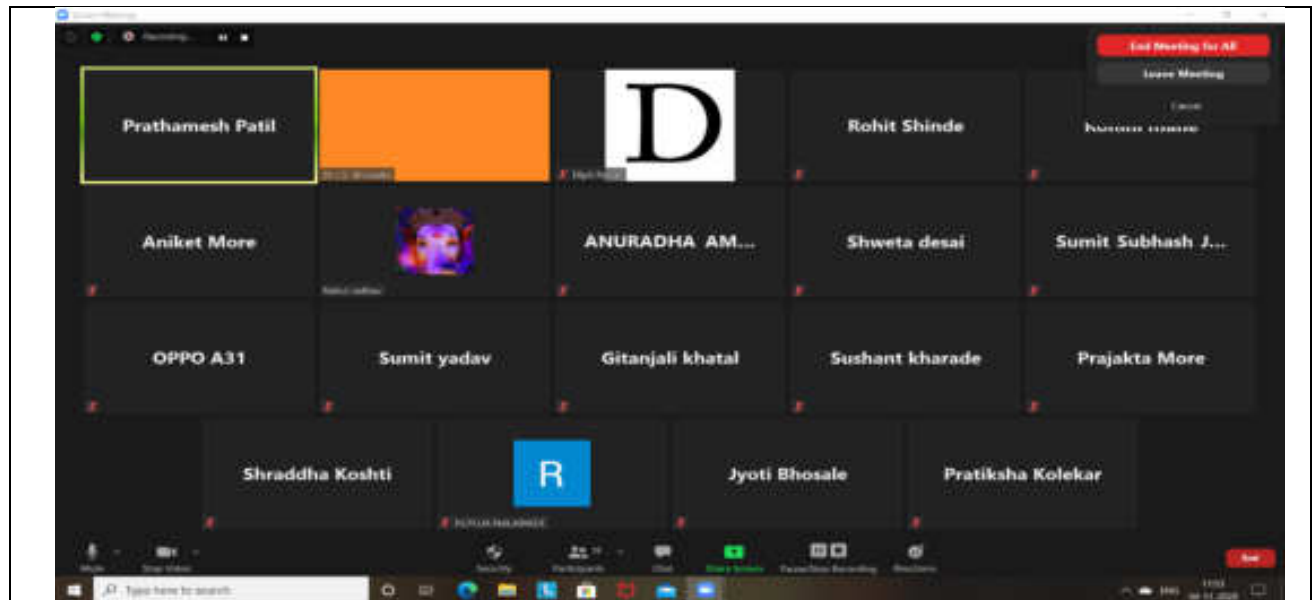
Time	Monday 14/12/2020	Tuesday 15/12/2020	Friday 18/12/2020	Saturday 19/12/2020
11.00am- 11.40am				I(JSG)
12.40pm- 01.20am	I(BSH)	I(BSH)	I(JSG)	

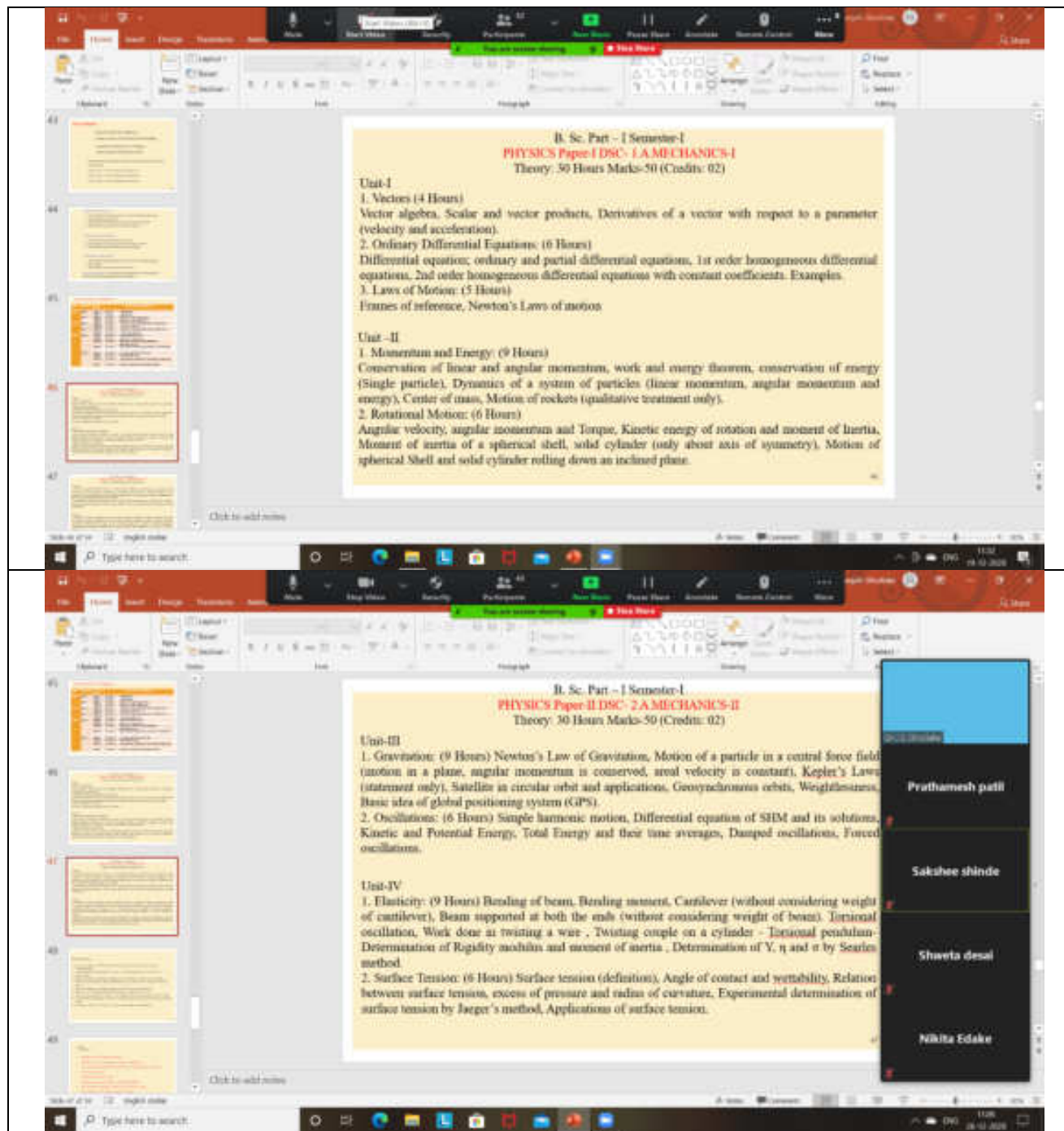
Time	Monday 21/12/2020	Tuesday 22/12/2020	Friday 25/12/2020	Saturday 26/12/2020
11.00am- 11.40am				I(JSG)
12.40pm- 1.200am	I(BSH)	I(BSH)	--	











Students Attendance

Timestamp	Name of Student	Attendance
11/4/2020 12:14:05	Trupti Pawar	Present
11/4/2020 12:14:26	Gitanjali khatal	Present
11/4/2020 12:14:37	kajal mane	Present
11/4/2020 12:14:54	Prashant Gurav	Present
11/4/2020 12:15:06	Nainavheleebirade	Present
11/4/2020 12:15:33	Jyotee Vijaykumar Patil	Present
11/4/2020 12:15:41	Sayalishankarpatil	Present
11/4/2020 12:16:10	Nilofar shahajan shaikh	Present
11/4/2020 12:17:12	Rohit shinde	Present
11/4/2020 12:18:32	Nikhil popatyadav	Present
11/4/2020 12:19:49	PrathameshUttam Patil	Present

11/4/2020 12:23:07	Shital Subhash Jadhav	Present
11/4/2020 12:23:17	Vijay More	Present
11/4/2020 12:26:54	Prajakta more	Present
11/4/2020 12:30:06	Anuradha Amrut Pawar	Present
11/4/2020 12:30:59	Sakshi prakashshete	Present
11/4/2020 12:31:07	Suvarna Shankar Shelke	Present
11/4/2020 12:32:45	Sushant Nivas Patil	Present
11/4/2020 12:33:04	Shweta desai	Present
11/4/2020 12:34:38	Tejas Sanjay Yadav	Present
11/4/2020 12:35:39	Pallavi Narayan Mali	Present
11/4/2020 12:35:42	Neha Jadhav	Present
11/4/2020 12:38:20	Riteshsasane	Present
11/4/2020 12:40:27	Duryodhan Jadhav	Present
11/4/2020 12:41:52	Tushar Shendage	Present
11/4/2020 12:43:12	Harshad Jadhav	Present
11/4/2020 12:43:12	Sushant adsul	Present
11/4/2020 12:43:26	Aishwarya kadam	Present
11/4/2020 12:43:55	Rutuja Vilas Nalawade	Present
11/4/2020 12:44:08	AnujaLalaso Jadhav	Present
11/4/2020 12:44:18	SnehalDinkar More	Present
11/4/2020 12:44:20	Sushil kamble	Present
11/4/2020 12:44:27	Shraddha Bajirao Koshti	Present
11/4/2020 12:44:39	HarshadaHanmantpatil	Present
11/4/2020 12:44:44	Abhishek Sutar	Present
11/4/2020 12:45:12	Prajakta Prakash More	Present
11/4/2020 12:46:25	Sanika Bhaskar Jadhav	Present
11/4/2020 12:51:02	PradnyaDadasoChandanshive	Present
11/4/2020 12:51:23	Prajakta Shinde	Present
11/4/2020 12:52:19	PratikshaDinkarKolekar	Present
11/4/2020 12:56:03	Manasvi Ashok Patil	Present
11/4/2020 13:03:36	Hrutuja Rajendra Patil	Present
11/4/2020 13:05:54	Rohini Rajendra Mane	Present
11/4/2020 13:12:55	Ranjit Desai	Present
11/4/2020 14:03:07	Abhijeet hanmantpatil	Present
11/4/2020 14:04:22	Anant hingmire	Present
11/4/2020 14:20:26	Nishant marutibhosale	Present
11/4/2020 14:32:06	Aishwarya Sanjay Jayappa	Present
11/4/2020 14:52:24	Sushant nivaspatil.	Present

Timestamp	Name of Student	Attendance
11/4/2020 13:14:46	B S Harale	Present
11/4/2020 13:16:23	Sakshi BalasoGhorapade	Present
11/4/2020 13:16:42	Nikita Sanjay Patil	Present
11/4/2020 13:16:47	SuryawanshiHarekrushnaDnyandeo	Present
11/4/2020 13:16:56	Omkar shinde	Present
11/4/2020 13:18:38	Vijay Dnyaneshvrsutar	Present
11/4/2020 13:21:35	Shivani pravinjamadade	Present
11/4/2020 13:22:39	Pratikshashinde	Present
11/4/2020 13:23:24	Pratik vinayakpetkar	Present
11/4/2020 13:23:40	Snehal. Jadhav	Present
11/4/2020 13:32:10	Swati Kisan Patil	Present
11/4/2020 13:44:28	YashrajNandkumar Shinde	Present
11/4/2020 14:15:08	TasnimRiyaj Sayyad	Present
11/4/2020 14:19:03	Sohelnaykawadi	Present
11/4/2020 14:36:52	Sahil Ramchandra patole	Present
11/4/2020 14:41:18	Ratnadip Sanjay jadhav	Present
11/4/2020 14:48:16	TejashriAdhikaraoDagade	Present
11/4/2020 15:09:03	Mayuri suniltakale	Present

11/4/2020 15:50:51	Edake Pooja Tukaram	Present
11/4/2020 16:09:17	Nikita sharad mane	Present
11/4/2020 18:32:35	Shubham Namdev Mane	Present
11/4/2020 19:03:10	Maya Ashok Waghmare	Present
11/4/2020 20:57:47	Anjali More	Present
11/4/2020 21:29:05	Rohan balasahebpatil	Present
11/4/2020 21:39:14	HarekushnaDnyandevSuryavanshi	Present
11/5/2020 8:41:11	Anuradha Ramesh Patil	Present
11/5/2020 11:45:28	vaishnavisanjaygavandi	Present
11/5/2020 14:12:11	Shubham Netaji Mane	Present
11/5/2020 14:46:06	SunayanaJaved Attar	Present
11/5/2020 16:43:35	Deepak Gosavi.	Present
11/6/2020 18:09:54	Pooja bhosale	Present

Timestamp	Name of Student	Attendance
11/6/2020 12:00:18	Sakshi prakash shete	Present
11/6/2020 12:00:22	Tejas Sanjay Yadav	Present
11/6/2020 12:00:35	Rohini Rajendra Mane	Present
11/6/2020 12:00:38	Aishwarya Sanjay Jayappa	Present
11/6/2020 12:00:59	Dipti Pawar	Present
11/6/2020 12:01:04	Prathmesh shinde	Present
11/6/2020 12:01:11	Jeevan jadhav	Present
11/6/2020 12:01:14	Sayali shankar patil	Present
11/6/2020 12:01:43	Prajakta Prakash More	Present
11/6/2020 12:02:09	Anuradha Amrut Pawar	Present
11/6/2020 12:02:38	Shraddha Bajirao Koshti	Present
11/6/2020 12:02:56	Nishant maruti bhosale	Present
11/6/2020 12:03:24	Trupti Pawar	Present
11/6/2020 12:03:40	Prajakta Madhukar Jadhav	Present
11/6/2020 12:07:02	Jadhav	Present
11/6/2020 12:07:12	Nainavhelee birade	Present
11/6/2020 12:07:32	Prashant Gurav	Present
11/6/2020 12:08:04	Prajakta Shinde	Present
11/6/2020 12:08:57	Soumaiya satish hate	Present
11/6/2020 12:09:52	Ankita ghodake	Present
11/6/2020 12:09:55	Shital Subhash Jadhav	Present
11/6/2020 12:11:36	Akshay patil	Present
11/6/2020 12:12:36	Akshay Dasharath Pawar	Present
11/6/2020 12:14:32	Rutuja Vilas Nalawade	Present
11/6/2020 12:22:09	Sumit yadav	Present
11/6/2020 12:24:36	Manasvi Ashok Patil	Present
11/6/2020 12:29:20	kajal mane	Present
11/6/2020 12:34:01	Hrutuja Rajendra Patil	Present
11/6/2020 12:34:19	Shirish mane	Present
11/6/2020 12:34:46	Sushil Suhas kamble	Present
11/6/2020 12:34:49	Snehal Dinkar More	Present
11/6/2020 12:35:03	Gitanjali khatal	Present
11/6/2020 12:35:07	Suvarna Shankar Shelke	Present
11/6/2020 12:35:27	Rohit shinde	Present
11/6/2020 12:35:54	Neha Jadhav	Present
11/6/2020 12:36:38	Aishwarya Nandkumar kadam	Present

11/6/2020 12:37:00	Pallavi Narayan Mali	Present
11/6/2020 12:37:59	Punam Chandrakant Pail	Present
11/6/2020 12:38:02	Nilofar shahajan shaikh	Present
11/6/2020 12:38:05	Anuja Lalaso Jadhav	Present
11/6/2020 12:38:27	Prajakta more	Present
11/6/2020 12:39:19	Pratiksha Dinkar Kolekar	Present
11/6/2020 12:42:12	Kajal baburao rasale	Present
11/6/2020 12:44:23	Abhijeet hanmant patil	Present
11/6/2020 12:47:54	Nikhil popat yadav	Present
11/6/2020 12:48:54	Jyotee Vijaykumar Patil	Present
11/6/2020 13:01:43	Pradnya Dadaso Chandanshive	Present

Timestamp	Name of Student	Attendance
11/7/2020 12:57:31	Pooja vijaykumar pol	Present
11/7/2020 12:57:31	Omkar shinde	Present
11/7/2020 12:57:40	Prachi Avadhut Patil	Present
11/7/2020 12:57:40	Reshma shinde	Present
11/7/2020 12:57:46	Patil Nikita Pradip	Present
11/7/2020 12:57:49	Trupti Lokhande	Present
11/7/2020 12:57:51	Harekrushna Dnyandev suryawanshi	Present
11/7/2020 12:58:03	Madhura manik jamdade	Present
11/7/2020 12:59:03	Sahil Ramchandra patole	Present
11/7/2020 12:59:20	Pallavi dilip bhosale	Present
11/7/2020 12:59:23	Dhanshri mainkar	Present
11/7/2020 13:00:12	Vaishnavi Sunil Gaikwad	Present
11/7/2020 13:02:05	Nikita sharad mane	Present
11/7/2020 13:02:12	Pratiksha Rajendra shinde	Present
11/7/2020 13:03:02	Maya Ashok Waghmare	Present
11/7/2020 13:04:12	harshada shinde	Present
11/7/2020 13:04:29	Tejashri Adhikarao Dagade	Present
11/7/2020 13:23:11	Anjali Jadhav	Present
11/7/2020 13:27:05	Prajakta Mahadev Patil	Present
11/7/2020 13:28:48	Pratik Vinayak Petkar	Present
11/7/2020 13:35:45	Kunal todkar	Present
11/7/2020 13:40:48	Rutuja dilip shinde	Present
11/7/2020 13:43:46	Pooja bhosale	Present
11/7/2020 14:06:48	Vishal Jadhav	Present
11/7/2020 14:08:14	Tasnim Riyaj Sayyad	Present
11/7/2020 14:34:47	Nikita Sanjay Patil	Present
11/7/2020 14:40:48	Edake Pooja Tukaram	Present
11/7/2020 15:16:25	Prathamesh chavan	Present
11/7/2020 16:01:13	Swati Kisan Patil	Present
11/7/2020 17:46:29	Harsh Dilip Kamble	Present

11/7/2020 18:32:43	Deepak Manik Gosavi.	Present
11/7/2020 18:45:34	Shubham Namdev Mane	Present


Timestamp	Name of Student	Attendance
12/18/2020 13:35:24	Tejashri adhikarao dagade	Present
12/18/2020 13:35:54	Nikita Pradip Patil	Present
12/18/2020 13:36:12	Dhanshree Dattatray Mainkar	Present
12/18/2020 13:36:27	harshada shinde	Present
12/18/2020 13:37:36	Harshvardhan sandip mane	Present
12/18/2020 13:39:35	Nikita Sanjay Patil	Present
12/18/2020 13:42:32	Omakar shinde	Present
12/18/2020 13:42:42	Pooja Edake	Present
12/18/2020 13:46:09	Pallavi dilip bhosale	Present
12/18/2020 13:46:12	Arpita vishwas more	Present
12/18/2020 13:54:54	Deepak Manik Gosavi.	Present
12/18/2020 13:57:12	Yash Ghatage	Present
12/18/2020 14:07:29	Vaishnavi Sunil Gaikwad	Present
12/18/2020 14:07:45	Patil pranali Sanjay	Present
12/18/2020 14:07:51	Sonali patil	Present
12/18/2020 14:08:10	Nikita sharad mane	Present
12/18/2020 14:09:01	Prachi Avadhut Patil	Present
12/18/2020 14:11:47	Mayuri takale	Present
12/18/2020 14:14:55	Edake rutuja jagannath	Present
12/18/2020 14:35:51	Pratiksha Rajendra shinde	Present
12/18/2020 15:01:33	Gitanjali shidu wagh	Present
12/18/2020 15:32:27	Harekrushna Dnyandev suryavanshi	Present
12/18/2020 18:45:22	Shubham Namdev Mane	Present
12/18/2020 20:07:19	SAHIL RAMCHANDRA PATOLE	Present
12/19/2020 11:08:13	Sohel naykawadi	Present
12/19/2020 11:50:04	Chavan Prathamesh pandurang	Present

Timestamp	Name of Student	Attendance
12/19/2020 12:23:15	Jyotee Vijaykumar Patil	Present
12/19/2020 12:23:31	Harshad jadhav	Present
12/19/2020 12:23:34	Sanika Patil	Present
12/19/2020 12:24:20	Sumit Subhash Jadhav	Present
12/19/2020 12:24:20	Trupti Pawar	Present
12/19/2020 12:24:22	Nainavhelee birade	Present
12/19/2020 12:24:43	Aniket more	Present
12/19/2020 12:27:05	Sumit yadav	Present

12/19/2020 12:27:36	sanket mule	Present
12/19/2020 12:32:43	Bhosale Jyoti Laxman	Present
12/19/2020 12:33:05	Rohini Rajendra Mane	Present
12/19/2020 12:33:07	Sakshi prakash shete	Present
12/19/2020 12:33:10	Hrutuja Rajendra Patil	Present
12/19/2020 12:33:10	Prathamesh balaso	Present
12/19/2020 12:33:11	Sayali shankar patil	Present
12/19/2020 12:33:15	Prajakta more	Present
12/19/2020 12:33:15	Jyoti pawar	Present
12/19/2020 12:33:17	Shweta Babaso salunkhe	Present
12/19/2020 12:33:19	Nishant Maruti Bhosale	Present
12/19/2020 12:33:19	Tushar Suresh shendage	Present
12/19/2020 12:33:20	Rutuja Vilas Nalawade	Present
12/19/2020 12:33:22	Aishwarya Sanjay Jayappa	Present
12/19/2020 12:33:26	Kajal Ankush mane	Present
12/19/2020 12:33:31	Nikhil popat Yadav	Present
12/19/2020 12:33:35	Rutuja Mohan Mane	Present
12/19/2020 12:33:42	Sushant Nivas Patil	Present
12/19/2020 12:33:45	Prajakta Prakash More	Present
12/19/2020 12:33:49	Kajal baburao rasale	Present
12/19/2020 12:33:50	Pawar anuradha amrut	Present
12/19/2020 12:34:22	Anuja lalaso jadhav	Present
12/19/2020 12:34:27	Shital Subhash Jadhav	Present
12/19/2020 12:34:30	Shraddha Bajirao Koshti	Present
12/19/2020 12:35:01	Ankita Anandrao Ghodake	Present
12/19/2020 12:35:05	Ritesh balaso sasane	Present
12/19/2020 12:35:55	Pradnya Dadaso Chandanshive	Present
12/19/2020 12:36:39	Tejas Sanjay Yadav	Present
12/19/2020 12:39:25	Megha mane	Present
12/19/2020 12:40:50	Swati ravindra waghmare	Present
12/19/2020 12:43:29	Neha Jadhav	Present
12/19/2020 12:47:57	Aishwarya Nandkumar kadam	Present
12/19/2020 12:48:50	Ashitosh rakshe	Present
12/19/2020 13:00:32	Vijay More	Present
12/19/2020 13:20:33	Samruddhi Rajaram Shinde	Present
12/19/2020 13:26:57	Snehal Dinkar More	Present
12/19/2020 13:27:35	Yadav Tejas Durvas	Present
12/19/2020 13:31:54	Manasvi Ashok Patil	Present
12/19/2020 13:34:28	Pallavi Narayan Mali	Present
12/19/2020 13:58:57	Sushant namdev adsul	Present
12/19/2020 14:30:08	Nilofar shahajan shaikh	Present
12/20/2020 17:04:59	Vishal Hanmant Jadhav	Present

Timestamp	Score	Name of Student	Attendance
12/26/2020 12:44:09		Nikita Edake	Present
12/26/2020 12:44:21		Sakshi prakash shete	Present
12/26/2020 12:44:29		Neha Jadhav	Present
12/26/2020 12:44:29		Rohit Shinde	Present
12/26/2020 12:44:32		Patil prajakta vijay	Present
12/26/2020 12:44:34		Swati waghmare	Present
12/26/2020 12:44:39		Aishwarya Sanjay Jayappa	Present
12/26/2020 12:44:43		Shweta Babaso salunkhe	Present
12/26/2020 12:44:45		Prashant Gurav	Present
12/26/2020 12:45:15		Tushar Suresh shendage	Present
12/26/2020 12:45:20		Ghodake J S	Present
12/26/2020 12:45:31		Trupti Pawar	Present
12/26/2020 12:45:46		Aniket more	Present
12/26/2020 12:45:54		Dipti Pawar	Present
12/26/2020 12:46:47		Tejas Sanjay Yadav	Present
12/26/2020 12:47:09		Kajal Ankush mane	Present
12/26/2020 12:47:12		Punam Chandrakant Patil	Present
12/26/2020 12:47:57		Aishwarya Nandkumar kadam	Present
12/26/2020 12:48:20		Rohit balaso patil	Present
12/26/2020 12:48:48		Shraddha Bajirao Koshti	Present
12/26/2020 12:49:07		Jeevan jalindar jadhav	Present
12/26/2020 12:49:32		Sayali shankar patil	Present
12/26/2020 12:50:21		Kajal baburao rasale	Present
12/26/2020 12:51:31		Nainavhelee birade	Present
12/26/2020 12:52:33		Prajakta Shinde	Present
12/26/2020 12:53:02		Duryodhan Ravasaheb jadhav	Present
12/26/2020 12:53:21		Shirish mane	Present
12/26/2020 12:56:49		Ritesh balaso sasane	Present
12/26/2020 12:57:09		Ankita Anandrao Ghodake	Present
12/26/2020 13:00:24		Sanika Patil	Present
12/26/2020 13:02:18		Shweta desai	Present
12/26/2020 13:02:55		Nishant maruti bhosale	Present
12/26/2020 13:05:22		Harshad jadhav	Present
12/26/2020 13:05:41		Sushant Nivas Patil	Present
12/26/2020 13:11:23		Shital Subhash Jadhav	Present
12/26/2020 13:11:32		Rohini Rajendra Mane	Present
12/26/2020 13:11:34		Hrutuja Rajendra Patil	Present
12/26/2020 13:11:53		Pratiksha Tanaji Patil	Present
12/26/2020 13:11:56		Anuja lalaso Jadhav	Present
12/26/2020 13:12:30		Sakshee shinde	Present
12/26/2020 13:12:37		Pawar anuradha amrut	Present
12/26/2020 13:12:40		Manasvi Ashok Patil	Present

12/26/2020 13:12:44	Nikita Nitin Patil	Present
12/26/2020 13:13:00	Rutuja Mohan Mane	Present
12/26/2020 13:13:27	Pradnya Dadaso Chandanshive	Present
12/26/2020 13:15:41	Prathamesh balaso patil	Present
12/26/2020 13:16:32	Samruddhi Rajaram Shinde	Present
12/26/2020 13:16:39	Rutuja Vilas Nalawade	Present
12/26/2020 13:20:44	Suvarna Shankar Shelke	Present
12/26/2020 13:23:13	Karan krushnarao ghotkar	Present
12/26/2020 13:24:31	Prajakta Prakash More	Present
12/26/2020 13:25:45	Jyoti pawar	Present
12/26/2020 13:26:43	Jyotee Vijaykumar Patil	Present
12/26/2020 13:47:57	Akshay Patil	Present
12/26/2020 13:59:53	Abhijeet patil	Present
12/26/2020 14:16:24	Vaishnavi balaso mane	Present
12/26/2020 14:18:21	Snehal Dinkar More	Present
12/26/2020 14:36:00	Soumaiya satish hate	Present
12/26/2020 14:49:17	Nilofar shahajan Shaikh	Present
12/26/2020 14:54:53	Rutuja Utale	Present
12/26/2020 15:30:11	Prathamesh swami	Present
12/26/2020 15:56:16	Vijay More	Present
12/26/2020 17:58:29	AKSHAY	Present
12/26/2020 18:17:34	Megha mane	Present


Dr. Ghodake J. S.
Head,
Department of Physics,
Padmabhushan Dr. Vasantodada Patil
Mahavidyalaya, Tasgaon, (Dist. Sangli)

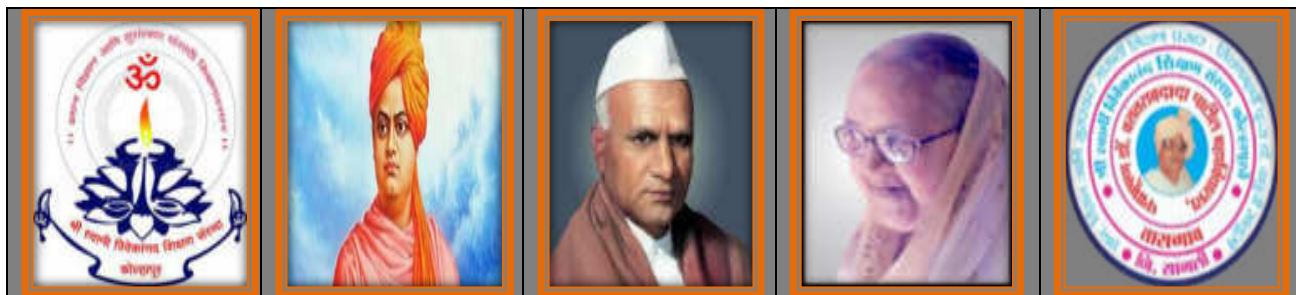
**PADMABHUSHAN DR. VASANTRAODADA
PATILMAHAVIDYALAYA, TASGAON DIST – SANGLI**

Department of Chemistry

BRIDGE COURSE REPORT

BRIDGE COURSE





"ज्ञान, विज्ञान आणि सुसंस्कार यासाठी शिक्षणप्रसार" - शिक्षणमहर्षी डॉ. बापूजी साळुंखे

Shri Swami Vivekanand Shiksha Sanstha Kolhapur

PADMABHUSHAN DR. VASANTRAODADA PATIL

MAHAVIDYALAYA, TASGAON DIST – SANGLI

(Affiliated to Shivaji University, Kolhapur)

Department of Chemistry

BRIDGE COURSE REPORT

**B.Sc. First Year
Chemistry**

BRIDGE COURSE



2020-21

BRIDGE COURSE

Every year our college conducted a bridge course for the newly admitted students of all subjects before the commencement of the first semester of university. The main objective of the course is to bridge up the gap between subjects they would be studying in 12th Standard and subjects they studied at university level. Bridge course syllabus is framed in such a way that equal importance is given to both the levels. The course duration is of 6 or 8 days. The students are skilled in their subjects to improve their confidence level. All the concepts they studied in 12th standard and other additional concepts required for the next levels are enclosed in the course framework. This course will help the students to have a smooth conversion to the next level. Our observation has shown that students who completed the bridge course have proceeded enormously well in university examinations. The strength of our relationship with the students is our ability to scale their strengths and weaknesses. Bridge courses conducted by our departments at the beginning of the academic year are an effort on our part to fill-in the gaps that exist between the shared knowledge of students at the entry-level and the course requirements. Their parents have also deeply appreciated the bridge course activity conducted by our college.

**Department of Chemistry
PDVP College, Tasgaon**

OBJECTIVES



Bridge course act as a buffer for new entrance

This course provides a better changeover to get ready for the next level in university

To bridge the gap between subjects studied at 12th class level and the subjects they would be studying in B.Sc. classes.

Bridge Course

Interactive & active students will be equipped with knowledge and confidence

Clear all basic concepts which required in university level

To provide adequate foundation in the basic science subject, so that students do not face any difficulty in university level

TIME-TABLE

Shri Swami Vivekanand Shikshan Sanstha's

Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon. Dist-Sangli.

Time-Table (Senior Science) 2020-2021

B. Sc. I

Time-Table for Online Bridge course Chemistry

Period	Time	Div.	Inauguration	Day 1	Day 2	Day 3	Day 4	Day 5
			28 th Oct 20	29 th Oct 20	31 st Oct 20	2 nd Nov 20	3 rd Nov 20	4 th Nov 20
			WED	THURS	SAT	MON	TUES	WED
1	11.00	A						
	to							
	11.40	B		Chem (MUP)				
2	12.00	A		Chem (ANA)				Chem (ASK)
	to							
	12.40	B	Chem (SSP)			Chem (SDJ)		

TIME-TABLE

Shri Swami Vivekanand Shikshan Sanstha's

**Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon. Dist-Sangli.**

Time-Table (Senior Science) 2020-2021

B. Sc. I

**Time-Table for Online Bridge course
Chemistry**

Period	Time	Div.	Day 6	Day 7	Day 8	Day 9	Day 10
			5 th Nov 20	6 th Nov 20	7 th Nov 20	9 th Nov 20	10 th Nov 20
			THURS	FRI	SAT	MON	TUES
1	11.00	A					
	to						
	11.40	B			Chem (ANA)		
2	12.00	A		Chem (MUP)	Chem (SDJ)		
	to						
	12.40	B				Chem (ASK)	

SYLLABUS

Inorganic Chemistry

➤ **Introduction to Chemistry**

Basics of Chemistry, sub divisions of Chemistry

➤ **Basic concepts in inorganic Chemistry**

Atoms, molecules, ions, types of bonding

➤ **Periodic table**

Basics of periodic table, groups, periods, periodic properties

➤ **Electronic configuration of elements**

Electronic configuration of elements

Organic Chemistry

➤ **Introduction to Chemistry**

Basics of Chemistry, sub divisions of Chemistry

➤ **Basic concepts in organic Chemistry**

Introduction to Organic reaction, Concept of Acid-base

➤ **Organic reaction transformation**

Introduction to chemical bond, breaking and making of bond, types of reactions.

➤ **Reaction Intermediates**

Introduction to reaction intermediates, Type of reagents

Physical Chemistry



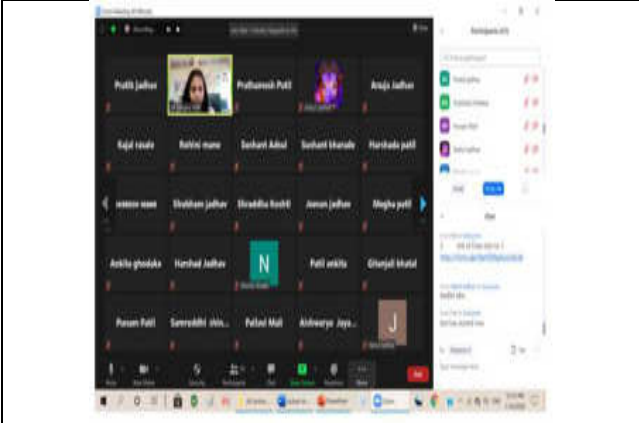
- **Introduction to Chemistry**
Basics of Chemistry, sub divisions of Chemistry
- **Basic concepts in Physical Chemistry**
Introduction to physical parameters
- **Chemical Kinetics**
Introduction to basic concepts of chemical kinetics
- **Thermodynamics**
Introduction to basic concepts of thermodynamics
- **Electrochemistry**
Introduction to basic concepts of electrochemistry

Analytical & Industrial Chemistry



- **Introduction to Chemistry**
Basics of Chemistry, sub classes of Chemistry
- **Basic concepts in Analytical Chemistry**
Introduction to analytical index
- **Introduction about Instrumentations and apparatus in lab**
Thermometer, PH meter, Conductometry, Viscosity,
- **Basic concepts in Industrial Chemistry**
Introduction to basic concepts of industrial chemistry
- **Introduction about nearby located the industries**
Introduction to sugar industries, pharmaceutical industries

PHOTO GALLERY



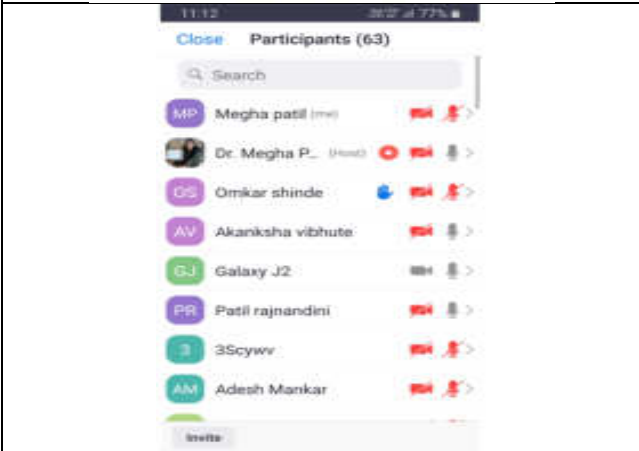
<https://forms.gle/9jahDR8gALzoS4ca8>




Welcome

BRIDGE COURSE
ON B. SC. I(A)

Prepared by, **Dr Megha U. Patil**
M.Sc. Chemistry, NET-JRF, GATE, Ph.D.
Assistant professor in Chemistry
PDVP College Tasgaon



B. Sc. I Division A...
Anil Manoj & Sound, Lij...

Topic: Basic concepts in Physical Chemistry (under bridge course)
Time: Nov 6, 2020 12:00 PM India

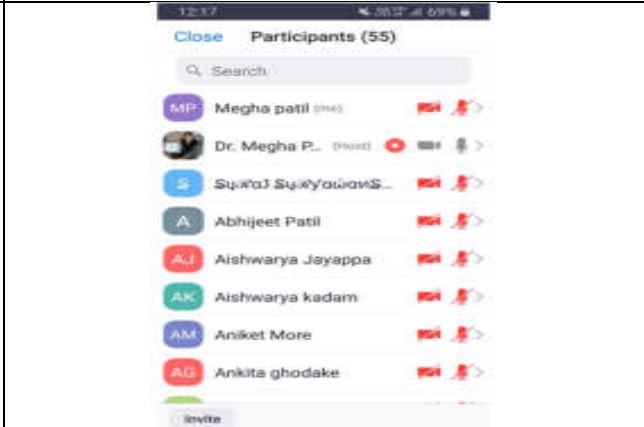
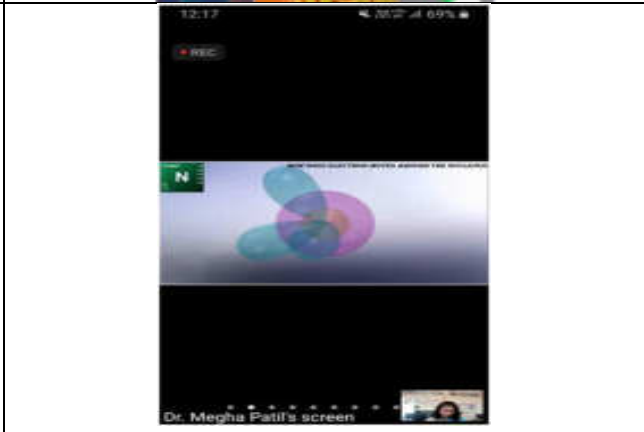
Join Zoom Meeting
<https://us04web.zoom.us/j/9303288182?pwd=ZkJSNy90UFNMZGxPMFpQMWVYYVZBxdz09>

Meeting ID: 930 328 8182
Passcode: 1111

Attendance report of Dr. Megha Patil's lecture...

Dear students,
I am sharing today's attendance link and class test link. Fill that attendance and solve test

1. Link for Attendance <https://forms.gle/XDWDvZZz5cf3EkiJe8>
2. link of Class test no 1 <https://forms.gle/9jahDR8gALzoS4ca8>



Attendance


Timestamp	Email Address	Write your name in detail	Lecture time	Write your contact number	
10/29/2020	11:49:50	Tajeta26@gmail.com	Tajeta Adhikara Dagit	11:00am - 12:00 pm	7620324198
10/29/2020	11:49:50	aytanore17@gmail.com	Arta vijaya more	11:00am - 12:00 pm	849834070
10/29/2020	11:50:17	rajandipat214@gmail.com	Pati Rajandin	11:00am - 12:00 pm	888020844
10/29/2020	11:50:20	maneshubhan7761@gmail.com	Shubhan Nitaj Mars	11:00am - 12:00 pm	902286441
10/29/2020	11:50:26	vjayastuti177@gmail.com	Vjay Dnyeeswar Sutar	11:00am - 12:00 pm	902526708
10/29/2020	11:50:35	np726719@gmail.com	Nitika Sanjay Patil	11:00am - 12:00 pm	762042230
10/29/2020	11:50:37	ghatgryahroya321@gmail.com	Yash Narayan Ghatage	11:00am - 12:00 pm	702048891
10/29/2020	11:50:40	shubhammarv3946@gmail.com	Shubham Namdev Mars	11:00am - 12:00 pm	989114055
10/29/2020	11:50:42	patiharewa17@gmail.com	Swati Balasubh Patihare	11:00am - 12:00 pm	782003947
10/29/2020	11:51:36	rajanyajadh324@gmail.com	Rajanya Jadhav	11:00am - 12:00 pm	932286978
10/29/2020	11:52:10	manopati12@gmail.com	Mansi patil	11:00am - 12:00 pm	899005058
10/29/2020	11:53:10	pankajpatil19@gmail.com	Pankaj Anu patil	11:00am - 12:00 pm	702042756
10/29/2020	11:53:15	sahipatole33@gmail.com	Sahil Ramchandra Patole	11:00am - 12:00 pm	775891688
10/29/2020	11:53:17	mowajpatil12@gmail.com	Angali more	11:00am - 12:00 pm	973035240
10/29/2020	11:53:19	np487522@gmail.com	Nitika Pradip Patil B.sc	11:00am - 12:00 pm	758484352
10/29/2020	11:53:57	sahinaykawa388@gmail.com	Sahil shabbir naykawa	11:00am - 12:00 pm	997041088
10/29/2020	11:54:09	pchawan719@gmail.com	Chawan pratishesh patil	11:00am - 12:00 pm	975062917
10/29/2020	11:54:11	sahinaykawa388@gmail.com	Sahil shabbir naykawa	11:00am - 12:00 pm	997041088
10/29/2020	11:54:58	Dhanayjayshinde457@gmail.com	Shreya Dhanayjay Shinde	11:00am - 12:00 pm	8970759107
10/29/2020	11:55:08	rajandipat214@gmail.com	Rajandini Patil A.p Rajaj	11:00am - 12:00 pm	888020844
10/29/2020	11:55:26	patilpradya967@gmail.com	Patil Pradya Anandaa	11:00am - 12:00 pm	762097374
10/29/2020	11:55:35	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:55:37	khamatayast07@gmail.com	Khamata Jyoti dpatil	11:00am - 12:00 pm	935622968
10/29/2020	11:56:21	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:57:18	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:57:33	an263967@gmail.com	Surya Nandkumar Mali	11:00am - 12:00 pm	907306230
10/29/2020	11:57:48	sumitshinde9961@gmail.com	Sunit Ravasabhi Shinde	11:00am - 12:00 pm	902147786
10/29/2020	11:58:01	DEEPAKGOSAVI.07@gmail.com	Deepak Gosavi	11:00am - 12:00 pm	716020854
10/29/2020	11:58:51	reddhan214@gmail.com	Dhanesh mankar	11:00am - 12:00 pm	716020854
10/29/2020	12:16:33	Rutajshinde3855@gmail.com	Rutaj dip shinde	11:00am - 12:00 pm	970293865
10/29/2020	12:32:54	harshadshinde1163@gmail.com	Harshada shinde	11:00am - 12:00 pm	8177866447
10/29/2020	12:51:19	patilakul16311@gmail.com	Gnate Rahul patil	11:00am - 12:00 pm	8421014796
11/6/2020	12:35:27	surajpatil2092@gmail.com	Suraj surajwarshi	11:00am - 12:00 pm	8200194388
11/6/2020	12:35:58	bhaskarshankar945@gmail.com	Shakti manohar bhaskar	11:00am - 12:00 pm	Yes
11/6/2020	12:45:25	ashwaryakadam1452@gmail.com	Ashwarya Nandkumar ka	11:00am - 12:00 pm	9889021452
11/6/2020	12:45:26	pr566132201@gmail.com	Jyotee Vijaykumar Patil	11:00am - 12:00 pm	953090118
11/6/2020	12:46:35	patilmal261@gmail.com	Malhar Narayan Mali	11:00am - 12:00 pm	946320219
11/6/2020	12:48:41	dnyudhanpatil4@gmail.com	Dnyudhan Ravasabhi jar	11:00am - 12:00 pm	7620961435
11/6/2020	12:45:58	sumitshinde154@gmail.com	Sumit Jadhav	11:00am - 12:00 pm	966526568
11/6/2020	12:46:19	sharda kishori11@gmail.com	Shardika Bajrai Kashi	11:00am - 12:00 pm	904619942
11/6/2020	12:46:19	sanmudhi197200@gmail.com	Sanmudhi Rajaram Shri	11:00am - 12:00 pm	7620072857
11/6/2020	12:46:36	mowajpatil12@gmail.com	Vjay Vishnu More	11:00am - 12:00 pm	796226944
11/6/2020	12:46:36	riampati1199@gmail.com	Sushant mas patil	11:00am - 12:00 pm	826381310
11/6/2020	12:46:46	jeenapatil1302@gmail.com	Jeevan jadhav	11:00am - 12:00 pm	938422788
11/6/2020	12:46:42	kajalshinde9@gmail.com	Kajal babasaheb shinde	11:00am - 12:00 pm	928459484
11/6/2020	12:46:43	khatajpatil12@gmail.com	Ganpati sajendra khataj	11:00am - 12:00 pm	751169365
11/6/2020	12:48:48	abhishek2001patil@gmail.com	Abhishek hamant patil	11:00am - 12:00 pm	989007769
11/6/2020	12:46:54	arunajadhav1211@gmail.com	Aruna Lakshmi Jadhav	11:00am - 12:00 pm	9316832196
11/6/2020	12:46:54	rutajpatil192001@gmail.com	Rutuja Vilas Hatavade	11:00am - 12:00 pm	935668998
11/6/2020	12:47:01	maneshubhan7761@gmail.com	Manesh shubhan	11:00am - 12:00 pm	Thanks
11/6/2020	12:47:19	np726719@gmail.com	Pratik Suresh jadhav	11:00am - 12:00 pm	9021913886
11/6/2020	12:47:36	svamishankar13@gmail.com	Shelika Sunana Shankar	11:00am - 12:00 pm	977097956
11/6/2020	12:47:41	rutajpatil192001@gmail.com	Rutuja Vilas Hatavade	11:00am - 12:00 pm	935668998
11/6/2020	12:48:53	shwetasakshikha12@gmail.com	Shweta Babasaheb sakshikha	11:00am - 12:00 pm	957934157
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10/29/2020	11:49:50	Tajeta26@gmail.com	Tajeta Adhikara Dagit	11:00am - 12:00 pm	7620324198
10/29/2020	11:49:50	aytanore17@gmail.com	Arta vijaya more	11:00am - 12:00 pm	849834070
10/29/2020	11:50:17	rajandipat214@gmail.com	Pati Rajandin	11:00am - 12:00 pm	888020844
10/29/2020	11:50:20	maneshubhan7761@gmail.com	Shubhan Nitaj Mars	11:00am - 12:00 pm	902286441
10/29/2020	11:50:26	vjayastuti177@gmail.com	Vjay Dnyeeswar Sutar	11:00am - 12:00 pm	902526708
10/29/2020	11:50:35	np726719@gmail.com	Nitika Sanjay Patil	11:00am - 12:00 pm	762042230
10/29/2020	11:50:37	ghatgryahroya321@gmail.com	Yash Narayan Ghatage	11:00am - 12:00 pm	702048891
10/29/2020	11:50:40	shubhammarv3946@gmail.com	Shubham Namdev Mars	11:00am - 12:00 pm	989114055
10/29/2020	11:50:42	patiharewa17@gmail.com	Swati Balasubh Patihare	11:00am - 12:00 pm	782003947
10/29/2020	11:51:36	rajanyajadh324@gmail.com	Rajanya Jadhav	11:00am - 12:00 pm	932286978
10/29/2020	11:52:10	manopati12@gmail.com	Mansi patil	11:00am - 12:00 pm	899005058
10/29/2020	11:53:10	pankajpatil19@gmail.com	Pankaj Anu patil	11:00am - 12:00 pm	702042756
10/29/2020	11:53:15	sahipatole33@gmail.com	Sahil Ramchandra Patole	11:00am - 12:00 pm	775891688
10/29/2020	11:53:17	mowajpatil12@gmail.com	Angali more	11:00am - 12:00 pm	973035240
10/29/2020	11:53:19	np487522@gmail.com	Nitika Pradip Patil B.sc	11:00am - 12:00 pm	758484352
10/29/2020	11:53:57	sahinaykawa388@gmail.com	Sahil shabbir naykawa	11:00am - 12:00 pm	997041088
10/29/2020	11:54:09	pchawan719@gmail.com	Chawan pratishesh patil	11:00am - 12:00 pm	975062917
10/29/2020	11:54:11	sahinaykawa388@gmail.com	Sahil shabbir naykawa	11:00am - 12:00 pm	997041088
10/29/2020	11:54:58	Dhanayjayshinde457@gmail.com	Shreya Dhanayjay Shinde	11:00am - 12:00 pm	8970759107
10/29/2020	11:55:08	rajandipat214@gmail.com	Rajandini Patil A.p Rajaj	11:00am - 12:00 pm	888020844
10/29/2020	11:55:26	patilpradya967@gmail.com	Patil Pradya Anandaa	11:00am - 12:00 pm	762097374
10/29/2020	11:55:35	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:55:37	khamatayast07@gmail.com	Khamata Jyoti dpatil	11:00am - 12:00 pm	935622968
10/29/2020	11:56:21	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:56:37	hambharkar2311@gmail.com	Harsh Dip Kambli	11:00am - 12:00 pm	970792291
10/29/2020	11:57:18	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:57:33	an263967@gmail.com	Surya Nandkumar Mali	11:00am - 12:00 pm	907306230
10/29/2020	11:57:48	sumitshinde9961@gmail.com	Sunit Ravasabhi Shinde	11:00am - 12:00 pm	902147786
10/29/2020	11:58:01	DEEPAKGOSAVI.07@gmail.com	Deepak Gosavi	11:00am - 12:00 pm	716020854
10/29/2020	11:58:51	reddhan214@gmail.com	Dhanesh mankar	11:00am - 12:00 pm	716020854
10/29/2020	12:16:33	Rutajshinde3855@gmail.com	Rutaj dip shinde	11:00am - 12:00 pm	970293865
10/29/2020	12:32:54	harshadshinde1163@gmail.com	Harshada shinde	11:00am - 12:00 pm	8177866447
10/29/2020	12:51:19	patilakul16311@gmail.com	Gnate Rahul patil	11:00am - 12:00 pm	8421014796
10/29/2020	11:51:36	rajanyajadh324@gmail.com	Rajanya Jadhav	11:00am - 12:00 pm	932286978
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10/29/2020	11:53:10	pankajpatil19@gmail.com	Pankaj Anu patil	11:00am - 12:00 pm	702042756
10/29/2020	11:53:15	sahipatole33@gmail.com	Sahil Ramchandra Patole	11:00am - 12:00 pm	775891688
10/29/2020	11:53:17	mowajpatil12@gmail.com	Angali more	11:00am - 12:00 pm	973035240
10/29/2020	11:53:19	np487522@gmail.com	Nitika Pradip Patil B.sc	11:00am - 12:00 pm	758484352
10/29/2020	11:53:57	sahinaykawa388@gmail.com	Sahil shabbir naykawa	11:00am - 12:00 pm	997041088
10/29/2020	11:54:09	pchawan719@gmail.com	Chawan pratishesh patil	11:00am - 12:00 pm	975062917
10/29/2020	11:54:11	sahinaykawa388@gmail.com	Sahil shabbir naykawa	11:00am - 12:00 pm	997041088
10/29/2020	11:54:58	Dhanayjayshinde457@gmail.com	Shreya Dhanayjay Shinde	11:00am - 12:00 pm	8970759107
10/29/2020	11:55:08	rajandipat214@gmail.com	Rajandini Patil A.p Rajaj	11:00am - 12:00 pm	888020844
10/29/2020	11:55:26	patilpradya967@gmail.com	Patil Pradya Anandaa	11:00am - 12:00 pm	762097374
10/29/2020	11:55:35	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:55:37	khamatayast07@gmail.com	Khamata Jyoti dpatil	11:00am - 12:00 pm	935622968
10/29/2020	11:56:21	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:56:37	hambharkar2311@gmail.com	Harsh Dip Kambli	11:00am - 12:00 pm	970792291
10/29/2020	11:57:18	adeshmankar99@gmail.com	Adesh Rajkumar Mankar	11:00am - 12:00 pm	973094775
10/29/2020	11:57:33	an263967@gmail.com	Surya Nandkumar Mali	11:00am - 12:00 pm	907306230
10/29/2020	11:57:48	sumitshinde9961@gmail.com	Sunit Ravasabhi Shinde	11:00am - 12:00 pm	902147786
10/29/2020	11:58:01	DEEPAKGOSAVI.07@gmail.com	Deepak Gosavi	11:00am - 12:00 pm	716020854
10/29/2020	11:58:51	reddhan214@gmail.com	Dhanesh mankar	11:00am - 12:00 pm	716020854
10/29/2020	12:16:33	Rutajshinde3855@gmail.com	Rutaj dip shinde	11:00am - 12:00 pm	970293865
10/29/2020	12:32:54	harshadshinde1163@gmail.com	Harshada shinde	11:00am - 12:00 pm	8177866447
10/29/2020	12:51:19	patilakul16311@gmail.com	Gnate Rahul patil	11:00am - 12:00 pm	8421014796

Class Test

11/6/2020 12:34:48	15/25	Jyoti Vijaykumar Patil	B.Sc.I	A	FALSE	TRUE	TRUE	T, C
11/6/2020 12:34:50	15/25	Hammad Jadhav	B.Sc.I	A	FALSE	TRUE	TRUE	All of these
11/6/2020 12:35:44	15/25	Rishabh Babasaheb Sasane	B.Sc.I	A	FALSE	TRUE	TRUE	All of these
11/6/2020 12:35:45	20/25	Pankaj Narayan Mali	B.Sc.I	A	TRUE	TRUE	TRUE	All of these
11/6/2020 12:36:01	25/25	Nehar Shahajan smashil	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 12:36:25	15/25	Aneja Lalaso Jadhav	B.Sc.I	A	FALSE	TRUE	FALSE	T, P
11/6/2020 12:36:33	10/25	Devashree Ramakrishna	B.Sc.I	A	FALSE	FALSE	TRUE	T, P
11/6/2020 12:36:47	25/25	Hannada Hanmant patil	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 12:37:47	20/25	Mansani Anshul Patil	B.Sc.I	A	TRUE	TRUE	TRUE	All of these
11/6/2020 12:40:37	20/25	Nikhil Manoj Shesat	B.Sc.I	A	TRUE	TRUE	TRUE	All of these
11/6/2020 12:41:33	20/25	Santosh Rajaram Sh B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 12:41:37	15/25	Girishaji sajanrao Ghosal B.Sc.I	B.Sc.I	A	TRUE	TRUE	FALSE	T, P
11/6/2020 12:42:19	15/25	Aishwarya Hanumantrao	B.Sc.I	A	FALSE	TRUE	TRUE	T, P
11/6/2020 12:42:13	20/25	Shrawaha Binasa Kote B.Sc.I	B.Sc.I	A	TRUE	TRUE	FALSE	T, P
11/6/2020 12:42:20	10/25	Shreshth shikant more	B.Sc.I	A	TRUE	FALSE	TRUE	T, C
11/6/2020 12:42:32	25/25	Mansu megha sanjay	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 12:44:12	10/25	Jayant jalindar jadhav	B.Sc.I	A	TRUE	FALSE	FALSE	All of these
11/6/2020 12:44:23	20/25	Neha Ragu Jadhav	B.Sc.I	A	FALSE	TRUE	TRUE	T, P
11/6/2020 12:44:39	15/25	Sushant ramdev adail	B.Sc.I	A	FALSE	TRUE	TRUE	All of these
11/6/2020 12:47:17	20/25	Prapada Shinde	B.Sc.I	A	FALSE	TRUE	TRUE	T, P
11/6/2020 12:48:32	15/25	Anant shant more	B.Sc.I	A	FALSE	TRUE	FALSE	T, P
11/6/2020 12:49:08	15/25	ANURADHA AMRIT Patil B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	All of these
11/6/2020 12:50:38	10/25	Pratiksha Dinkar Kulkarni B.Sc.I	B.Sc.I	A	TRUE	FALSE	FALSE	T, P
11/6/2020 12:50:45	15/25	Sushant Nitesh Patil	B.Sc.I	A	TRUE	TRUE	FALSE	All of these
11/6/2020 12:52:01	25/25	Shival Dinkar More	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 12:52:19	15/25	Shikha Savarna Shant B.Sc.I	B.Sc.I	A	FALSE	TRUE	TRUE	All of these
11/6/2020 12:56:28	0/25	Prakrta Prakash More	B.Sc.I	A	FALSE	FALSE	FALSE	C, P
11/6/2020 13:00:37	20/25	Sushil Suhas kamble	B.Sc.I	A	TRUE	TRUE	TRUE	All of these
11/6/2020 13:02:41	20/25	Shweta Babasa salunkh B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	T, C
11/6/2020 13:04:01	25/25	Dr Megha Uday Patil	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 13:13:05	15/25	Rohit Shinde	B.Sc.I	A	TRUE	TRUE	FALSE	T, P
11/6/2020 13:13:48	25/25	Abhishek Anandrao Suti B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 13:20:13	15/25	Hrutuja Rajendra Patil	B.Sc.I	A	FALSE	TRUE	FALSE	T, P
11/6/2020 13:25:14	25/25	Sumit annu yadav	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 13:29:27	25/25	Patil Prathamesh Uttam B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 13:30:09	15/25	Rutuja Vilas Halawade	B.Sc.I	A	FALSE	TRUE	TRUE	All of these
11/6/2020 14:03:31	20/25	Punam Chandrakant Patil B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	All of these
11/6/2020 14:47:46	15/25	Akshay patil	B.Sc.I	A	TRUE	TRUE	FALSE	All of these
11/6/2020 15:53:45	20/25	Suraj vikas suryawansh B.Sc.I	B.Sc.I	A	TRUE	FALSE	TRUE	T, P
11/6/2020 16:30:00	25/25	Aishwarya Sanjay Jayap B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 17:33:40	15/25	Kajal baburao rasale	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 19:12:52	25/25	Ashitosh jalindar raksh B.Sc.I	B.Sc.I	A	TRUE	TRUE	TRUE	T, P
11/6/2020 19:22:35	10/25	Ankita anandrao ghodal B.Sc.I	B.Sc.I	A	TRUE	FALSE	FALSE	C, P


Dr. Ajay N. Ambhore


Head
 Dept. of Chemistry
 P.D.V.P. College, Tasgaon


Prin. Dr. Milind S. Hujare
Principal
 Padmabhushan Dr. Vasantao Dada Patil
 Mahavidyalaya, Tasgaon (Sangli).

"Dissemination of Education through Knowledge, Science and Culture".

--Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

Padmabhushan Dr. Vasantodada Patil

Mahavidyalaya, Tasgaon

Tal. Tasgaon. Dist. Sangli-416 312 (Maharashtra)

(Affiliated to Shivaji University, Kolhapur)

**DEPARTMENT OF
COMPUTER-SCIENCE**

(2020-21)

BRIDGE COURSE
IN

BASIC FUNDAMENTAL OF COMPUTER SYSTEM



"Dissemination of Education through Knowledge, Science and Culture".

--Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

Padmabhushan Dr.Vasatraodada Patil

Mahavidyalaya, Tasgaon

Tal. Tasgaon. Dist. Sangli-416 312 (Maharashtra)

(Affiliated to Shivaji University, Kolhapur)

Title of the Bridge Course

"fundamental's of computer"

Organizer:Department Of Computer Science
and

Internal Quality Assurance cell (IQAC) PDVP College,
Tasgaon

Total no.of students participated:-36

Total Faculty :-2

Notice

B. Sc. I Bridge Course Schedule

All science faculty members are hereby informed that **B. Sc. I** online classes will be started from **28th October 2020** as a bridge course.

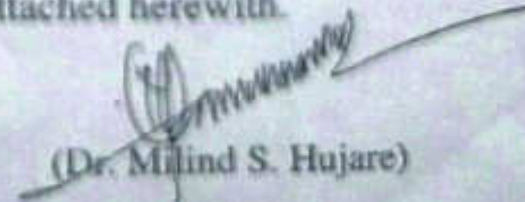
❖ On **28th October 2020** at **11.00 am** the bridge course will be inaugurated by the address of **Prin. Dr. M. S. Hujare**.

- Whatsapp groups are created for Division A and Division B. Teachers should communicate their links to the students on these groups, before scheduled time.
- Faculty members should use their mobile phones or laptops and engage lectures using **Zoom app / Google meet**.
- The faculty members should maintain the record of online bridge course.
- Collect the students' attendance using **Google form**.
- All faculty members should engage their lectures in the allotted classrooms as following:
 - **B. Sc. I Division A : Mathematics Department**
 - **B. Sc. I Division B : Physics Lecture Hall**
- The detailed schedule of bridge course is attached herewith.


(Dr. S. D. Jadhav)

Incharge,

Time-Table Committee


(Dr. Milind S. Hujare)

Principal

**Shri Swami Vivekanand Shikshan Sanstha's
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya, Tasgaon. Dist-Sangli.
Time-Table Link (Senior Science) 2020-2021
B. Sc. I (Computer Science)
Online Bridge course**

Instuctions for students

1) All computer science students of A and B Group informed that use google meet application for computer science bridge course

Download google meet app from play store

[https://play.google.com/store/apps/details?
id=com.google.android.apps.meetings&hl=en](https://play.google.com/store/apps/details?id=com.google.android.apps.meetings&hl=en)

2) Join for lecture before 5 min .

3) When we start lecture mute ur video and audio.

Time-Table and Link:

Sr.no	Period	Date	Time	Link
1	2	29 th oct	12.00pm	https://meet.google.com/ibt-inyg-bqw
2	2	4 th nov	12.00pm	https://meet.google.com/bms-ijoo-ivr
3	2	6 th nov	12.00pm	https://meet.google.com/ftm-tmah-zyp
4	2	7 th nov	12.00pm	https://meet.google.com/ngs-pwen-qfy
5	2	9 th nov	12.00pm	https://meet.google.com/ibt-inyg-bqw
6	2	27 th nov	12.00pm	https://meet.google.com/bms-ijoo-ivr
7	2	30 th nov	12.00pm	https://meet.google.com/ftm-tmah-zyp
8	2	4 th dec	12.00pm	https://meet.google.com/ngs-pwen-qfy
9	2	7 th dec	12.00pm	https://meet.google.com/ftm-tmah-zyp
10	2	11 th dec	12.00pm	https://meet.google.com/ftm-tmah-zyp

Note:Only click link and join

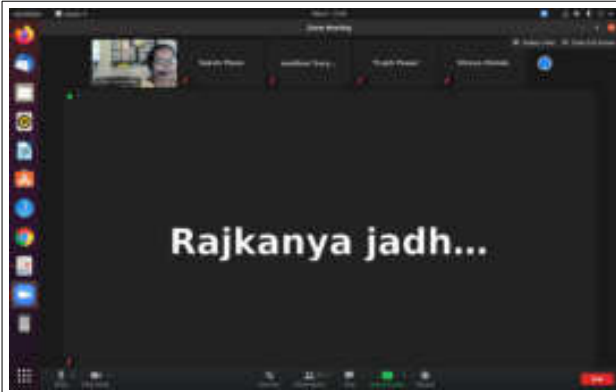
**Shri Swami Vivekanand Shikshan Sanstha's
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya, Tasgaon. Dist-Sangli.
Time-Table Link (Senior Science) 2020-2021
B. Sc. I (Computer Science)**

Online Bridge course

Syllabus

Sr.no	Topic	Content
1	Hardware	Input ,output ,system unit
2	Software	Open source and closed source
3	Networking	LAN ,MAN and WAN
4	Carrier Oportunity in C.S.	-----

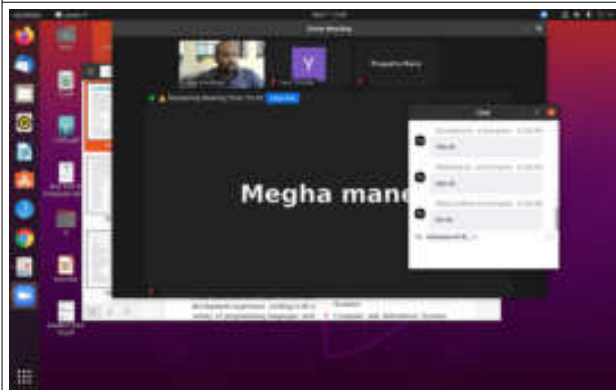
SCREEN SHOT



1



2



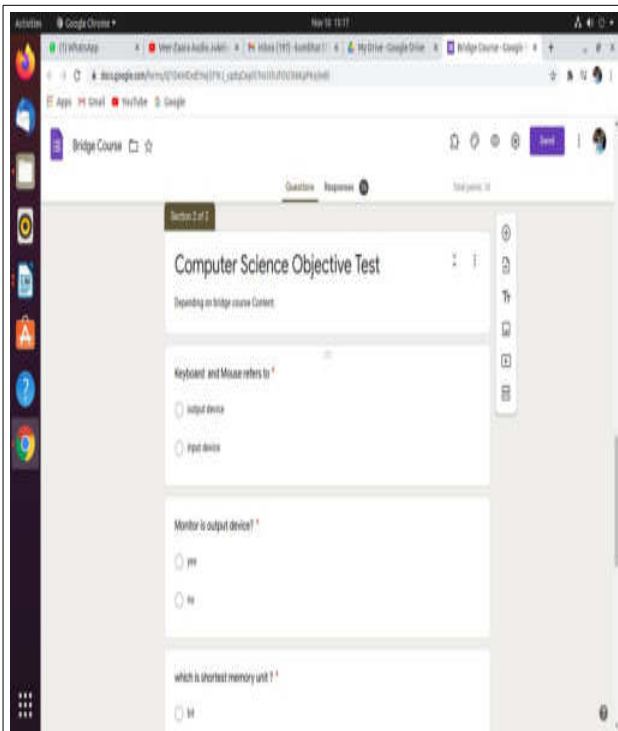
3



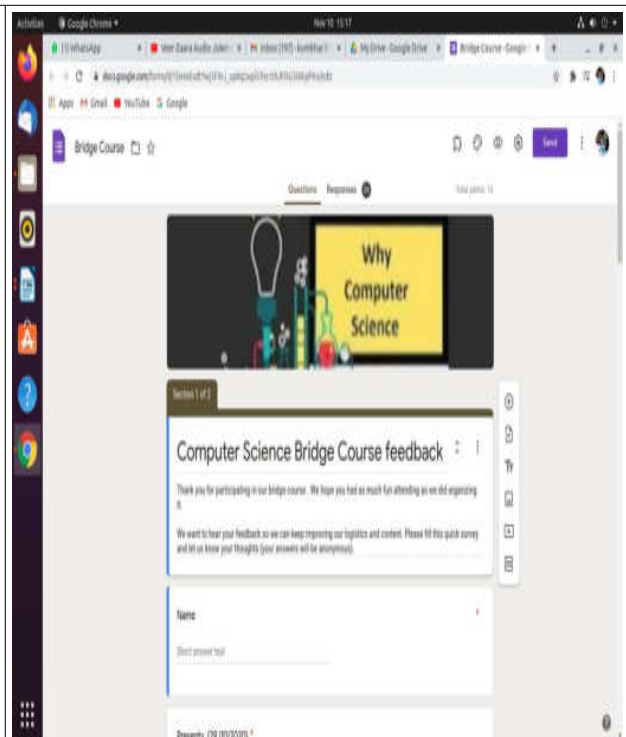
4



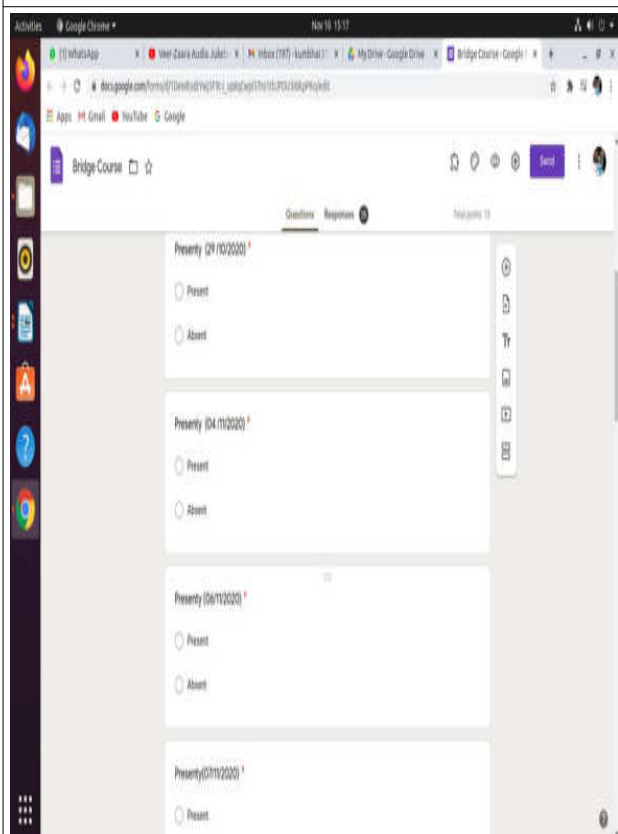
ONLINE TEST, PRESENTY , ATTENDANCE AND FEEDBACK



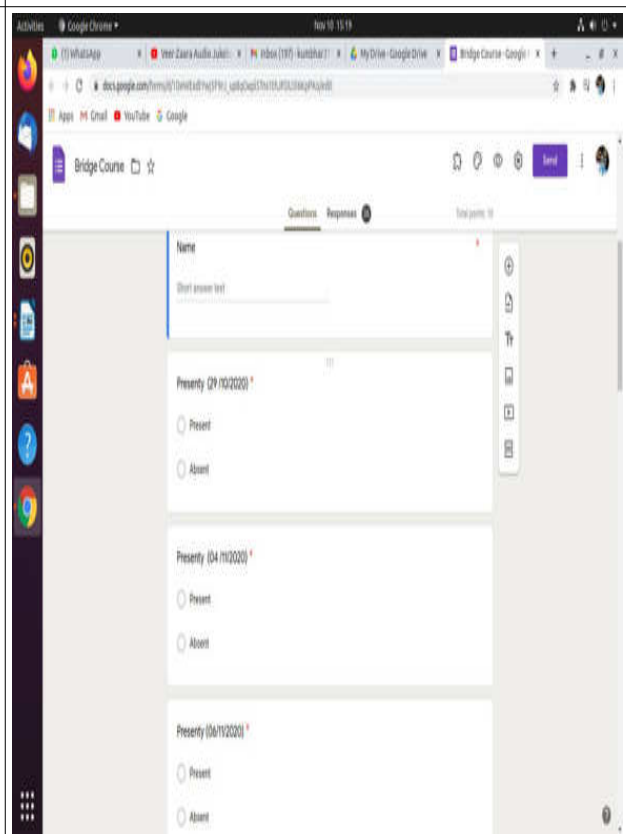
1



2



3



4

ONLINE TEST, ATTENDANCE, FEEDBACK, REPORT

1

	A	B	C	D	E	F	G	H	I
	11/17/2023	13:42:08	Nikhil pappu yadav	No	Priority (D1)	Priority (D2)	Priority (D3)	Priority (D4)	Priority (D5)
	11/17/2023	13:42:08	Shreya Chaturya Shriv	No	Priority (D1)	Priority (D2)	Priority (D3)	Priority (D4)	Priority (D5)
	11/17/2023	13:42:08	Saurabh Kumar	Yes	Priority (D1)	Priority (D2)	Priority (D3)	Priority (D4)	Priority (D5)

2

	K	L	M	N	O	P	Q	R
	Priority(D1)	Priority(D2)	Priority(D3)	Priority(D4)	Priority(D5)	Keyboard and Mouse is setup device?	Which is shortest memory unit?	Types of networking
	Priority	Priority	Priority	Priority	Priority	input device	yes	kb
	Priority	Priority	Priority	Priority	Priority	input device	yes	bit

3

	Q	P	Q	R	S	T	U	V
1	Keyboard and Mouse is output device?	Priority	which is shortest memory unit ?	Priority	which is open source operating system?	Priority	Email Address	Score
2	output device	yes	kb	wan	linux	unix		2 / 10
3	input device	yes	bit	can	linux	unix		8 / 10
4	input device	yes	bit	man, wan	mac	chrome		6 / 10
5	output device	yes	byte	man, wan	linux			2 / 10
6	input device	yes	kb	lan, man, wan	linux	chrome		6 / 10
7	input device	yes	byte	man, wan, can	linux	unix		6 / 10
8	output device	yes	kb	can	linux			2 / 10
9	input device	yes	bit	lan, man, wan	linux	unix		10 / 10
10	input device	yes	bit	lan, man, wan	linux	unix		10 / 10
11	input device	yes	bit	lan, man, wan	linux	unix		10 / 10
12	output device	yes	byte	lan, man, wan	linux	unix, chrome		4 / 10
13	output device	yes	byte	man	linux			2 / 10
14	output device	yes	kb	lan, man, wan	linux	mac, chrome		4 / 10
15	input device	no	kb	man	linux			2 / 10
16	output device	yes	byte	wan	linux			2 / 10
17	input device	yes	bit	lan, man, wan	linux	unix		10 / 10
18	input device	yes	bit	lan, man, wan	linux	unix		10 / 10
19	input device	yes	byte	lan, man, wan	linux	mac		6 / 10
20	input device	no	bit	wan, can	linux			4 / 10
21	input device	yes	kb	lan	linux			4 / 10
22	input device	yes	kb	lan	linux			4 / 10
23	input device	yes	bit	lan, man, wan	linux			8 / 10
24	input device	yes	bit	lan, man, wan	linux			8 / 10
25	output device	yes	byte	man, wan	linux	unix, mac		2 / 10
26	output device	yes	byte	man, wan	linux	unix, mac		2 / 10
27	input device	no	gb	lan, man, wan	linux			4 / 10

Sample Certificate

		Header (Default Style) ▾
<p><i>'Dissemination of Education through Knowledge, Science and Culture' - Shikshamaharshi Dr. Bapuji Sahankhe</i> <u>Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's</u></p> <p>PADMABHUSHAN DR. VASANTRAODADA PATIL MAHAVIDYALAYA, <u>Tasgaon, Dist.: Sangli.</u></p> <p>Organized By</p> <p>DEPARTMENT OF COMPUTER SCIENCE Bridge COURSE IN <i>fundamental's of Computer</i></p> <p>CERTIFICATE</p>		
<p>This is to certify that Mr./Miss/Mrs.----- of Arts/Commerce/Science I/II/III has competition <i>fundamental's of Computer</i> course.</p>		
<p><u>Mr. Kumbhar N.V.</u> Coordinator</p>	<p><u>Mr. Kumbhar V.T.</u> Organizing Secretary</p>	<p><u>Dr. Milind S. Hujare</u> Principal</p>

Alka Inamdar
IQAC Co-Ordinator,
PD V P Mahavidyalaya,
Tasgaon.

Milind S. Hujare
Prin. Dr. Milind S. Hujare
Principal
Padmabhushan Dr. Vasanttraodada Patil
Mahavidyalaya, Tasgaon (Sangli).

“ज्ञान, विज्ञानाणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. बापूजीसाकुंखे,
Shri Swami Vivekanand Shikshan Sanstha's Kolhapur

**PADMABHUSHAN DR. VASANTRAODADA PATIL
MAHAVIDYALAYA, TASGAON**

(AFFILIATED TO SHIVAJI UNIVERSITY, KOLHAPUR)

Department of English

**BRIDGE COURSE IN TRANSLATION OF OTHER
REGIONAL LITERATURE INTO ENGLISH**

Academic Year 2020-21

Mr. A. R. Patil
Co-coordinator,
Bridge Course in Translation of other Regional
Literature into English

1st July 2020

To,
The Principal,
P D V P College, Tasgaon
Tal: Tasgaon, Dist: Sangli - 416 312.

Sub: - Regarding Permission to commencement of Bridge Course in Translation of other
Regional Literature into English

Respected Sir,

With reference to above subject, our college is going to start a Graduate Bridge Course entitled Bridge Course in Translation of other Regional Literature into English for B. A. PART-I in this current academic year. So, I request you to kindly give me permission to commencement above said course.

Please consider the same and oblige.

Thanking you

Yours faithfully



(Mr. A. R. Patil)

**BRIDGE COURSE IN TRANSLATION OF OTHER REGIONAL
LITERATURE INTO ENGLISH**

DEPARTMENT OF ENGLISH

Academic year 2020-21


NOTICE

1st July, 2020

All the students of B. A. PART-I(English optional) are hereby informed that, our college is going to start a Bridge course entitled **Bridge Course in Translation of other Regional Literature into English** for this current academic year. All the students should take admissions as earlier.



Mr. A. R. Patil
Co-ordinator,
Bridge Course in Translation of other
Regional Literature into English


Mr. A.S. Pachore
Head
HEAD

Department of English
P.D.V.P.College, Tasgaon.

BRIDGE COURSE IN TRANSLATION OF OTHER REGIONAL LITERATURE INTO ENGLISH

TIME TABLE

Sr. No	Date	Room No	Name of Chapter	Name of Faculty	Remark
1	09/07/2020	Room No-05	A Short Story as a Minor Form of Literature Forms of Literature Short Story :Minor form of literature	Mr. A. S. Pachore	
2	10/07/2020		Some types of Short Story An Anecdote A Legend	Mr. P.R. Khade	
3	11/07/2020		Premchand The Greatest Writer in Hindi Literature Short Story Writer and Dramatist	Mr.A. R. Patil	
4	12/07/2020		Sarat Chandra Chatterji Prominent Bengali Writer	Dr. D. B. Thorbole	
5	13/07/2020		A Short Story as a Minor Form of Literature Character Plot	Mr. A. S. Pachore	
6	14/07/2020		Some types of Short Story A Parable	Mr. P.R. Khade	
7	16/07/2020		Premchand A Famous Novelist	Mr.A. R. Patil	
8	17/07/2020		Sarat Chandra Chatterji Short Story Writer of 20th Century	Dr. D. B. Thorbole	
9	18/07/2020		Sarat Chandra Chatterji Village Life and Traditions	Dr. D. B. Thorbole	

A.R.P.

Mr. A. R. Patil
Co-ordinator,
Bridge Course in Translation of other
Regional Literature into English

Mr. A.S. Pachore
[Signature]
HEAD
Department of English
P.D.V.P.College, Tasgaon,

**TITLE: Bridge Course in Translation of other Regional literature into
English**

YEAR OF IMPLEMENTATION: 2020-21

CONDUCTING DEPARTMENT: ENGLISH

PREAMBLE:

This bridge course provides information about the translation in Indian writing in English. Though, these bridges course to aware the students about greatness of regional literature.

TEACHING OBJECTIVES:-

1. To acquaint the students with translated Modern Indian literature in English.
2. To introduce the students to short story as a form of literature with reference to the texts prescribed.
3. To develop literary competence among students.

LEARNING OUTCOMES:

1. The students will understand the importance of translation.
2. The students will get acquainted with functioning of regional literature into English.

DURATION

1. The Bridge Course shall be a Graduate.
2. The duration of Bridge Course shall be of 10 days.

MEDIUM OF INSTRUCTION:

The medium of instruction shall be in English.


EVALUATION SYSTEM:

All the students will be continuously evaluated by,

- | | |
|---|-----------------|
| a. Attendance | 05 M arks |
| b. Final examination (One Theory paper) | 15 Marks |
| <hr/> | |
| | 20 Marks |

Azath

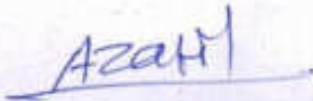
Mr.A. R. Patil
Co-ordinator,
Bridge Course in Translation of
other Regional Literature Into English

Mr.A.S. Pachore

HEAD
Department of English
P.D.V.P.College, Tasgaon.

NOTICE

19th July, 2020

All the students are informed that the Bridge Course in Bridge Course in Translation of other Regional literature Into English Examination will be held on July 22, 2020 from 12: 00 to 1: 00am. However, everyone should be present on time.



Mr. A.R. Patil
Co-ordinator,
Bridge Course in Translation of
other Regional Literature Into English



HEAD
Department of English
P.D.V.P. College, Tasgaon.

Total Hours of Teaching: 10

Syllabus Contents:

Unit No	Contents	Period
1	A Short Story as a Minor Form of Literature <ol style="list-style-type: none"> 1. Forms of Literature 2. Short Story :Minor form of literature 3. Character 4. Plot 	02
2	Some types of Short Story <ol style="list-style-type: none"> 1. An Anecdote 2. A Legend 3. A Parable 	03
3	Premchand <ol style="list-style-type: none"> 1. The Greatest Writer in Hindi Literature 2. Short Story Writer and Dramatist 3. A Famous Novelist 	03
4	Sarat Chandra Chatterji <ol style="list-style-type: none"> 1. Prominent Bengali Writer 2. Short Story Writer of 20th Century 3. Village Life and Traditions 	04

References:

Venugopal, C. V. The Indian Short Story in English: A Survey. Bareilly: Prakash Book Depot. 1976.

Prasad, B. A. A Background to the Study of English Literature. Madras: MacMillan India, 1965.

Mehta, Kamal. The Twentieth Century Indian Short Story in English. New Delhi: Creative Books, 1997.

Hunter, Adrian. *The Cambridge Introduction to the Short Story in English*. New Delhi: Cambridge University Press, 2007.

Shaw, Valerie. *The Short Story: A Critical Introduction*, New York: Longman Group Limited. 1983.

Warren, Robert Penn, Erskine, Albert. (Ed.). *Short Story Masterpieces*, New York: Dell Publishing Co., 1954.

B.A.PART I Roll Call (2020-21) Eng.Opt.

Sr. No	Roll No.	Gender	NAME OF THE STUDENTS	Subject
				ENG.
1	2004	MISS	ATTAR ALISHA RAJU 9172641680	Eng.Opt.,
2	2011	SHRI	BHOSALE NUTANKUMAR KIRAN 9960759959	Eng.Opt.,
3	2013	MISS	BHOSEKAR SWAPNALI POPAT 9373557882	Eng.Opt.,
4	2014	MISS	BODAKE VAISHNAVI UTTAM 9322940179	Eng.Opt.,
5	2025	MISS	DESHMUKH DHANASHRI MARUTI 8482020695	Eng.Opt.,
6	2027	SHRI	DEVKULE ASHISH DIPAK 8554974272	Eng.Opt.,
7	2032	SHRI	DHANGOLI DIPAK VYANKAPPA 8121501214	Eng.Opt.,
8	2035	SHRI	GAIKWAD ANIKET JIJABA 9766896501	Eng.Opt.,
9	2043	MISS	GHODAKE VAISHNAVI VINOD 9322602431	Eng.Opt.,
10	2046	SHRI	GURAV POONAM PRAKASH 7028746675	Eng.Opt.,
11	2055	MISS	JADHAV NEHA RAMESH 8329947988	Eng.Opt.,
12	2074	MISS	KADAM VRUSHALI GOVIND 9096420517	Eng.Opt.,
13	2075	MISS	KALE DNYANESHWARI SANJAY 9021391222	Eng.Opt.,
14	2080	MISS	KAMBLE PRITI POPAT 7720913401	Eng.Opt.,
15	2083	MISS	KAMBLE SNEHAL VINOD 7219645223	Eng.Opt.,
16	2091	MISS	KENGAR RUTUJA RAJKUMAR 9158663781	Eng.Opt.,
17	2097	SHRI	KHARAMATE VISHAL RAMESH 9673426483	Eng.Opt.,
18	2099	MISS	KHARMATE TANUJA BANDU 9689812310	Eng.Opt.,
19	2111	MISS	MADANE KOMAL RAMESH 9823482558	Eng.Opt.,
20	2112	MISS	MADANE NIKITA RAJENDRA 9049511842	Eng.Opt.,
21	2113	MISS	MADANE POOJA CHANDRAKANT 9730679868	Eng.Opt.,
22	2119	SHRI	MALI ATUL VITTHAL 9370809742	Eng.Opt.,
23	2123	MISS	MALI PRANALI PRAMOD 9765160360	Eng.Opt.,
24	2130	MISS	BALLAL PALLAVI SHANKAR 7292039406	Eng.Opt.,
25	2141	MISS	MANE BHARATI ARJUN 7218165017	Eng.Opt.,
26	2144	MISS	MANE PRATIBHA POPATRAO 7767882947	Eng.Opt.,
27	2149	MISS	MANE SNEHA DIPAK 8830779352	Eng.Opt.,
28	2150	MISS	MISAL SUMITRA RAJKUMAR 9172642928	Eng.Opt.,
29	2152	MISS	MOHITE RESHMA RAMESH 9975681302	Eng.Opt.,
30	2164	MISS	MULANI SANIYA FIROJ 9096460038	Eng.Opt.,
31	2168	MISS	MULLA RIZA KALANDAR 9096420386	Eng.Opt.,
32	2171	SHRI	NADAF JUBER AJIJ 9579414468	Eng.Opt.,

B.A.PART I Roll Call (2020-21) Eng.Opt.

Sr. No	Roll No.	Gender	NAME OF THE STUDENTS	Subject
				ENG.
33	2180	MISS	NYAYNIT SAKSHI GOUTAM 9767523639	Eng.Opt.,
34	2182	SHRI	PATHAN AAJIM YUSFKHAN 9858988008	Eng.Opt.,
35	2189	MISS	PATIL AISHWARYA VILAS 7219785868	Eng.Opt.,
36	2192	MISS	PATIL DIPTI DILIP 9923267090	Eng.Opt.,
37	2194	MISS	PATIL MADHURA SATISH 9372438565	Eng.Opt.,
38	2195	MISS	PATIL MAYURI SANDIP 7775921862	Eng.Opt.,
39	2204	SHRI	PATIL PRATIKSHA HANMANT 8767101777	Eng.Opt.,
40	2206	MISS	PATIL RADHIKA SANDIP 9356683853	Eng.Opt.,
41	2209	SHRI	PATIL RITESH RAJESH 9325049607	Eng.Opt.,
42	2219	MISS	PATIL SANJIVANI DHONDIRAM 9021802930	Eng.Opt.,
43	2227	MISS	PATIL SNEHAL SUDHAKAR 7758873948	Eng.Opt.,
44	2235	SHRI	PATIL VISHAL BHAUSO 9172586636	Eng.Opt.,
45	2236	MISS	DEVKULE UTKARSHA RAKESH 9322933007	Eng.Opt.,
46	2246	MISS	SALUNKHE AMISHA ANKUSH 8975929060	Eng.Opt.,
47	2250	MISS	SALUNKHE SNEHAL NANDKUMAR 9421223915	Eng.Opt.,
48	2255	MISS	SAWANT DIPALI GAUTAM 7058315391	Eng.Opt.,
49	2259	MISS	SHAIKH YASMIN HARUN 7558631607	Eng.Opt.,
50	2265	MISS	SHINDE AKANKSHA HARIDAS 9405477969	Eng.Opt.,
51	2267	SHRI	SHINDE AKSHAY BABURAO 9322191138	Eng.Opt.,
52	2268	SHRI	SHINDE ANIKET VISHNU 9096993125	Eng.Opt.,
53	2269	MISS	SHINDE ANKITA TANAJI 8956449926	Eng.Opt.,
54	2274	MISS	SHINDE PRIYANKA SANJAY 9370003075	Eng.Opt.,
55	2277	MISS	SHINDE VAIBHAVI DILIP 7387571573	Eng.Opt.,
56	2280	MISS	SURVE KIRTIKA RAJARAM 8412961985	Eng.Opt.,
57	2284	MISS	SURYAWANSHI VAISHANVI PRAKASH 9096264097	Eng.Opt.,
58	2298	MISS	YADAV PALLAVI DIPAK 9503644205	Eng.Opt.,
59	2305	SHRI	AWASARE OMKAR PRAMOD 9511229273	Eng.Opt.,
60	2307	SHRI	BANDGAR JIVAN TANAJI 9356822717	Eng.Opt.,
61	2310	MISS	BODAKE AKANKSHA APPASO 9503483083	Eng.Opt.,
62	2324	MISS	ERANDOLE RUTUJA DAGADU 9022900071	Eng.Opt.,
63	2326	SHRI	GAIKWAD VAIBHAV VITTHAL 8530187776	Eng.Opt.,
64	2328	MISS	GARALE DHANSHRI UTTAM 7774994751	Eng.Opt.,

B.A.PART I Roll Call (2020-21) Eng.Opt.

Sr. No	Roll No.	Gender	NAME OF THE STUDENTS	Subject
				ENG.
65	2335	SHRI	GURAV MILAN SHANKAR 8624825817	Eng.Opt.,
66	2350	SHRI	KALE SHUBHAM MAHADEV 7821949502	Eng.Opt.,
67	2368	SHRI	MALI SURAJ HANMANT 7057712640	Eng.Opt.,
68	2392	MISS	PATIL DHANASHRI TANAJI 9529193290	Eng.Opt.,
69	2406	SHRI	PATOLE AMIT BALASAHEB 9975329064	Eng.Opt.,
70	2430	SHRI	YALMAR PRASAD LALASO 8806493640	Eng.Opt.,
71	2433	SHRI	PHALLE SATYAJEET GANPAT 8600275192	Eng.Opt.,


HEAD
Department of English
P.D.V.P.College, Tasgaon.

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PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASAGAON

POST GRADUATE
DEPARTMENT OF CHEMISTRY

Date: -05/07/2016

Notice

All the students of M. Sc. I and II (Chemistry) are here by informed that, Seminars will start from 16th July 2016. The schedule of seminar is displayed on notice board.



Incharge Teacher



Head,
Dept. of Chemistry

HEAD
DEPARTMENT OF CHEMISTRY
P.D.V.P. COLLEGE, TASGAON.


PG DEPARTMENT OF CHEMISTRY
SEMINAR SCHEDULE 2016-17

M. Se. II

Sr. No.		Seminar Date
1	BholePallaviJalindar	16-07-2016
2	Chandanshive Mahesh Ajit	
3	ChandanshiveUmesh Siddhartha	
4	ChavanMohiniVinayak	
5	Desai ShubhamKrushnat	23-07-2016
6	GosaviShashankPrمود	
7	HonutagiVidyaChandrakant	
8	JadhavAratiBalaso	
9	JadhavSurajRangrao	30-07-2016
10	Jadhav Vijay Appasaheb	
11	Jagtap Mahesh Ankush	
12	JagtapRanjitShivaji	
13	JagtapShivanjaliRajendra	06-08-2018
14	KambleBipinHarishchandra	
15	Kamble Pankaj Maruti	
16	KambleSomeshwariGunwant	
17	KambleYogesh Suresh	13-08-2016
18	KarandeAjitkumarRanoba	
19	Khandekar Ramesh Murgyappa	
20	LipareKomalManohar	
21	LoharPoojaSambhaji	20-08-2016
22	MaladeManubai Krishna	
23	Mali AmolGurubal	
24	Mane ShaileshGhanhasham	

25	Mangsule Pankaj Maruti	27-08-2016
26	Mankapure Namrata Ashok	
27	Mohite Vishal Sadashiv	
28	Nadaf Suraj Alam	
29	Nimbalkar Hitesh Rajendra	03-09-2016
30	Padaval Dhanashri Laxman	
31	Patil Kajal Shivaji	
32	Patil Priyanka Ramchandra	
33	Patil Revati Jitendra	10-09-2016
34	Patil Rupesh Chandrakant	
35	Patil Vishal Kisan	
36	Patil-Dhavale Amol Anandrao	
37	Pawar Abhijeet Hanmant	17-09-2016
38	Pawar Gitanjali Chandrakant	
39	Pawar Indrajit Yashwant	
40	Sadakale Rahul Sambhaji	
41	Shetage Mukesh Shravan	24-09-2016
42	Suryawanshi Sujata Suresh	
43	Sutar Ravindra Namdev	
44		


Teacher Incharge


Head,
Dept. of Chemistry
HEAD
DEPARTMENT OF CHEMISTRY
P.D.V.P. COLLEGE, TARGAON

PG DEPARTMENT OF CHEMISTRY
SEMINAR TOPIC 2016-17

M. Sc. II

Sr. No.	Name of Student	Seminar Topic	Seminar Date
1	Bhole Pallavi Jalindar	Addition of HBR on carbon-carbon double bonds	16-07-2016
2	Chandanshive Mahesh Ajit	Identification of polarity of solvents	
3	Chandanshive Umesh Siddhartha	Nature of Nucleophiles	
4	Chavan Mohini Vinayak	Nucleophilic addition of Grignard reagent on carbonyl compounds	
5	Desai Shubham Krushnat	Chirality of organic compounds	23-07-2016
6	Gosavi Shashank Pramod	Inductive effect in acidic compounds	
7	Honutagi Vidya Chandrakant	Process of recrystallization	
8	Jadhav Arati Balaso	Stereospecific addition reactions	
9	Jadhav Suraj Rangrao	Orientation of di-substituted aromatic compounds	30-07-2016
10	Jadhav Vijay Appasaheb	Generation of carbocation	
11	Jagtap Mahesh Ankush	Pyrolytic elimination	
12	Jagtap Ranjit Shivaji	Umpolung	
13	Jagtap Shivanjali Rajendra	Isotopic labeling	06-08-2018
14	Kamble Bipin Harishchandra	Identification of order of reaction	
15	Kamble Pankaj Maruti	Molecularity of the aldol condensation reaction	
16	Kamble Someshwari Gunwant	Electrophilic substitution of carbon-carbon double bond	

17	Kamble Yogesh Suresh	Chemio and regioselectivity of Diels-Alder reaction	13-08-2016	
18	Karande Ajitkumar Ranoba	Tapping of the intermediates	20-08-2016	
19	Khandekar Ramesh Murgyappa	Effect of resonance on basicity		
20	Lipare Komal Manohar	Identification of reactive intermediates		
21	Lohar Pooja Sambhaji	Energy of activation		
22	Malade Manubai Krishna	Catalysis		
23	Mali Amol Gurubal	Aromatic SN¹ reaction		
24	Mane Shailesh Ghanhasham	Prochiral relationship		
25	Mangsule Pankaj Maruti	Chemical shift		
26	Mankapure Namrata Ashok	Testing of the possible intermediates		
27	Mohite Vishal Sadashiv	Geometrical isomerism		
28	Nadaf Suraj Alam	Shielding mechanism in NMR	27-08-2016	
29	Nimbalkar Hitesh Rajendra	Reactivity of compound by steric effect		
30	Padaval Dhanashri Laxman	Pressional frequency		
31	Patil Kajal Shivaji	Nature of leaving groups		
32	Patil Priyanka Ramchandra	Splitting of signal in NMR		
33	Patil Revati Jitendra	Aromatic SN² reaction		
34	Patil Rupesh Chandrakant	Competition between Elimination Vs. Substitution		
35	Patil Vishal Kisan	Ipsso attack		
36	Patil Dhavale Amol Anandrao	Effect of hydrogen bonding on IR stretching frequency		10-09-2016

37	PawarAbhijeetHanmant	Rearrangement of carbocation	17-09-2016
38	PawarGitanjaliChandrakant	Reactivity of compound by field effect	
39	PawarIndrajitYashwant	Phase transfer catalyst	
40	Sadakale Rahul Sambhaji	Molecular orbital symmetry	
41	ShetageMukeshShravan	Substitution reaction on SP^2 carbon atom	
42	SuryawanshiSujata Suresh	Jablonski diagram	
43	SutarRavindraNamdev	Cross-over experiments	
44			24-09-2016



Teacher Incharge



Dept. of Chemistry

HEAD

DEPARTMENT OF CHEMISTRY
P.D.V.P. COLLEGE, TARGAON

PC DEPARTMENT OF CHEMISTRY
Seminar Completion Report 2016-17
M. Sc. II

No.	Name of Students	Name of Seminar Topic	Synopsis	Date	Sign. of Student	Remark
1	Bhole Pallavi/Alindar	Addition of HBr on C=C bond	✓	16-7-2016	Bhole	Completed + C=O
2	Chandanshive Mahesh Ajit	Identification of polarity of solvent	✓	16-7-2016	Chandanshive	(R) → Slike
3	Chandanshive Umesh Siddhartha	Nature of Nucleophile	✓	16-7-2016	Chandanshive	(B) Slike
4	Chavan Mohini Vinayak	Nucleophilic addn of Gr. R. on carbonyl comp.	✓	16-7-2016	Chavan	(A) Slike
5	Desai Shubham Krushnati	Chirality	✓	23/7/16	Desai	(B) Slike
6	Gosavi Shashank Pramod	Inductive effect in acidic compounds	✓	23/7/16	Gosavi	(B) Slike
7	Honutagi Vidya Chandrakant	Process of Rehybridization	✓	23/7/16	Honutagi	(C) Slike
8	Jadhav Arati Balaso	Stereospecific addition reactions	✓	23/7/16	Jadhav	(B) Slike
9	Jadhav Suraj Rangrao	Orientation of disubstituted to. Comp.	✓	30-7-16	Jadhav	(C) Slike
10	Jadhav Vijay Appasaheb	Generation of carbocation	✓	30-7-16	Jadhav	(C) Slike
11	Jagtap Mahesh Ankush	Synthetic elimination	✓	30-7-16	Jagtap	(C) Slike
12	Jagtap Ranjit Shivraji	Umpolung	✓	06-08-16	Jagtap	(C) Slike
13	Jagtap Shivraj Rajendra	Isotopic labelling	✓	06-08-16	Jagtap	(C) Slike
14	Kamble Bipin Harishchandra	Identification of order of reaction	✓	06-08-16	Kamble	(C) Slike
15	Kamble Pankaj Maruti	Molecularity of aldol condensation	✓	06-08-16	Kamble	(C) Slike
16	Kamble Somyeshwari Gunwant	Electrophilic Aromatic Substitution	✓	06-08-16	Kamble	(C) Slike
17	Kamble Yogesh Suresh	Chromoselectivity & Regioselectivity of DAR	✓	13-8-16	Kamble	Repeat
18	Karande Ajit Kumar Ranoba	Drugging of Intermediate	✓	13-8-16	Karande	(B) Slike
19	Khandekar Ramesh Murgyaappa	Effect of resonance on basicity	✓	13-8-16	Khandekar	(A) Slike
20	Lipare Komal Manohar	Identification of reactive intermediate	✓	13-8-16	Lipare	(B) Slike
21	Lohar Pooja Sambhaji	Energy of Activation	✓	22/8/16	Lohar	(B) Slike
22	Malade Manubai Krishna	Catalysis	✓	22/8/16	Malade	(B) Slike

PG DEPARTMENT OF CHEMISTRY
Attendance Report 2016-17

Name of Teacher: Mr. Shinde S.K.

M. Sc. II

Name of Topic/Practical: Seminar

Sr. No.	Name of Students	16/7/16	23/7/16	30/7/16	6/8/16	13/8/16	22/8/16	29/8/16	24/8/16	3/9/16	10/9/16
1	Bhole Pallavi Jalindar	Bhole	Bhole	Bhole	Bhole	Bhole	Bhole	Bhole	Bhole	Bhole	Bhole
2	Chandanshive Mahesh Ajit	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive
3	Chandanshive Umesh Siddhartha	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive	Chandanshive
4	Chavan Mohini Vinayak	Murug	Murug	Murug	Murug	Murug	Murug	Murug	Murug	Murug	Murug
5	Desai Shubham Krushnat	Desai	Desai	Desai	Desai	Desai	Desai	Desai	Desai	Desai	Desai
6	Gosavi Shashank Pramod	Gosavi	Gosavi	Gosavi	Gosavi	Gosavi	Gosavi	Gosavi	Gosavi	Gosavi	Gosavi
7	Honutagi Vidya Chandrakant	Off	Off	Off	Off	Off	Off	Off	Off	Off	Off
8	Jadhav Arati Balaso	Atkulkar	Atkulkar	Atkulkar	Atkulkar	Atkulkar	Atkulkar	Atkulkar	Atkulkar	Atkulkar	Atkulkar
9	Jadhav Suraj Rangrao	Bhadwal	Bhadwal	Bhadwal	Bhadwal	Bhadwal	Bhadwal	Bhadwal	Bhadwal	Bhadwal	Bhadwal
10	Jadhav Vijay Appasaheb	Desai	Desai	Desai	Desai	Desai	Desai	Desai	Desai	Desai	Desai
11	Jagtap Mahesh Ankush	Mhatre	Mhatre	Mhatre	Mhatre	Mhatre	Mhatre	Mhatre	Mhatre	Mhatre	Mhatre
12	Jagtap Ranjit Shivaji	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap
13	Jagtap Shivanjali Rajendra	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap	Jagtap
14	Kamble Bipin Harishchandra	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble
15	Kamble Pankaj Maruti	Pankaj	Pankaj	Pankaj	Pankaj	Pankaj	Pankaj	Pankaj	Pankaj	Pankaj	Pankaj
16	Kamble Sureshwar Gurwant	Skamble	Skamble	Skamble	Skamble	Skamble	Skamble	Skamble	Skamble	Skamble	Skamble
17	Kamble Yogesh Suresh	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble	Kamble
18	Karande Ajitkumar Ranoba	Karande	Karande	Karande	Karande	Karande	Karande	Karande	Karande	Karande	Karande
19	Khandedkar Ramesh Murgyaappa	Ramesh	Ramesh	Ramesh	Ramesh	Ramesh	Ramesh	Ramesh	Ramesh	Ramesh	Ramesh
20	Lipare Komal Manohar	Lipare	Lipare	Lipare	Lipare	Lipare	Lipare	Lipare	Lipare	Lipare	Lipare
21	Lohar Pooja Sambhaji	Lohar	Lohar	Lohar	Lohar	Lohar	Lohar	Lohar	Lohar	Lohar	Lohar
22	Malade Manubai Krishna	Malade	Malade	Malade	Malade	Malade	Malade	Malade	Malade	Malade	Malade

Name of Teacher: Mr. Shinde S. K.

M. Sc. II

Name of Topic/ Practical:

Seminar's

Sr. No.	Name of Students	17/9/16	24/9/16	01/10/16					
1	Bhole Pallavi Jalindar	Bhole	Bhole	Bhole					
2	Chandanshive Mahesh Ajit	Chandanshive	(A)	(A)					
3	Chandanshive Umesh Siddhartha	Chandanshive	(A)	(A)					
4	Chavan Mohini Vinayak	Chavan	(A)	(A)					
5	Desai Shubham Krushnat	Desai	Desai	Desai					
6	Gosavi Shashank Pramod	Gosavi	Gosavi	Gosavi					
7	Honutagi Vidya Chandrakant	Honutagi	Honutagi	Honutagi					
8	Jadhav Arati Balaso	(A)	(A)	(A)					
9	Jadhav Suraj Rangrao	Jadhav	Jadhav	Jadhav					
10	Jadhav Vijay Appasaheb	Jadhav	Jadhav	Jadhav					
11	Jagtap Mahesh Ankush	Mahesh	Mahesh	Mahesh					
12	Jagtap Ranjit Shivaji	Jagtap	Jagtap	Jagtap					
13	Jagtap Shivaraj Rajendra	Jagtap	Jagtap	Jagtap					
14	Kamble Bipin Harishchandra	Kamble	Kamble	Kamble					
15	Kamble Pankaj Maruti	Kamble	Kamble	Kamble					
16	Kamble Sureshwar Gurwant	Kamble	Kamble	Kamble					
17	Kamble Yogesh Suresh	Kamble	Kamble	Kamble					
18	Karande Ajitkumar Ranoba	Karande	Karande	Karande					
19	Khandekar Ramesh Murgyaappa	Khandekar	Khandekar	Khandekar					
20	Lipare Komal Manohar	Lipare	Lipare	Lipare					
21	Lohar Pooja Sambhaji	Lohar	Lohar	Lohar					
22	Malade Manubai Krishna	Malade	Malade	Malade					

17/12/16 24/12/16 11/10/16

23	Mali Amol Gurubal	Amal	Amali	Amali
24	Mame Shailesh Gnanhasham	Sgnan	Sgnan	Sgnan
25	Mangule Pankaj Maruti	Pankaj	Pankaj	Pankaj
26	Mankapure Namrata Ashok	Namrata	Namrata	Namrata
27	Mohite Vishal Sadashev	Vishal	Vishal	Vishal
28	Nadaf Suraj Alam	Suraj	Suraj	Suraj
29	Nimbalkar Hitesh Rajendra	Hitesh	Hitesh	Hitesh
30	Padaval Dhanashri Laxman	Dhanashri	Dhanashri	Dhanashri
31	Patil Kajal Shivaji	Kajal	Kajal	Kajal
32	Patil Priyanka Ramchandra	Priyanka	Priyanka	Priyanka
33	Patil Revati Jitendra	Revati	Revati	Revati
34	Patil Rupesh Chandrakant	Rupesh	Rupesh	Rupesh
35	Patil Vishal Kisan	Vishal	Vishal	Vishal
36	Patil Dhavale Amol Anandrao	Dhavale	Dhavale	Dhavale
37	Pawar Abhijeet Hamant	Abhijeet	Abhijeet	Abhijeet
38	Pawar Gitarjali Chandrakant	Gitarjali	Gitarjali	Gitarjali
39	Pawar Indrajit Yashwanth	Indrajit	Indrajit	Indrajit
40	Sadakale Rahul Sambhaji	Rahul	Rahul	Rahul
41	Shetage Mukesh Shrawan	Mukesh	Mukesh	Mukesh
42	Suryawanshi Sujata Suresh	Sujata	Sujata	Sujata
43	Sutar Ravindra Namdev	Ravindra	Ravindra	Ravindra

44] Ghadage Puja Sangam
 45] Ghadage Mahesh Mohan

Completed
 01/10/2016
 S. Ghadage

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DIST- SANGLI

PG Department of Chemistry

SEMINAR

Date: 1/10/2016

Name of Student : Ghadage Mayuri Mohan
Class : M.Sc-II Roll No: 44
Subject : Organic chemistry
Title : sigmatropic rearrangement.

SYNOPSIS

sigmatropic Rearrangement

1] Definition

2] Types of sigmatropic Rearrangement.

- ① 1,3 sigmatropic rearrangement.
- ② 1,5 sigmatropic rearrangement.
- ③ 1,7 sigmatropic rearrangement.
- ④ 3,3 sigmatropic rearrangement.
- ⑤ 5,5 sigmatropic rearrangement.

Ghadage
Signature of Student

Shilpa
Incharge Teacher

Coordinator

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DIST- SANGLI

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SEMINAR

Date: 4/10/2016

Name of Student : Ghadage Puja Sangam
 Class : M.Sc II Roll No:
 Subject : ~~Norrish~~ reaction organic chemistry
 Title : ~~Norrish~~ reaction

SYNOPSIS

- 1] Introduction
- 2] Norrish type I - α
- 3] Norrish type II
- 4] Norrish I - β cleavage
 - 2] decarbonylation
 - 3] Intramolecular hydrogen abstraction
- 5] Norrish II - β cleavage of α & β carbon
 - 6] α substituted ring closing

Ghadage
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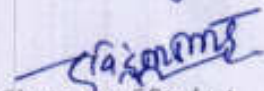
SEMINAR


Date: 1/10/2016

Name of Student : Ravindra Namdev Sutar
 Class : MSc II Roll No:
 Subject : Organic chemistry
 Title : Ionisation Method in Mass spectrometry

SYNOPSIS

- Introduction
- What is mean by the ionisation?
- Ionisation Method
 - 1) Electron ionisation Method
 - 2) chemical ionisation Method
 - 3) Desorption Ionisation Method
 - i) Secondary ion mass spectrometry
 - ii) Fast Atom bombardment
 - iii) MALDI
 - 4) Electrospray ionisation.


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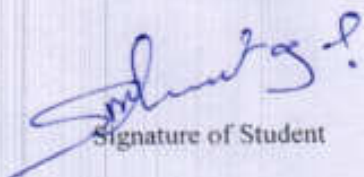
Date: 1/10/2016

Name of Student : Shetage Mukesh Shrawan
 Class : M.Sc. II Roll No:
 Subject : organic chemistry
 Title : Substitution Reaction on sp^2 carbon atom

SYNOPSIS

- i) Introduction - i) Combination of substitution
 - ii) origin of substitution
 - iii) condition for substitution
- ii) Type of sp^2 carbon atom
 - i) aliphatic ii) aromatic
- iii) substitution Rean on aliphatic carbon (sp^2) atom
 - i) carbonyl comp. and it's derivative
 - ii) carbonyl oxygen lost example.
- iv) aromatic sp^2 carbon atom :-

Electrophilic Reaction <ol style="list-style-type: none"> i) nitration ii) Sulphonation iii) Friedel craft Rean 	Nucleophilic substi. <ol style="list-style-type: none"> i) addition, elimination ii) diazonium ion Reaction
--	---


 Signature of Student


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SEMINAR

Date: 17/9/2016

Name of Student : Miss. Gitanjali Chandrakant pawar.
 Class : M.SC-II Roll No:
 Subject : Organic chemistry
 Title : Reactivity of comp. by field effect.

SYNOPSIS

Introduction :-

Field effect depends on two effects.

a) Inductive effect (I)

i) Donating inductive effect (+I)

ii) Withdrawing inductive effect (-I)

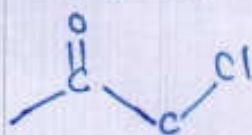
b) mesomeric effect (M) / Resonance effect (R)

i) Donating (+R)

ii) Withdrawing (-R)

Field effect :- It is also called as Direct effect, through space.

Example :-



Field effect

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No Field effect

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SEMINAR

Name of Student : Sadakale Rahul Sambhaji
Date: 24/9/16
Class : MSc-II Roll No:
Subject : Chemistry
Title : Symmetry in π molecular orbital.

SYNOPSIS

Symmetry in π molecular orbital.

- Introduction: -

C_2 -Symmetry - e

m-Symmetry -

examples,

(i) Ethylene

(ii) 1,3 dibutadiene

(iii) 1,3,5 hexatriene

R. S. Sadakale.
Signature of Student

S. Shinde
Incharge Teacher

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DIST- SANGLI

PG Department of Chemistry

SEMINAR

Date: 24-9-16

Name of Student : Surjawanshi Sujata Suresh
Class : MSc-II Roll No: 4193
Subject : organic chemistry
Title : Jablonski diagram

SYNOPSIS

- ① Introduction of Jablonski diagram
- ② Spin orientation of absorption of a light photon
- ③ Fluorescence -
 - (a) Non radiative and radiative transition
 - (b) spectrum of Absorption and Fluorescence
 - (c) characteristics
 - (d) Factor
 - (e) Application
- ④ phosphorescence -
 - (a) characteristics
 - (b) Factor
 - (c) Application
- ⑤ Biomolecular quenching

Signature of Student

Sujata

Incharge Teacher

Sujata

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DIST- SANGLI

PG Department of Chemistry

SEMINAR

Date: 17/19/16

Name of Student : Patil Revati Jitendra
Class : M.Sc II. Roll No:
Subject :
Title : Aromatic S_N2 reactⁿ.

SYNOPSIS

1. Debinatⁿ
2. Type of reaction :- Aromatic S_N2 reactⁿ
Aliphatic S_N2 reactⁿ.
3. Aromatic S_N2 reactⁿ occurs in two conditⁿ.
 - 1) Attaching strong electron withdrawing gr.
 - 2) providing a better leaving gr.
- 4) Applicatⁿ
- 5) Limitatⁿ.

Revati
Signature of Student

S. Shik
Incharge Teacher

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DIST- SANGLI

PG Department of Chemistry

SEMINAR

Date: 24/9/2018

Name of Student : Pawar Imbrajit Yashwant
 Class : M.Sc-II Roll No:
 Subject : organic chemistry
 Title : phase transfer catalyst

SYNOPSIS

Phase transfer catalyst

- 1) Introduction
- 2) type
 - a) quaternary salt
 - b) crown ether
- 3) Applications

Imbrajit

Signature of Student

[Signature]

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Shri Swami Vivekanand Shikshan Santha's
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SEMINAR

Date: 10/09/2016

Name of Student : Patel Amal Anandrasao
 Class : M.Sc. II Roll No:
 Subject : Chemistry
 Title : effect of H-bonding on IR stre. freq.

SYNOPSIS

- ① Factors responsible for shifting ~~freq~~ of frequency
- ② What is H-bond?
- ③ Two types
 - Ⓐ Intramolecular
 - Ⓑ Intermolecular
- ④ Examples.


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SEMINAR

Date: 10/9/2016

Name of Student : Patil Vishal Kisan
Class : M.Sc. II Roll No:
Subject : Chemistry
Title : IPso Attack.

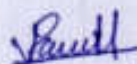
SYNOPSIS

1) Introduction :- general information
definition.

2) Types of IPso substitution ;

- 1) Electrophilic
- 2) nucleophilic

3) Examples.



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SEMINAR

Date: 10/09/2016

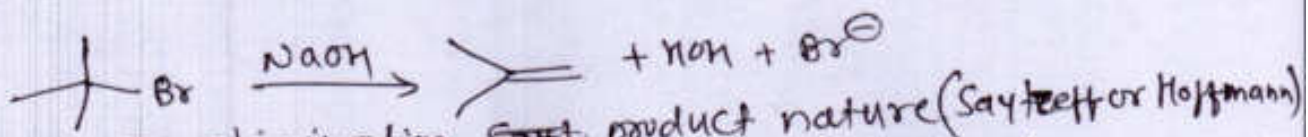
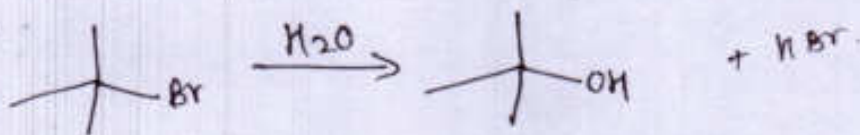
Name of Student : Patil Rupesh Chandrakant
Class : M.Sc.II Roll No:
Subject : Organic Chemistry
Title : Competition betⁿ Substitution & Elimination.

SYNOPSIS

① Introduction - ① Defⁿ of Substitution & Elimination
② example.

② How alkyl halide gives nucleophilic substitution as well as elimination reactⁿ.

eg.



③ ~~Rate~~, In elimination ~~sub~~ product nature (Saytzeff or Hoffmann)

④ How to nucleophilic affect elimination versus substitution

- ① Basicity
- ② Size
- ③ Temperature
- ④ Solvent effect.

Signature of Student

Patil

[Signature]

Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 3/09/16

Name of Student

: Padaval Dhanashri Laxman.

Class

: MSc-II

Roll No:

Subject

:

Title

: precessional frequency.

SYNOPSIS

- ① Defination of precessional frequency.
- ② Nuclear magnetic moment.
- ③ precessional motion.
- ④ factor affecting precessional frequency
- ⑤ Example.



Signature of Student



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Coordinator

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SEMINAR

Date: 3-9-16

Name of Student : Patil Priyanka Ramchandra.
Class : M.Sc-II Roll No:
Subject :
Title : splitting of signal in NMR.

SYNOPSIS

- ① spin-spin splitting
- ② $n+1$ rule
- ③ The origin of spin-spin splitting
- ④ Pascals triangle

Signature of Student

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Coordinator

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SEMINAR


Date: 3-09-16

Name of Student : Patil Kajal Shivaji
Class : M.Sc II Roll No:
Subject :
Title : Nature of leaving group.

SYNOPSIS

- ① Introduction
- ② Definition
- ③ Properties of leaving group
- ④ Types of leaving group.
 - stable leaving group
 - unstable leaving group.


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Shri Swami Vivekanand Shikshan Santha's
PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASGAON
DIST- SANGLI

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SEMINAR

Date: 27-08-2016

Name of Student : Nadaf Suraj Alam
Class : M.Sc-II Roll No: .
Subject : Organic Chemistry
Title : Shielding Mechanism in NMR spectroscopy

SYNOPSIS

- Shielding Mechanism.
 - Inductive & Resonance effect and electronegativity.
 - Effect of Halogen.
 - No. of 'H' on carbon.
 - Molecular stru. and chemical shift
 - Hydrogen on Aromatic ring.
 - Anisotropic effect-
 - Alkyne
 - Aldehyde
 - Confirmation and configuration.

Nadaf
Signature of Student

Shikha
Incharge Teacher

Coordinator

Shri Swami Vivekanand Shikshan Santha's
PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASGAON
DIST- SANGLI

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SEMINAR

Date: 3/9/016

Name of Student Nimbalkar Hitesh Rajendra

Class M.Sc II

Roll No:

Subject Chemistry

Title Reactivity due to steric effect

SYNOPSIS

- ① What is steric effect
- ② How steric effect occurs
- ③ Reactivity due to steric effect in S_N1 & S_N2
- ④ Reactivity $-||-$ in $C=O$
- ⑤ $-||-$ in acidity & basicity
- ⑥ $-||-$ in axial & eq. conformaⁿ
- ⑦ How to remove steric strain.

[Signature]
Signature of Student

[Signature]
Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 27/8/16

Name of Student : Mankapure Namrata Ashok
Class : M.Sc - II Roll No:
Subject :
Title : Testing of possible intermediate


SYNOPSIS

1) Introduction

Three methods for detection of intermediate.

- i) Isolation
- ii) Physical method
- iii) Trapping.

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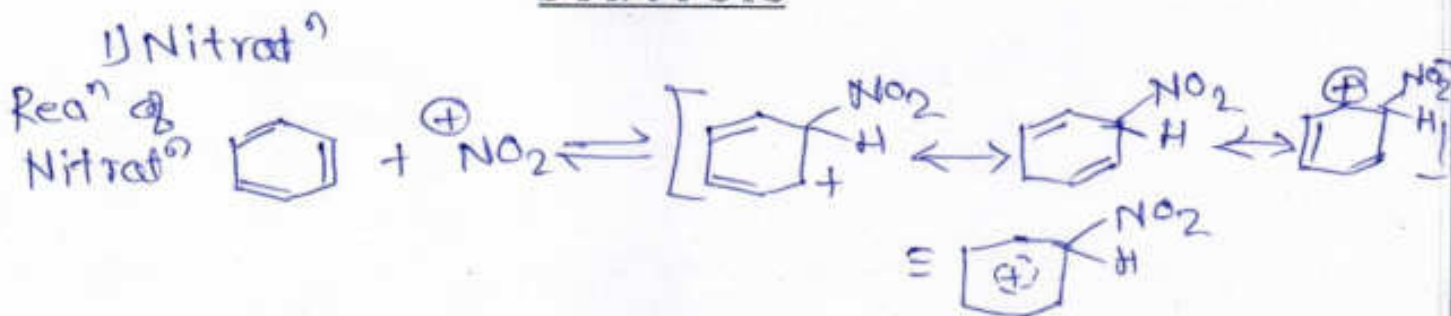
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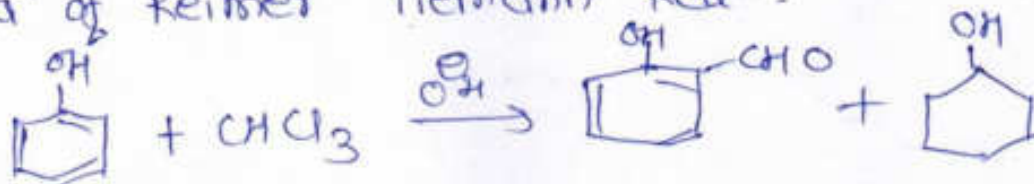
Date: 6/8/16

Name of Student : Kamble Someshwari Gunward
Class : M.Sc II Roll No:
Subject : Organic Chemistry .
Title : Aromatic substitutⁿ reaⁿ .

SYNOPSIS



2) Reaⁿ of Reimer Tiemann Reaⁿ .



Mechanism - 1) Generatⁿ of dichlorocarbene
2) formatⁿ of salicylaldehyde

Kamble
Signature of Student

Shinde
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SEMINAR

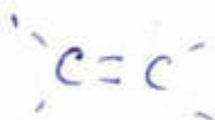
Date: 27/08/2016

Name of Student : Mohite Vishal Sadashiv
Class : M.sc-II Roll No:
Subject : Organic Chemistry
Title : Geometrical Isomerism

SYNOPSIS

- Geometrical Isomerism
- Introduction
- Classification
- Nomenclature

eg



- Determination of Configuration of geometrical Isomerism
- (a) melting point
- (b) NMR spectra

Signature of Student

Mohite

Shik

Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 27/8/2016

Name of Student : Mane shailesh Ghanasham
Class : M.Sc-II Roll No:
Subject : Organic chemistry
Title : Prochiral Relation ship

SYNOPSIS

Prochiral Relationship

- 1) Introduction
- 2) Definition
- 3) Type :
 - 1) Homotopic gr./ligand/atom:-
 - 2) Heterotopic gr./ligand/atom:-
 - i) Enantiotopic -:
 - ii) Diastereotopic -:


Signature of Student


Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 24-8-16

Name of Student : Mali Amol Gurubal
Class : M.Sc-II Roll No:
Subject Title : Aromatic S_N1 Reaction
Title subject : Organic Chemistry

SYNOPSIS

Introduction.

- 1) Defⁿ - Aromatic nucleophilic substitution - Replacement of Hydrogen
- 2) Types of Aromatic nucleophilic substitution
- 3) Aromatic S_N1 - Reaction
- Mechanism.
- 4) Evidences of Aromatic S_N1 Reaction
- 5) Examples of Aromatic S_N1 Reaction
- 6) Reactivity of Aromatic S_N1 Reaction
- 7) Stability of Aromatic S_N1 Reaction

Amali

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S. D. Dite

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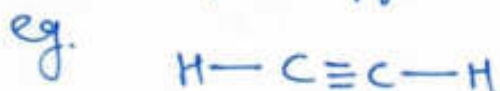
SEMINAR

Date: 27/08/2016

Name of Student : Pankaj Manuti Mangsule
Class : M.Sc. II Roll No:
Subject : Organic Chemistry
Title : Chemical Shift

SYNOPSIS

- ① Introduction
- ② The mechanism of absorption
- ③ Shielding and De-Shielding
- ④ Anisotropic effect



- ⑤ Factor affecting chemical shift
 - i) Electronegativity
 - ii) Hybridisation
 - iii) Resonance
 - iv) Inductive effectwith suitable example

P. Mangsule
Signature of Student

S. Shinde
Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 22/8/2016

Name of Student : Maode Mandabai Krishna.

Class : M Sc. II Roll No:

Subject :

Title : Catalysis.

SYNOPSIS

- ① Introduction
- ② Catalysis
- ③ Type of catalysis.
- ④ Application.

Maode
Signature of Student

S. Shinde
Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 22 - 8 - 16

Name of Student : Lohar Pooja Sambhaji
Class : M.Sc - II Roll No:
Subject :
Title : Energy of Activation.

SYNOPSIS

- 1) Introduction.
- 2) Energy profile
- 3) Energy profile for rea^\ddagger with single transition state.
- 4) Energy profile for rea^\ddagger with an intermediate.
- 5) Exothermic & Endothermic rea^\ddagger .


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SEMINAR

Date: 6/8/16

Name of Student : Jagtap shivanjali Rajendra
Class : M.Sc II Roll No:
Subject : organic chemistry
Title : Isotopic Labeling

SYNOPSIS

Isotopic Labeling

- 1) Introduction
- 2) Hydrolysis of ester
- 3) claisen Rearrangement
- 4) Discovery of Reaction Intermediate
- 5) Uses.

Jagtap

Signature of Student



Incharge Teacher

Coordinator

PG Department of Chemistry

SEMINAR

Date: 30/07/2016.

Name of Student : Jadhav Vijay Appasaheb.
Class : M.Sc. II Roll No:
Subject : Organic chemistry.
Title : Generation of carbocation.

SYNOPSIS

Generation of carbocation -

- 1] Introduction -
- 2] stability of carbocation -
- 3] factors affects stability of carbocation -
 - i] Hyper-conjugation effect -
 - ii] Resonance effect -
 - iii] +I Inductive effect -
 - iv] mesomeric effect -
- 4] Generation of carbocation -
 - i] from alcohol -
 - ii] from alkyl halide -
 - iii] from acyl halide -
 - iv] from alkene -
 - v] from diazonium ion -
- 5] Reaction of carbocation -

Jadhav.
Signature of Student

S. Patil
Incharge Teacher

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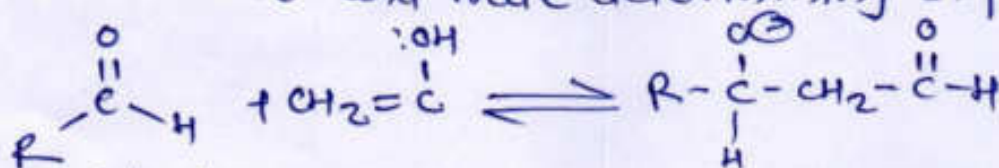
SEMINAR

Date: 06/08/2016

Name of Student : Kamble Pankaj Maruti
 Class : M.Sc -II Roll No:
 Subject : organic chemistry
 Title : Molecularity of aldol condensation.

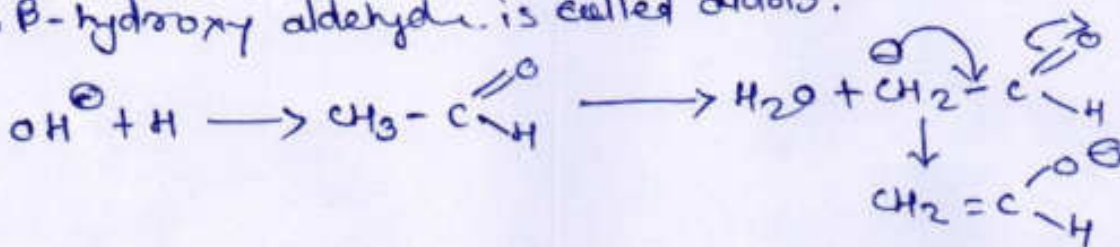
SYNOPSIS

① Introduction, ② molecularity, and ③ mole. of aldol condensation
 - The molecularity of is defined as the number of atoms or molecules which collide together at one and the same time and rate determining step.



aldehyde order = 2, molecularity = 2

The aldehyde having α -hydrogen undergoes a self condensation on warming with dil. or mild base to give in β -hydroxy aldehyde. is called aldols.



Rate = $k_2 [\text{CH}_3\text{CHO}] [\text{OH}^\ominus]$
 molecularity = 3, order = 2.

Pankaj
 Student

J. Patil
 Incharge Teacher

Coordinator

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SEMINAR

Date: 06/08/2016

Name of Student : Kamble Bipin Harishchandra
Class : M. Sc II Roll No:
Subject : Organic chemistry
Title : Identification of Oxides of carbon

SYNOPSIS

- 1) Introduction -
- 2) What is Oxides of carbon
- 3) How to identify the oxides of carbon.
- 4) Oxides of carbon
 - (i) 1st oxide (ii) 2nd oxide (iii) 3rd oxide
 - (iv) 2^{co} oxide (v) Pseudo-first oxide,
 - (vi) Fraction of oxide (vii) Mixed oxide
 - (viii) Negative oxide of carbon
- 5) Examples of oxides of carbon (mainly name of carbon - Carbohydrates)


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SEMINAR

Date: 6-8-2016

Name of Student : Jagtap Ranajit Shivaji
Class : M.Sc II Roll No: -
Subject : Organic chemistry
Title : Umpolung.

SYNOPSIS

Umpolung

- 1) Introduction
- 2) Examples / Reagents
 - I) Epoxide as a umpolung reagent.
 - II) 1,3 Dithianes as a umpolung reagent.
 - III) cyanide as a umpolung reagent
 - IV) Nitro compounds as umpolung reagent.


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SEMINAR

Date: 30/09/16.

Name of Student : Jagtap Mahesh Ankuash
Class : M.Sc-II Roll No:
Subject : organic chemistry
Title : pyrolytic elimination

SYNOPSIS

Pyrolytic Elimination :-

- 1) Introduction.
- 2) conditions
- 3) examples -
 - 1) Xanthate
 - 2) ester
 - 3) 3° amine oxide
- 4) Applications.


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Date: 13/8/2016

Name of Student : Komal Manohare Lpate
Class : MSc II Roll No:
Subject : organic chem.
Title : Identification of Reactive intermediate

SYNOPSIS

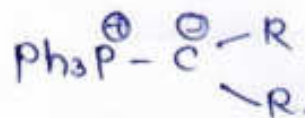
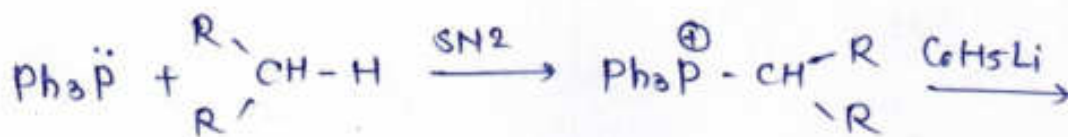
* Ylide -

1) phosphorous ylide

1) Reaⁿ

2) Mechanism

3) Stability.



phosphorous ylide.


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SEMINAR

Date: 13-8-2016

Name of Student : Karande Ajitkumar Ranoba.
Class : MSc-II Roll No: -
Subject : organic chemistry.
Title : Trapping of intermediate.

SYNOPSIS

- ① Definition of intermediate.
- ② Trapping of intermediate and examples.
 - ① Trapping of carbocation
 - ② Trapping of carbenes
 - ③ " " nitrenes
 - ④ " " Benzenes
- ③ Examples.


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SEMINAR

Name of Student : Kamble Yogesh Suresh
Class : M.Sc - II Roll No:
Subject : organic chemistry.
Title : Stereochemistry & Regioselectivity. D.A.R.
Date: 13-8-2016

SYNOPSIS

- 1) Regioselectivity.
 - 1) Diels-Alder Reaction.
- 2) Stereochemistry of Diels-Alder Reaction.
 - 1) Regioselectivity.


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SEMINAR

Date: 13/08/2016

Name of Student : Khandekar Ramesh Murgappa
Class : M.Sc-II Roll No:
Subject : organic chemistry
Title : Effect of Resonance on Basicity.

SYNOPSIS

Effect of Resonance on Basicity.

Introduction of Basicity.

Basic compounds.

what is the effect of Resonance on Basic Compound.

Effect of Substitutions on Basic compound.

i) electron donating group

ii) electron withdrawing group.

with Examples.



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SEMINAR

Date: 30/07/016

Name of Student : Jadhav Suraj Ramrao
Class : M.Sc-II Roll No: -
Subject : organic chemistry
Title : Orientation of disubstituted aromatic comp.

SYNOPSIS

① Substitution

- i) Nucleophile
- ii) electrophile

② orientation of the second group.

- i) electron donating group
- ii) electron withdrawing group.

③ orientation of the third incoming group.

case i) when both gr. are electron donating

case ii) one electron donating & one group of the other electron withdrawing.

case iii) when both gr. are electron withdrawing.
i) with examples.

S. Jadhav

Signature of Student

S. Jadhav

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Date: 16/7/2016

Name of Student Chandanhive Umesh Siddhartha

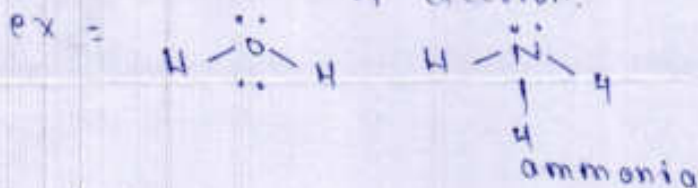
Class : M.Sc-II Roll No:

Subject : Organic Chemistry

Title : Nature of Nucleophiles.

SYNOPSIS

Nucleophiles :- nucleus having are electron donor species. they may be have neutral having non bonding lone pair of electron.



then attack of substrate is also called as ambident nucleophiles
ex cyanide

TYPE of nucleophiles

- ① Hard & soft nucleophiles
- ② π - bond
- ③ σ - bond

Chandanhive

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
SEMINAR


Date: 16-7-17

Name of Student : Bhole Pallavi Jalindar
Class : M.Sc II Roll No:
Subject : Organic chemistry
Title : Addition of HBr on C=C bond

SYNOPSIS

Introduction about addition reaction
Addition reaction by Markownikoff's rule
Addition reaction by Peroxide effect
i.e. Anti-Markownikoff's rule.
Mechanism:


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SEMINAR

Date: 16-7-2016

Name of Student : Chavan Mohinee Vinayak
 Class : M.Sc II Roll No:
 Subject : Org. Chem.
 Title : Nucleophilic additⁿ of G.R. on carbonyl comp.

SYNOPSIS

1. Prepⁿ of G.R.
2. Nucleophilic additⁿ Rⁿ
 Additⁿ to formaldehyde, aldehyde, ketone
3. Effect of steric hindrance.
4. Additⁿ to cyclic ketone
5. Additⁿ to ketenes
6. Additⁿ to Isocyanates
7. Additⁿ to CO₂



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SEMINAR

Date: 16/7/2016

Name of Student : Chandanshive mahesh. Ajit
 Class : msc - II Roll No:
 Subject : organic Chemistry
 Title : Identification of polarity of solvent.

SYNOPSIS

- 1) solvent - definition.
- 2) Type of polar solvent.
- 3) Physical property of ^{common} polar solvent.

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SEMINAR

Date: 23/07/2016

Name of Student : Desai Shubham Krishnat
 Class : M.sc-II Roll No:
 Subject : Chirality organic chemistry
 Title : Chirality

SYNOPSIS

Chirality

- 1) Introduction of chiral & achiral molecule
- 2) Chiral centre without carbon atoms like
 - 1) N
 - 2) P
 - 3) Si etc
- 3) molecular symmetry
 - 1) simple / axis of symmetry
 - 2) plane of symmetry
 - 3) Centre of symmetry
 - 4) improper axis of symmetry

Desai
Signature of Student

Shik
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SEMINAR

Date: 23/7/2016

Name of Student : Homnabagi. vidya. chandrakant
Class : M.Sc-II Roll No:
Subject : Organic chemistry
Title : Process of Recrystallization

SYNOPSIS

Title: Process of Recrystallisation

- 1) Introduction
- 2) Recrystallisation Process
- 3) Types of Recrystallisation
- 4) characteristics
- 5) common solvents
- 6) solvent selection
- 7) Recrystallisation in low temp
- 8) Use of decolourising carbon
- 9) common Problems are observed in Recrystallisation Process.



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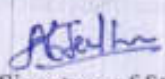
SEMINAR

Date: 23/7/16

Name of Student : Jadhav Arti Balasa
 Class : MSc-II Roll No:
 Subject : Organic Chemistry
 Title : stereospecific addition Reaⁿ.

SYNOPSIS

- ① Introduction
- ② Stereospecific addition Reaⁿ.
- ③ Bromination of alkene (anti addition)
E & Z Butene
- ④ Examples.


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SEMINAR

Date: 23/7/2016

Name of Student : Gosavi Shashank Pramod
Class : M.Sc II Roll No: .
Subject : Organic chemistry.
Title : Inductive effect in acidic compound

SYNOPSIS

* Inductive effect :-

Defn -
characteristics -

* Types of inductive effect :-

- i) - I effect -
- ii) + I effect -

* Applications :-

- i) Acid strength of Aliphatic carboxylic acid -
Acetic acid, formic acid, Monochloroacetic acid,
trichloroacetic acid
- ii) Aromatic carboxylic acid -
Benzoic acid.

Gosavi
Signature of Student

Shik
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Synthesis, Properties and structure of Boranes-

Boron is one of the III A group element. It is p block element. Its ground state E.C is $1s^2 2s^2 2p^1$.

1) Abundance in earth crust is 8×10^{-6}

2) Atomic radius = 0.80 \AA

3) Electronegativity = 2.0

4) Ionisation potential = 8.33

5) M.P = 2300°C ; B.P = 2500°C

6) Density = 2.4 gm/lit .

Source - Borax = $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$

Keratin = $\text{Na}_2\text{O}_4\text{H}_7 \cdot \text{H}_2\text{O}$

Cafemanik = $\text{Ca}_2\text{B}_4 \cdot 5\text{H}_2\text{O}$

Boron is non-metallic due to its small size and comparatively possess high ionisation potential. The M.P and B.P are extremely high due to strong bonding in between individual in solid state. Boron can not combine directly with hydrogen to form hydrides but no. of boron hydrides are known. These are called "Boranes." These are gaseous substance at ordinary temp. B_2H_6 i.e. diborane is most common boron hydride. The boranes are given by formula B_nH_{n+4} nido boranes and B_nH_{n+6} arachano boranes and closo boranes.

Boranes were first prepared by Alfred Stock in 1936-72.

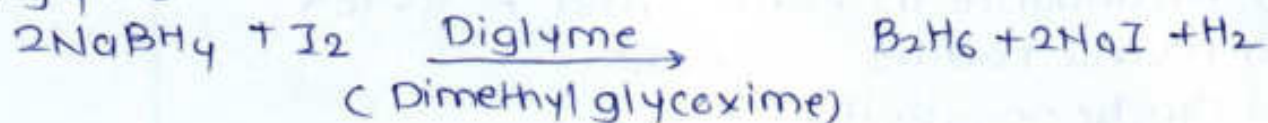
The most important hydride of boron is diborane i.e. B_2H_6 .

• Synthesis of boranes-

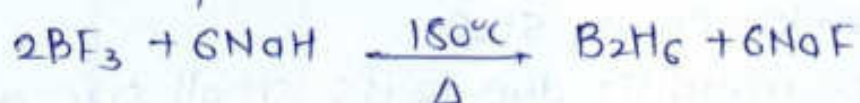
1) Magnesium boride on treatment with dil HCl or phosphoric acid a volatile mixture of several boranes obtained. These borane hydride are very sensitive & hence scientist Stock used high vacuum technique for the separation of boranes and studying their properties. Boranes are much sensitive to air and moisture. Diborane can be isolated in very small amount.



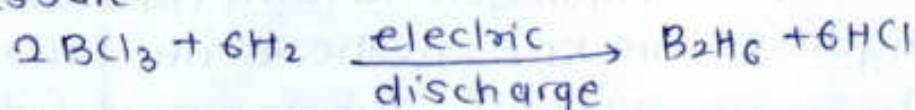
2) By the reaqⁿ of iodine on sodium borohydride in diglyme.



4) By reducing gaseous boron trifluoride with sodium hydride at 180°C



5) By passing silent electric discharge through a mixture of hydrogen and boron trichloride at a low pressure.



Properties -

Diborane is colourless gas and must be handled with care as it is highly reactive. It catches fire spontaneously in air & explodes with dioxygen.

i) It is stable only at low temp when heated at 100°C to 250°C in closed vessel, it gives higher borane.



ii) Action of oxygen - Large amount of energy is released.



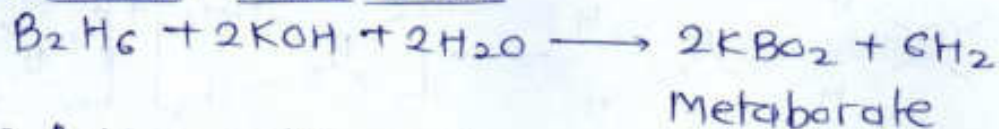
$$\Delta H = -2165 \text{ kJ/mole}$$

iii) Hydrolysis - It gets hydrolysed readily to give

boric acid.



iv) Action with alkali -

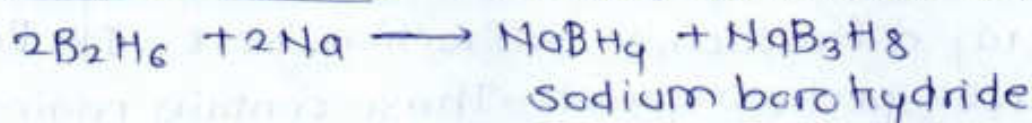


v) Action with carbon monoxide -

It reacts with Co at 100°C & 20 atm pressure to form carbonyl H_3BCo .



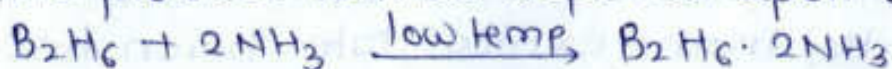
vi) Action of metal - It reacts with Na, Ca, K



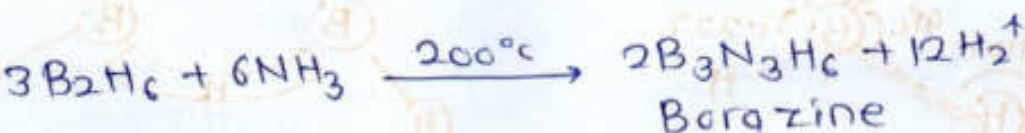
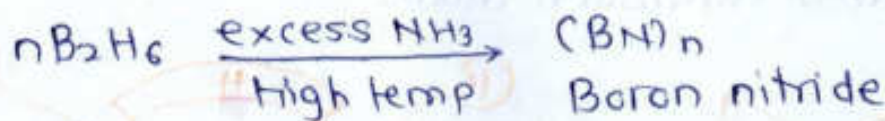
vii) Action of halogens - It reacts with halogens under different condition to form halodiboranes.



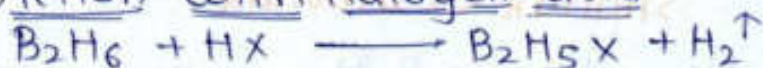
viii) Action of ammonia - Diborane reacts with NH_3 , the product formed depends upon condition of reaction.



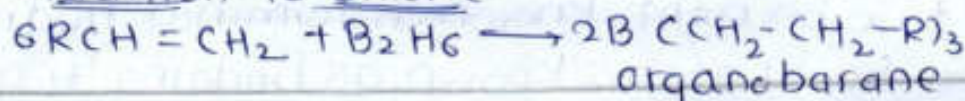
diammoniate of diborane



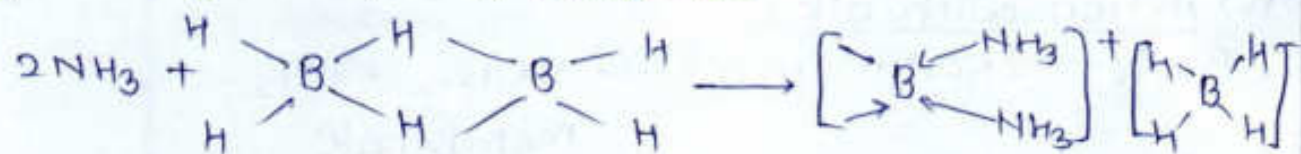
ix) Action with halogen acid



x) Addition to alkene -



The compound $B_2H_6 \cdot 2NH_3$ obtained is ionic & exist as, $[H_3N \rightarrow BH_2 \leftarrow NH_3]^+ [BH_4]^-$. It's formation can explained by assuming unsymmetrical cleavage of BH_3 bridges of B_2H_6 as-

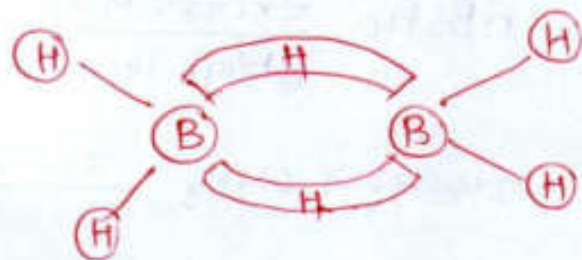
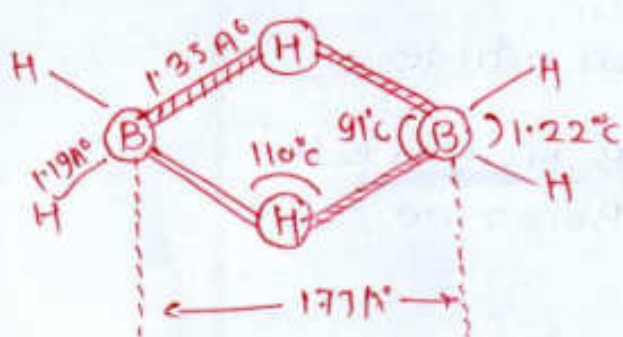


Structure-

The boranes are electron deficient molecule. They have insufficient no. of e^- to permit the linking of two boron and 6 hydrogen atom in diborane by e^- pair bond. The structure of diborane is mostly studied with x-ray diffraction, e^- diffraction, NMR, IR, Raman microwave spectroscopy etc. These contain mainly following types of bonding in boranes.

- 1) Normal $2c-2e^-$ B-H terminal bond
- 2) Direct $3c-2e^-$ hydrogen bridge bond
- 3) open $3c-2e^-$ Boron bridge bond
- 4) closed $3c-2e^-$ Boron bridge bond

Boron hydrides are all e^- deficient molecule in which one valence shell has more atomic orbitals. In diborane (B_2H_6) in the excited state boron has 3 half filled orbitals. It has $3e^-$ available for sharing. It can link to 3 hydrogen atoms.



Structure of diborane

It is diamagnetic, all e^- in B_2H_6 must paired. The 4H atom two on right known as terminal hydrogen. The other two H atoms known as bridging 'H' atom.

There are not enough valence e^- to form conventional two e^- bonds betn all the adjacent pairs of atom & so this comp is termed e^- -deficient. In diborane there are 12 valence e^- , three from each boron & 6 from the hydrogen atoms. The terminal B-H bond are normal covalent bonds. i.e. 2c-2e-bonds. 3c-2e- include B-H-B.

Cluster Valence Electron Theory-

This is a method for determining the geometry borane. To use this method, the total no of valence e^- used in cluster binding must be determined. There are

3 valence e^- for each B atom,
1 valence e^- for each H atom and
the total charge on the complex must be added in. The structure is determined according to the following eqn

- closo - $4n + 2$ valence electron
- nido - $4n + 4$ valence electron
- arachno - $4n + 6$ valence electron

$n = \text{no. of boron atom in the cluster}$

e.g. B_2H_6

$$\text{No of B} = 2 \times 3 = 6$$

$$\text{No. of H} = 6 \times 1 = 6$$

$$\text{Total no. of B and H} = 6 + 6 = 12$$

$$\text{This follows, } \therefore 4n + 4 = 4(2) + 4 = 12$$

\therefore The B_2H_6 is nido borane.

2) B_4H_{10}

$$\text{No of B} = 4 \times 3 = 12$$

$$\text{No of H} = 10 \times 1 = 10$$

$$\text{Total no. of B & H in } B_4H_{10} = 12 + 10 = 22$$

According to cluster valence e^- -theory it follows, $4n + 6 = 4(4) + 6 = 22$

\therefore The B_4H_{10} is arachno borane.

Wade's Rule-

According to Wade's rule,
Each B-H unit contributes $2e^-$.
negative charge contributes $1e^-$
and for additional H atom $1e^-$.
According to Wade's rule, there are 3 types.

1) Closoboranes - $(m+1)e^-$ pairs

m = total no. of B atom
e.g. $B_6H_6^{2-}$ = $m=6$
= $(6+1)e^-$ pair = $7e^-$ pair

2) Nido borane $(m+2)e^-$

e.g. B_5H_9 - B-H unit = $5 \times 2 = 10$
Additional H = $4 \times 1 = \frac{4}{14}$
 $(m+2)e^- = (5+2)e^- = 7e^-$ pairs

3) Arachnorbanes $(m+3)e^-$

e.g. B_5H_{11} = B-H unit = $5 \times 2 = 10$
Additional H = $6 \times 1 = \frac{6}{16}$
 $(m+3) = (5+3) = 8e^-$ pair

Synthesis, Properties and Structure of carborane-

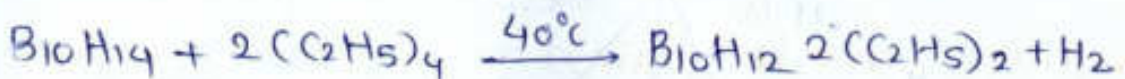
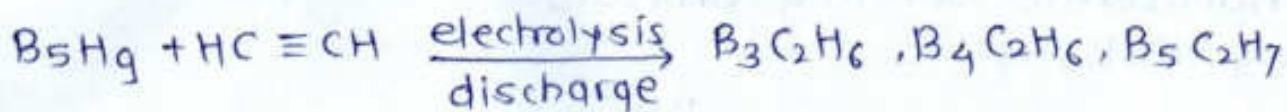
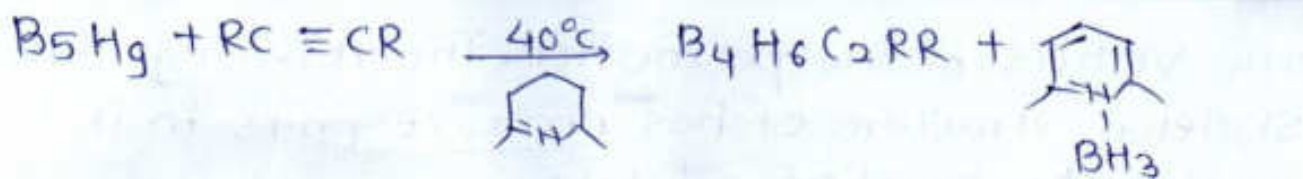
A carborane may be defined as a "polyhedral type containing B, C, & H atom in the polyhedral framework." It may be derived from a known boron hydride anion by replacing $B-H^{\ominus}$ with C-H.

e.g. $B_{10}C_2H_{12}$ is analogous to $[B_{10}H_{12}]^{2-}$ or from a boron hydride by replacing B-H bridging H with C-H.

e.g. $B_4C_2H_8$ analogous to B_6H_{10}

Synthesis-

Carborane generally been prepared by $1e^-$ of acetylene or substituted acetylene with boron hydrides.



Structural conformation for these compound comes from NMR spectra. Thus, for $B_3C_2H_6$ appears to a trigonal bipyramidal while $B_4C_2H_6$ have distorted octahedral skeleton.

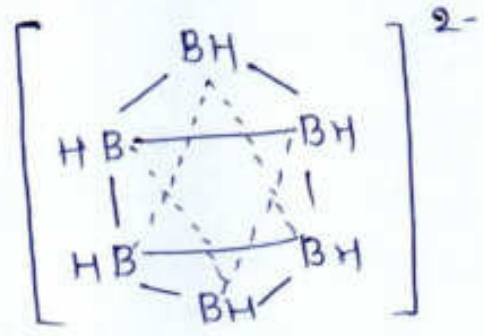
Three structurally different types of carboranes are closo, nido and arachano.

Structure-

- In carboranes, according to wade's rule
- Each B-H unit contributes $2e^-$
- negative charge $1e^-$
- Additional H atom $- 1e^-$
- Each C-H unit $3e^-$

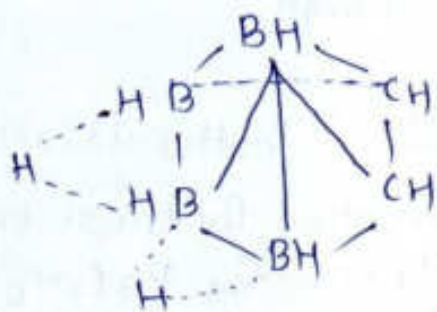
1) closo carboranes- They have a completely closed triangular polyhedral structure in which all the vertices of the triangular polyhedron are occupied by 'B' and 'C' atoms.

eg. $B_6H_6^{2-}$ anion



2) Nido carboranes- These carboranes may be assumed to be derived from the closo triangular polyhedral of a closo carborane by the removal of one

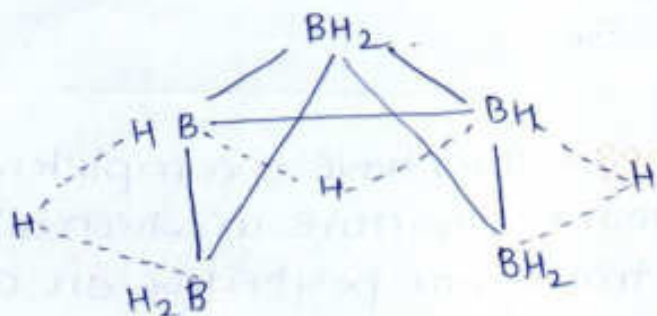
the vertices of the polyhedron. The resulting nido skeleton structure of has $(m+2)$ e^- pairs in its multicentre bonding orbitals.
 e.g. $C_2B_4H_8$



e) Archano Carboranes.

The structure of these type of carboranes may be assumed to have derived from nido structure by the removal of the vertices of these structure, there are $(m+3)$ e^- pairs in the multicentre bonding orbitals of carboranes

e.g. $C_2B_7H_{13}$



nido carboranes

These carboranes may be derived from the nido structure by the removal of a vertex of the polyborane core.

Polymorphism / Allotropy

The existence of substance in a more than one crystalline form possessing different physical properties is known as polymorphism. This phenomenon occurs in both elements and comp. when it is present in elements, it is known as allotropy. Each polymorphic form constitutes a separate phase. The Temp. at which one crystalline form changes into another is known as the transition temperature.

Polymorphism of Carbon.

Carbon is the second most known element except hydrogen. It is a p-block element and its chemistry is extensive and important such that it constitutes a special branch of chemistry known as organic chem.

The electronic structure of carbon in ground state is $1s^2, 2s^2, 2p^2$ with the two 2p unpaired electrons following Hund's rule of max. multiplicity in order to account for normal four covalent bonds of carbon. We may consider that it is promoted to valence state based on configuration $1s^2, 2s^1, 2p_x^1, 2p_y^1, 2p_z^1$. Naturally occurring carbon in which isotopic composition $^{12}\text{C} = 98.89\%$ and $^{13}\text{C} = 1.11\%$. In addition the traces of ^{14}C are observed which are useful for providing basis for radioactive dating techniques.

Carbon exists into two ways well known forms i.e. Diamond and Graphite. which are always Allotropes of each other. α -graphite, β -graphite, diamond carbon. A rare hexagonal form of diamond and a variety of discrete molecules such as C_{60} which are clusters of carbon atoms and are collectively called as fullerenes. Diamond and graphite are structurally very much different from each other, while diamond has three dimensional tetrahedral structure and Graphite has a two dimensional layer structure.

Diamond is extremely unreactive while graphite is quite reactive. These two polymorphs differ in their physical as well as chemical properties. Diamond is denser than graphite (Diamond = 3.51 gm/cm^3) & (Graphite = 2.22 gm/cm^3) But graphite is more stable by 0.69 Cal/mole at 0 K and at 1 atm. pressure .

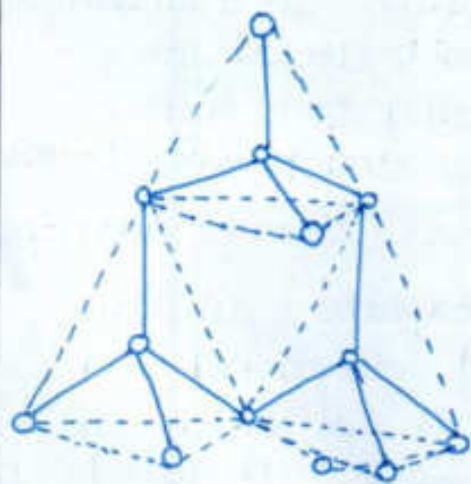
Graphite $\xrightarrow[150 \text{ atm. press.}]{300 \text{ K temp.}}$ Diamond.

But diamond can not be converted into graphite. We seen that graphite it converts or transform into diamond. press. must be applied is 150 atm. at applied 300 K tem. equilibrium is slowly attained.

Q. Draw the structure of graphite & diamond & conductivity by different gr.

§ Diamond

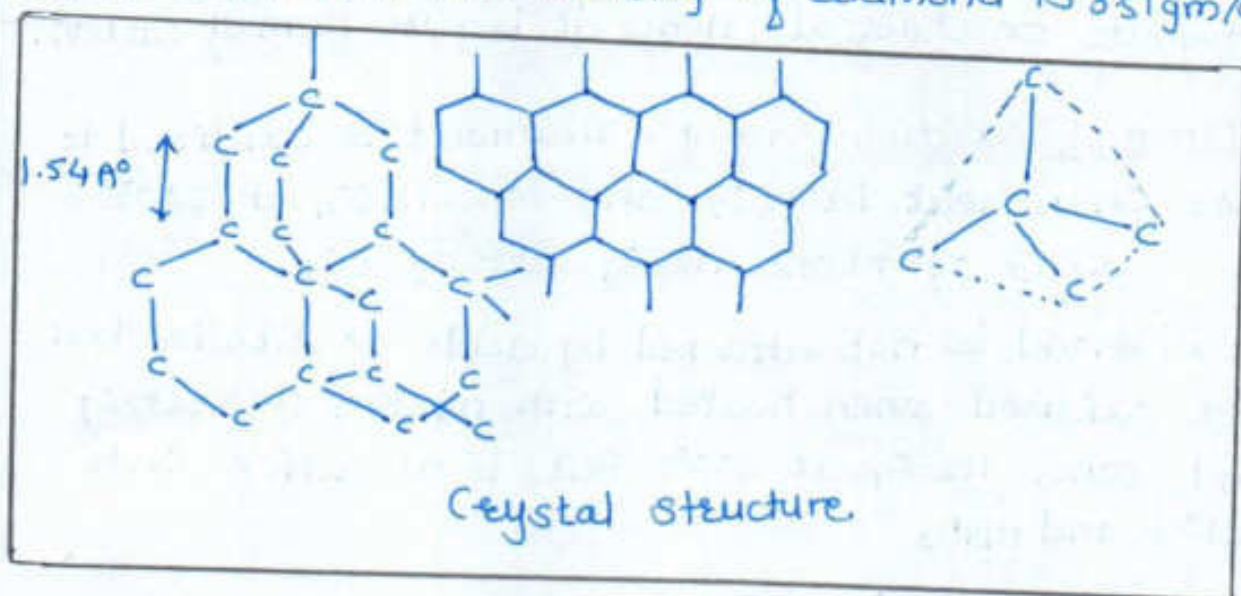
Diamond is a big three dimensional polymer comprising a large no. of carbon atoms arranged in tetrahedrally by utilising their sp^3 hybrid orbitals.



(Structure of diamond)

The diamond is the one hardest and known with the highest density which is explained in terms of structure with the help of x-ray diffraction studies. In the diamond lattice, each carbon atom at a distance of 1.54 \AA . This peculiar structure is extended in the throughout, entire crystal form the giant molecule. This structure accounts for many other properties of the diamond. m.p & b.p. are unknown.

But extra ordinary high M.P and B.P. Diamond is poor conductor of heat and electricity. since, all the e^- in diamond lattice are firmly held in carbon-carbon bonds. Diamonds appear to consists of series of interlocked hexagones where all the carbon atoms are tetrahedrally disposed to form a big three dimensional polymer. Due to high strength of covalent bonds holding numerous carbon atom together, diamond is very hard and high density. To break these bonds high temp. is required hence M.P. is 3500°C . The unit cell is cubic one. The density of diamond is 3.51 gm/cm^3



1) Physical Nature.

Diamond is hard and brittle crystalline solid parent to x-ray. Pure diamond is colourless or many times with yellow sometimes it is coloured green, pink, red, blue, brown, violet or even less gray or black due to presence of metallic impurities.

$$C=C = 1.41 \text{ \AA}$$

2) Hardness.

It is naturally occurring hardest substance. In Mohs of hardness it's hardness is 10 units & 5 unit Mohs

3) Density

Density of diamond is 3.51 gm/ml & it has very high room temperature. has R.I. = 2.417

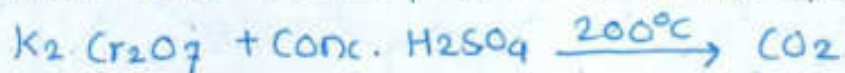
4) Melting point - Due to very strong covalent bonding its M.P. is extremely high i.e. 3930°C

5) Conductivity - Bonding is essentially (non-polar covalent) electron delocalised hence diamond are poor conductor of heat and electricity.

6) Chemical Nature - Diamond is chemically unreactive. It is insoluble in all liquids. There is no effect of corrosive substance on diamond.

7) Action of heat - It burns in air with difficulty at 900°C or in at 700°C to form CO_2 . Only pure diamonds does not leaves ash or does give out water vapour like graphite or charcoal. Hence it is pure form of Carbon.

8) Action of oxidising agent - Diamond is oxidised to CO_2 when heat $\text{K}_2\text{Cr}_2\text{O}_7$ and conc. H_2SO_4 at 200°C



Diamond is not attacked by acids or alkalis but get oxidised when heated with mixture of $\text{K}_2\text{Cr}_2\text{O}_7$ and conc. H_2SO_4 at 200°C But is no action with KClO_3 and HNO_3

Uses of Diamond.

1) Perfectly Crystalline transparent and coloured diamonds are used previous beautiful and for decorative purposes. Deep black and diamond are called carbonates and are used for industries

2) Diamonds are used for drilling

3) Diamonds are used for cutting hard substances like glass, hard rubber and very hard tools & for gem engraving.

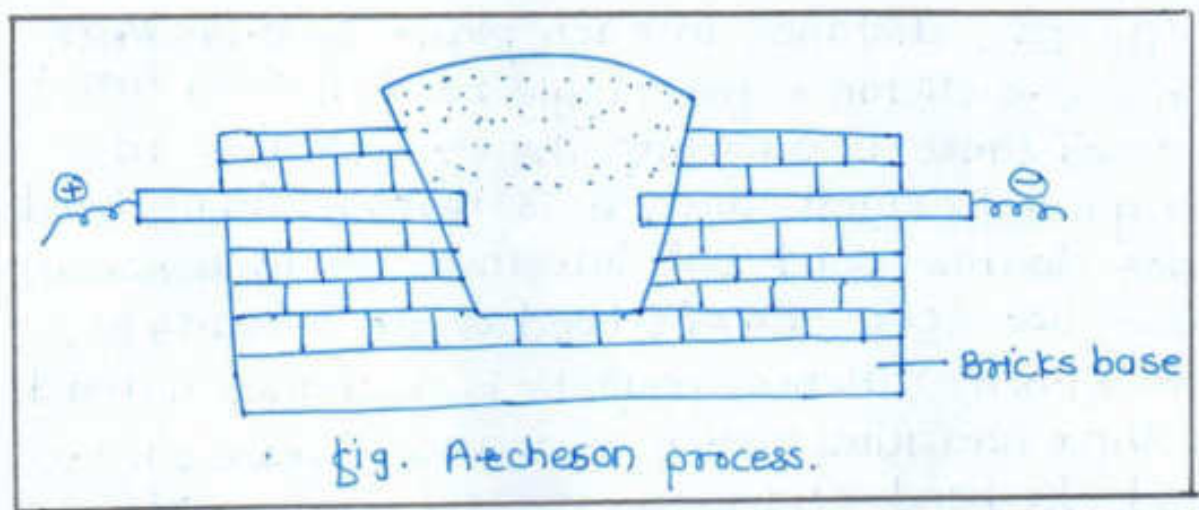
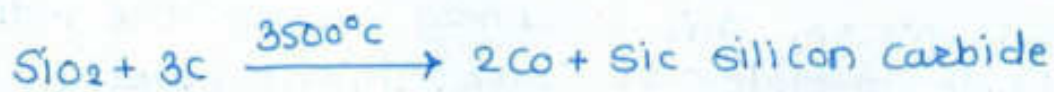
4) Diamond powder is used for abressive for polishing diamond and other gem stones

5) Diamond dyes are used in drawing wire like tungsten filaments used in electric bulbs.

Graphite

Graphite occurs in Native state in Sri Lanka, India, Canada, USA, Mexico etc. In small quantities. It occurs in India at places Rajasthan, Vishakhapatnam and West Bengal.

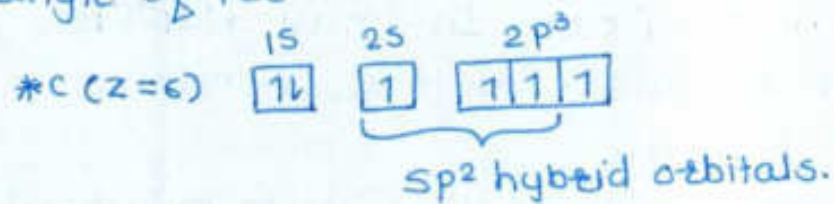
In large scale pure graphite is manufactured by one "Acheson process" mixture of sand and anthracite (purest form of coal) or petroleum coke together with a little binding material such as coal and is placed into a rectangular fire brick furnace provided with heavy carbon electrodes. The mixture is heated to about 3500°C for 24-30 hours. Silicon carbide (SiC) formed decomposes to graphite and silicon where silicon volatilises off leaving behind graphite.



x-ray analysis shows that graphite has some different structures than diamond. It is the two dimensional sheet like polymeric structure.

In graphite only three of the valence electrons of each carbon structure. In graphite only three of the valence electrons of each carbon involved in

Sigma bonding by sp^2-sp^2 overlap with the free neighbouring carbon atom in plane giving C-C-C bond angle of 120°

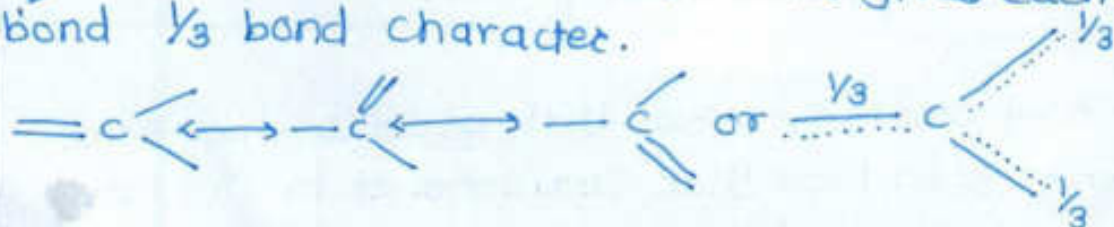


1) Thus, the structure of graphite is two dimensional and it consists of no. of benzene rings fused together and hence it has layer structure.

2) In graphite individual layer is parallel to each other and held together not by valence bond but by weaker electrostatic force called van der Waals.

3) The C-C distance in each layer is 0.143 nm while the distance between the two successive layer is 0.335 nm . This wide separation shows that there is not covalent bonding but there is presence of weak van der Waals forces. This weakness enable the individual layer to slide over other. This structure gives softness and excellent lubricating character to graphite.

4) The C-C distance in each plane is 0.142 nm . Since C-C distance for single bond is 0.154 nm & C=C distance is 0.134 nm . The C-C distance in graphite suggest that, it is neither single bond nor double bond but intermediate in between these two (C-C = 1.42 \AA ; bond length C-C = 1.54 \AA , C=C = 1.31 \AA). Hence graphite is resonance hybrid of three structures. such resonance gives each C-C bond $\frac{1}{3}$ bond character.



Graphite exists in two crystalline - @ α -graphite

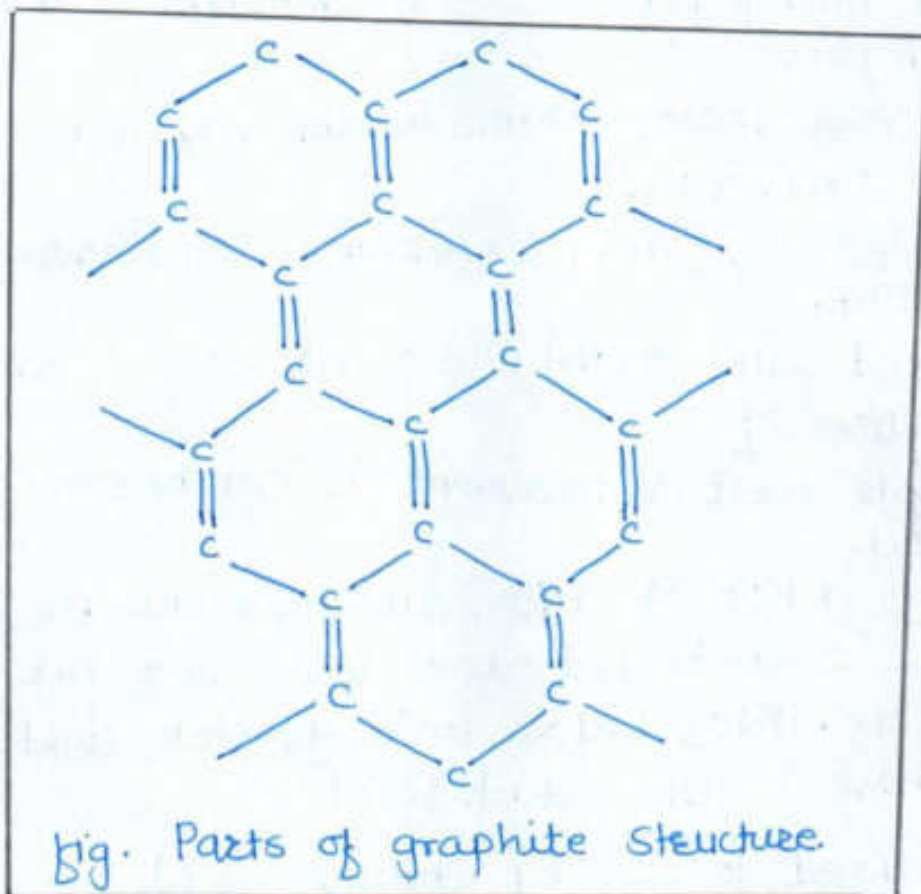
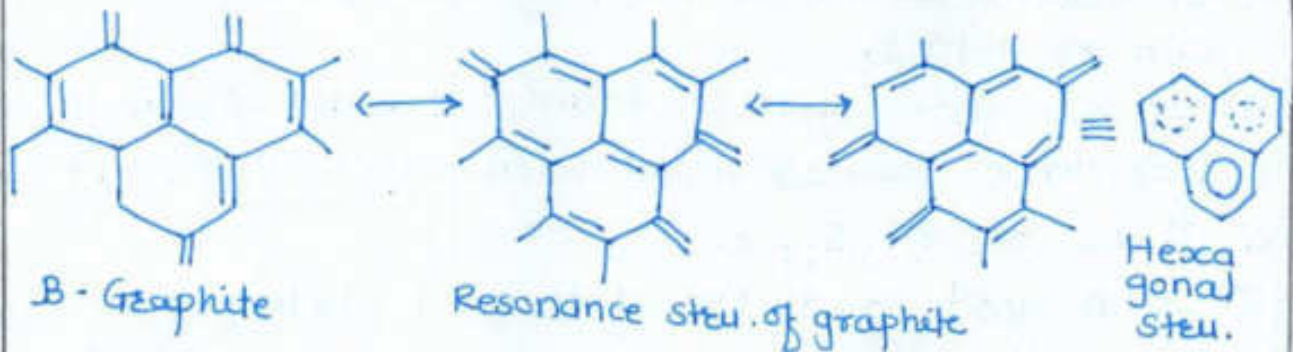
ⓑ β -graphite α -graphite $\xrightarrow{500^\circ\text{C}}$ β -graphite

α - has ABAB type of layers. and β - form has ABCABC type of layers.

* α - graphite - These are two dimensional layer lattice structure sp^2 hybridised

* β - graphite - layer lattice area.

The 'H' bonding may be accounted from $2p_z$ orbital of each carbon which are excluded from hybridisation. The fourth valence electron which remains ~~is~~ unpaired permits the passage of electric current through layer of structure of graphite and this make is excellent conductor of electricity



- a) Two rings stacked each other by Van der Waals forces
b) Across parallel layer no free electrons.



α - graphite

Properties.

- 1) Graphite is dark gray crystalline solid with a little metallic lustre.
- 2) It crystallises in the form of hexagonal plates with gray lustre.
- 3) It has more open crystalline structure which is less dense than diamond (density - 2.25 gm/ml)
- 4) It is soft & slippery.
- 5) It is good conductor of heat & electricity
- 6) On rubbing with papers leaves black marks used in carbon pencil prop.
- 7) Chemically very inert, thermodynamically it is stable by 1.9 kJ/mole .
- 8) It is an opaque, blacky crystalline solid having very high m.p.
- 9) When heated with strongly in O_2 at 700°C it burns with formation of
- 10) By suitable heat & treatment it can be converted into diamond.
- 11) There is no action of acids, alkalies & chlorine on graphite but when heated above 100°C with mix. of KClO_3 & conc. HNO_3 below 100°C graphite acid is formed ($\text{C}_{11}\text{H}_{40}\text{O}_5$) yellow acid. solid
- 12) It is oxidised to CO_2 by chromic acid.

Uses of Graphite

- 1) It produces black streak on paper hence mixed with some plastic or clay used in manufacture of lead pencils.
- 2) Used in manufacture of graphite crucibles.
- 3) Since, It is chemically inert refractory used excellent conductor of electricity. Graphite is used as graphite electrodes in electrochemical industries for electric lamp, furnace lamp, dry cells etc.
- 4) As the powder or as the fine dispersion in the graphite is used as lubricant especially for high temp. high speed machinery
- 5) Being chemically inert used as protecting coating for iron to prevent its corrosion.
- 6) Used as pigment in dyes.
- 7) Used as moderator in nuclear reaction.
- 8) It is used in making electrodes, carbon rods etc.

Polymorphism of Phosphorous.

Phosphorous is one of the V A group element
Its ~~is~~ Electronic Configuration is $1s^2, 2s^2, 2p^6, 3s^2, 3p^3$
Hence, phosphorous is 'p' block element.

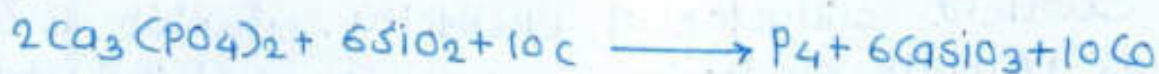
- 1) Sum of first three ionisation potential is 60.4 eV
- 2) Electronegativity = 2.04
- 3) Radii a) ionic radius = 2.12 \AA
b) Atomic radius = 1.06 \AA

In case of V A group elements ionisation energy decreases from top to bottom.

- 4) M.P. of white phosphorous = 44.1°C & B.P = 280°C
phosphorous occurs in various orthophosphates minerals generally fluorapatite $3\text{Ca}_3(\text{PO}_4)_2(\text{O}(\text{FCl})_2$

Preparation.

Elemental phosphorous is obtained by reduction of phosphorous rock with coke & silica in an electric furnace. phosphorous volatilises as P_4 molecule. (Partly dissociated above $800^\circ C$ into P_2) and is condensed under water as white phosphorous.



Phosphorous exists in different allotropic forms such as white, red, scarlet, metallic or α -black phosphorous, β -black phosphorous, β -black, violet phosphorous etc.

The main allotropic forms are white, red and black phosphorous.

a) white phosphorous.

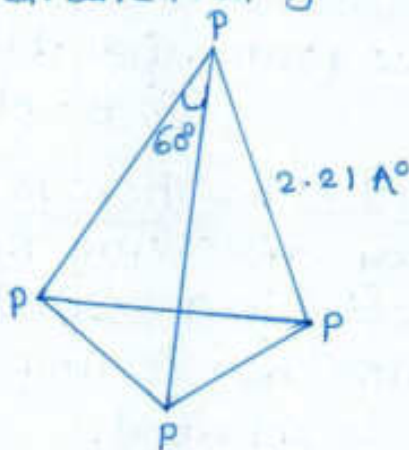
It is obtained by rapidly cooling the vapours of phosphorous. When pure, it is translucent, waxy solid. White phosphorous is rapidly soluble in CS_2 and evaporation of the solution yields the octahedral crystals of phosphorous below $800^\circ C$. The vapour density corresponds to P_4 . In liquid & solid white phosphorous are tetrahedral P_4 molecule. takes place. The P-P distance is 2.21 \AA . The P-P angles are 60° . The low angles in P-P indicates considerable strain & strain energy of six P-P bonds of same length formed by phosphorous atoms with normal bond angles. Thus, the structure of molecule is having high reactivity. It is observed that, pure sp orbitals are involved even though the bonds are bent as hybridisation sp^2 which gives 60° angles will require rather large promotion energy.

Special Property

It glows in the dark on account of its slow oxidation. This phenomenon is known as "phosphorescence". During oxidation it gives a smell like that of garlic. White phosphorus is extremely poisonous. The vapours of phosphorus if continuously inhaled causes necrosis of the bones, particularly those of the jaws and nose. In very small doses, phosphorus acts as a stimulant. White phosphorus is readily soluble in turpentine oil & ether.

Structure

Four phosphorus atoms lie at the corners of a regular tetrahedron. Each P atom is linked to each of the other three phosphorus atoms by covalent bonds. It has tetrahedral geometry.

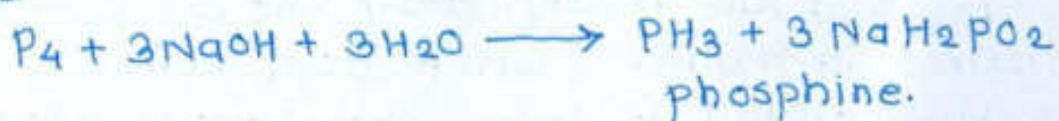


b) Red Phosphorus.

Red phosphorus is obtained in "crystalline form" by heating white phosphorus at very high pressure.

Preparation.

It is prepared by heating white phosphorus at 253°C in a cast iron egg-shaped vessel in which air is replaced by CO₂ which contains some white phosphorus which is eliminated by treating with alkali.



Red phosphorous is not attacked by alkali. This is polymeric solid which is much less reactive than with white phosphorous. It is stable in air & does not ignite. Unless it is heated to 400°C . It is insoluble in organic solvents. Need not to be store under water. Three dimensional pentagonal arrangement. It is not poisonous. It sublimes on heating giving vapours as same given by the white phosphorous. When these vapours are condensed white phosphorous is obtained

Red phosphorous melts at 610°C (M.P.) under pressure. giving a yellow liquid which on heating cooling gives white phosphorous.

c) Black Phosphorous.

It exists in two forms. (i) α -black phosphorous
(ii) β -black phosphorous.

i) Metallic or α -Black Phosphorous.

It is obtained by dissolving Red phosphorous in molten lead at 400°C . in a sealed tube for sufficiently long time on cooling crystals of black phosphorous is obtained.

It is very stable allotrope of phosphorous & does not oxidises in air unless heated very strongly. It can not conduct electricity.

white P $\xrightarrow[\text{or } 220-370^{\circ}\text{C for 8 days in Hg.}]{\text{either hig press.}}$ Crystalline black P.

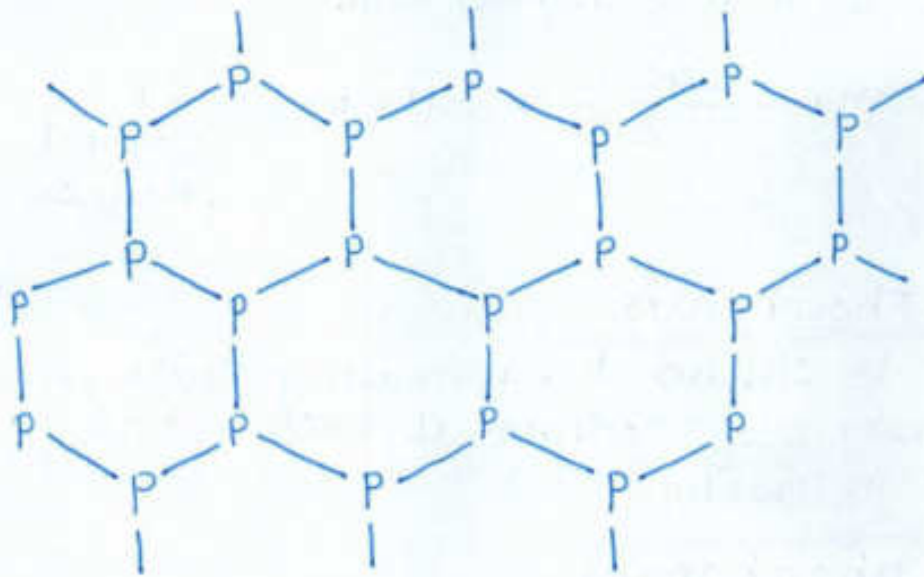
ii) β -Black Phosphorous.

The structure is definitely known. It is crystalline in nature. Each P atom being covalently bonded to three phosphorous atoms.

The P-P-P angle is 99° The P-P distance is 2.18\AA
M.P = 58° It has specific gravity 2.39.

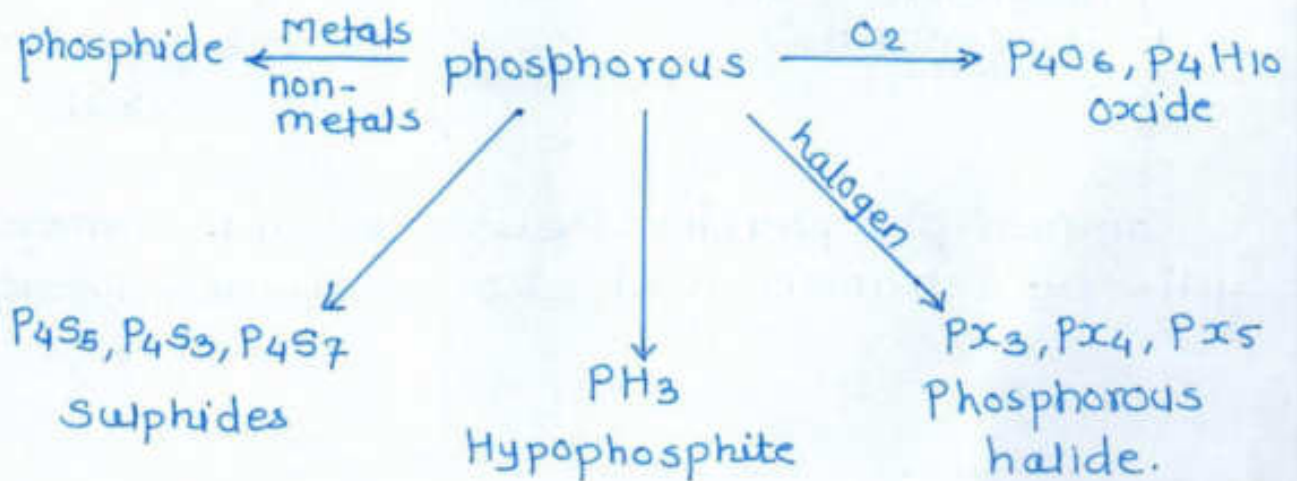
Structure.

It is consist of corrugated sheets.



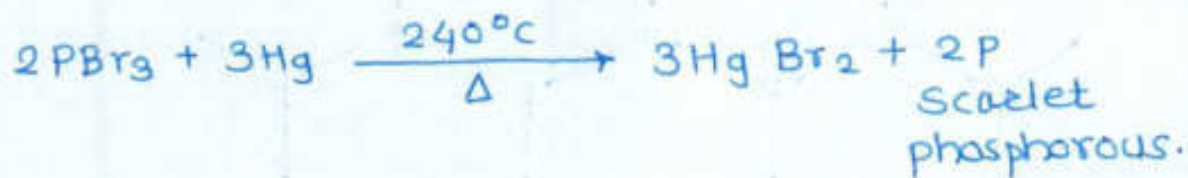
Layer structure of β -Black Phosphorous.

The main form of phosphorous are shown chemical activity. white phosphorous is more reactive while black phosphorous is least reactive white phosphorous is stored under water to protect from white. red and black are stable in air. Black phosphorous can be ignited with difficult while white phosphorous inflames. in air and soluble organic solvents. such as CS_2 & become some reactions of both red and white is shown in below.



iii) Scarlet phosphorous.

It may be prepared by heating phosphorus tribromide with mercury at 240°C



iv) Violet Phosphorous.

It is obtained by heating white phosphorous in presence of sodium at 250°C . It is Crystalline in nature.

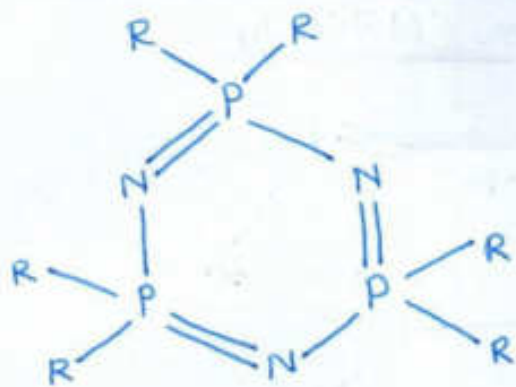
Uses of phosphorous.

- i) It is mostly used in manufacture of matches
- ii) It is used as rat poison.
- iii) It is used in making phosphorous bronze which is hard and not corroded by water.
- iv) It is used in manufacture of tracer bullets
- v) It is employed in preparation of compounds, phosphoric acids, hypophosphites, phosphorous chlorides etc.

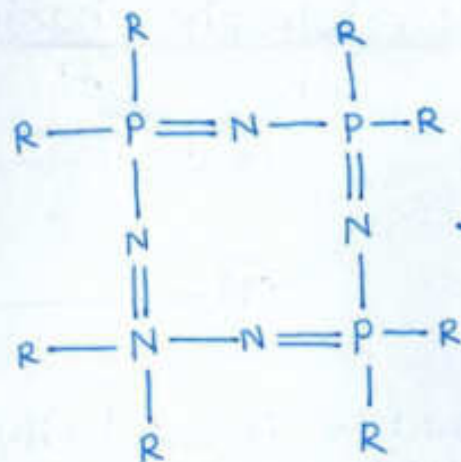
Phosphazines.

Phosphazines are cyclic or chain polymers which contains, $\left(-\text{N}=\underset{\text{R}}{\text{P}}-\text{R} \right)$ as the repeating units

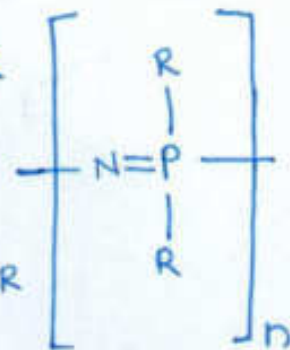
Common phosphazines exists as cyclic trimers cyclic tetramers and also as linear polymers.



Cyclic trimer



Cyclic tetramer.



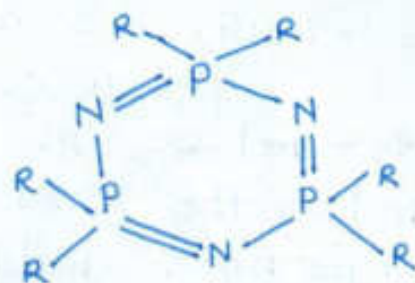
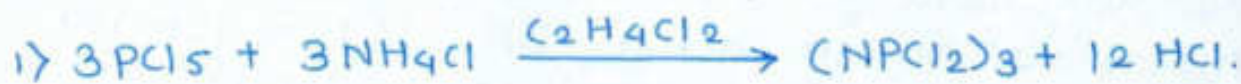
Linear Polymer.

Chlorophosphazenes

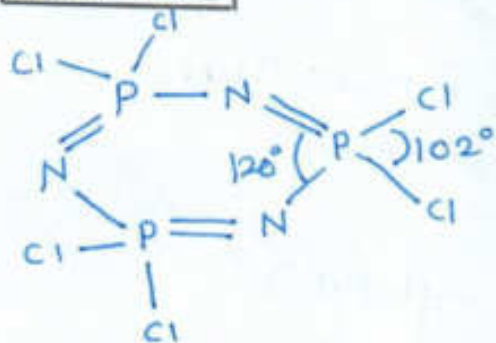
Two imp. chlorophosphazenes are -

i) Hexachlorophosphazenes (NPCl₂)₃

It is prepared by reaction of PCl₅ with NH₄Cl in the presence of ~~under~~ C₂H₄Cl₂ under controlled condition.

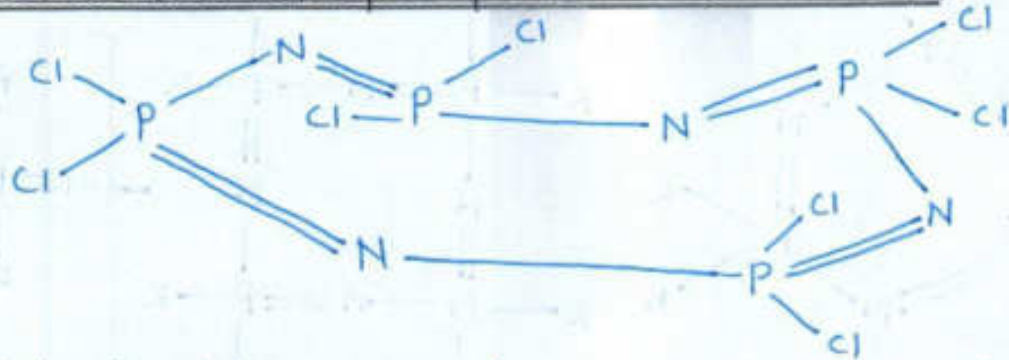


Structure



Benzene involves $\text{PTT} \rightarrow \text{PTT}$ bonding but phosphazenes involves $\text{PTT} \rightarrow \text{dTT}$ bonding. The $\text{Cl}-\text{P}-\text{Cl}$ bond angle is 102° and $\text{N}-\text{P}-\text{N}$ bond angle is 120° .

ii) Octachlorotetraphosphazine $(NPCl_2)_4$



This structure is evidently non-polar phosphazenes of high molecular masses come under inorganic polymers.

Polymorphism in Sulphur.

Sulphur is one of the element VI A group element. The E.C. of Sulphur is $1s^2, 2s^2, 2p^6, 3s^2, 3p^4$. It is 'p' block element. Sulphur is yellow solid without taste and odour. It is freely soluble in carbon disulphide (CS_2) but insoluble in water and sparingly soluble in alcohol & ether. Its vapour are poisonous for bacteria and fungi but not for animals or human beings. Its electronegativity is -2.44. Its B.P = $444.6^\circ C$ & M.P = $119^\circ C$.

Sulphur occurs widely in nature as elements H_2S and SO_2 in many sulphides, ores of metals and in the form of various sulphides such as Gypsum & unhydrated magnesium sulphate etc.

Sulphur occurs in different allotropic i.e. polymorphic forms.

1) Solid sulphur - (A) Rhombic (B) Monoclinic

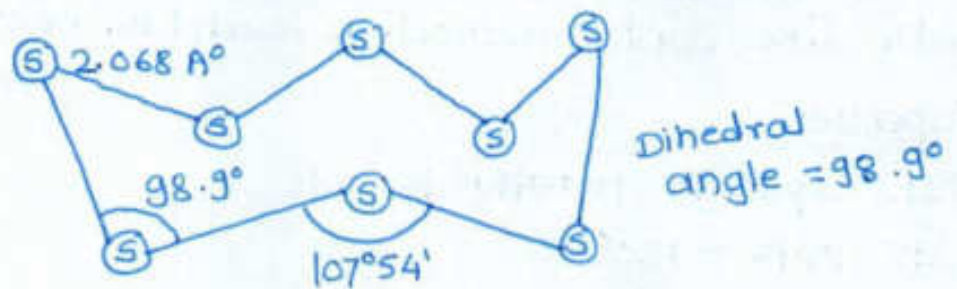
(C) Angles sulphur.

2) Liquid sulphur -

3) Gaseous sulphur (vapour sulphur)

1) Rhombic & monoclinic Sulphur

Rhombic (octahedral) sulphur is stable below temp. 95.6°C while another form monoclinic sulphur is stable above 95.6°C . The enthalpy of this transition is small i.e. 0.096 K cal and the process is slow. Hence, it is possible to convert rhombic sulphur by heating up to 112.8°C into monoclinic sulphur as M.P. is 119°C but it breaks down at low temp. than the M.P. due to break down at S_8 molecule in liquid sulphur. Both rhombic and monoclinic sulphur contains puckered ring with crown conformation S_8 molecule which is cyclic



Both forms shows highly soluble in organic solvents like CS_2 molecular weight measurements shows that S_8 ring persist in the solⁿ at ordinary temp. when sulphur solⁿ in iodoforn is crystallised needles of charge transfer. Comp. $\text{CHI}_3 \cdot S_8$ are obtained in which $\text{PI}_3, \text{AsI}_3, \text{SbI}_3$ are known.

Crystalline modification of 'sulphur' is Engels Sulphur i.e. which is obtained by purifying $\text{Na}_2\text{S}_2\text{O}_3$ solⁿ in conc. HCl at 0°C and extracting S with toluene from this Engels sulphur crystallises in short time. the hexagonal crystals are not stable. The crystal contains S_6 rings closely packed with chair configuration. The S-S bond dist. is $2.06 \pm 0.02 \text{ \AA}$ identical with rhombic sulphur in S_8 ring. The conversion of plastic & rhombic sulphur proceed by opening of S_8 ring & partial fragmentation of to forms chain & S_8 rings.

Properties - Rhombic Sulphur

- 1) In Rhombic Sulphur, S-S-S angle is 108°
- 2) It is stable below 95.6°C
- 3) Its M.P. = 114°C
- 4) Crystal structure contains cyclic S₈ ring in puckered manner.

Monoclinic Sulphur (prismatic or β -sulphur)

It is prepared by melting ~~the~~ rhombic sulphur in a dish and allowing it to cool until a crust is formed at surface. The two holes are pierced into the crust. Liquid sulphur lying below the crust is poured through the holes. On removing crust small needle like crystal monoclinic sulphur become formed.

Properties.

- 1) Its specific gravity is 1.96
- 2) Its M.P. = 120°C

2) γ sulphur.

This form is obtained by decomposing ethylene xanthate of copper in pyridine or by chilling hot solⁿ of sulphur in CS_2 . It exists in the form pale yellow needles. Its specific gravity is 2.18

3) colloidal or δ -sulphur

When H_2S is passed through HNO_3 sulphur separates out in colloidal state.



It is also obtained by passing H_2S through water containing dissociation.



Colloidal sulphur.

Plastic or Amorphous Sulphur.

When molten sulphur heated to about 350°C and poured into cold water. rubber like mass called plastic sulphur is obtained. It is dark in colour. and has specific gravity 1.95. It is soluble in CS_2 . plastic sulphur is considered to be so supercooled liquid "x-ray" studies shows that the fibre contains helical chain of sulphur atoms. with about ten atoms per 3 cycles of the helix. Amorphous sulphur can be made in other ways. but in all cases.

Liquid Sulphur

Liquid sulphur is yellow at its M.P. of little above further of its temp. is increased it becomes increasingly viscous until temp. is 200°C viscosity again decreases with further increase in temp. At its B.P. 444°C It is rather mobile liquid viscosity changes. The rings which originate in the crystalline solid becomes unstable as the temp. raised. It was assumed that 6 membered & 4 membered rings are formed in eqⁿ with 8 membered rings. The formation of open chains by homolytic S-S bonds in is most important. such chains so that at any temp. an eqⁿ betⁿ rings and chains of many lengths will eventually set up.

Vapour Sulphur

Vapour sulphur also known to contain various sp. in temp. dependant eqⁿ various species $\text{S}_8, \text{S}_6, \text{S}_4$ & S_2 are the greatest imp. Assuming that there are the sp. presents the vapour density data requires that angles of 430 and 500 mm pressure. eg. amount of these species are $\text{S}_8 = 54\%$, $\text{S}_6 = 37\%$, $\text{S}_4 = 5\%$ and $\text{S}_2 = 4\%$.

-: Oxides and non-transition elements:-

★ Oxides and non-transition elements:-

I Group IA:-

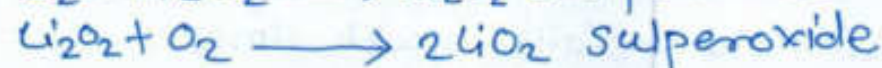
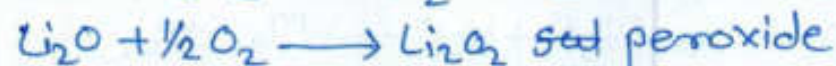
H	Li	Na	K	Rb	Cs	Fr
1	3	11	19	37	55	87

properties:-

H is present in gaseous form. if from covalent bonding. However, Li, Na and K not forms covalent bonding halogens do form corresponding acids, Li, K, Na reacts with VIII A group to form covalent comp. I values decreases from top to bottom $Li > Na > K > Rb > Cs > Fr$. reactivity in water and air increases down the group all the metals are highly electropositive B.P and M.P increases with increase in atomic no. and ionisation potential.

a) Oxides:- These I group metals form oxide, peroxide or superoxide by dissolving the metal in liq. NH_3 . Li_2O and Na_2O are pure white solid as expected but K_2O is pale yellow.

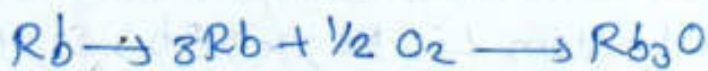
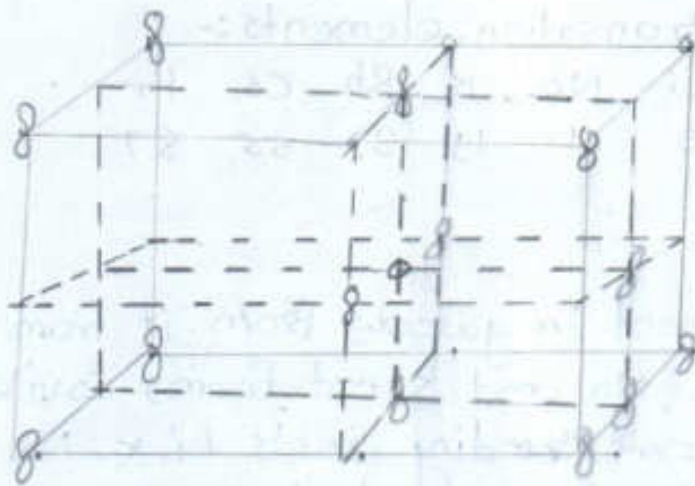
Monoxides are ionic e.g. $2Li^+$ and O_2^{2-} Rb_2O is bright yellow Cs_2O is orange



peroxide, superoxide. used as biological killers.

K_2O_2, KO_2 are added in tooth paste

e.g. Na_2O_2



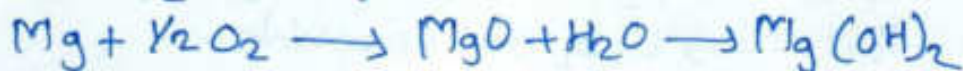
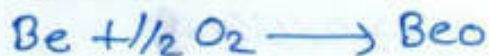
K_2 is used in capsules submarines because both products dioxygen and remove CO_2

★ Group -II A — Be, Mg, Ca, Sr, Ba, Ra

4 12 20 38 56 81

a) Oxides:- Be \rightarrow BeO, Mg \rightarrow MgO, Ca \rightarrow CaO, Sr \rightarrow SrO, Ba \rightarrow BaO

All this elements of this group burn in O_2 or form monoxide. 'Be' is relatively unreactive in massive form and it does not reacts below $600^\circ C$. But the powder is much reactive.



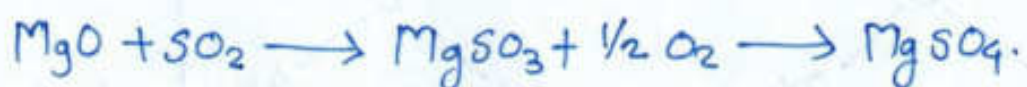
$Be(OH)_2$ is amphoteric, but the hydroxide of Mg, Ca, Ba and Sr are basic. The basic strength increases from Mg to Ba.



$Mg(OH)_2$ is sparingly soluble in water solubility is 1.4×10^{-9} gm/lit

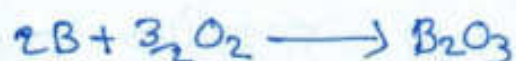
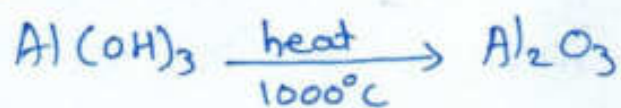
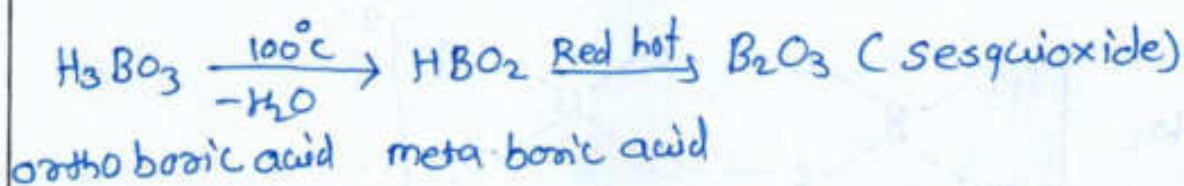
BeO is covalent in nature. while other are ionic in nature all having NaCl structure

MgO - Fabrication of furnaces solubility goes on increasing group. CaO is imp industrially used in cement, ceramic industry. Superoxide are not form of alkaline earth metals. used for removal of toxic gases



★ Group - III A -

	oxd ⁿ state	oxides
⁵ B	+3	B ₂ O ₃ nonmetallic & acidic
¹³ Al	+1, +3	Al ₂ O ₃ acidic
³¹ Ga	+1, +3	Ga ₂ O ₃ amphoteric
⁴⁹ In	+1, +3	In ₂ O ₃
⁸¹ Tl	+1, +3	Tl ₂ O ₃ basic

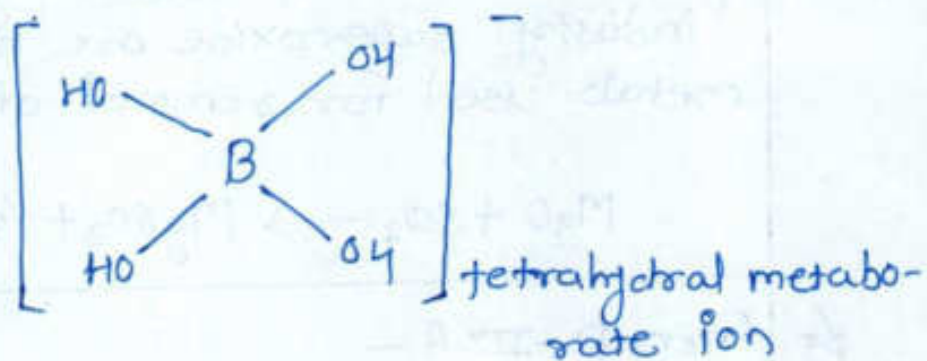
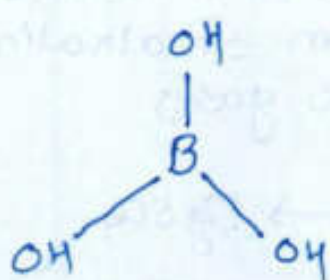
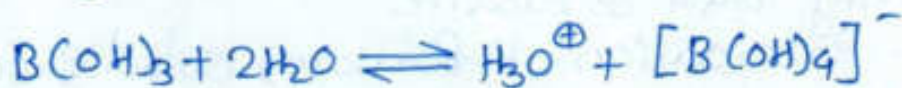


$\alpha\text{-Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3 + \text{Cr}_2\text{O}_3 \longrightarrow$ emery stone. used for polishing purpose of Al₂O₃ is less stable

Sesqui means one and half, so the oxide should have formula

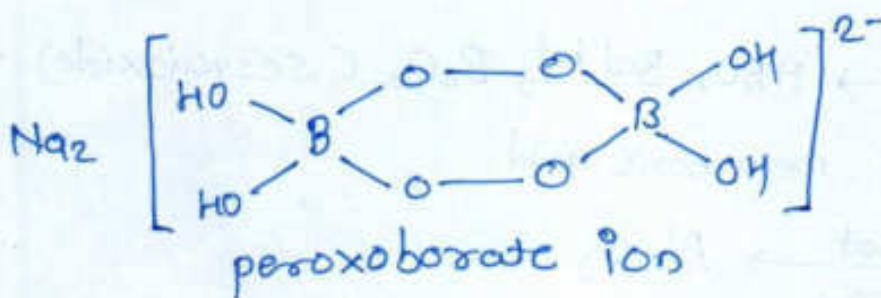
All the em elements in the group forms of sesquioxides when the dioxygen.

B_2O_3 is typical non-metallic oxide is acidic in its properties



Sodium peroxoborate is used a brightener in washing powders. It is compatible with enzymes which are added to some biological powders.

The peroxide linkage $O-O$ break down to give H_2O_2



* other oxides of group III A:-

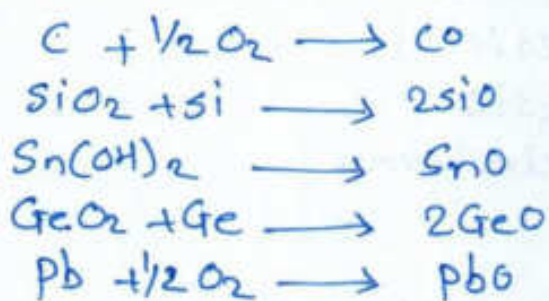
Alumina Al_2O_3 exists in two crystalline forms α - Al_2O_3 called corundum and Al_2O_3 . corundum is found as a mineral and is extremely hard and is used as jewelry rouge to polish glass

Ga_2O_3 and $Ga(OH)_3$ are both amphoteric $Ga(OH)_3$ is white and dissolve in alkali. In contrast Thallous hydroxide $TlOH$ is a strong base and soluble in water.

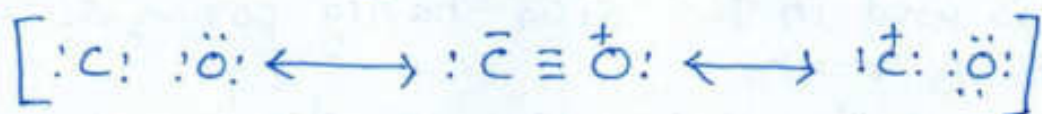
★ Group - IV A elements:-
oxides -

${}^6\text{C} \longrightarrow \text{CO}, \text{CO}_2$	} Acidic	oxid ⁿ -state +4
${}^{14}\text{Si} \longrightarrow \text{SiO}, \text{SiO}_2$		+2, +4
${}^{32}\text{Ge} \longrightarrow \text{GeO}, \text{GeO}_2$	} Amphoteric	+2, +4
${}^{50}\text{Sn} \longrightarrow \text{SnO}, \text{SnO}_2$		+2, +4
${}^{82}\text{Pb} \longrightarrow \text{PbO}, \text{PbO}_2$		+2, +4

|
strong oxidising agent



- 1) CO - Incomplete combustion of carbonaceous material. Tasteless gas Bond dissociation energy 428.5 kJ/mole $\text{C} \equiv \text{O}$ stable species combines with transition metal and forms metal carbonyl imp ligand in CO-oxidation chemistry CO is used in manufacturing of powder gas. It is poisonous. gas reacts with Cl and forms COCl_2



- 2) CO_2 - It is linear molecule obtained by complete combustion of carbonaceous material or fossils. 188 million year tons. But it is the byproduct of respiration. It is essential for plants.



CO_2 is green house gas!



(i) CO can be prepared by dehydrating formic acid with conc. H_2SO_4



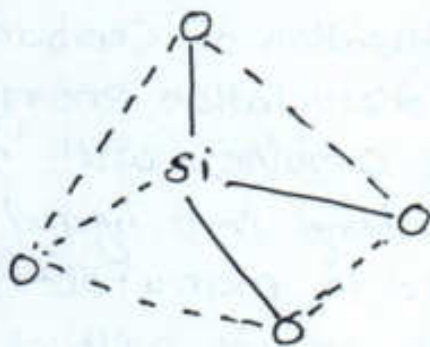
(ii)



(iii) SiO_2 is naturally occurs which on heat at high temp gives SiO

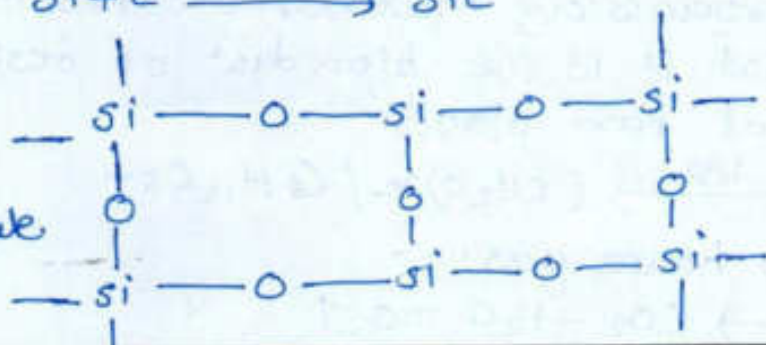


Structure:-



SiO_2 is used in TLC SiO_4^{4-} having porous structure acts as resin.

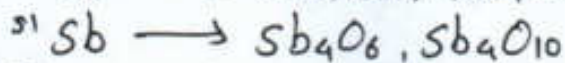
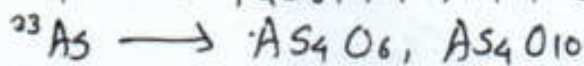
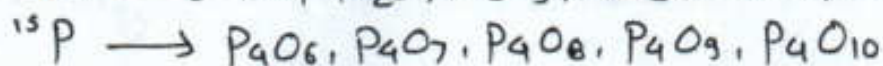
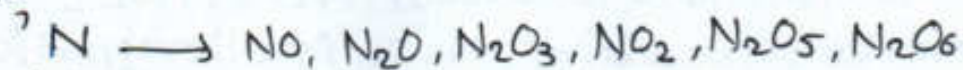
SiO_2 used for extraction of silicon metal



SiO_2
giant molecule

★ Group II A :-

* Oxides :-

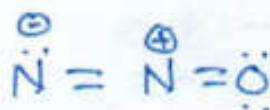


N_2O is prepared by thermal decomposition of molten ammonium nitrate



N_2O is stable relatively nonreactive (colourless). It is neutral oxide N_2O is used as propellant. It is non-toxic. It is used as anaesthetic.

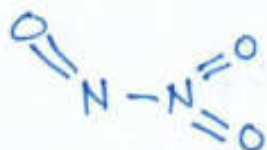
The molecular is linear.



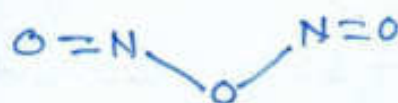
NO (nitric oxide) - NO is colourless gas and is an important intermediate in the manufacture of nitric acid by catalytic oxidation of ammonia. NO is neutral oxide.



NO readily co-ordinates with transition metal ions. The complexes nitrogen sesquioxide (N_2O_3) are called nitrosyls e.g. $[\text{Cr}(\text{NO})_9]$. It is an acidic oxide. It reacts with conc. acids to form nitrosyl salts.



asymmetrical form



symmetrical form.

NO_2 and N_2O_4 - NO_2 is red brown poisonous gas. This NO_2 dimerises into colourless N_2O_4 . NO_2 is paramagnetic and very reactive. N_2O_4 is diamagnetic



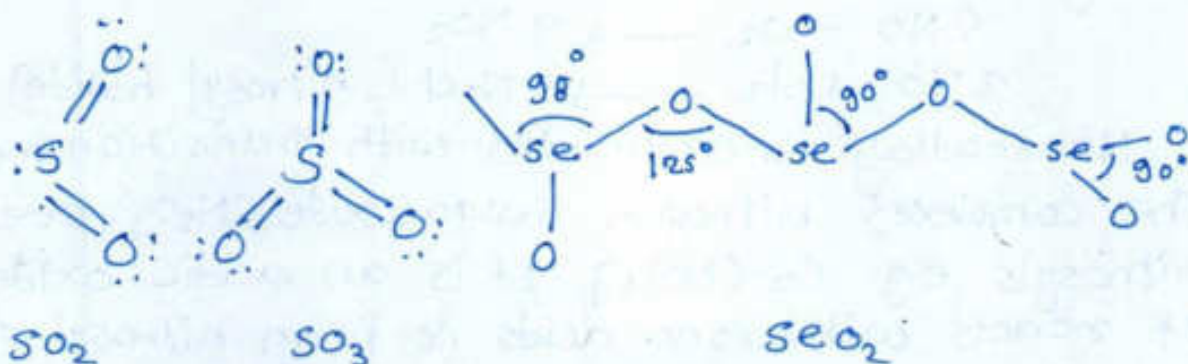
NO_2 is angular with an O-N-O angle of 132° . The bond length of O-N is 1.20 \AA . The solid N_2O_4 has planer st.



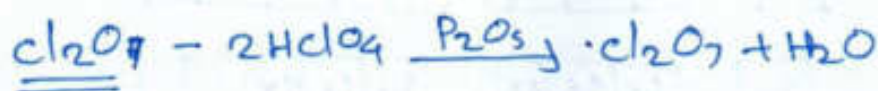
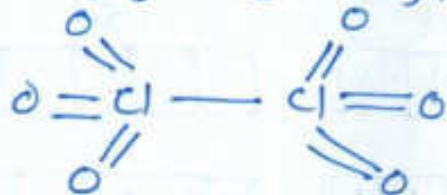
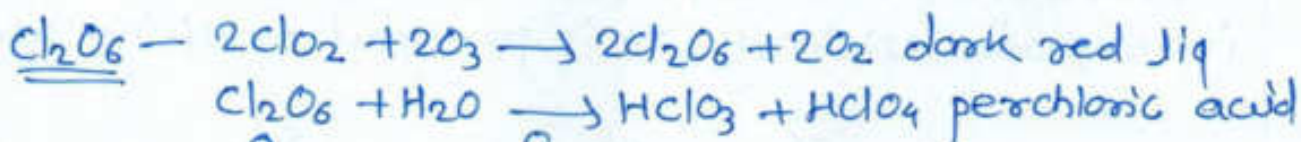
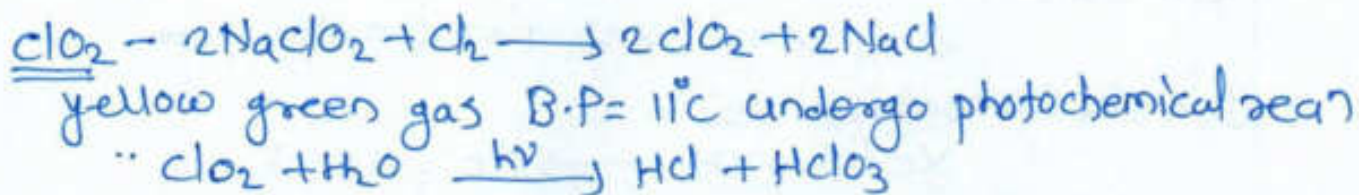
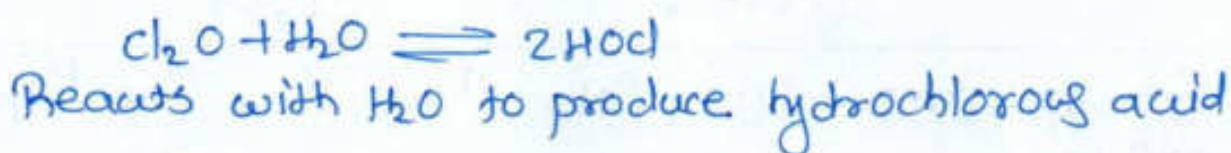
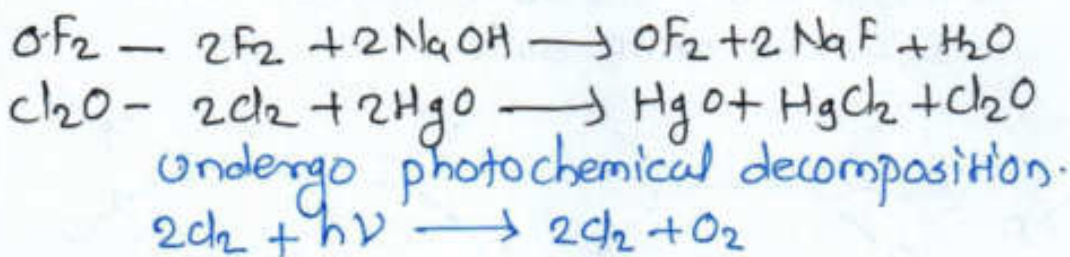
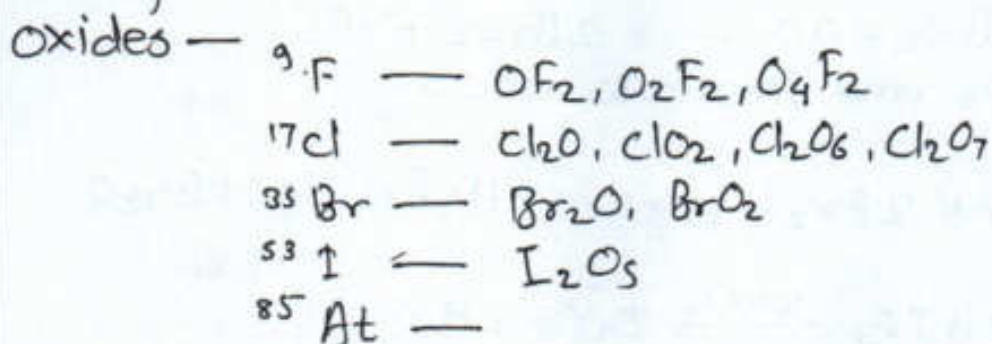
★ Group VI A -

Oxides -	8 O	-2, -1	O_2O_3
	^{16}S	-2, 4, 6	SO_2, SO_3
	^{34}Se	-2, 2, 4, 6	$\text{SeO}_2, \text{SeO}_3$
	^{52}Te	2, 4, 6	$\text{TeO}_2, \text{TeO}_3, \text{TeO}$
	^{84}Po	2, 4	$\text{PoO}_2, \text{PoO}_3$

They are called as chalcogens because are forming

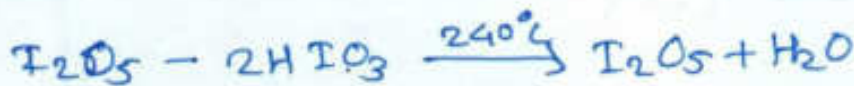
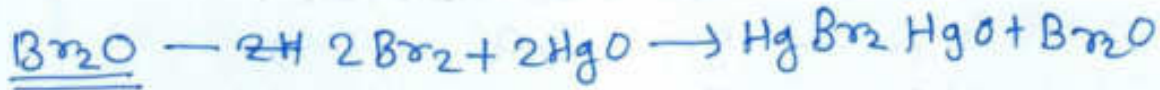


★ Group - VII A -

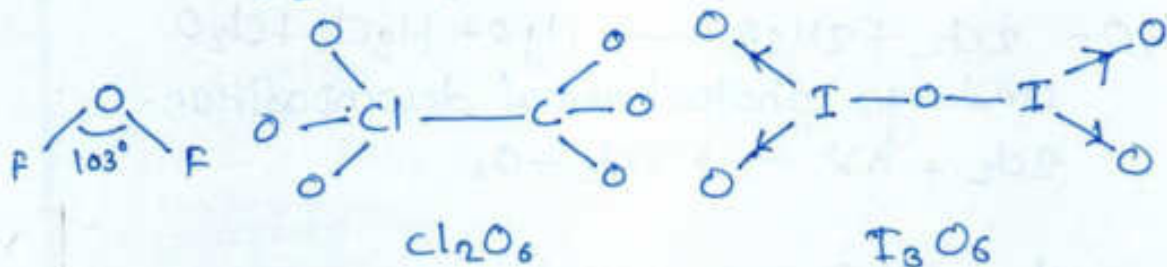




BrO_3 and BrO also exists.



I_2O_5 is white solid.



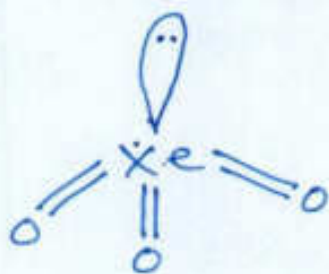
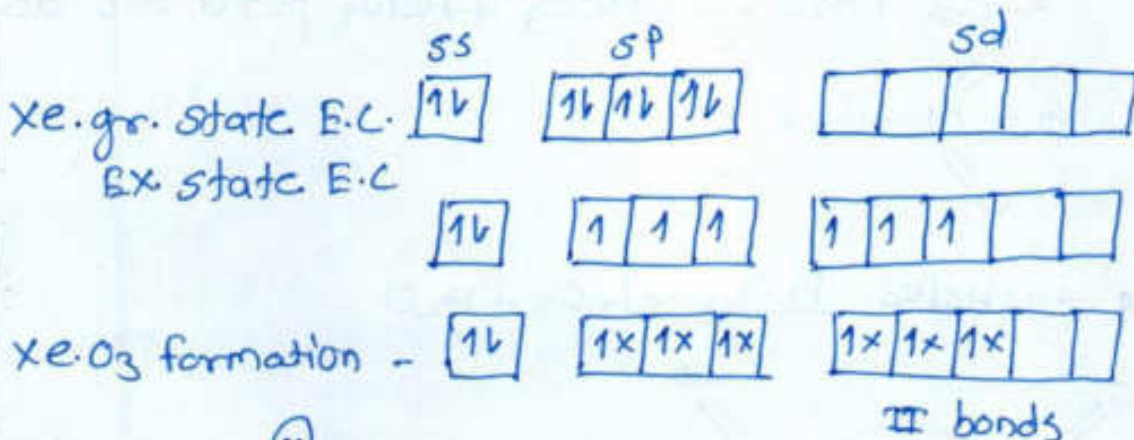
★ Group - VIII - ^2He ; ^{10}Ne ; ^{18}Ar ; ^{36}Kr ; ^{54}Xe ; ^{86}Rn
oxide -

only xenon (Xe) forms oxides etc

① XeO_3 (Xenon trioxide) -



White, non-volatile, water soluble comp. In dry condition it explodes violently. It is powerful oxidising agent



② Xenon tetra-oxide (XeO_4) -



Barium perxenate

XeO_4 is not stable like XeO_3 Decomposes to give $\text{Xe} \& \text{O}$

fundamental group frequencies of various diatomic groups:-

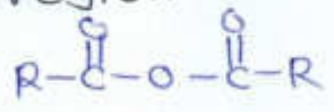
frequency range ν cm ⁻¹	vibration due to bond.	Types of Compound
--	------------------------	-------------------

Bonds of hydrogen (H) atoms region:-

3600-3650	-O-H free	Alcohols, phenols
3200-3500	-O-H bonded	H bonded alcohols & phenols
3450-3600	sharp	Intramolecular H bonding
3200-3550	Broad	Intramolecular H bonding
3300-3350	-N-H	Amines, Amides
3200-3310	\equiv C-H	Acetylenic
3000-3100	=C-H	Ethylenic & aromatic
2850-2950	-C-H	Saturated alkane
2700-2900	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-H} \end{array}$	Aldehyde
2500-3600	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-O-H} \end{array}$	carboxylic acids
2550-2900	-S-H	Thiols.

Double bond region:-

1740-1850



anhydride

1770-1815



acyl halides.

1770 - 1815	$R-\overset{\text{O}}{\parallel}{C}-X$	acyl halides
1730 - 1750	$R-\overset{\text{O}}{\parallel}{C}-O-R$	esters
1730 - 1740	$R-\overset{\text{O}}{\parallel}{C}-H$	aldehyde
1705 - 1720	$R-\overset{\text{O}}{\parallel}{C}-R$	ketones
1700 - 1725	$R-\overset{\text{O}}{\parallel}{C}-OH$	carboxylic acids
1370 - 1540	$-\overset{\text{O}}{\parallel}{N}-O$	Nitro group.
1680 - 1700	$R-\overset{\text{O}}{\parallel}{C}-N$	amide
1590 - 1690	$-C=N-$	imines
1620 - 1680	$-C=C-$	alkene
1500 - 1600	$>C=C<$	aromatics.

Triple bond region:-

2240 - 2260	$-C\equiv N$	non-conjugated nitriles
2215 - 2240	$-C\equiv N$	conjugated nitrile
2100 - 2140	$-C\equiv C-H$	terminal alkynes
2200 - 2260	$-C\equiv C-C$	non-terminal alkynes.

Single bond region:-

1000-1300	$-\overset{ }{\underset{ }{\text{C}}}-\text{O}-$	alcohols, phenols, esters, ethers.
1100-1120	$-\overset{ }{\underset{ }{\text{C}}}-\text{C}-$	saturated alkanes
500-750	$-\overset{ }{\underset{ }{\text{C}}}-\text{X}$	halo alkanes.

characteristic I.R. absorption of functional group:-
Hydrocarbons (Alkanes)

Group	Type of vibration	Band position in cm^{-1}
$-\text{CH}_3$ (methyl)	C-H stre.	2960-2850 (s)
$-\text{CH}_2-$ (methylene)	(asym. & sym.) C-H def.	two to three bands. 1470-1430 (m)
$-\text{CH}_3$	sym def.	1390-1370 (m)
$-\overset{ }{\underset{ }{\text{C}}}-\text{H}$ (methine)	C-H stre.	2890-2880 (w)
C-C	C-C stre.	1200-800 (w)

b) Alkenes:-

$\text{H}-\overset{ }{\underset{ }{\text{C}}}=\text{CH}_2$	=C-H stre. (CH ₂)	3095-3075
	C-H stre. (C-H)	3040-3010
cis-CH=CH-	=C-H out of plane, detom ⁿ	725-675
trans-CH=CH-	, , ,	970-960
-C=C- (isolated)	C=C ster.	1680-1640
-C=C- (conjugat ⁿ)		1645-1580

c) Alkynes : terminal ($R-C\equiv CH$) & non-terminal ($R-C\equiv C-R$) alkynes:

$\equiv C-H$	C-H stre.	3310 - 3300
	CH def.	650 - 600
$-C\equiv CH$	$C\equiv C$ stre.	2140 - 2050
$-C\equiv C-$	$C\equiv C$ stre.	2260 - 2190

2) Alcohols & phenols :- They produce characteristic IR-bands due to stretching of O-H & C-O groups.

O-H free	O-H stre.	3650 - 3590 (v) Sharp
O-H (hydrogen bonded)	O-H stre.	3600 - 3200 (s) broad
	O-H def. (in plane)	1410 - 1260
	i) 1°, 2° alcohols	1350 - 1260
	ii) 3° alcohol, phenol	1410 - 1310
C-O	C-O stre.	1230 - 1000
	i) phenols	1280 - 1140
	ii) 1° alcohol	1085 - 1053
	iii) 2° alcohol	1125 - 1087
	iv) 3° alcohol	1205 - 1124

3) Ethers & epoxides :- Both of them carry C-O-C system. strong broad band at $1230 - 1000 \text{ cm}^{-1}$ due to asymmetrical C-O-C site. can hint an ether if it lacks carbonyl ($C=O$) & hydroxy (O-H) vibrational frequencies.

Alkyl ethers : $1060 - 1150 \text{ cm}^{-1}$

Aryl & vinyl ethers : $1200 - 1275 \text{ cm}^{-1}$

4) Carbonyl Compounds:- The most characteristic & easily recognizable band of all carbonyl compounds lies between 1870-1540 cm⁻¹. The exact position of this band varies with the nature of atom/group to which C=O group is attached.

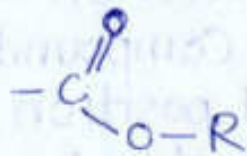
ca) Aldehyde & ketones:-

Aliphatic aldehyde	>C=O stre.	1740-1720 (s)
Aromatic aldehyde		1710-1690
Unsat. aldehyde & ketone		1685-1675
Dialkyl ketone	>C=O stre.	1730-1705
Alkyl aryl ketone		1690-1680
Diaryl ketone		1670-1645
Aldehyde	-C(=O)-H stre.	2880-2650 (975-780)

b) Carboxylic Acids:-

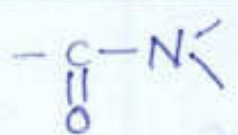
$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C} \\ \backslash \\ \text{O-H} \end{array}$	O-H stre. (H-bond)	3300-2500 (broad)
	O-H def. (inplane)	1440-1395
	O-H out of plane	920
	C=O stre.	1720-1706 1320-1210
$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C} \\ \backslash \\ \text{O-H} \end{array}$	C-O stre.	1400

c) Esters & lactone:-

	$C=O$ str. $C-O$ str. asy. $O-CH_2$ asy str.	1780-1710 1300-1160 1150-1035
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5) Amides:- All amide show carbonyl stretching band Near 1670 cm^{-1} It is called amide band-I The lower frequency value is due to enhancement of resonance due to lone pair of e^- of Nitrogen.

The 1° & 2° amide show additional band in the carbonyl region ($1650-1515\text{ cm}^{-1}$) It is known as amide-II band & arises due to N-H deformation.

	$C=O$ str. (amide-I)	1670
1° & 2° amides	$N-H$ def. (amide-II) $C-N$ str.	1650-1515 1400
1° amide (-CONH ₂)	$N-H$ str. (asym) $N-H$ str. (sym)	3350 3180
2° amide (-CONH-R)	$N-H$ str.	3333-3268

6) Amine:- The 1° amines usually show two peak bands one at $3550-3350\text{ cm}^{-1}$ & other at $3450-3250\text{ cm}^{-1}$ due to a asymmetrical & symmetrical stretching of N-H bonds.

The 2° amine give only one band bet^o $3550-3350\text{ cm}^{-1}$

The 3° amine lacks N-H group so lack this band.

$\begin{array}{c} -NH_2 \\ (1^\circ \text{ amine}) \end{array}$	N-H asym str. N-H Sym str. N-H def. C-N str.	3550-3390 3450-3250 1650-1580 1220-1020 (aliphatic) 1340-1250 (aromatic)
$\begin{array}{c} -N-H \\ \\ (2^\circ \text{ amine}) \end{array}$	N-H str. N-H def. C-N str.	3550-3350 1650-1550 1350-1280 (aromatic)
$\begin{array}{c} -N- \\ \\ (3^\circ \text{ amine}) \end{array}$	C-N str.	1380-1260 (aromatic)

Aromatic hydrocarbons:- IR spectra of aromatic comp. is distinct from alkenes. To diagnose a comp. to be aromatic the below given bands are to be found in the IR spectra.

Aromatic Hydrocarbon	-C-H stre. C-H def. (out of plane) C-H def. (in plane) C=C stre.	3080-3030 900-650 1300-1000 1600, 1585, 1500, 1450
Monosubstituent	C-H def. (out of plane)	710-690 770-730
Disubstituent i) ortho ii) Meta iii) para	C-H def. (out of plane)	770-735 725-680 810-750 850-790.

Fundamental Modes of vibrations

At all temperatures incld including absolute zero the constituents atoms of the molecule vibrate and rotate in a variety of ways. The motions leading to change in bond distance (stretching) and bond angles (bending) are called vibrational motions. The stretching and bending vibration are known as fundamental modes of vibrations.

The number of fundamental modes for polyatomic molecule can be calculated from the number of atoms (N) present in the molecule and their degrees of freedom. The number of degrees of freedom is equal to the sum of coordinates necessary to locate all the atoms of the molecule in the space.

For a non-linear molecule having N number of atoms
No. of fundamental modes of vibration = $3N - 6$

$3N$ is degrees of freedom because three cartesian coordinates are needed to locate each other atom in space. These atomic degrees of freedom, three describe translation of the molecule as whole and three describe rotation of the molecule as a whole. These six degrees cannot lead to internal vibrations so, they are subtracted. For linear molecule there are only two axes (instead of three) about which the molecule as whole can rotate so,

The number of modes = $3N - 5$

Carbon dioxide is linear triatomic molecule, so use this formula, we get,

$$\begin{aligned}\text{No. of fundamental Modes of vibration} &= \\ &= 3 \times 3 - 5 \\ &= 4\end{aligned}$$

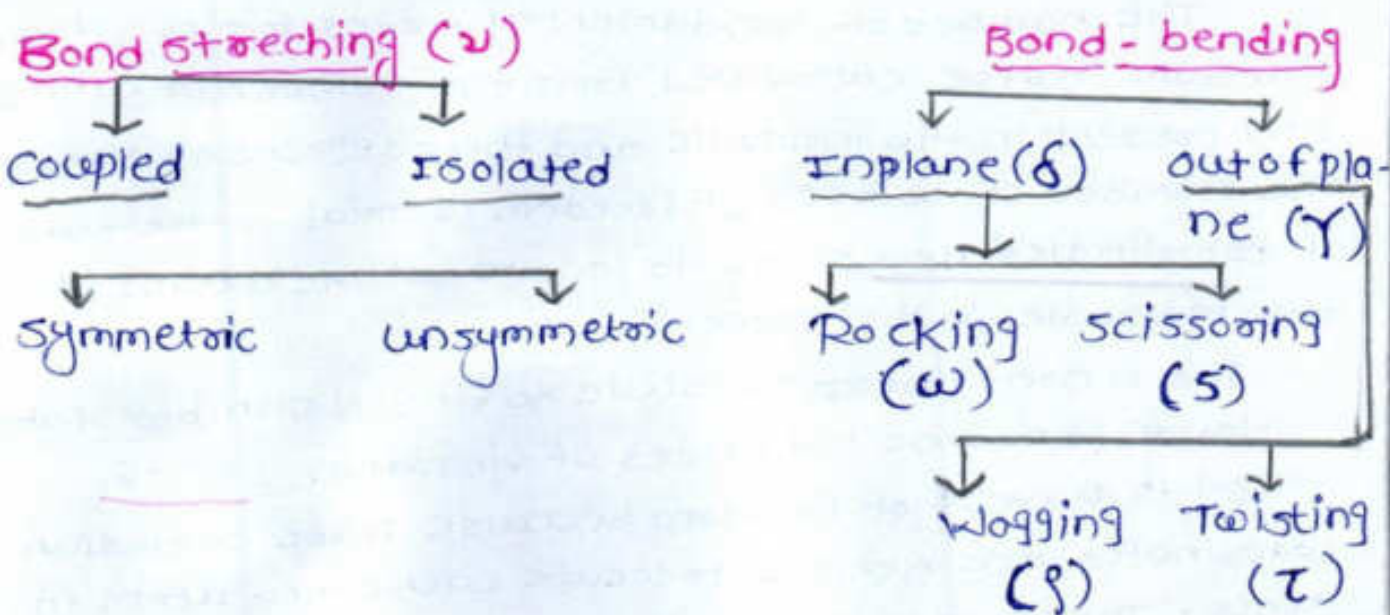
These four modes can be represented as below,



The arrows indicate movement of atoms along bond axis whereas + and - signify vibrations perpendicular to the plane of paper.

The fundamental vibrations are grouped and subgrouped as under.

Vibrations



① Bond stretching vibrations

They are vibrations wherein a pair of atoms constantly vibrate along their bond axis changing the bond length but maintaining constant bond angle. They can be of two types.

① Isolated - Two atoms undergo bond stretching vibrations with respect to each other e.g. C=O, O-H bonds.



① Isolated



Symmetric



Unsymmetric

② Coupled

Bond stretching vibrations

2
⑥ Coupled :- They are stretching vibrations of two atoms with respect to common central atom. They can be two types.

i] Coupled symmetric :- Both the atoms simultaneously move away or towards the central atom.

ii] Coupled unsymmetric - when one atom moves towards the other moves away from central atom.

stretching vibrations are represented by $\nu(\nu)$ followed by chemical group in bracket e.g. $\nu(\text{C=O})$ 1600cm^{-1} implies stretching vibrations of carbonyl group is observed at 1600cm^{-1}

The stretching vibrations generally require higher energy than the bending vibrations. So give the absorption band at higher frequencies ($4000-1400\text{cm}^{-1}$)

② Bond bending vibrations :- The central atom remains fixed to a point on a fixed plane but two atoms bonded to it vibrate to change the bond angle keeping the bond length constant. So, the shape of molecule changes hence it is also called deforming vibrations. They are of two types.



Rocking



Scissoring



Wagging



Twisting

(a) In plane.

(b) out of plane.

Bending vibrations

① In-plane bending vibrations (δ) - - ; coupled ②

The vibrating atoms remain in same plane along with the common atom. They can be of two types.

i] Rocking vibrations (ω) - The vibrating atoms move simultaneously back (\leftarrow) or forth (\rightarrow)

ii] scissoring vibrations (ς) - The vibrating atoms move back and forth towards each other.

② out of the plane bending vibrations (γ) -

The vibrating atoms move out of the plane of central atom. They can be of two types.

i] wagging vibrations (ρ) - The vibrating atoms simultaneously move back (-) or forth (+) of the common plane.

ii] Twisting vibrations (τ) - - ; antisymmetric vibrations ③
The vibrating atoms move in opposite direction to common plane.

The bending vibrations require lesser energy than stretching vibrations so, they give the absorption band at lower frequencies ($1550 - 500 \text{ cm}^{-1}$)

The intensity of absorption due to stretching vibration is more than that due to deforming vibrations.

In addition to these fundamental modes of vibrations, combinations of vibrations may occur.

The actual number of fundamental frequencies depend on the symmetry of molecule.

Conditions for the absorption of IR-radiation

The necessary conditions for the absorption of infrared radiation are as under.

① Natural frequency of vibration of the molecule must be equal to frequency of incident radiation - only those IR frequencies which match exactly with the vibrating frequency of group of atoms are absorbed and the amplitude of vibration is increased when absorbed energy is lost as heat, the molecule reverts back to ground state and amplitude is restored.

② changes in the vibrations must simulate changes in dipole moment of molecule - when a polar covalent bond undergo stretching vibration along the bond axis the electron distribution changes, so also the dipole moment changes.

periodic changes in the dipole moment accompanying the vibration generate secondary electric field which interacts with electric vector of electromagnetic radiation leading IR-absorption.

Non-polar molecules like N_2 , O_2 , Cl_2 cannot absorb IR-radiation as they do not have permanent dipole moment.

polyatomic molecule like CO_2 has no permanent dipole moment but vibrations give rise to changes to dipole moment. Therefore it absorbs IR radiation.

③ Intensity of absorption must be proportional to rate of change of dipole -

IF the rate of change of dipole is slow absorption of radiation is weak and if it is fast absorption of radiation is strong.

Selection rule:-

The selection rule for IR-spectrum says that, only those vibrations are effective in causing IR-absorption which are not centro-symmetric.

(i.e. vibrations are not symmetric about the center of the molecule). So we have —

i] IR-active transitions:-

The vibrations that generate oscillating dipole moment absorb in IR and are known as IR-active transitions.

e.g. - H-Cl , I-Cl , $\text{C}\equiv\text{O}$, CHCl_3

ii] IR-inactive transitions - The vibrations that ~~occurs~~ doesn't occur generate an oscillating dipole moment do not absorb in IR are known as IR-inactive transitions.

e.g. - symmetrical stretching of $\text{C}=\text{C}$ bond in ethylene (centrosymmetric molecule) will not produce change in dipole moment of the molecule. Therefore these mode of vibration is infrared inactive.

e.g. - H_2 , N_2 , Cl_2 , CCl_4 etc.

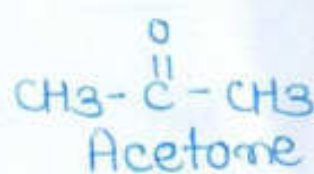
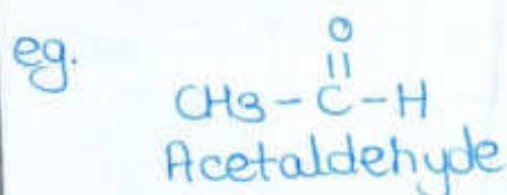
iii] Allowed and Forbidden transitions:- only the transitions from ν_0 to the next higher level ν_1 ($\nu_0 \rightarrow \nu_1$) or $\nu_1 \rightarrow \nu_2$ etc. are allowed for idealised system such perfectly harmonic oscillator. These are called fundamental transitions and only these transitions are allowed.

Generally molecular vibrations are not strictly harmonic, so overtones band are also seen and their frequency will be 2ν , 3ν etc. These are known as Forbidden transitions detected as weak overtones in the spectrum.

Factors Affecting IR absorption Frequency

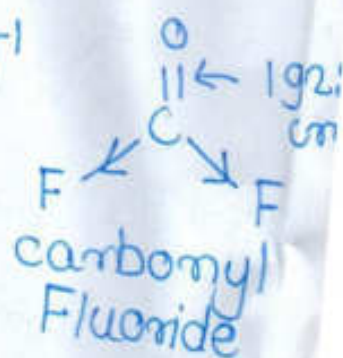
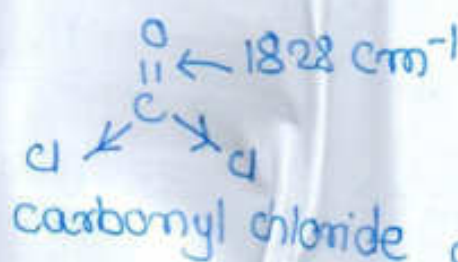
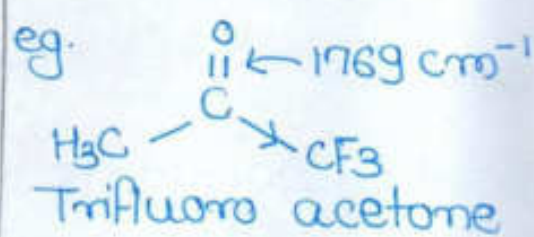
1) Inductive Effect -

The electron releasing group (+I effect) attached to functional group decreases stretching frequency & shift IR band to lower side.



An extra methyl group (exerts +I) in acetone decreases the double bond character of $>\text{C}=\text{O}$ group. The decreased bond strength raises the vibrational frequency & push IR absorption to lower side (1710 cm^{-1}).

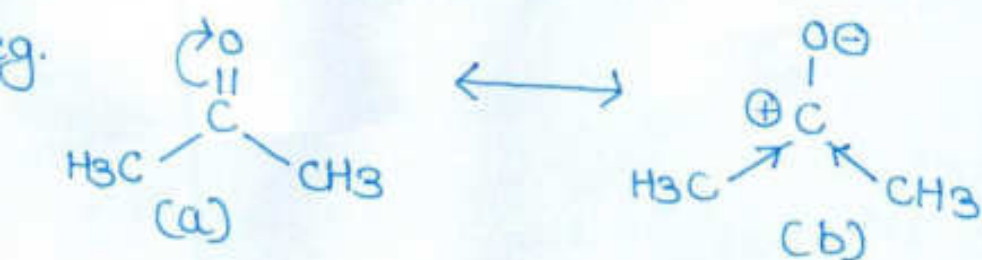
The electron withdrawing (-I effect) group stabilise double bond character of $\text{C}=\text{O}$ group & increases stretching frequency to shift IR band to higher side.



Fluorine & chlorine are highly electronegative atoms exerts highest inductive effect. when they are attached to carbonyl group extreme shift is observed.

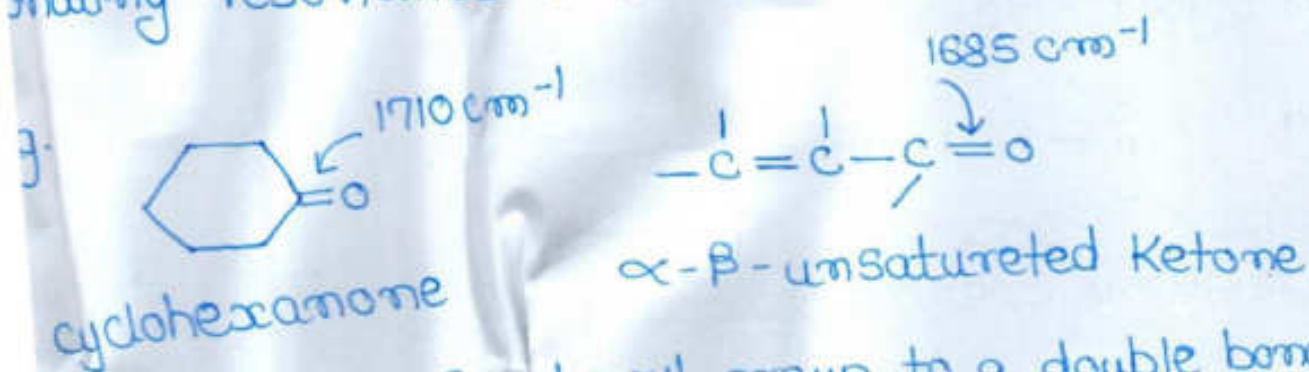
2) Resonance (R) Effect -

The stretching frequency of functional group is decreased by electron donating resonance (+R) effect. & increased by electron withdrawing resonance (-R) effect.



The +I inductive effect of two methyl groups helps to stabilize the single bonded resonance form (b). As single bond energy is less than for double bond energy the stretching frequency decreases from 1720 cm^{-1} (acetaldehyde) to 1710 cm^{-1} (in acetone).

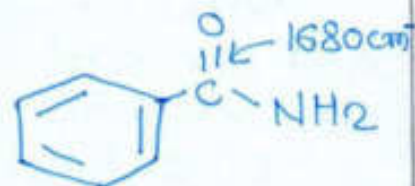
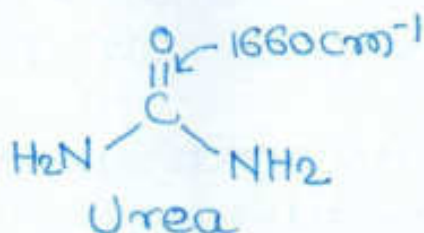
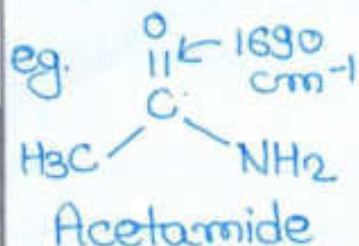
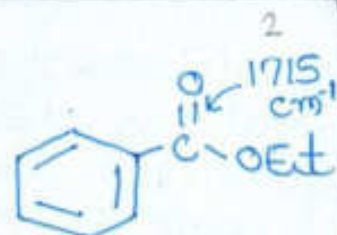
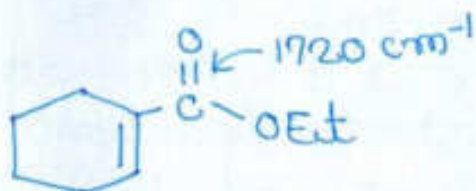
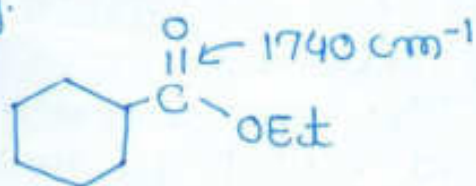
Conjugated aldehydes & ketones show electron donating resonance (+R) effect.



Conjugation of carbonyl group to a double bond exerts (+R) effect. This decreases double bond character of $>C=O$ group decreasing stretching frequency.

In amines +I effect decreases stretching frequency considerably to as low as 1690 cm^{-1} compared to all other carbonyl compounds.

eg.



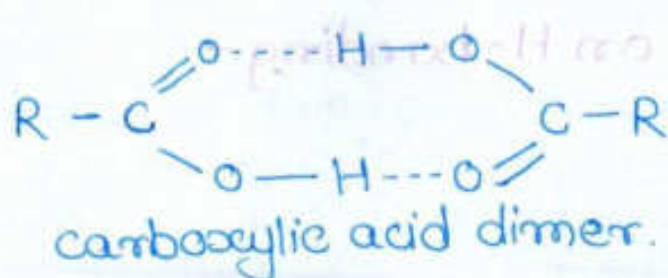
All acid derivatives (Amides, esters, acid chlorides) show prominent resonance effect. If there is additional conjugation effect carbonyl stretching frequency decreases.

3) Hydrogen bonding -

Hydrogen bonding lowers the absorption frequency of functional group involved. The hydrogen bonding weakens the bond by increasing bond length & pushes absorption to lower side. Thus, it affects stretching frequencies of group like $-\text{COOH}$, $-\text{OH}$, $-\text{NH}_2$ etc.

a) Carboxylic acids -

In solid or pure liquid form strong hydrogen bonding occurs & exists as dimers. This affects stretching frequencies of both carbonyl & hydroxyl group.



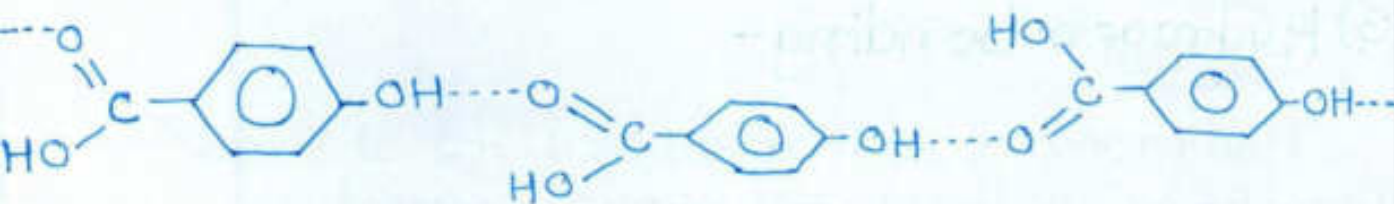
The involvement of $>C=O$ bond in H-bonding weakens the $C=O$ bond thereby lowering the carbonyl stretching frequency. The appearance of peak at 1721 cm^{-1} indicates acid exist as dimer.

H bonding stretches $O-H$ bond decreasing the bond strength. This shifts absorption to lower side.

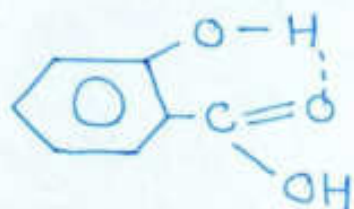
O-H	H-bonded OH stretching frequency	$2700-2500\text{ cm}^{-1}$
O-H	Free OH stretching frequency	3500 cm^{-1}

Intramolecular hydrogen bonding shows greater effect on $C=O$, $O-H$ stretching than intermolecular H-bonding.

eg. p-Hydroxy benzoic acid & o-Hydroxy benzoic acid.



p-hydroxy benzoic acid (intermolecular H-bonding)



o-Hydroxy benzoic acid (intramolecular H-bonding)

Effect of dilution on H-bonding -

Dilution with aprotic solvent prevent intermolecular H-bonding. This stabilizes $O-H$ bond & shifts IR absorption to lower side. So, it is called dilution effect.

eg. Lack of H-bonding shifts $>C=O$ & $-O-H$ group frequencies to higher side 1760 cm^{-1} & 3550 cm^{-1} respectively



Both $C=O$ & free $O-H$ groups are free from H-bonding.

b) Alcohols & Phenols -

Strong H-bonding effects are also observed in alcohols & phenols. Alcohol show peak in $3200-3500\text{ cm}^{-1}$ range due to H-bonded $O-H$ group. But in gas phase or in extreme dilution in aprotic solvents H-bonding is missing & stretching frequency shift to 3650 cm^{-1} .



Alcohols & Phenols (intermolecular H bonding)



Alcohols & Phenols (intramolecular H bonding)

Dilution effect is not observed if H-bonding is intramolecular. Thus comparison of IR spectra before & after dilution can be used to assess nature of H bonding.

c) Amines & Amides -

Strong H-bonding effects are observed in $1^\circ/2^\circ$ amines & $1^\circ/2^\circ$ amides. The H-bonding effect is not shown by 3° amines & amides as they lack N-H bond.

4) Effect of Strains -

Angle ring strain in cyclic compound causes large shifts in frequency. Thus, as ring size decreases angle strain increases this causes increase in carbonyl stretching frequencies!



cyclohexanone 1710 cm^{-1}



1740 cm^{-1}

cyclopentanone



1775 cm^{-1}

cyclobutanone



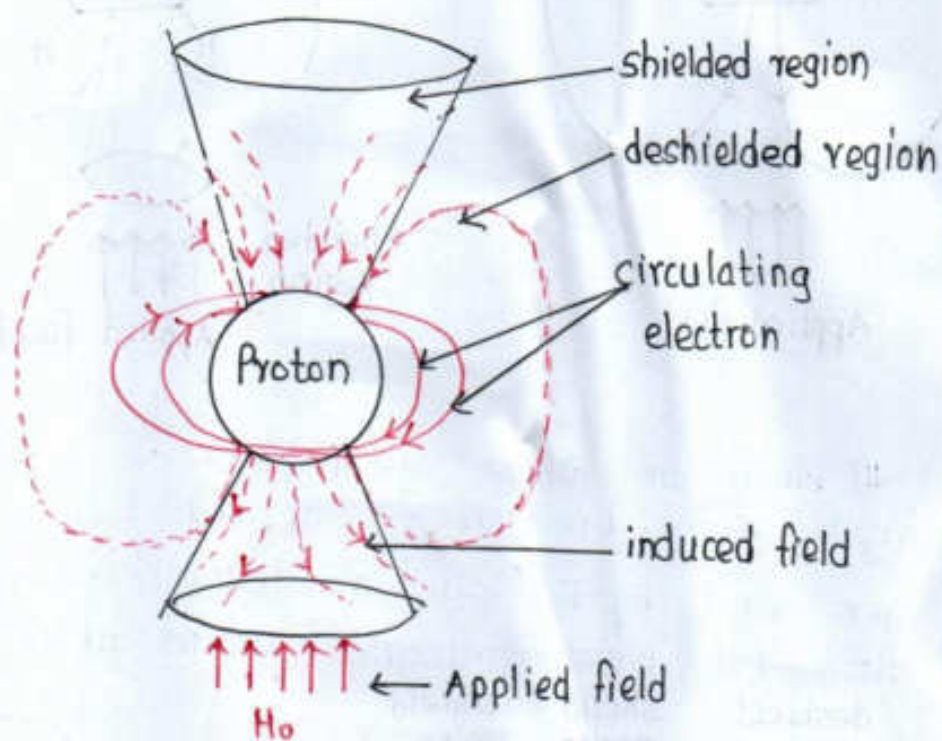
1818 cm^{-1}

cyclopropanone

Shielding, Deshielding Anisotropic effect :-

Electrons surrounding a nucleus, under the influence of magnetic field will circulate and they generate their own magnetic field. This is opposed to applied field at the nucleus as show in fig. below. Because of this induced field opposes to applied field, electrons are said to be diamagnetic effect on the nucleus is called diamagnetic shielding.

Because the nucleus experiences weaker magnetic field than that of applied externally it is said to be shielded, and this type of shielding called as diamagnetic shielding. And when induced field augment or reinforce the applied field it is called as paramagnetic deshielding. In the deshielding, nucleus experience same magnetic field as applied externally.



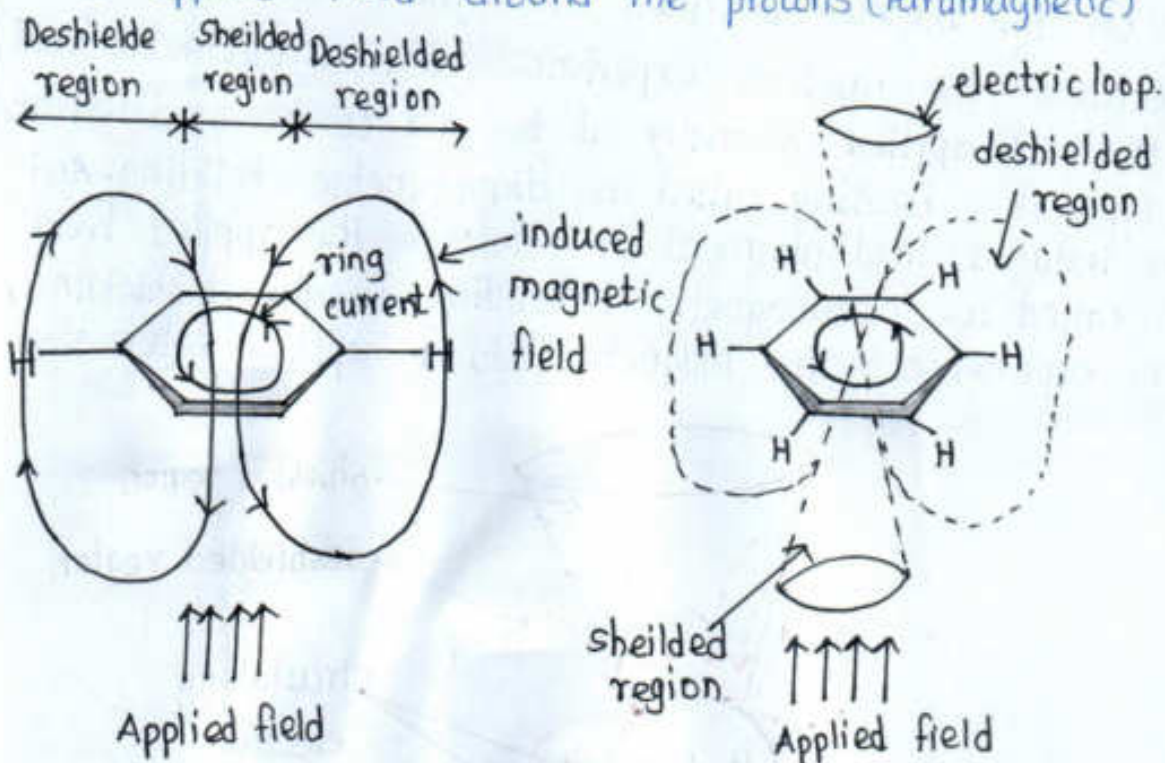
Anisotropic effect:-

In the presence of applied magnetic field circulation of π - e^- and generate diamagnetic current which produce its counter magnetic field called induced magnetic field which oppose the applied magnetic field shows diamagnetic effect or Anisotropy.

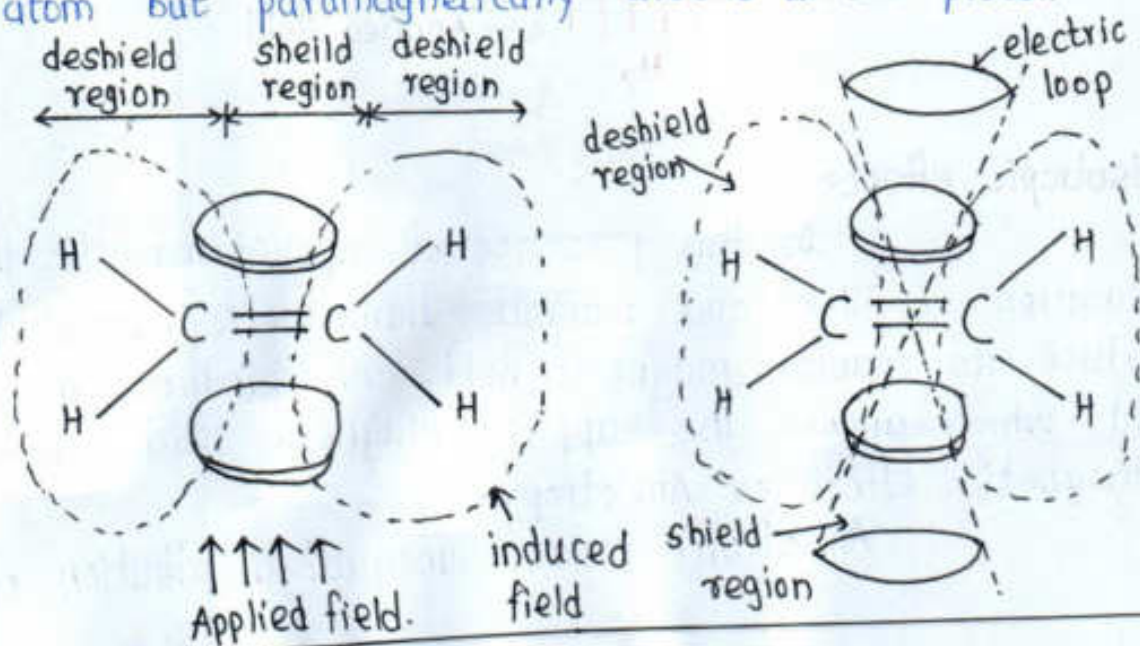
Anisotropy means unequal distribution of

electron because circulation of $\pi-e^-$ in applied magnetic field. If protons are present in electric loop called shielded proton surrounded by electron. Proton from outside electric loop called deshielded proton.

i] In benzene the induced magnetic field generated by the circulation of π -electron opposes the applied field (diamagnetic) in the centre of the ring. But it reinforces the applied field around the protons (Paramagnetic)



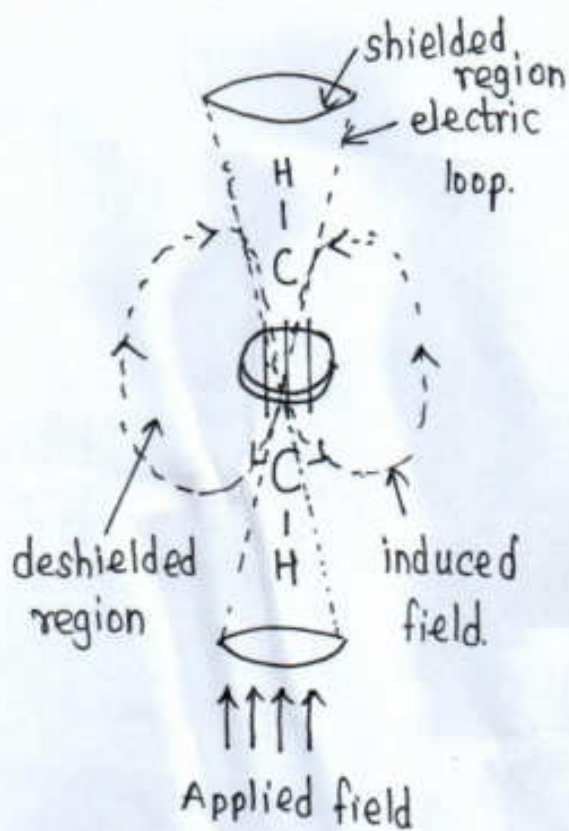
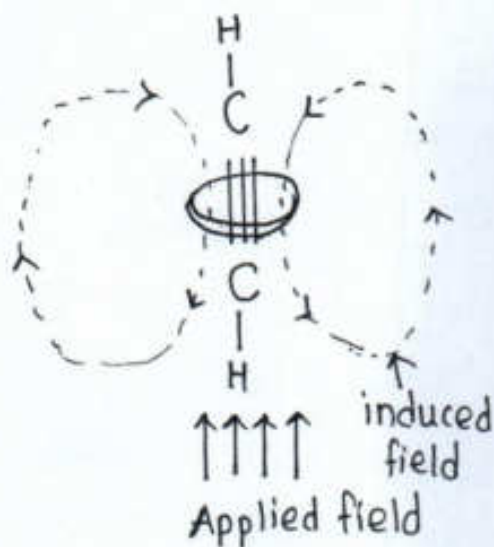
ii] When an alkene group ($-C=C-$) is oriented at right angle direction of applied field, the induced field generated by $\pi-e^-$ act as diamagnetically around carbon atom but paramagnetically around alkene proton.



2

iii] The electron distribution in triple bond of acetylene is symmetric about the axis. When axis is parallel to applied field, electronic circulation perpendicular to the axis will be induced. This causes shielding of acetylenic protons. So acetylenic proton absorb further upfield than vinylic protons.

deshielded region shielded region deshielded region

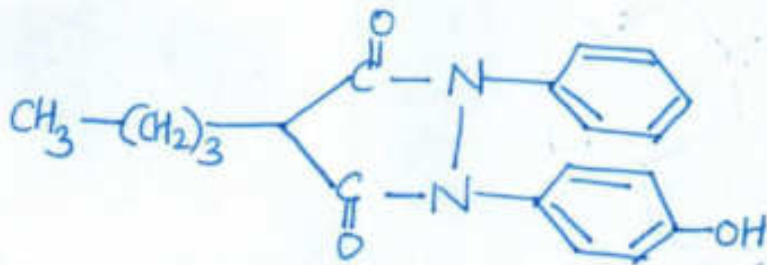


Drug

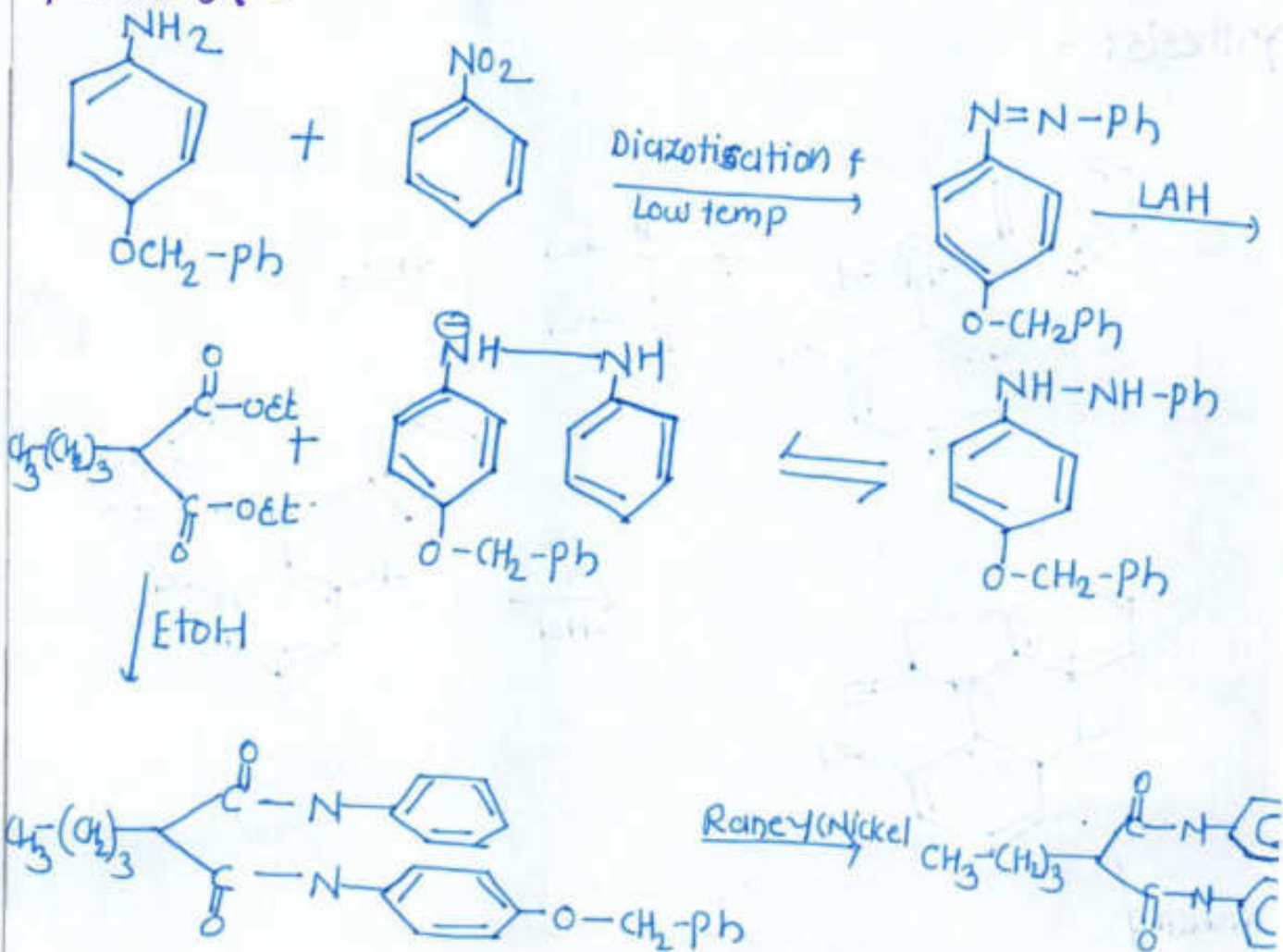
Anti inflammatory drug

1) oxyphenyl butazone

Structure: -

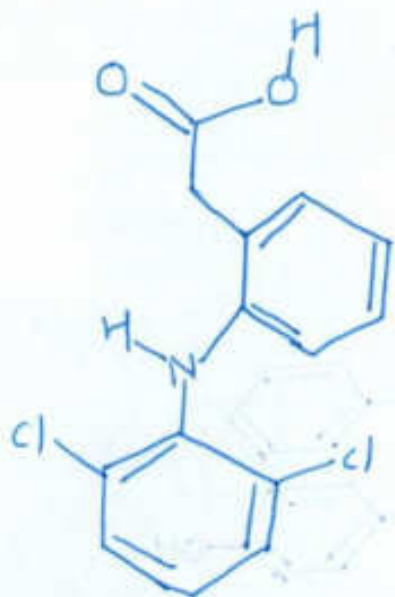


Synthesis: -

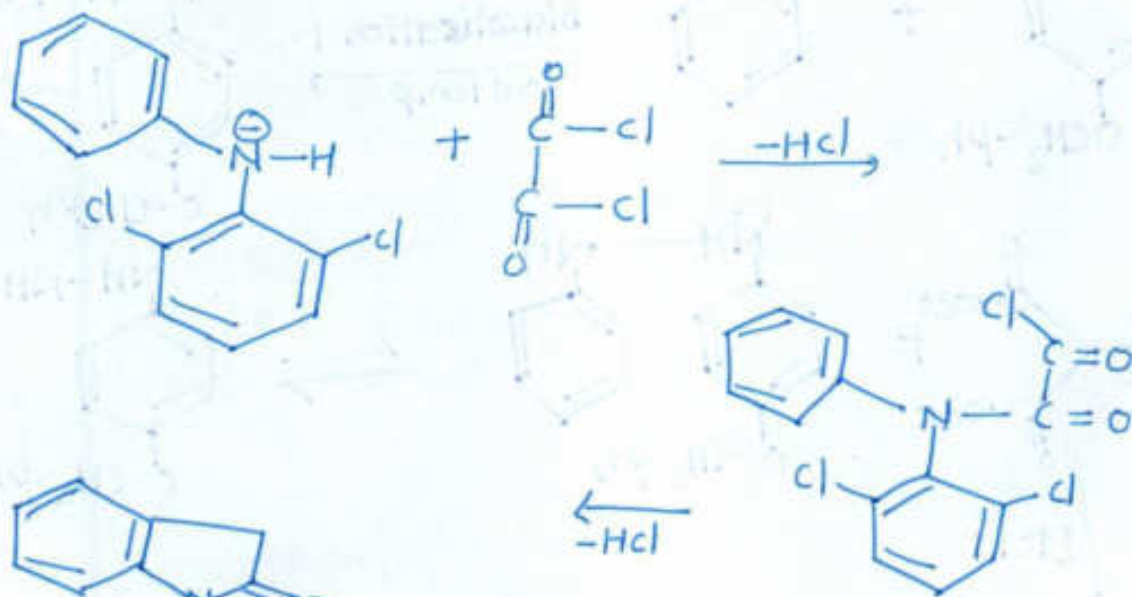


2) Dichlophenac: -

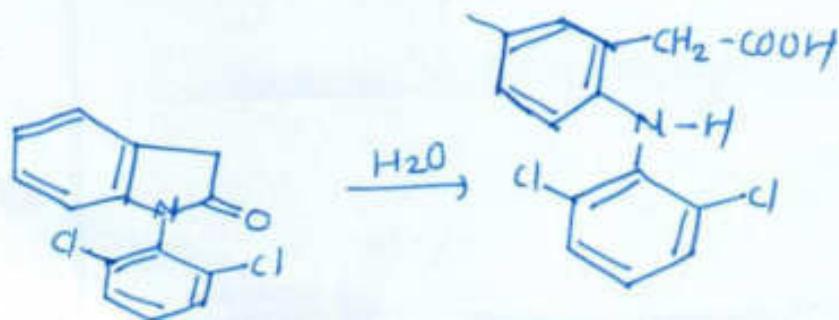
Strukturze: -



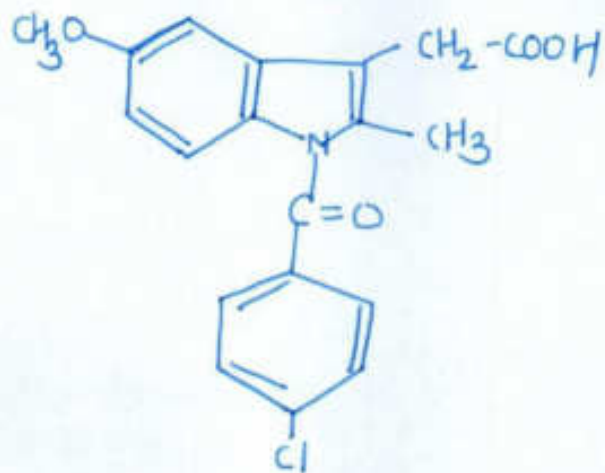
Synthesis: -



Reduction

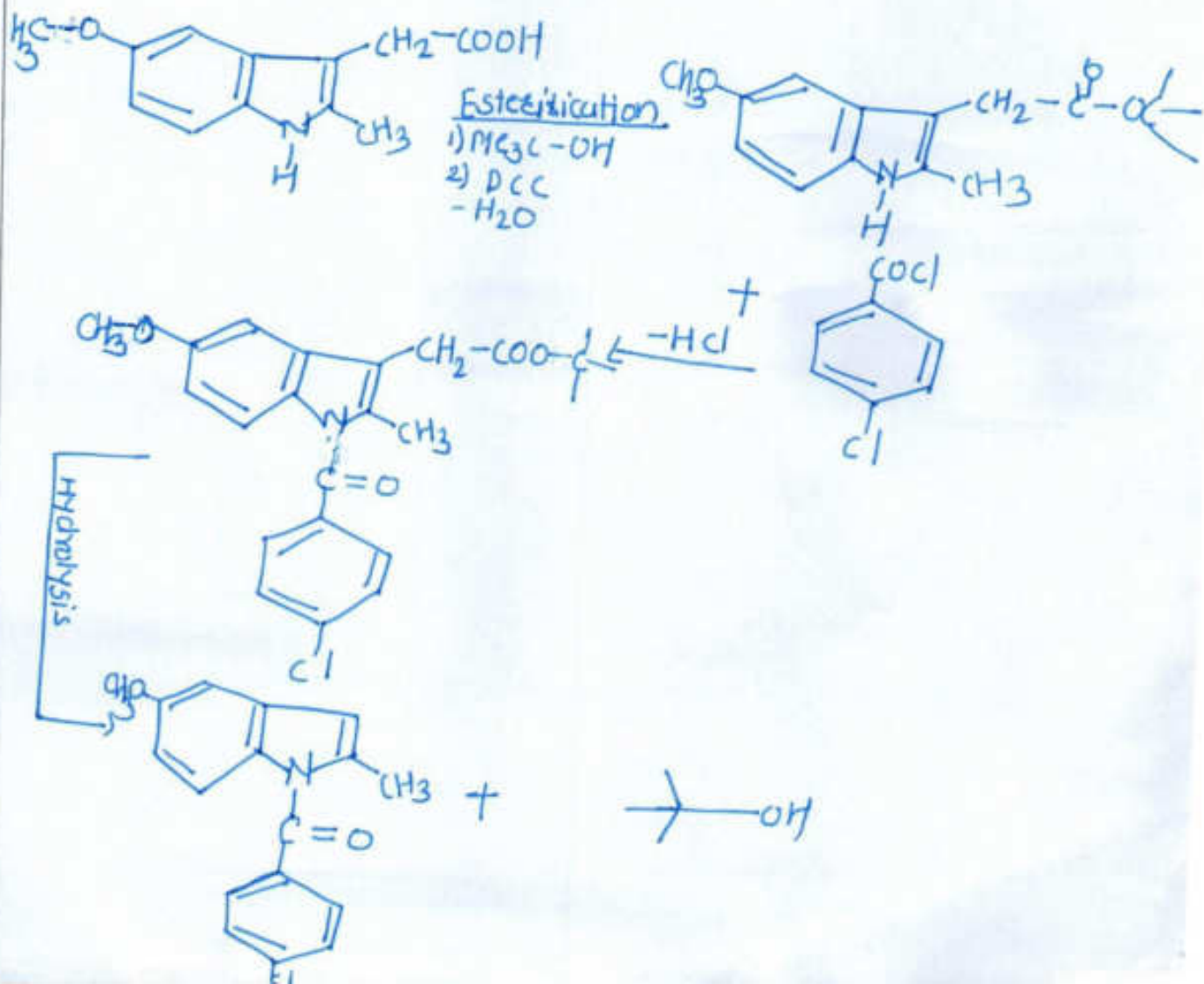


Indomethacin: -



It is used as standard anti-inflammatory its activity short duration. The main disadvantage of this drug is its little solubility in water.

Synthesis: -



Drug & Heterocycles

Introduction of Drugs -

" pharmaceutical chemistry is a branch of organic chemistry which deals with the study of various drugs".

Pharmaceuticals or drugs are used for the diagnosis or diseases so as to increase the life span of human and animal. Life is a chemical process in which major change that occurs are due to chemicals. Hence its of the body must be cured by chemicals.

"Paul Ehrlich" in 1909 discovered the term chemotherapy. "The chemical compound which kills the microorganism of the body and these chemicals compound have therapeutic value are called as drugs". The Scientist Ehrlich discovered many drugs. The first drug is methylene blue in 1909 for treatment of disease called malaria. Then he propose a theory called drug action on a body". Hence due to this he was called as father of chemotherapy. gave the characteristics of ideal drug.

- i) The drug should have less side effects.
- ii) It must act on specific system with efficiency safely.
- iii) It should be localised at the site where it is eq".
- iv) The drug should be bacterioside rather than bacteriosite.

↳ It should destroy pathogenic state micro organism without affecting host tissue.

Drug Design -

1) Scientist Hippocrates (Greek physician scientist).

Disease is a pathological process.

2) Paracelsus (G.P) - Ill of body can be overcome chemicals.

e.g - chloroform

hypnotics - (Sulphur)

analgesic -

antihistatic

Concept of pro-drug -

All therapeutic agent possess various physico-chemical & biological property which includes the desirable & undesirable property.

Researchers in pharmaceuticals are minimising the magnitude & no. of undesirable property of a drug while retaining enhancing desirable i.e. therapeutic property.

Def - prodrug - A prodrug is a pharmacological substance or prodrug i.e. administer drug in an inactive form & get metabolised in vivo into the active comp.

• Drug derivative -

Irreversible - Development new drug molecules.

- To overcome drawbacks.

- To improve potency

- To incorporate some other desirable property in the existing of parent drug mol.

Reversible - This comp upon introduction to the appropriate biological env revert to the enzymatic & non enzymatic means & elisite expands at specific site.

Albert - 1950 are discovered.



Prodrug

enzyme or
Hydrolysis



release of drug at
site of action.

purpose - Improve patients Acceptability.

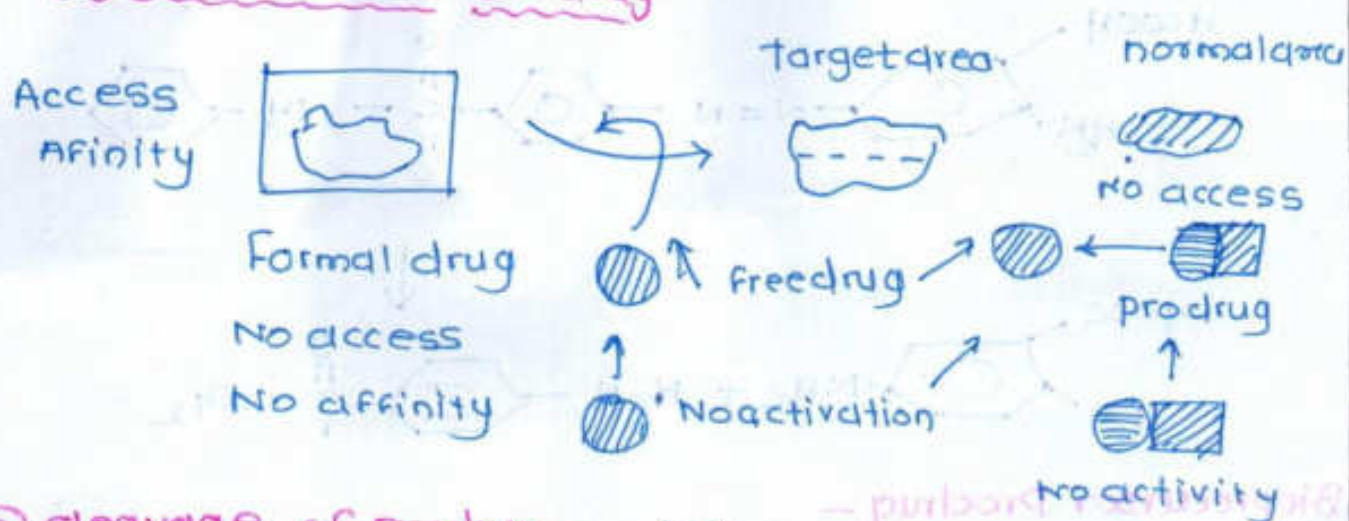
- After / improve absorption.
- After biodistribution.
- After elimination.
- Reduce toxicity or Advert effect.

Characteristics -

- low oral absorption property.
- lack of site specificity
- chemical instability
- bad test.
- odour

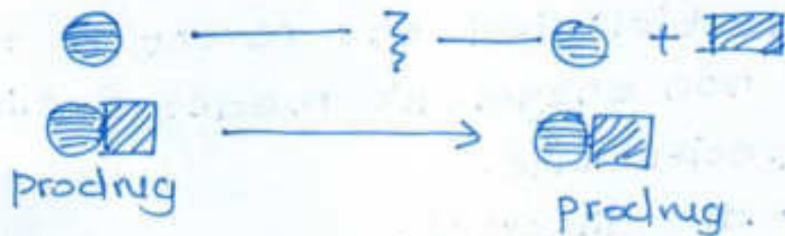
Classification of drug -

① Transformation of drug -



② cleavage of prodrug - Active site.

3) Retention of drug -



To improve the ch. is three type.

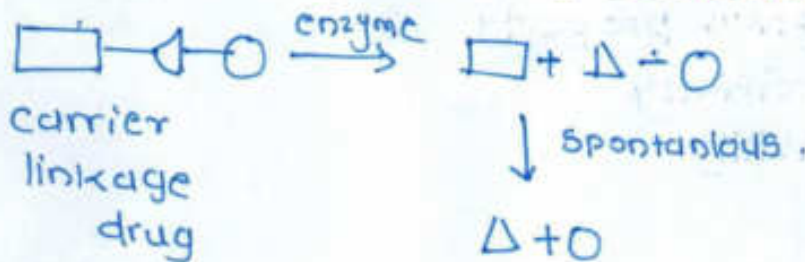
Classification of prodrug -

Carrier linked prodrug -

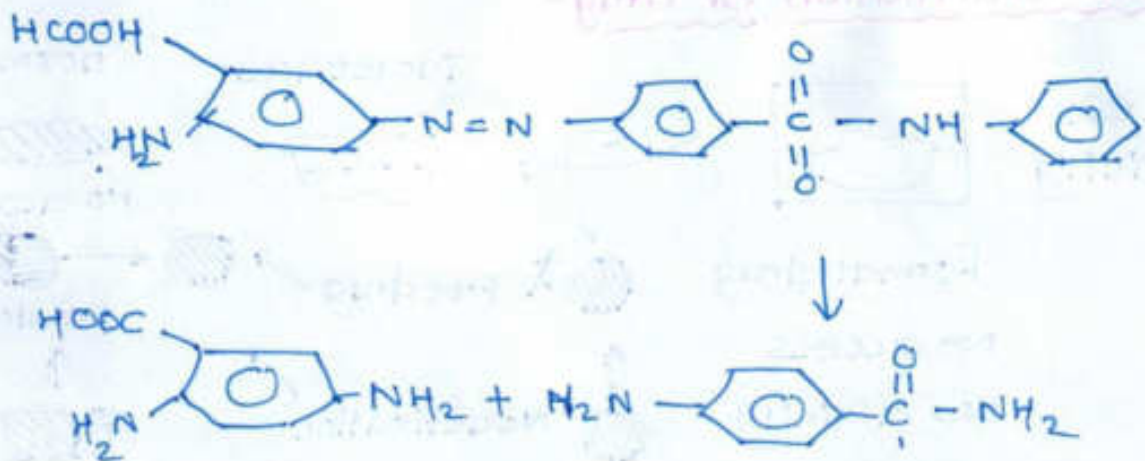


- ① Bipartite - one linkage
- increase in stability in H₂O improve absorption & distribution.

- ② Tripartite - carriers not connected directly.



- ③ Mutual prodrug -



Bioprecursor prodrug -

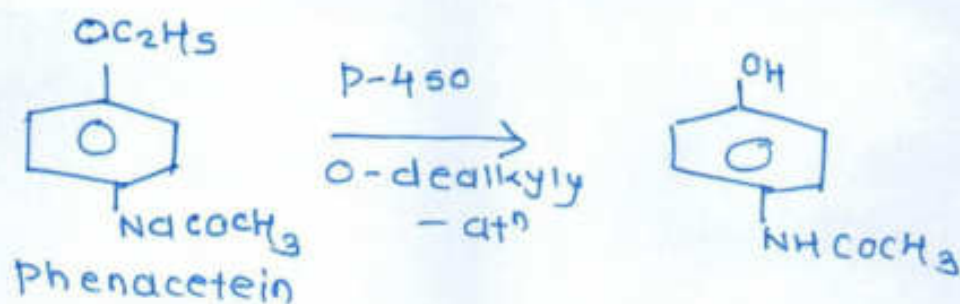
Oxidative
Activation

Reduction
Activation

Nucleotide
Activation

polyphosphory-
lation

eg



Type of drug -
functional drug -

Central nervous system - (brain & spinal cord).
Peripheral Nervous system - (Nervous & ganglia).
Used to stimulate reduce pain & suffering

- 1) CNS - stimulant e.g. caffeine.
- 2) Transquilisers - improve mood & behaviour.
e.g. - chlorpromazine.
- 3) Hypnotics - (Luminal)
- 4) Sedatives - (Luminal).
- 5) Anticonvulsant (Noctall)
- 6) Analgesics - Relieve the pain.

② i) Analgesic - e.g. - Aspirin, paracetamol.

- ii) Antipyretics
- iii) Antihistamine -
- iv) Antidiabetics
- v) Anti-inflammatory agents.
- vi) Cardiac drug.

These are four types as follows.

- | | | | |
|------------------------|-------------------------------|-----------------------|--------------------|
| ① Cariotonic drug | ② Antipyretic - lessive Agent | ③ Antipyretic chloral | ④ stereosing agent |
| ⑤ Antihistaminic agent | ⑥ Vasodilator. | (Reduce the Choice) | |

Singlet Molecular Oxygen Reactions.

1) The scientist Kaustkey discovered reactivity of singlet oxygen.

2) The oxygen (O_2) molecule called as dioxygen & dioxidene ($O=O$) oxygen molecule is gases in organic chemical compound. its reactivity depends on ground state & excited state nature of oxygen.

1) Oxygen in its Ground State:

Molecular oxygen (O_2 atomic No. 16) Ground state Electronic Configuration.

$$1s^2, 2s^2, 2p^6, 3s^2, 3p^4$$

$$6(1s)^2, 6^*(1s)^2, 6(2s)^2, 6^*(2s)^2, 6(2p_z)^2, \pi(2p_x)^2 = \pi(2p_y)^2, \pi^*(2p_x)^1, \pi^*(2p_y)^1, 6^*(2p_z)^0$$

$$= \begin{array}{|c|c|c|} \hline 2p_x & 2p_z & 2p_y \\ \hline \uparrow & & \uparrow \\ \hline +1 & 0 & -1 \\ \hline \end{array}$$

$$M_L = 0 = S = \Sigma$$

$$M_S = 1, (2S+1) = 3$$

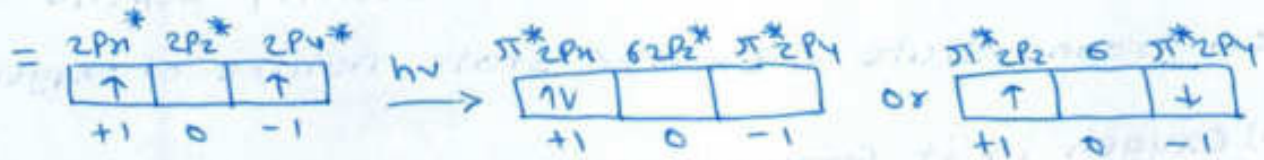
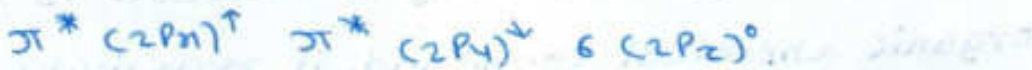
$$^3 \Sigma_g^-$$

Ground state = 3O_2 triplet



2) Oxygen in Excited state:

Excited state:



$$M_L = 0 = S = \Sigma$$

$$M_L = 2 = d = \Delta \quad M_L = 0 = S = \Sigma$$

$$M_S = 1, (2S+1) = 3$$

$$M_S = 0, 2S+1 = 1 \quad M_S = 0, 2S+1 = 1$$

$${}^3 \Sigma_g^-$$

$${}^1 \Delta_g$$

$${}^1 \Sigma_g^+$$

Excited = 3O_2 triplet state = 1O_2 singlet state



1) The ${}^1\Delta_g$ singlet oxygen is mostly responsible for photo-oxidation reaction.

2) Triplet to ~~to~~ singlet energy difference is 22-23 kcal/mol.

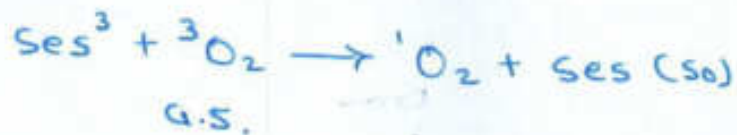
3) life time of singlet oxygen depends on solⁿ phase in water 2×10^{-6} sec. Deuterium 20×10^{-6} sec.

$$CHCl_3 = 60 \times 10^{-6} \text{ sec. } CCl_3 = 300 \times 10^{-6} \text{ sec.}$$

$$CFCl_3 = 1000 \times 10^{-6} \text{ sec. indicate lifetime.}$$

Synthesis of singlet oxygen:

1) sensitized oxygenation:



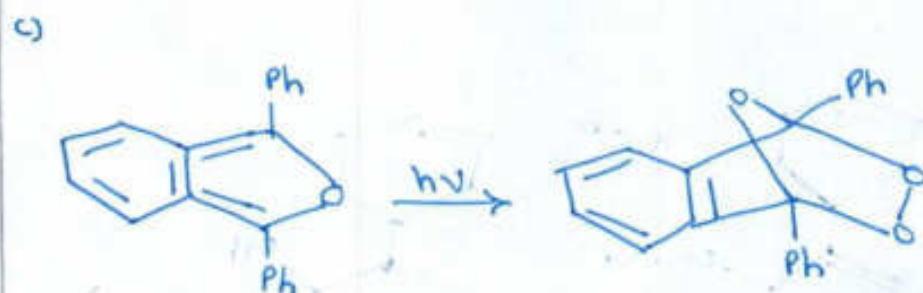
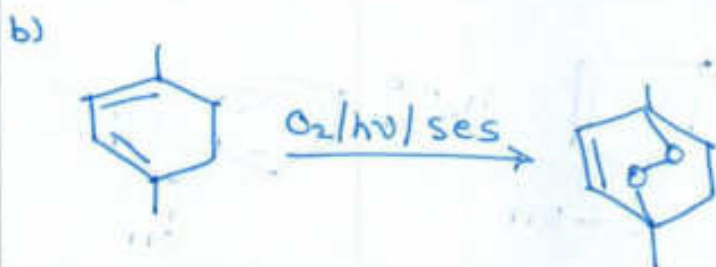
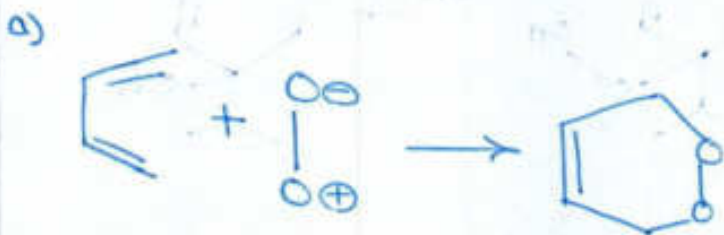
2) Thermal decomposition:

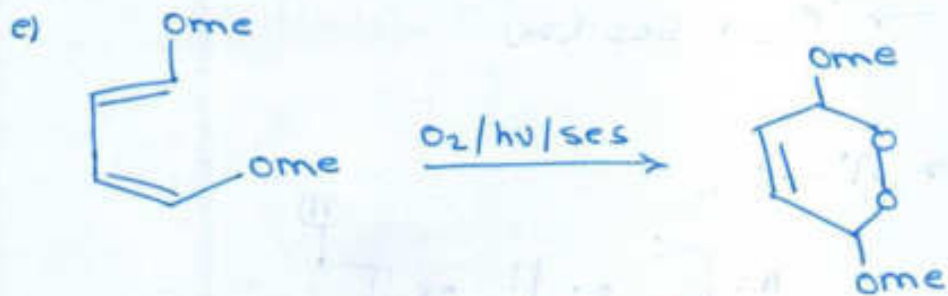
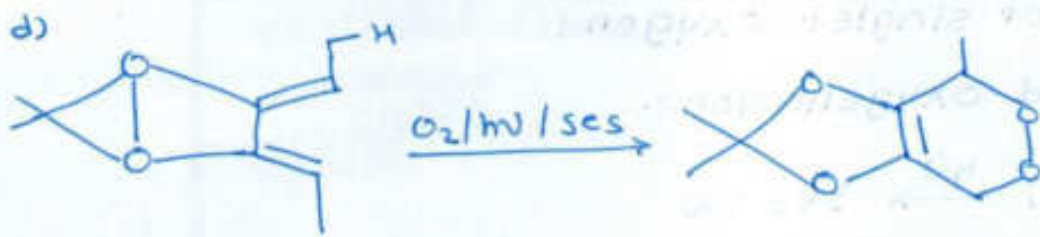


Reactions of singlet oxygen:

1) 4+2 cycloaddition:-

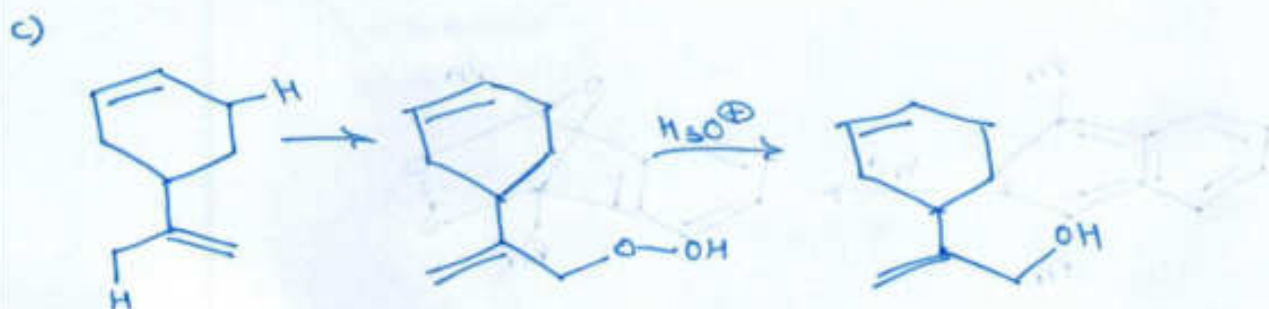
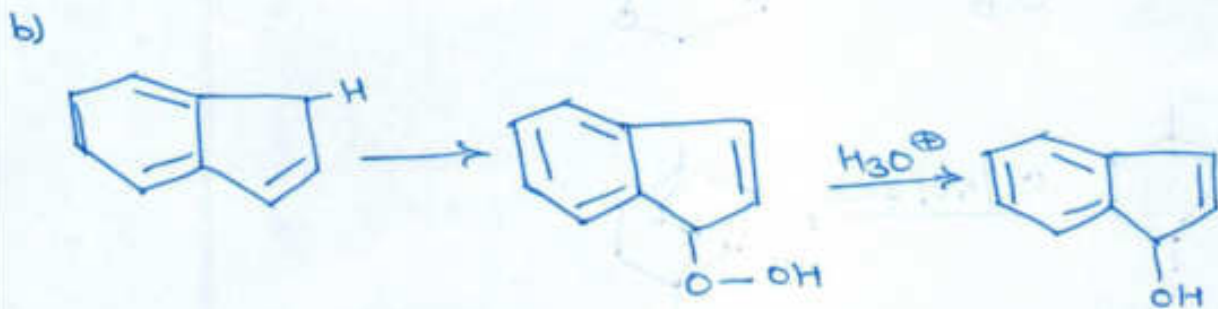
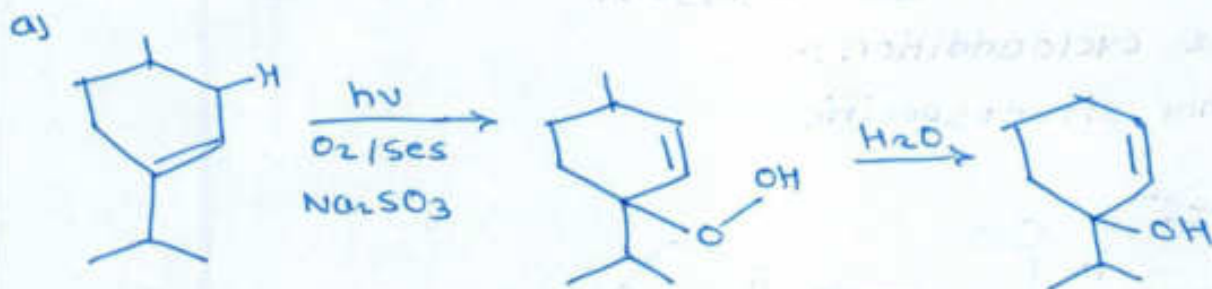
Highly stereospecific



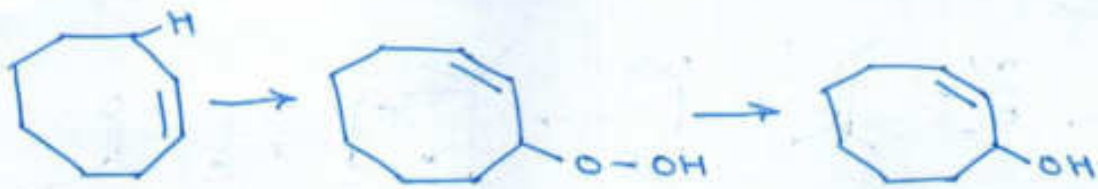


2) Ene type reaction:

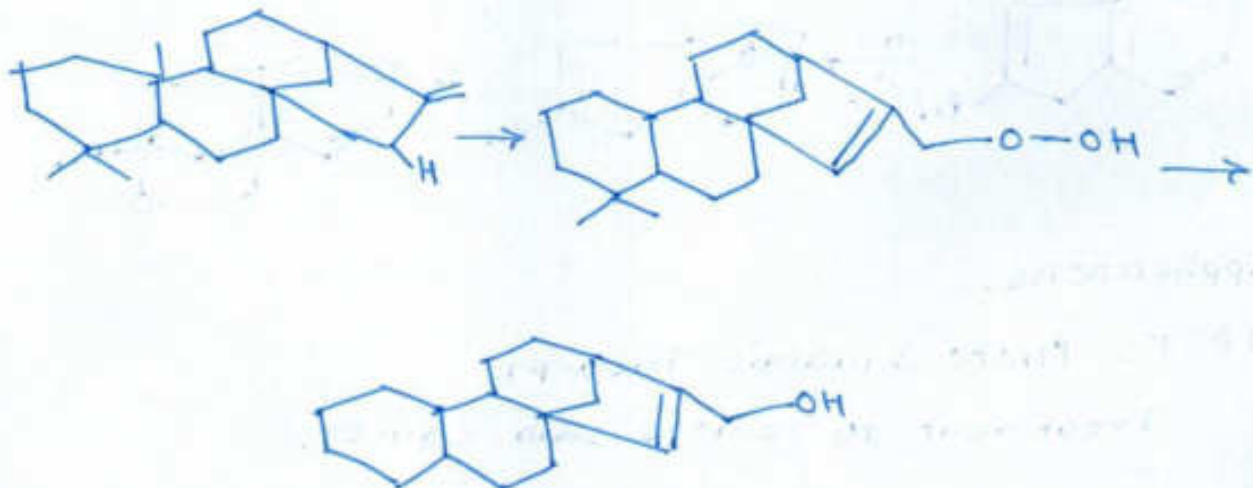
Alkene having allylic Hydrogen



d)



e)

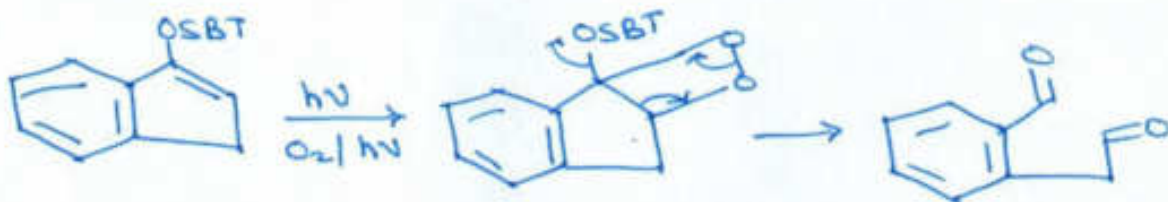


3) 2 + 2 cycloaddition:-

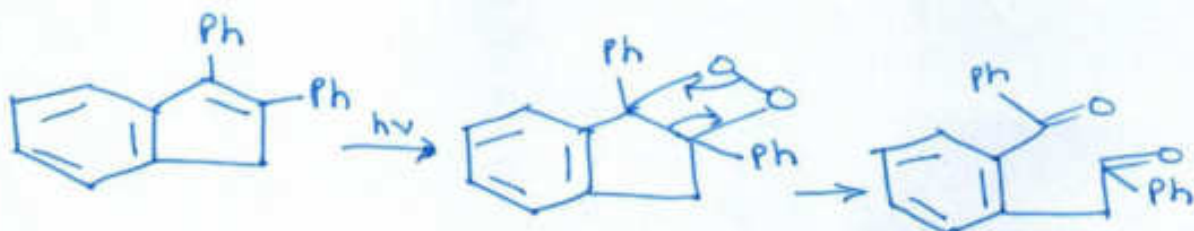
e^- rich alkene gives stereospecific addition but decomposes easily.

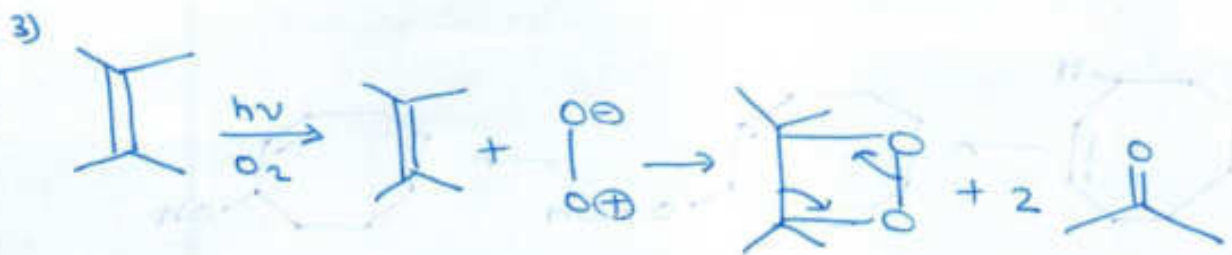


1)



2)

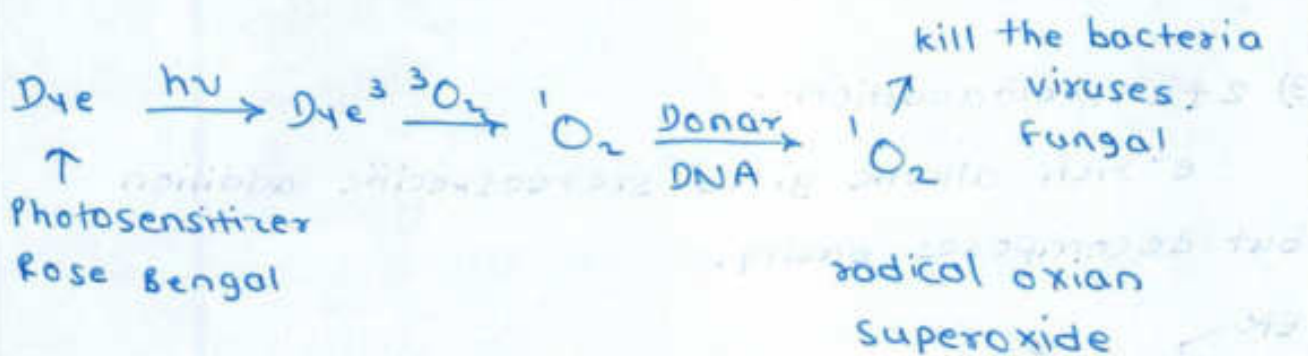




Applications:

1) PDT: Photo Dynamic Therapy

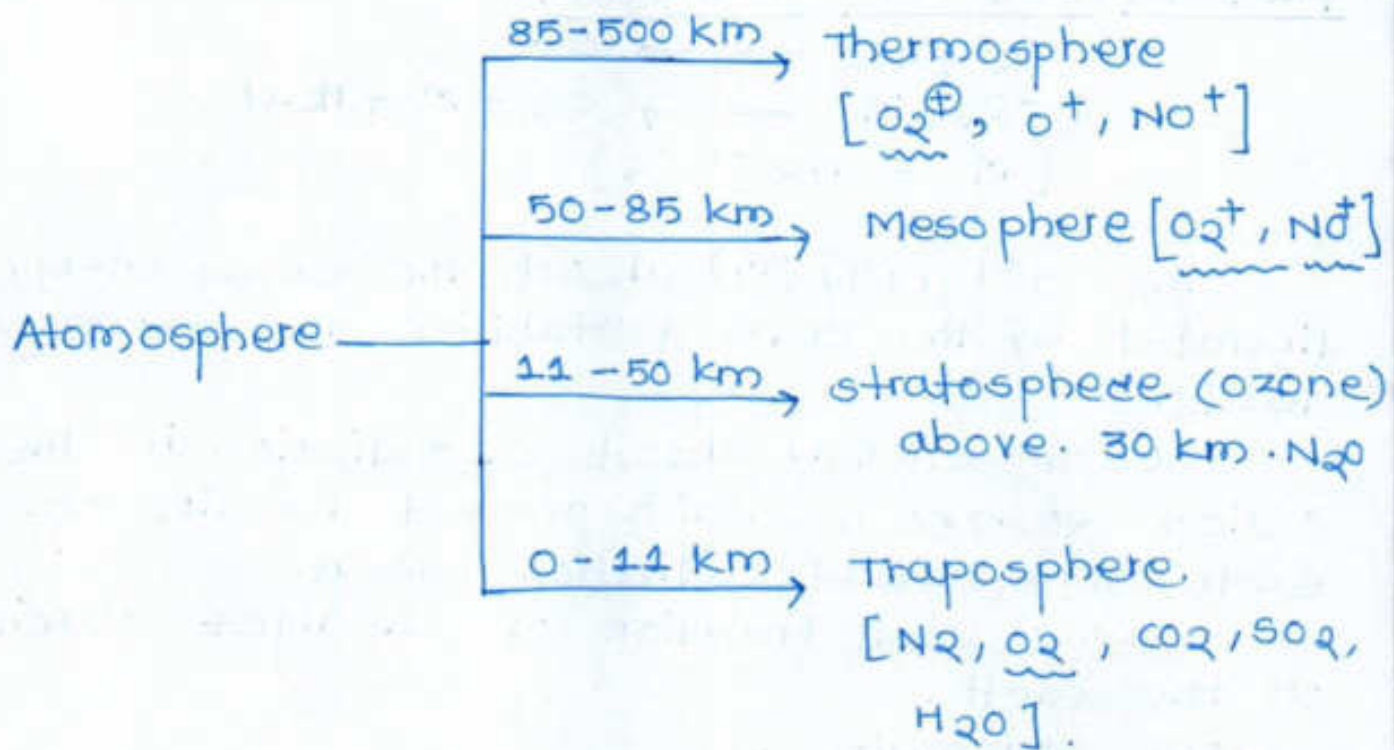
Treatment to remove skin cancer.



Photodissociation Gas phase photolysis

"The study of photochemical reaction in atmosphere in presence of sunlight called as gas phase dissociation."

The gas phase photochemical reaction depends on the atmospheric structure



oxygen play imp role in Troposphere where as ozone (O_3) play key role in stratosphere.

1) Dissociation of oxygen :-

A chemical activity of oxygen plays an imp role in lower atmosphere. The stable form of almost all elements are oxides.

UV light causes photochemical dissociation & ionisation.



2) Dissociation of ozone :-

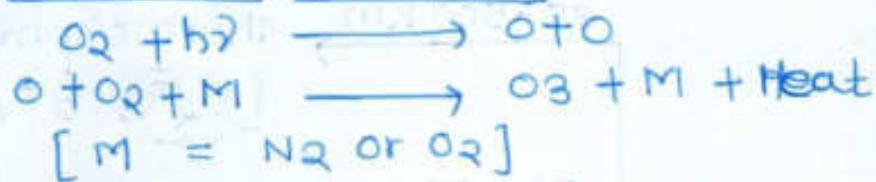
ozone is the imp species in the stratosphere

phere. acting as protecting radiation sheet for leaving organisms in earth. the max concⁿ of ozone around 10 ppm in stratosphere at an altitude of 25-30 km.

Formation of ozone:

ozone is form by photochemical deⁿ of oxygen followed by their 3rd body deⁿ.

photochemical reanⁿ of oxygen:

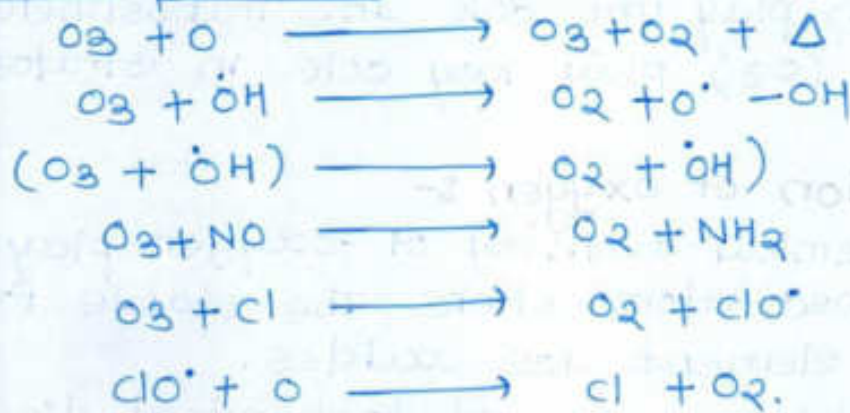


The 3rd body (M) absorb the excess energy liberated by the deⁿ. & stabilise the ozone molecule.

ozone strongly absorb uv radiation in the region 220-350 nm which protect the life on earth from severel radiation damage

only small fraction of uv light reaches on the earth.

ozone depletion :-



(recyclisation)

The mechanism of ozone removal is not well known.

3) Dissociation of Nitrogen oxides.

The nitrogen oxide in the atmosphere are Nitrous oxide (N₂O), Nitric oxide (NO) & Nitrogen dioxide (NO₂)

N₂O is laughing gas,



1) N_2O originates from microbiological process & occurs in unpolluted air.

2) Nitric oxide (NO) & NO_2 are imp constituents of polluted air means these are the man made sources. i.e. combustion of fossil & fuels.

4) Dissociation of chlorofluoro carbon. [$\text{Cl}_2\text{F}_2/\text{CFCl}_3$]

The chlorofluorocarbons are used as refrigerant, cooling agent, aerosol spray, propellant.

They are present in troposphere but slowly diffuse into stratosphere in deplete the ozone layer.



1 chlorine molecule deplete 1,00,000 molecules of O_3

In a cyclic chain 1 Cl can initiate a series of identical chain which lead to deplete upto 1,00,000 molecules of ozone without being destroyed itself in the process.

5) Dissociation of Hydrocarbon :-

A natural sources particularly trees, domestic, animal, anaerobic decomposition of bacteria in which large quantity of hydrocarbon in the atmosphere.

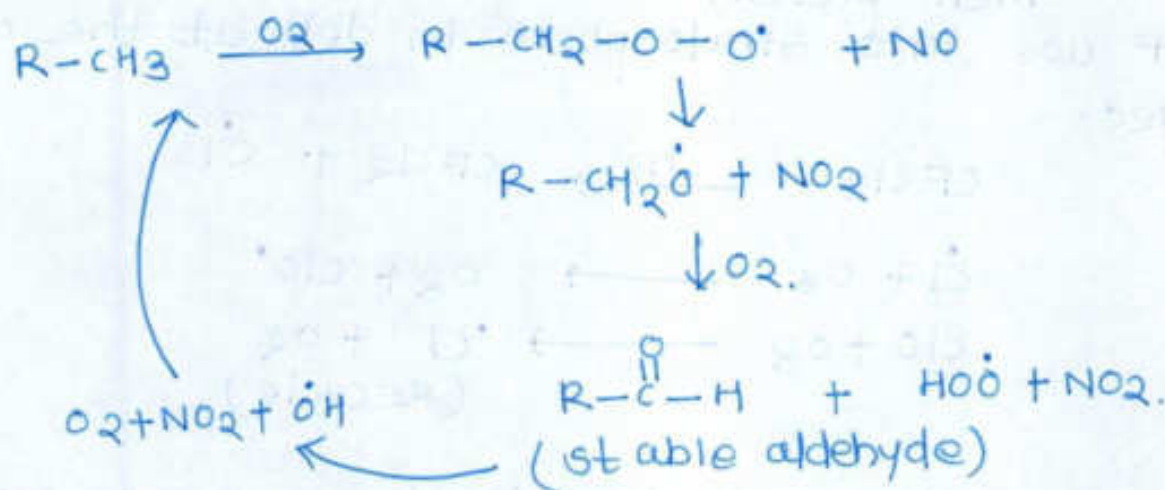
Methane is naturally major occurring hydrocarbon. Domestic animal contribute about 85 million tons of methane to the atmosphere each year.

About 20 diff hydrocarbon were identified in the atmosphere. among them are methane, ethane, ethene, ethyne, propane, n-butane, isobutane, n-pentane, isopentane, toluene, m-xylene etc.

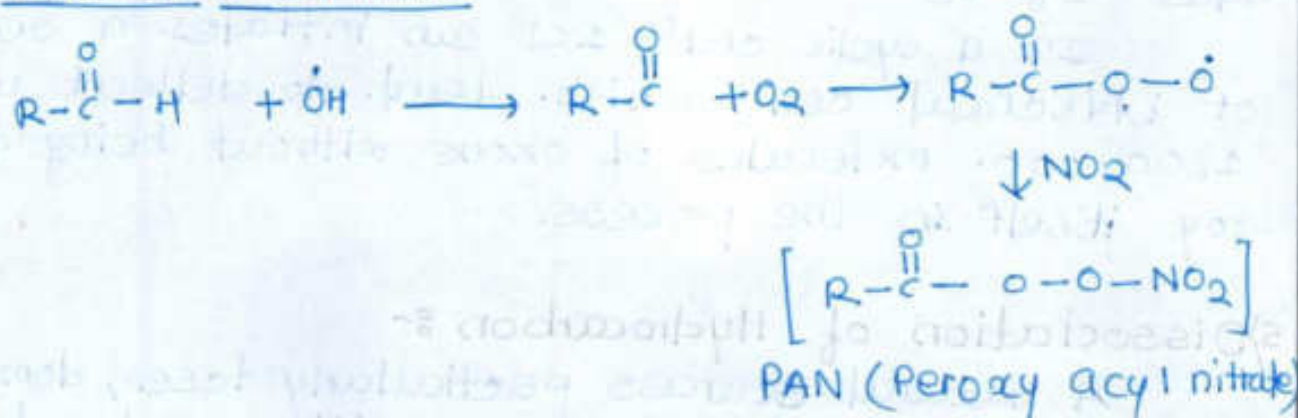
Mechanism: Smog formation mechanism

photochemical smog is high concⁿ of oxidant (O₃ & NO₂) Hydrocarbon do not react rapidly with sunlight but they are reactive towards other substance produced by photochemical.

Reaction :- Hydrocarbon.



Formation of PAN:



PAN is an irritant to respiratory system & eyes & has also be found to be highly toxic to the plant.

PHOTOCHEMISTRY OF VISION

"The most precious gift given by God is VISION".
Scientifically, eye is a biorganic instrument which is able to convert radiation energy [visible region of electromagnetic radiation] into chemical energy then into electrical energy.

The procedure of vision starts with the absorption of photon [Radiation particle], followed by isomerisation (chemical change) which creates electrical pulses. neurons conduct these pulses to brain where brain read it.

The eye of vertebrate consists of lens & retina. The retina is made up of two types of light sensitive cell.

- a) Rods
- b) cones.

The rods & cones, present in the retina act as the receptor of the eye. The rods are very sensitive therefore, they can function at low light intensity. The rods, are also responsible for vision even in dim light.

The cones have been found to have a comparatively low sensitivity. The cones are responsible for colour selector cones are much less in no. than the rods & provide complete

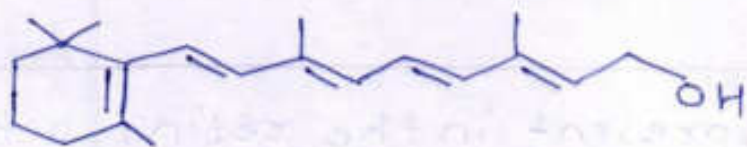
colour vision in sufficient light. The centre of the vision field of the retina consists of only cones.

• Sensitivity of eyes :-

The eye is an extraordinary sensitive instrument. It can respond betⁿ the wavelength of 400 to 800 nm. i.e. visible region of spectrum. The degree of sensitivity of eye is such that fully dark-adapted eye can clearly detect an object in light so dim as to be equivalent to a light input over retina of only about 10,000 quanta (second per light quantum per three min. per receptor cell in the retina. The eyes of a human & most other mammals are made of a photosensitive comp. protein. This protein is of two type.

① Opsin

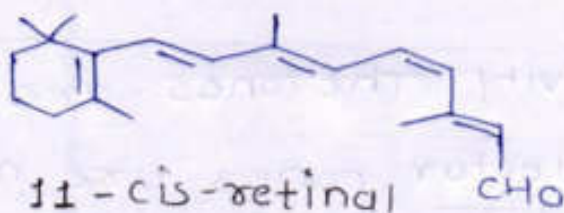
② Retinal.



Structure of Retinal.



All trans retinal



11-cis-retinal

• Photosensitive pigment :-

The photosensitive pigment in the rods is known as rhodopsin or visual purple pigments.

Its retinal is 11-cis-retinal and its opsin is scotopsin. Rhodopsin shows absorption maximum at ≈ 500 nm. Human rhodopsin has molecular weight 41000 and 348 amino acid residues.

The 90% of the total protein in the membrane of rod disk are made up of rhodopsin. It is coupled with G-protein as:

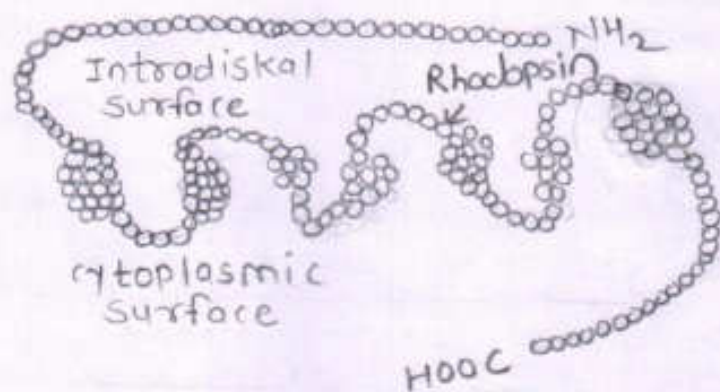
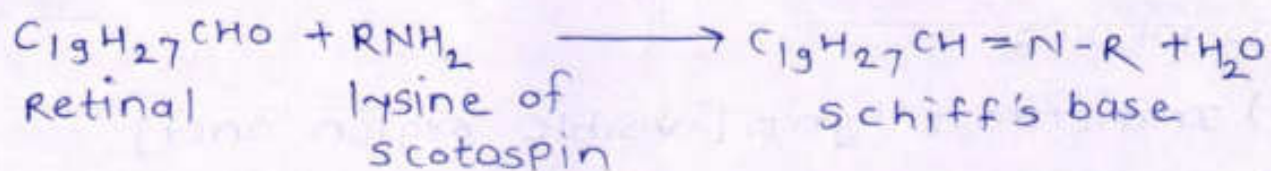


Fig 7.3 Coupling of G-proteins to Rhodopsin.

Retinal is parallel to the surface of the membrane & is attached to a lysine residue at position 296 in the seventh transmembrane domain.



The linkage betⁿ the scotopsin back & retinal results from addition reaction followed by elimination reaction betⁿ aldehyde group (-CHO) of retinal & an α -amino group of a lysin of scotopsin.

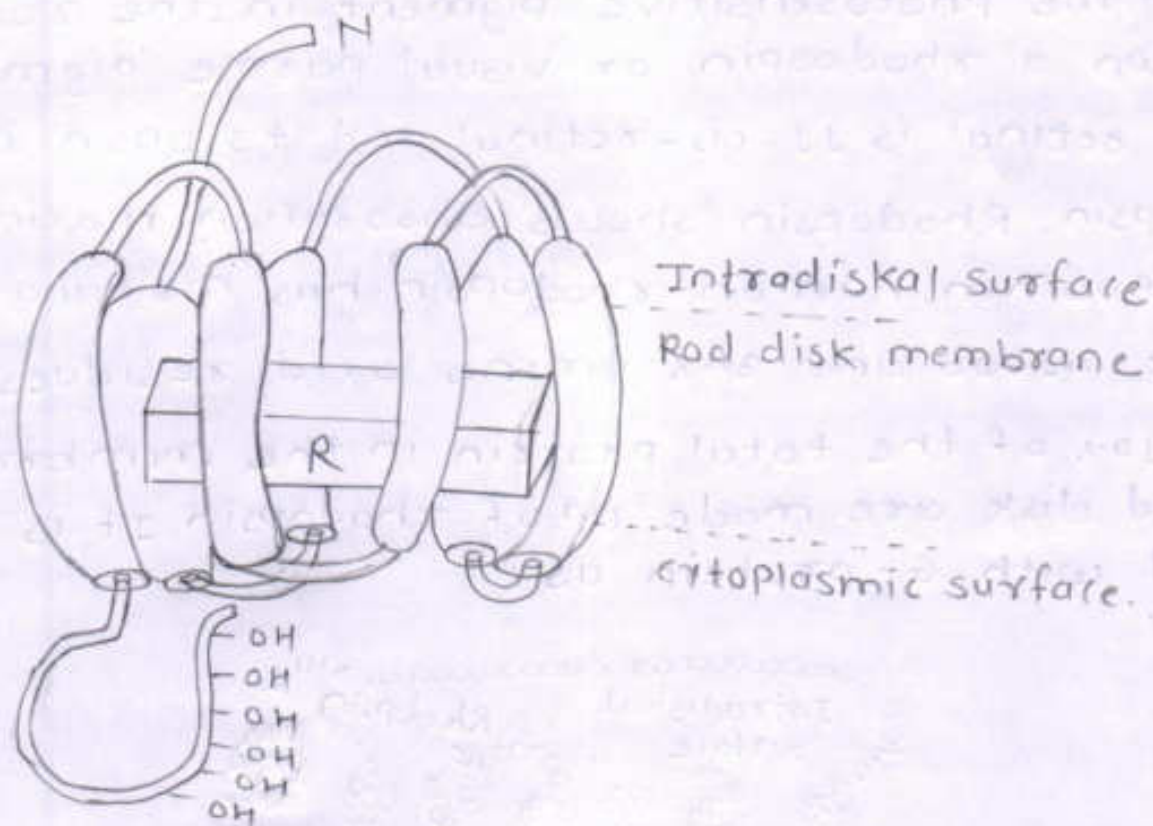


Fig : Diagrammatic representation of the surface of Rhodopsin, showing the position of 11-cis-retinal (R) in the rod disk membrane.

The process of vision is a sequence of changes in phototransduction of photosensitive pigments [rods and cones]. These events may be summarise as follows.

- i) Incident light [visible region only]
- ii) structural change in the 11-cis-retinal of photopigment
- iii) conformational changes in photopigments.
- iv) Activation of transduction.
- v) Activation of phosphodiesterase.

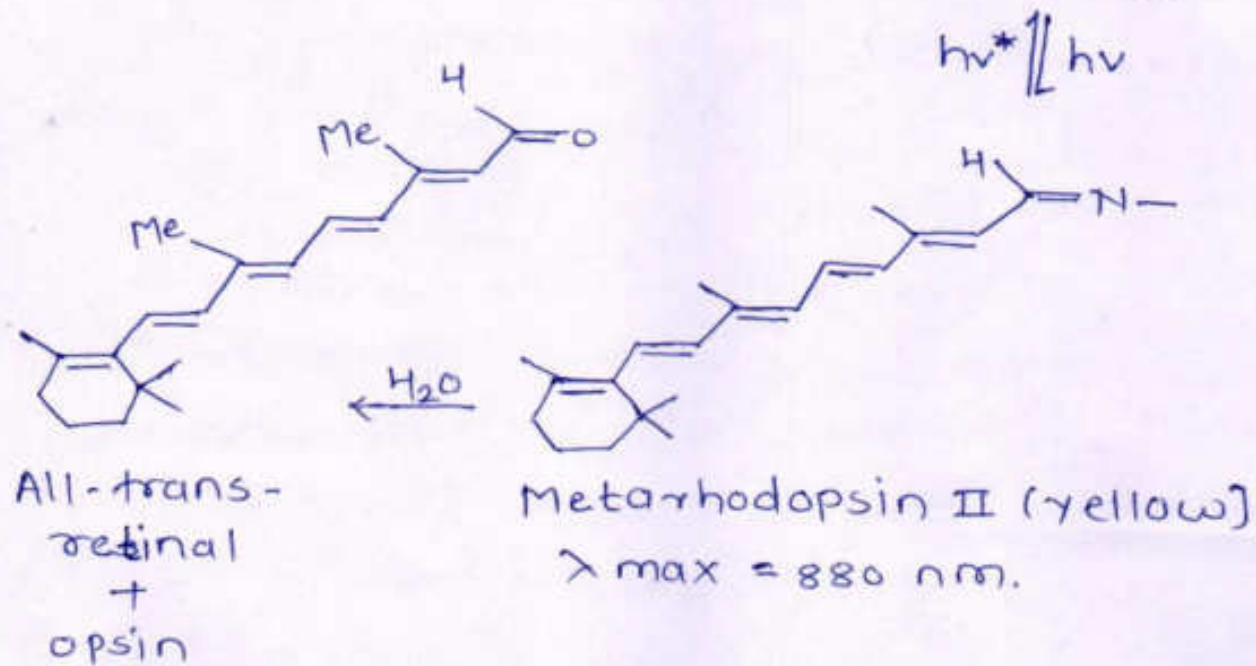
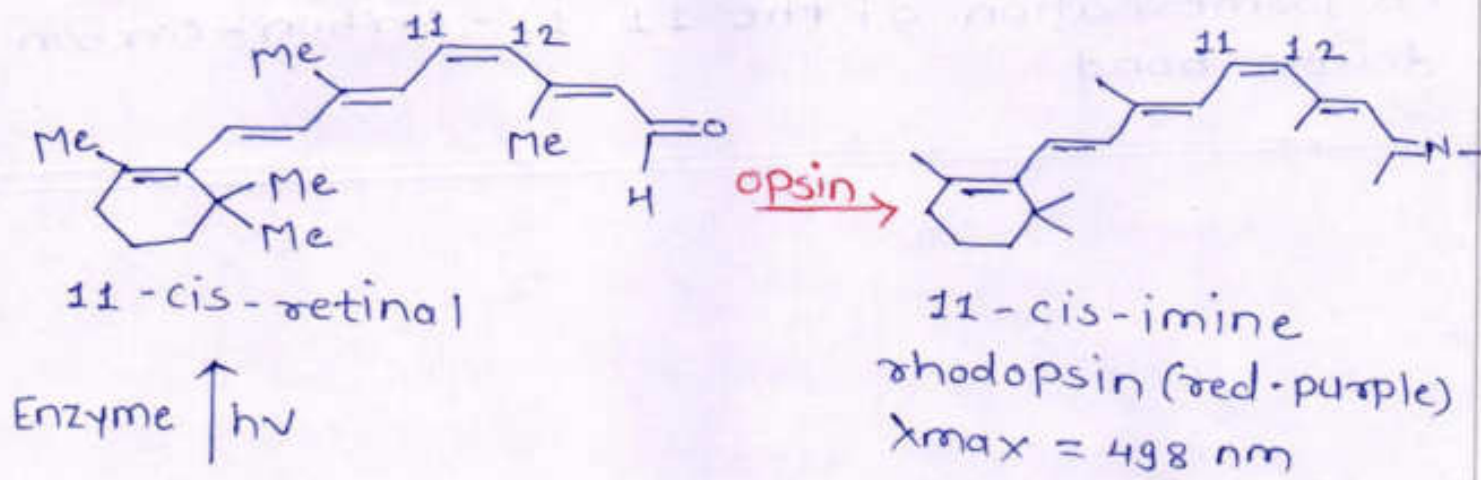
- vi] Reduction in concentration of intercellular cGMP
- vii] closure of Na^+ channels.
- viii] Hyper polarisation.
- ix] decrease in release of synaptic transmitter
- x] Resonance in bipolar cells and other neural elements.

Rods are very sensitive towards the light and the main research interest is towards the photochemistry of transduction of rods. The following sequence of steps occurs during the phototransduction of rods.

- i] Light activates rhodopsin.
- ii] Activated rhodopsin activates transduction.
- iii] Activated transduction binds with GTP [phosphodiesterase].
- iv] Activated phosphodiesterase catalyses the conversion of cyclic GMP [cGMP into 5'-GMP.
- v] This makes reduction in protoplasmic concentration of a GMP.
- vi] This reduction causes closure of a cGMP-gated ion channels.

In the dark retinal in rhodopsin is in the 11-cis-configuration. Irradiation of rhodopsin leads to a series of conformational changes which can be noted by the disappearance and appearance of various intermediates of different colours.

The photochemistry of vision is triggered by absorption of a photon and induces cis-trans isomerisation. The conjugated polyenal, 11-cis-retinal and the protein opsin combine in retina to give the red-purple 11-cis-imine, rhodopsin.



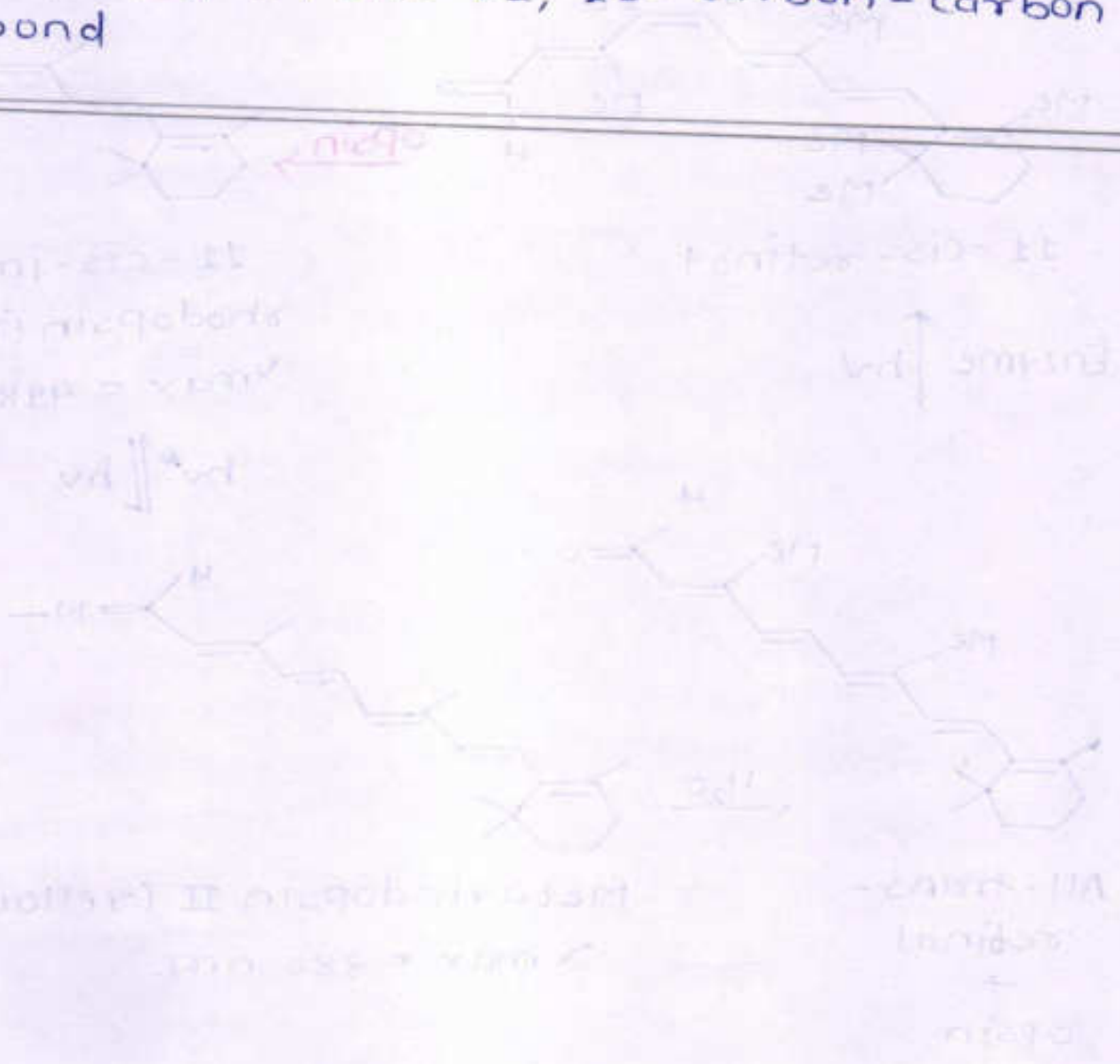
When the absorption of light occurs the cis-double bond is isomerized to give the yellow all-trans metarhodopsin II and a nerve impulse is triggered.

Unlike its 11-cis-imine isomer the all-trans product does not fit to the site on the protein surface & the carbon-nitrogen double bond is exposed and

hydrolysed to give all-trans-retinal and opsin.

The all-trans retinal can be converted back to the 11-cis-isomer in the retina by enzyme and light of much shorter wavelength.

In bright light metarhodopsin II can be converted to 11-cis-imine rhodopsin, by trans-cis isomerisation of the 11, 12-carbon-carbon double bond



When the absorption of light occurs the chromophore is isomerized to give the yellow all-trans metarhodopsin II and a nerve impulse is triggered. Unlike the 11-cis-isomer, the all-trans product does not fit in the site of the protein surface. The carbon-carbon double bond is oriented...



“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार”- शिक्षणमहर्षी डॉ.वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit

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Tasgaon, Dist- Sangli, Pin-416312
DEPARTMENT OF HISTORY

Student Seminar

Date: 27 & 28 March 2021.

Submitted to
IQAC, Internal Quality Assurance Cell
Padmabhushan Dr.Vasantodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

Organized by
Department of History

Head
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Department of History
Padmabhushan Dr.Vasantodada Patil Mahavidyalaya,
Tasgaon
2020-2021

"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार" -शिक्षणमहर्षी डॉ.बापूजी साळुंखे
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DEPARTMENT OF HISTORY
T. Y. B. A.
(2020-2021)

सूचना

दिनांक : २२/ ०३/२०२१

इतिहास विभागातील बी.ए.भाग ३ या वर्गातील सर्व विद्यार्थ्यांना सूचित करण्यात येते की, शैक्षणिक वर्ष २०२०-२०२१ सत्र ०५ मध्ये आपल्याला देण्यात आलेल्या विषयाप्रमाणे विभागांतर्गत घेण्यात येणारे (१० मार्क) सेमिनार २६ व २७ मार्च २०२१ रोजी विभागात घेण्यात येणार आहेत. तरी अनुपस्थित असणाऱ्या विद्यार्थ्यांचे मार्क विद्यापीठास पाठविले जाणार नाहीत याची सर्वांनी नोंद घ्यावी.

Dr. Haji Nadaf. → Dha.
Dr. V. D. Chorge → V. D. Chorge
Ms. K. R. Kelap → K. R. Kelap
Mr. R. S. Kamble → R. S. Kamble

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DEPARTMENT OF HISTORY

(2020-2021)

SEMINAR OF T.Y.B.A.STUDENTS

Paper XI: History : Its Theory

Date:27 /03/2021

Sr.No.	Name of The Student	Subject
1.	Aiwale Rushikesh Amar	संशोधनाचा आराखडा
2.	Chavan Ganesh Pruthviraj	संदर्भ संकलन म्हणजे काय
3.	Chavan Vaibhav Hanmant	इतिहासलेखन संबंधी संदर्भ ग्रंथ
4.	Chavan Vijay Prakash	संदर्भ साधने म्हणजे काय
5.	Chavan Vishal Vitthal	बहिरंग परीक्षण
6.	Chougule Jotiram Gajanan	अंतरंग परीक्षण
7.	Deshmukh Ranjit Rajaram	इतिहासाचे स्वरूप
8.	Deshmukh Sonali Maruti	इतिहासलेखनशास्त्र म्हणजे काय
9.	Dugam Shekhar Mahadev	सर्व इतिहास हा समकालीन असतो
10.	Fakir Alfaj Iqbal	इतिहासाची गती सरळ रेषेत की चक्रार
11.	Garande Akanksha Ashok	इतिहास शास्त्र की कला
12.	Godase Ranjeet Ananda	इतिहासात मूल्यमापन असतो कि नाही
13.	Gurav Shubham Shivaji	कालक्रमण
14.	Hajare Akshay Kisan	परिशिष्ट
15.	Irale Shirish Ramchandra	सूची
16.	Jadhav Megha Raju	संदर्भ सूची व संदर्भ ग्रंथ सूचीतील फरक
17.	Javale Rohit Ramesh	मुलाखत पद्धत
18.	Javir Priyanka Mohan	तळ टिपा
19.	Kamble Mangal Suresh	नोट पद्धत
20.	Kharapude Rushikesh Keraba	संदर्भ पद्धत
21.	Kharat Swapnil Sambhaji	सादरीकरण
22.	Kokate Pravin Vaman	साधनांचे मूल्यमापन
23.	Kumbhar Tushar Dipak	प्रश्नावली पद्धत
24.	Lohar Suraj Bhimrao	प्रकरण योजना
25.	Mali Mayuri Jaykumar	पूर्व संशोधन
26.	Mali Omkar Balasaheb	वाङ्मय चौक
27.	Mandle Abhijeet Rajkumar	विषयाची निवड
28.	Mane Vishwjeet Dipak	लघु पट
29.	Patil Adesh Tanaji	आकाशवाणी
30.	Patil Amit Sukhadev	टेलिव्हिजन
31.	Patil Chandrasen Chandrakant	रेडिओ व इतिहास
32.	Patil Dipali Jagannath	इंटरनेट एक ऐतहासिक साधन
33.	Patil Mayur Arvind	वर्तमानपत्रे
34.	Patil Tejas Navanath	इतिहासाची व्याख्या
35.	Patil Vaibhav Vishrant	पुराभिलेखागर
36.	Pawar Somnath Shahaji	सर्वेक्षण पद्धत
37.	Salunkhe Sandeep Vijay	इतिहासाची सहाय्यकारी शास्त्रे
38.	Shaikh Samir Jamal	इतिहासाचे प्रकार
39.	Yedake Amol Kisan	इतिहासाचे महत्व व व्याप्ती

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DEPARTMENT OF HISTORY

(2020-2021)

SEMINAR OF T.Y.B.A.STUDENTS

Paper X: Political History of the Maratha

Date:27 /03/2021

Sr.No.	Name of The Student	Subject
1.	Aiwale Rushikesh Amar	छ. शाहू व मुघल संबंध
2.	Chavan Ganesh Pruthviraj	बालाजी विघ्नराव
3.	Chavan Vaibhav Hanmant	छत्रपती पराक्यातील बादवी
4.	Chavan Vijay Prakash	कोल्हापूर राज्यक्रांती
5.	Chavan Vishal Vitthal	महाराणी ताराबाई
6.	Chougule Jotiram Gajanan	पेशव्यांच्या उदात्ती करणे
7.	Deshmukh Ranjit Rajaram	मराठा राजमंडळ
8.	Deshmukh Sonali Maruti	बारणेचा तह
9.	Dugam Shekhar Mahadev	पानिपतची लढाई
10.	Fakir Alfaj Iqbal	मोहम्मदशाह अब्दाली
11.	Garande Akanksha Ashok	महाराणी ताराबाई व मुघल संबंध
12.	Godase Ranjeet Ananda	निजाम व पेशवे संबंध
13.	Gurav Shubham Shivaji	पहिला बाजीराव पेशवा
14.	Hajare Akshay Kisan	मराठ्यांचा सागरी संबंध
15.	Irale Shirish Ramchandra	नादीरशाहाची स्वारी
16.	Jadhav Megha Raju	बाजीराव व पोर्तुगीज संबंध
17.	Javale Rohit Ramesh	मराठेकालीन शैलीव्यवस्था
18.	Javir Priyanka Mohan	मराठेकालीन धार्मिक स्थळे
19.	Kamble Mangal Suresh	मराठेकालीन कुशीव्यवस्था
20.	Kharapude Rushikesh Keraba	मराठेकालीन समाजजीवन
21.	Kharat Swapnil Sambhaji	मराठा सत्तेच्या नासाची कारणे
22.	Kokate Pravin Vaman	इब्राहिमखान गारवी मराठे व हैदर आली
23.	Kumbhar Tushar Dipak	पेशवे अंग्रेज संबंध
24.	Lohar Suraj Bhimrao	शाहूंचे मूल्यमापन
25.	Mali Mayuri Jaykumar	बालाजी विघ्नरावाचे मूल्यमापन
26.	Mali Omkar Balasaheb	मना फडणवीस
27.	Mandle Abhijeet Rajkumar	बारभाई कारस्थान
28.	Mane Vishwjeet Dipak	बारभाई मंडळ
29.	Patil Adesh Tanaji	माधवराव पेशवा
30.	Patil Amit Sukhadev	बाजीराव दुसरे
31.	Patil Chandrasen Chandrakant	पेशवे इंग्रज संबंध
32.	Patil Dipali Jagannath	पेशवा मुघल संबंध
33.	Patil Mayur Arvind	मराठ्यांची आर्थीची लढाई
34.	Patil Tejas Navanath	कोल्हापूर संस्थान
35.	Patil Vaibhav Vishrant	सातार राज्य
36.	Pawar Somnath Shahaji	बाजीराव निजाम संबंध
37.	Salunkhe Sandeep Vijay	मराठेकालीन सांस्कृतिक जीवन
38.	Shaikh Samir Jamal	मराठेकालीन आर्थिक जीवन
39.	Yedake Amol Kisan	मराठेकालीन स्त्रीजीवन

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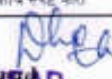
(2020-2021)

SEMINAR OF T.Y.B.A.STUDENTS

Paper IX: Age of Revolution

Date:26 /03/2021

Sr.No.	Name of The Student	Subject
1.	Aiwale Rushikesh Amar	प्रोटेस्ट धर्मसुधारणा क्रांतीची कारणे
2.	Chavan Ganesh Pruthviraj	प्रोटेस्ट धर्मसुधारणा क्रांतीचे परिणाम
3.	Chavan Vaibhav Hanmant	धर्मसुधारणा चळवळीतील मार्टिन ल्यूथर चे कार्य
4.	Chavan Vijay Prakash	धर्मसुधारणा चळवळीतील जॉन विकीलीक्स चे कार्य
5.	Chavan Vishal Vitthal	धर्मसुधारणा चळवळीतील इरसमस कार्य
6.	Chougule Jotiram Gajanan	स्विट्झर्लंडमधील धर्मसुधारणा चळवळीचा आढावा
7.	Deshmukh Ranjit Rajaram	इंग्लंडमधील धर्मसुधारणा चळवळीचा आढावा
8.	Deshmukh Sonali Maruti	धर्मसुधारणा चळवळीचे स्वरूप
9.	Dugam Shekhar Mahadev	सोसायटी ऑफ वीक्स संपटनेचे कार्य
10.	Fakir Alfaj Iqbal	धर्मसुधारणा चळवळीचे अणिष्ट परिणाम
11.	Garande Akanksha Ashok	औद्योगिक क्रांतीची कारणे
12.	Godase Ranjeet Ananda	औद्योगिक क्रांतीचे स्वरूप स्पष्ट करा
13.	Gurav Shubham Shivaji	औद्योगिक क्रांतीचे आर्थिक परिणाम
14.	Hajare Akshay Kisan	औद्योगिक क्रांतीचे सामाजिक परिणाम
15.	Irale Shirish Ramchandra	औद्योगिक क्रांतीचे राजकीय परिणाम
16.	Jadhav Megha Raju	औद्योगिक क्रांतीचे दुष्परिणाम
17.	Javale Rohit Ramesh	इंग्लंडची कापड उद्योगातील क्रांती
18.	Javir Priyanka Mohan	कोळसा व लोखंड उद्योगातील क्रांती
19.	Kamble Mangal Suresh	बाहतुकीच्या साधनांतील क्रांती
20.	Kharapude Rushikesh Keraba	बाण शक्ती व विद्युत् शक्ती
21.	Kharat Swapnil Sambhaji	कृषी क्षेत्रातील क्रांती
22.	Kokate Pravin Vaman	अमेरिकन स्वातंत्र्ययुद्ध ची कारणे
23.	Kumbhar Tushar Dipak	अमेरिकन स्वातंत्र्ययुद्धात चे स्वरूप स्पष्ट करा
24.	Lohar Suraj Bhimrao	अमेरिकन स्वातंत्र्ययुद्धात चे परिणाम
25.	Mali Mayuri Jaykumar	स्टॅम्प ॲक्ट चा आढावा
26.	Mali Omkar Balasaheb	बॉस्टन टी पार्टी चे वर्णन
27.	Mandle Abhijeet Rajkumar	पॅरिस कराराचा आढावा
28.	Mane Vishwjeet Dipak	अमेरिकन वसाहतीच्या विजयाची कारणे
29.	Patil Adesh Tanaji	इंग्रजांच्या 13 वसाहतींची माहिती द्या
30.	Patil Amit Sukhadev	अमेरिकेच्या विजयाची कारणे
31.	Patil Chandrasen Chandrakant	रसो चे कार्य स्पष्ट करा
32.	Patil Dipali Jagannath	मानवी हक्काच्या जाहीरनाम्याची माहिती
33.	Patil Mayur Arvind	राष्ट्रीय सभेचे कार्य
34.	Patil Tejas Navanath	फ्रेंच राज्यक्रांतीची कारणे
35.	Patil Vaibhav Vishrant	फ्रेंच राज्यक्रांतीच्या बाटचालीचा आढावा
36.	Pawar Somnath Shahaji	फ्रेंच राज्यक्रांती चे परिणाम
37.	Salunkhe Sandeep Vijay	4 ऑगस्ट ची सामाजिक क्रांती
38.	Shaikh Samir Jamal	वॅस्टील च्या पाडाचा चा आढावा
39.	Yedake Amol Kisan	मॉन्टेसरी चे कार्य स्पष्ट करा व कार्य स्पष्ट करा


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SEMINAR OF T.Y.B.A.STUDENTS

Paper- VIII: History of Medieval India (1206-1526)

Date:26 /03/2021

Sr.No.	Name of The Student	Subject
1.	Aiwale Rushikesh Amar	इब्राहीम आदिलशाह व मराठा संस्कृतीचे समन्वय
2.	Chavan Ganesh Pruthviraj	महमूद गवान
3.	Chavan Vaibhav Hanmant	फिरोजशाह तुघलक
4.	Chavan Vijay Prakash	सलतनात कालीन पुतलीय साधने
5.	Chavan Vishal Vitthal	सलतनात कालीन केंद्रीय प्रशासन
6.	Chougule Jotiram Gajanan	विजयनगर कालीन केंद्रीय प्रशासन
7.	Deshmukh Ranjit Rajaram	विजयनगर कालीन प्रांतीय प्रशासन
8.	Deshmukh Sonali Maruti	सुलाह ए कुल नीती
9.	Dugam Shekhar Mahadev	सलतनातकालीन इत्हा पद्धत
10.	Fakir Alfaj Iqbal	सलतनातकालीन प्रशासन
11.	Garande Akanksha Ashok	मुहम्मद बिन तुघलकाचे कृषि धोरण
12.	Godase Ranjeet Ananda	फिरोजशाह तुघलकाचे कृषि धोरण
13.	Gurav Shubham Shivaji	सलतनात कालीन प्रांतीय प्रशासन
14.	Hajare Akshay Kisan	मुहम्मद बिन तुघलकाचे मूल्यमापन
15.	Irale Shirish Ramchandra	मुहम्मद बिन तुघलकाचे धार्मिक धोरण
16.	Jadhav Megha Raju	मुहम्मद बिन तुघलकाचे चलन प्रयोग
17.	Javale Rohit Ramesh	मुहम्मद बिन तुघलकाचे राजधानी बदल प्रयोग
18.	Javir Priyanka Mohan	मुहम्मद बिन तुघलकाची साम्राज्य विस्तार नीती
19.	Kamble Mangal Suresh	अलाउद्दीन खिलजीचे मूल्यमापन
20.	Kharapude Rushikesh Keraba	अलाउद्दीन खिलजीचे मंगोलविषयक धोरण
21.	Kharat Swapnil Sambhaji	अलाउद्दीन खिलजीचे दक्षिण भारतातील साम्राज्य विस्तार
22.	Kokate Pravin Vaman	अलाउद्दीन खिलजीची बाजार नियंत्रण व्यवस्था
23.	Kumbhar Tushar Dipak	अलाउद्दीन खिलजीचे सैन्य सुधारणा
24.	Lohar Suraj Bhimrao	ऐबक वंश
25.	Mali Mayuri Jaykumar	खिलजी वंश
26.	Mali Omkar Balasaheb	तुघलक वंश
27.	Mandle Abhijeet Rajkumar	सय्यद वंश
28.	Mane Vishwjeet Dipak	लोदी वंश
29.	Patil Adesh Tanaji	दोमिनो पेस
30.	Patil Amit Sukhadev	इब्राहीम आदिलशाह
31.	Patil Chandrasen Chandrakant	मलिक कफुर
32.	Patil Dipali Jagannath	झिपाउद्दीन बरनी
33.	Patil Mayur Arvind	अमुक्तमाल्यदा
34.	Patil Tejas Navanath	बलबन
35.	Patil Vaibhav Vishrant	इब्र बतुता
36.	Pawar Somnath Shahaji	सलतनात कालीन व्यापार व उद्योग
37.	Salunkhe Sandeep Vijay	विजयनगर कालीन व्यापार व उद्योग
38.	Shaikh Samir Jamal	कुषावैवराय
39.	Yedake Amol Kisan	बहामनी कालीन व्यापार व उद्योग

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SEMINAR OF T.Y.B.A.STUDENTS

Paper VII: Early India (from beginning to 4th C. BC)- Date:26 /03/2021

Sr.No.	Name of The Student	Subject
1.	Aiwale Rushikesh Amar	मौर्य प्रशासन
2.	Chavan Ganesh Pruthviraj	अशोकाचा धम्म
3.	Chavan Vaibhav Haumant	हदप्पा संस्कृतीचे ठिका
4.	Chavan Vijay Prakash	मगध अशोक
5.	Chavan Vishal Vitthal	आचार्य चाणक्य
6.	Chougule Jotiram Gajanan	अश्वमेधीन संस्कृती
7.	Deshmukh Ranjit Rajaram	नावाश्वमेधीन संस्कृती
8.	Deshmukh Sonali Maruti	हदप्पाकालीन धार्मिक जीवन
9.	Dugam Shekhar Mahadev	हदप्पाकालीन सामाजिक जीवन
10.	Fakir Alfaj Iqbal	हदप्पा संस्कृतीतील स्वळे
11.	Garande Akanksha Ashok	चंद्रगुप्त
12.	Godase Ranjeet Ananda	मौर्यकालीन राज्यव्यवस्था
13.	Gurav Shubham Shivaji	नंद घराणे
14.	Hajare Akshay Kisan	बौद्ध धर्माची तत्वे
15.	Irale Shirish Ramchandra	बौद्ध धर्माच्या ज्ञानाची करणे
16.	Jadhav Megha Raju	बौद्ध धर्म परिषद
17.	Javale Rohit Ramesh	बौद्ध चिकीत्से नियम
18.	Javir Priyanka Mohan	बौद्ध धर्म संघ
19.	Kamble Mangal Suresh	गौतम बुद्ध
20.	Kharapude Rushikesh Keraba	त्रिरत्न व चार शिक्षावृत्त
21.	Kharat Swapnil Sambhaji	पंचअनुवृत्ते व पंच महावृत्ते
22.	Kokate Pravin Vaman	भगवान महावीर
23.	Kumbhar Tushar Dipak	२८ तीर्थंकर
24.	Lohar Suraj Bhimrao	जैन धर्म
25.	Mali Mayuri Jaykumar	१६ महाजनपद
26.	Mali Omkar Balasaheb	महाशिला संस्कृती
27.	Mandfe Abhijeet Rajkumar	आधुनिक व्यवस्था
28.	Mane Vishwjeet Dipak	वैदिककालीन धार्मिक जीवन
29.	Patil Adesh Tanaji	वैदिककालीन साहित्य
30.	Patil Amit Sukhadev	उत्तर वैदिक संस्कृती
31.	Patil Chandrasen Chandrakant	वैदिक संस्कृती
32.	Patil Dipali Jagannath	हदप्पा संस्कृतीचा वारसा
33.	Patil Mayur Arvind	हदप्पा संस्कृतीच्या ज्ञानाची करणे
34.	Patil Tejas Navanath	पहिले नागरीकरण
35.	Patil Vaibhav Vishrant	द्वितीय नागरीकरण
36.	Pawar Somnath Shahaji	ताम्रपाषाण संस्कृती
37.	Salunkhe Sandeep Vijay	दशराज बुद्ध व राजकीय स्थित्यंतर
38.	Shaikh Samir Jamal	अशोकाचे शिलालेख
39.	Yedake Amol Kisan	प्राचीन विद्यापीठे


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T. Y. B. A.

(2020-2021)

STUDENT PARTICIPATION

Student Seminar presentation

Day & Date : *26/03/2021 to 27/03/2021.*

Sr. No.	Roll No.	Gender	Name of the Student	Signature
1	3616	M	AIWALE RUSHIKESH AMAR	<i>Rushikesh</i>
2	3617	M	CHAVAN GANESH PRUTHVIRAJ	<i>Chavan</i>
3	3618	M	CHAVAN VAIBHAV HANMANT	<i>V.H.</i>
4	3619	M	CHAVAN VIJAY PRAKASH	<i>Chavan</i>
5	3620	M	CHAVAN VISHAL VITTHAL	<i>V.V. Chavan</i>
6	3621	M	CHOUGULE JOTIRAM GAJANAN	<i>Chougule</i>
7	3622	F	DESHMUKH RANJIT RAJARAM	<i>Deshmukh</i>
8	3623	F	DESHMUKH SONALI MARUTI	<i>Ab</i>
9	3624	M	DUGAM SHEKHAR MAHADEV	<i>Dugam</i>
10	3625	M	FAKIR ALFAJ IQBAL	<i>A.S. Fakir</i>
11	3626	M	GARANDE AKANKSHA ASHOK	<i>Garande</i>
12	3627	M	GODASE RANJEET ANANDA	<i>Ab</i>
13	3628	F	GURAV SHUBHAM SHIVAJI	<i>G.S.</i>
14	3629	F	HAJARE AKSHAY KISAN	<i>Hajare</i>
15	3630	M	IRALE SHIRISH RAMCHANDRA	<i>Irale</i>

16	3631	M	JADHAV MEGHA RAJU	Ab
17	3632	F	JAVALE ROHIT RAMESH	Ab
18	3633	M	JAVIR PRIYANKA MOHAN	Ab
19	3634	M	KAMBLE MANGAL SURESH	M.S. Kamble
20	3635	M	KHARAPUDE RUSHIKESH KERABA	Rushikesh
21	3636	M	KHARAT SWAPNIL SAMBHAJI	SSK
22	3637	F	KOKATE PRAVIN VAMAN	Ab
23	3638	M	KUMBHAR TUSHAR DIPAK	Krushnar
24	3639	M	LOHAR SURAJ BHIMRAO	Shital
25	3640	M	MALI MAYURI JAYKUMAR	M.J. Mali
26	3641	M	MALI OMKAR BALASAHEB	Omkar
27	3642	M	MANDLE ABHIJEET RAJKUMAR	Abhijeet
28	3643	M	MANE VISHWJEET DIPAK	Omkar
29	3644	M	PATIL ADESH TANAJI	Adesh
30	3729	M	PATIL AMIT SUKHADEV	A.S. Patil
31	3730	M	PATIL CHANDRASEN CHANDRAKANT	Chandrakant
32	3731	M	PATIL DIPALI JAGANNATH	Dipali
33	3732	M	PATIL MAYUR ARVIND	Mayur
34	3733	M	PATIL TEJAS NAVANATH	Tejas
35	3734	M	PATIL VAIBHAV VISHRANT	Vaibhav
36	3735	M	PAWAR SOMNATH SHAHAJI	Ab
37	3736	M	SALUNKHE SANDEEP VIJAY	Salunkhe
38	3737	M	SHAIKH SAMIR JAMAL	Shaikh
39	3738	M	YEDAKE AMOL KISAN	Amol

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नाव- पारिल दिपाली जगन्नाथ

BA-III विषय - इतिहास सेमिनार

Roll No - 3731

विषय :- इंटरनेट मुक्त ऐतिहासिक साधन

२। वे शतक हे माहितीच्या मटाजालाचे युग म्हणून ओळखले जाते. इंटरनेटच्या माध्यमातून माहितीची अनेक साधने उपलब्ध होत आहेत. आय. टी. मुळे माहिती देणाऱ्या साधनांचा विस्फोट झाला आहे. त्यामुळेच या युगातील इंटरनेटला माहितीचे माधाजाल असे संबोधले जाते. आज कोणतीही माहिती आपणास द्याव्यात, मिळत आहे इंटरनेटमुळे संवाद, मनोरंजन व माहिती यांची उपलब्धता फार मोठ्या प्रमाणात झाली आहे. इंटरनेट हा शब्द 'वर्ल्ड वाइड वेब' म्हणून ओळखला जातो. वेगवेगळ्या माध्यमांद्वारे आपणास माहिती मिळविणे, त्याची नोंद वेगळे सुलभ झाले आहे.

इंटरनेटचा वापर प्रथम अमेरिकेत २० व्या शतकाच्या अगदी शेवटी झाला. भारतात १९९५ मध्ये इंटरनेटची सेवा, व्ही. एस. एन. एल. या कंपनीने सुरु केली. आज त्याचा स्पाट्याने प्रसार होतो आहे. कोट्यवधी लोक या माध्यमाद्वारे जोडले जात आहेत. या माध्यमाद्वारे आपण पडद्यावर जगाच्या कानाकोपऱ्यातील माहितीची देवाण-घेवाण करू शकतो. जगभरातील ग्रंथालये, कारखाने, डॉक्टर्स, इंजिनिअर्स यांच्याकडून त्या क्षेत्रातील आवश्यक माहिती घेऊन संपर्क साधून मिळू शकते. इंटरनेटद्वारे पुढील सेवा उपलब्ध होतात.

1) ई-मेल (ई-टपाल) :- ई-मेलद्वारे संदेश ही टपालाचे वहन करणारी आधुनिक सेवा आहे. आज अनेक कंपन्या त्यात काम करत आहेत ज्याला माहिती रावघाची आहे व जो माहिती देणारा आहे त्यांचा ई-मेल अॅड्रेस आवश्यक असतो. त्यानंतर संगणक प्रमाणातीद्वारे, तो मेल क्षणार्थात पोहोचतो. यामुळे इतिहासाचे दाखले, मूल साधने, नोंदी, प्रतिमा, नकाशे इत्यादी फोटो स्कॅन करून पाठविले जातात. याद्वारे क्षणार्थात या ठिकाणे टपाल त्या ठिकाळा पाठविले जाते.

2) वेब जर्नॅलिझम :- (वेब पत्रकारिता) :- इंटरनेटच्या माध्यमातून जी पत्रकारिता केली जाते त्याला वेब जर्नॅलिझम म्हणतात. यात वेब चॅनल्स, सोशल मीडिया, ट्विटर, फेसबुक, यु.ट्युब इत्यादी वेब माध्यमातून परस्पर सहकार्य करणारी पत्रकारिता केली जात आहे. भारतात विविध भाषांच्यामध्ये वेब पत्रकारिता व सोशल मीडिया कार्यरत आहे.

3) ई-वृत्तपत्र :- जगातील बहुसंख्य वृत्तपत्रे आज ई-स्वरूपात उपलब्ध करून दिली जात आहेत. मुद्रित माध्यमावरील ई-सेवा ही अधिक गतिमान झाली आहे. ई-वृत्तपत्रांना महत्वाचे स्थान प्राप्त झाले.

याशिवाय व्हिडिओफोनद्वारा व्हिओ सेवा दिली जाते त्यामुळे प्रत्यक्ष संवाद साधणे शक्य झाले आहे. संदेश पाठविण्यासाठी फेजर सेवा दिली जाते. आज मोबाईलच्या या सर्व सेवा उपलब्ध आहेत. आज सर्व क्षेत्रांत इंटरनेटच्या मोठ्या प्रमाणात वापर व उपयोग केला जातो.

इंटरनेटची कार्य प्रणाली :- इंटरनेटच्या नेटवर्किंगमुळे आपण घरवसल्या आपला संगणक, लॅपटॉप, पामटॉप इत्यादींच्या पडद्यावर जगाच्या कानाकोपऱ्यात माहितीची देवाणघेवाण करू शकतो. जगभरातील अंचालये, कारखाने डॉक्टर्स, इंजिनिअर्स यांच्याकडून त्या-त्या क्षेत्रातील आवश्यक माहिती त्यांच्याशी चेट संपर्क साधून घेऊ शकतो आपले नोटेबुक, मित्र-मैत्रीणी यांच्याशी सधज गप्पा मारू शकतो. इंटरनेटने आपणारत सुविधा पुरविल्या आहेत.

अशा प्रकारे अशिक्षित, शिक्षण, व्यापार, अर्थकारण, राजकारण, समाजकारण, कृषिक्षेत्र यात इंटरनेटचा मोठ्या प्रमाणात उपयोग करून घेतला जात आहे.

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नाव - रणजीत राजराज देशमुख

BA-III HISTORY

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परिष्ठा PRN-2018038432

Roll No-3621

विजयनगर कालीन प्रांतीय प्रशासन

प्राक्सनाच्या सोईसाठी विजयनगराच्या विशाल साम्राज्याचे सहा प्रांतांमध्ये विभाजन केले होते. डोमिंगो पॅसने ही संख्या 200 पर्यंत दिलेली आहे. दोन्ही आकड्यात पुढे अंतर असल्याचे कारण किट्टिचित व्यंजने प्रांतिक सुभेदार व मंडळीक संस्कार या श्रेणीतही पुढे मानलेले असोव. प्रांतांना राज्य, मंडळ, चावडी अशी नावे होती. राज्य वा मंडळ नावाने ओळखल्या जाणाऱ्या प्रांतांचे विभाजन अनेक जिल्ह्यांमध्ये करण्यात आले होते. जिल्हा 'कोट्टम' किंवा 'वल्लनडू' या नावाने संबोधला जात असे. कोट्टमचे विभाजन तालुक्यांमध्ये (परगाणा) करण्यात आले होते. प्रशासनातील सर्वात लहान एकक ग्राम हा होता.

प्रांताधिकारी व त्यांचे अधिकार - विजयनगर साम्राज्याचे सहा प्रमुख प्रांत होते. प्रांतप्रमुखाना राज्यपाल किंवा नायक म्हणत असत. राज्यपाल हा पुढेतर राज्यवाण्यातील किंवा अत्यंत प्रभावशाली संस्कार असे. जुलमी व वंडखोर अधिकाऱ्यांना फेरेव्युत करणाऱ्याचे अधिकार सम्राटाला होते. मुल्की लखरी आणि न्याय-विषयक अधिकारी राज्यपालाच्या हाती केंद्रित झाले होते. राज्यपाल राज्यप्रतिनिधी या नात्याने राजाच्या वतीने प्रांताचा राज्यकारभार पहात असे. व्याला प्राप्त असलेल्या अंतर्गत स्वायत्तेनुसार तो आपल्या प्रांतातील लष्करीय प्रमाणित नाणी पाडू शकत असे. प्रांताच्या पूर्ण उत्पन्नापैकी $\frac{1}{3}$ भाग केंद्रीय सरकारकडे रवाना केला जाई. व राहिलेल्या $\frac{2}{3}$ भागात प्रांतीय शासनाच्या सर्व खर्च चालविण्यात येत असे त्या मोहल्यात सम्राट व्यंजना श्रेणी हेत असत.

विजयनगराच्या प्रकट व सामर्थ्यशाली राजांचे प्रांतीय सुभेदानांबर पोलखी नियंत्रण होते. विजयनगराच्या गाईवर आरोप्य व दुर्बल शासक येताच प्रांतीय सुभेदांनाच्या बंडखोर व फुटीरवादी प्रवृत्तीस उधाण येत असे.

नायकर व्यवस्था - विजयनगराच्या प्रांतीय प्रशासन-व्यवस्थेतील नायकर व्यवस्था विविष्ट परिस्थितीत व तत्कालीन आवश्यकतेनुसार अस्तित्वात आली असावी. सुकवातीच्या काळात ही व्यवस्था विजयनगर साम्राज्यासाठी हितावहू खली असावी परंतु फुटे ची विजयनगराच्या नाझास कारणावळी खली. या व्यवस्थेत पाळेगाव किंवा नायक यांचा का उद्भव आला होता. विजयनगराच्या साम्राज्याची पुढ्यसंधर्षात ज्यांना पुराभूत केले तसेच अकिंत केले असेले ती त्यांना प्रेक्षा त्यांच्याकडे परत देण्यात आला होता. राजा हा सैन्यातील वरिष्ठ अधिकार्यांना 'अमरम' म्हणजे निश्चिती म्हणुनची प्राप्ती करणारा प्रेक्षा देत असे या सैन्याधिकार्यांना पालियागर वळले जात असे. हे सैन्याधिकारी शक्तिशाली बनले होते. राज्याच्या सैन्यासि पारदळ, घोडदळ, व इत्ली हे विविष्ट संख्येने वाळ्यात असत. या पाळेगावाना राजकीयात एक विविष्ट स्वकम भूभावी लागत असे. ते शक्तिशाली झाल्यामुळे शासनाला त्यांच्यावर नियंत्रण ठेवणे अवघड जात असे. या आंतर्किक दोषामुळे व अणिवामुळे विजयनगराच्या राज्याच्या वकडहीच्या लवकरत पनाभव झाली व नंतर त्यांचे विषय घडून आले. अमरम भूभावीचा उपभोग घेणाऱ्या या नायकांना 'अमर नायक' म्हणूनही संबोधले जात होते. सोळाव्या शतकाच्या विजयनगर साम्राज्यातील या अमर नायकांची संख्या जवळपास 200 होती. आणि त्यातील सर्वाधिक नायकांची नियुक्ती तामीळनाडू प्रदेशात झालेली असल्यामुळे तामीळनाडूतील अधिकंदा भूमी 'अमरम'च्या कपाने नायकांना देण्यात आली होती.

नायकांच्या अधिकाऱ्याखाली भूभेदात शीतता सुव्यवस्था, सुव्यवस्था व व्यवस्थाचे राज्य राखण्याचे कार्य नायकाला तत्परतेने पार पाडणे लागे. राक्षिया चोर, घेरेडोर, लुटक यांचा बंदोबस्त करावा लागे.

जिल्हा व ग्राम प्रशासन - प्रत्येक प्रांताचे विभाजन अनेक जिल्हात केलेले होते. जिल्हांना 'नाडू' किंवा 'कोहम' असे म्हणत. नाडूत अनेक खेडी व शहरांचा समावेश होता, व त्याचे प्रशासन वेगळे होते. विजयनगरच्या साम्राज्यातील व शासनव्यवस्थेतील सर्वांत लहान घटक ग्राम किंवा खेडे हा होय. ग्रामपंचायत खेड्याचा सर्व कारभार पाहत असे. ग्रामपंचायतीच्या प्रमुखाला 'अख्येगार' म्हणत. या अधिकार्याला पठानच्या मोबद्दल्यात काही जमीन व उत्पादनाचा काही भाग दिला जात असे. गावातील सुकल्याचे निरयि देणे, सरकारी कसुली व गावात शांतता प्रस्थापित करण्याचे काम अख्येगाराला करावे लागत होते. गावप्रमुखाला गौडा असे म्हटले जाई. खेड्यातील कुटूंब लोकांच्या मदतला महाजन म्हणत.

पोलीस व गुप्तहेर विभाग - प्रांताच्या प्रशासनात या विभागाला महत्त्वाचे स्थान होते. प्रांतात शांतता, सुव्यवस्था व सुवर्धितता ठेवण्यासाठी हा विभाग अत्यंत अग्रगण्य अशा प्रकारचा होता. या विभागाकडून परचक्र, आक्रमण इत्यादी माहिती उपलब्ध होत होती. प्रांतातील व इतर प्रांतातून आलेल्या चोर, दरोडेखोर, लुटकर याना पकडून त्यांना कडक शिक्षा करण्याचे काम पोलीस विभागाचे होते. यामुळे ते रात्रभर प्रांतातील नगरांमध्ये पहारा देत असत. पोलीस विभागावर नगराच्या संरक्षणाची जबाबदारी असे. विजयनगर काळातील गुप्तहेर विभाग प्रभावशाली व संघर्षीत होता. सम्राटला हजेरी राज्यातील माहिती देण्याचे काम गुप्तहेर विभाग करित असे.

फारकस्य प्रवाशांनीही विजयनगरच्या साम्राज्याचा गौरवपूर्ण उल्लेख केलेला असून प्रांतीय प्रशासनाचे विवेचन केलेले दिसून येते.

देशाचे स्वरूप प्राप्त झाले. पुढे अनेक राज्ये त्यांच्या अंकात आली आणि मगध सामर्थ्यात सत्ता बनली.

3) काशी:

वाराणसीचे वाराणसीजवळ काशीचे राज्य होते. वाराणसी ही या राज्याची राजधानी होती. या राजधानीच्या उत्तरेला व दक्षिणेला जिल्हा वाटत होत्या. जेणे करिता तीर्थकार पांढरनाथ यांचे जेथे अश्वमेध येथे राज्यकर्ते हेडिंग गेले तेथे राज्या कोसलची जिंकणी घेतले.

4) कोसल

उत्तर प्रदेशातील अवधच्या आसपास हे राज्य पसरले होते. या राज्याचे राश्ट्र नदीचे दोन किावळी वारं होती. यापैकी उत्तर कोसलची जाववस्ती व दक्षिण कोसलची कुशावती आर्याना दिन राजधानी होत्या. तसेच आर्योध्य हे एक महत्त्वाचे शहर होते. या राज्याने काशी राज्य जिंकून घेतले. प्रसेनजित हा बौद्ध समकालीन बलाढ्य राजा होता. त्याने मगध वशावरुवर वेढा घेऊन संघर्ष प्रस्थापित केले होते. बौद्ध वाइमयाग त्याच्याबद्दल वराच ठेल्लेख आढळतो. मगधच्या पुढे याही राज्याचा विकासास रागला नाही.

5) वज्जी

बिहार राज्याचा उत्तर भागात हे राज्य आहे. तेथे आठ वेगवेगळ्या जमाती होत्या. त्यातला विदेह वज्जीचा महत्त्वाच्या होत्या. वैशाली हे राजधानीचे ठिकाण होते. ते या जमातीचे सधराज्य होते. तेथे एक प्रतिनिधिक सभा होती. त्यात राजे, उपराजे, सेनापती, आडाकागणिक, श्वादेया, समावेरा होता. दैनंदिन कामकाज पाहण्यासाठी आढळले होते.

6) मल्ल

सुझ्याच्या गोरखपूर जिल्हाच्या जळपास हे मल्ल जमातीचे राज्य होते. महाभारतातही या जमातीचा उल्लेख आढळतो. या जमातीचे राज्य उ विभागात वज्जीला वेढा घेतले होते. कुशीनगरात पावा ही देण महत्त्वाची ठिकाण होती. प्रसंगी तेथे शमशाही सत्ता होती. परंतु पुढे त्यांनी प्रजासत्ताक शासनपद्धतीचा स्विकार केला. मल्ल लोक शूर व लढवळे होते. बौद्ध धर्माबद्दल

त्यांचा फार उपयोग हीडे लागला. शेतमान, कुषपणा, मासा काठिणी आशा नोंव्यांचा वापर सुरु झाला. लोकांना वचत करणे, सुलभ शाले. त्यातूनच काय सावकाशेचा काय उदरगण झाला.

भारतीय स्वयंसेवक व परकीय साम्राज्याची शक्ती न राखिल्यामुळे संस्कृतीचे नाशिकीकरण सुरु झाले. त्यांना जीवणारे सुते तयार झाले. व्यापारवृद्धी झाली. वेगवेगळ्या नगरातील लोकांचे वेगवेगळे संवय पंगीमण झाले. त्यातूनच ज्या बाणात लीजनेटिका स्थिरावले त्याला 'मजपदे' असे संबोधले. ही मजपदे क्षत्रियांच्या नेतृत्वाखाली असल्यामुळे त्यांच्या नावानी ती निगेडील राहिले. पुढे ज्या मजपदाचा सर्वत क्षेत्रांत विस्तार झाला त्यांना महाजमपदे हे संबोधना दिले गेले. त्यात सत्तास्थिती सुरु झाली. त्यामुळे काही नामशेष झाली. मगधचा साम्राज्यविस्तार सुरुच राहिला. परिणामी काही ठाणराज्येही त्यात नष्ट झाली. त्यातून मगध साम्राज्याचा उदय झाला.

७६. महाजमपदांची चोडक्यात माहिती पुढीलप्रमाणे :-

* ७६. महाजमपदे :-

१) उगम :-

इ.स. पूर्व सहाव्या शतकात उत्तर भारतात अस्तित्वात असलेले व ७६ महाजमपदांपैकी अंग हे एक राज्य होते. हल्लीच्या बिहार प्रांतातील भागलपुर, मोधीर हे जिल्हे या राज्यात मोडत होते. चंपानदी ही अंग व मगध राज्यांमधून वाहत असल्यामुळे ती त्यांची सिमा होती. अंगच्या पूर्वेकडे होती. चंपानगरी ही या राज्याची राजधानी होती. हे शहर व्यापारासाठी प्रसिद्ध होते. अंग राज्यावर मगधचा प्रमुख विस्तार याने वरिष्ठ प्रस्थापित केले.

२) मगध :-

मगधाच्या बिहार राज्यातील पाटणा व गंधा या दोन प्रांतांचा समावेश मगधमध्ये होता. तेव्हा आजाद्यांची सत्ता होती. वाडमयातही मगधचा तिरस्कारपुन उल्लेख आलेला आहे. त्या वेळाची राजधानी गिरिपुत्र किंवा राजकुंड हीच महाभारत काळात येथे बुद्धा धरणाची सत्ता होती. नरासध हा नवे राज्य करित होता. विविध आंगि आजातशत्रूच्या काळात त्याला मगध

त्यांना ओढा होता. कुड्डांचे महानिर्वाण. क्षुरिनभेरे येथेच झाले
पुढे हे राज्य मगधानी जिल्हा घेतले.

6) चिदी! चिदी वंगीयाच्या दोन कोंकणच्या वसाहती
अबुक्रमी नेपाळ व बुलखंड येथे होत्या. बुलखंडातील राजा
चेच मगधानीचे ठिकाण. काशीची मुनी तसेच शांभुमहारी
ही नेपाळच्या राज्याची राजधानी होती. कालिगात ही
त्याची एक शाखा होती. भगवान श्रीकृष्णांनी ज्या शिशु-
पालाचा वध केला ती चिदी राज्याची होती.

7) वत्स! उत्तर भारतातील बहुतेकचे राज्य म्हणून गणिते
जाले असलेले वत्स हे एक होते. त्याची राजधानी काशीची
होती. ती एक व्यापारी केंद्र होते. कांडाविषयी त्याची
ख्याती होती. भगवान कुड्डाकाल उदयन हा प्रजापतिपद
राज्य हीडून गेला. तो संस्कृत नाटक कांदळच्या त नायक
होता. यावरून त्याची लोकप्रियता लक्षात येते. त्याने
अवंतीचा राजा प्रद्योत याच्या मुलेशिषी विवाह केला
होता. तो अत्यंत पराक्रमी होता. त्याने ससुराळीत
अनेक किल्ले बांधले. हत्तीद्वाराकून सुसज्ज सैन्यदल
उभारले होते. वेद ह्यांविषयी त्याला फार आदर होता.
त्याने आपल्या राज्यात अनेक विहार बांधले.

8) कुशा! हे राज्य दिल्ली व मीरतच्या आसपास होते. ते
इतिहासप्रसिद्ध होते. त्याची राजधानी हंपूरुय होती. कौरव
राजा हा वेद सभेकालीन होता. सुरवातीला राजसत्ता
होती. मात्र नंतरच्या काळात प्रजासत्ताक शासनपद्धती
चढू लागली.

9) पांचाल! शैलीखंडात हे राज्य होते. या राज्यासमूह
गंगा नदी वाहते होती. त्यामुळे उत्तर व दक्षिण पांचाल
असे दोन भाग होते. उत्तर पांचालाची राजधानी
अहिलक्षत्र तर दक्षिणेची कापिल्य अशा राजधान्या
होत्या. प्राचीन तर्ष राजसत्ताक पद्धती होती. मात्र
नंतर प्रजासत्ताक शासनपद्धती चढू लागली.

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Department of History
SEMINAR OF B.A.P.V.

Paper: VII :- Early India (from beginning to 4th C.B.C)

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गोपालपातुन ५१ किमी. अंतरावर भीमबेटका आहे. युनेस्को वर्ल्ड हेरिटेज या संस्थेने मान्यता दिलेले हे प्रासिद्ध. या ठिकाणी अश्वमेधीन संस्कृती पहावयास मिळते. भारतीय पुराण वस्तु संग्रहालय विभागाचे ह्या जंगलात असलेल्या अश्वमेधीन जंगलाची शोध लावला. भियापुरा ह्या गावाच्या जवळ हे ठिकाण आहे. भियापुर हे किन्दासल पर्वतांच्या रांगा भाजुबाकुला भादित. भाजुबाकुला दारु जंगला, डोंगरकपारीत झसलेल्या गुहा या अतिप्रचलु हंगडाच्या राशीत भादित.

१७५५-५४ ई. व्ही. मस. वाळणकर या प्रासिद्धाच्या विचारु इतिहासकारांचे ह्या पर्वतरांगातील अश्वमेधीन गुहांचा शोध लावला. किमान इ.स. पूर्व ७५० इतार वरीपुर्वी येथे मानवी संस्कृती असल्याचे साहजुन झाले तसे त्याचे पुरावे मिळाले. भारतातील अश्वमेधीनाच्या मानवी संस्कृतीचे ठळक पुरावे येथे मिळाले. त्यांचा ह्या पर्वत रांगात झनेक गुहा सापडल्या आणि त्या गुहांमधुन ५०००० वर चित्रकृती सापडल्या. ह्या चित्रमाणेसुन माणुस मुळेकांची जिवत असावा. ही चित्रे भाणि त्या चित्राचे रंग आज ही व्यवस्थित आदित. कारु कुन भाणि पडुस येथे पोहोचत नाही. स्थानि जगा त्या चित्रकारांचे लुहामधुने गोळाशी भादित. हे वेगळ्या किमान १० इतार वरीपुर्वी मानव कसा रडान होता. त्याची पुढी कल्पणा ह्या गुहा पाहिल्यावर येते. ही मनोखी चित्रकला. त्यावेळी झसलेली पशुपक्षी भाणि मानवी संस्कृती ह्या चित्रातुन विसतात. येथील गुहा कुा पत्त प्रमुन भागात येथील गुहा ह्या भादित

① विनेका

② झोनरातली

१५६८ मध्ये म्हा फर्नान्डोवारे ह्युजेनॉट पंधीयांना धार्मिक
श्रद्धा, पूजा, आचार वाचनीत काही प्रमाणात स्वातंत्र्य दिले.
पुढील काळात कॅथॉलिकमध्ये कार्य प्रारंभही टिकले.

* स्पेनमधील धार्मिक संघर्ष →

१६ व्या शतकाच्या उत्तरार्धात स्पेनच्या राजा
कुसरा फिलिप याने कॅथॉलिकच्या प्रभाव वाढवा म्हणून प्रोटेस्टंट
पंधीयांविरुद्ध आर्वाइ केसी अनेक प्रोटेस्टंट पंधीयांना पाखंडी
व धर्मद्रोही यांना या सवयीबाबी ठार केले. तसेच अमेरिकेतील
स्पॅनिश वसाहतीमध्ये कॅथॉलिकांनाच राहण्यास अनुमती दिली.
फ्लिझाबेथ राठव्या कारबिर्दित इंग्लंडमधील कॅथॉलिकांच्या
उच्चवत्ताच्या मोहिमेवेळी स्पेन राजाने आपले द्वारमर पाहून
त्याला विरोध पुरवण्याचा प्रयत्न केला. मात्र त्यात त्याला यश
प्राप्त करता आले नाही.

* नेदरलँडमधील धार्मिक संघर्ष →

क्यलिन या धर्मसुधारकाच्या प्रयत्नांमुळे नेदरलँडमध्ये
प्रोटेस्टंट पंथाचा विकास झाला. मात्र फिलिप या स्पेनच्या
राजाने या पंथावर बरी दारुणता केली. कॅथॉलिक पंथ साधने
लापण्याचा प्रयत्न केला. राजाच्या या कुलभावाचा प्रतिकार
करण्यासाठी ज्या लोकांनी म्हा गुप्त संघटना स्थापन केली. त्यांनी
स्पेनच्या अनेक जहाने बुडवली. धार फिलिप राजाने ड्युक
ऑफ ब्राव्हाच्या नेतृत्वात सगिती नेमली. हा कुमिटीच्या
अहवालांमुळे अनेक ड्युक लोकांना जिवंत जाळण्याची शिक्षा
दिली. त्यामुळे तेथे स्थानिरा व स्व यांच्यात संघर्ष होवून अनेक
ज्यांची विकासे भुक्ता केली. तसेच राजाच्या धार्मिक बंधनातून
मुक्तता मिळविली.

* जर्मनीतील धार्मिक संघर्ष →

जर्मनीत मार्शल ल्युथरने कॅथॉलिकांना जो विरोध
केला त्याला काही जर्मन राजांनी, सरदारांनी व सर्वसाधारण
शेतकरी व मध्यमवर्गीयांनी पाठिंबा दिला. परंतु त्याला
यश आले नाही. पुढील शत्राने लोकांना धार्मिक स्वातंत्र्य
दिल्याने पोपची म्हासदारी नष्ट झाली. पुढे १६१८ ते ४८
काळात तेथे धार्मिक संघर्षांना सुरुवात झाली. फ्रिडनॉडने
प्रोटेस्टंट पंधीयांचा छळ सुरु केला.

* धर्मसुधारणा चळवळीचे अणिष्ट परिणाम →

धर्मसुधारणा चळवळीमुळे रोमन कॅथोलिक्स व प्रोटेस्टंट असे दोन पंथ निर्माण झाले. त्यांच्यात फ्रॅक्चरेशन आसूया वाढलेली होती. त्यातून मोठ्या प्रमाणावर संघर्ष झाले. त्यामुळे परस्परबंधा धार्मिक छळ, नृभ्रम करव्याच्या प्रवृत्तीत वाढ झाली. इंग्लंडमधील स्टुअर्ट राजांच्या कारकिर्दीत प्युरिटन्स फार छळ झाला. त्यामुळे त्यांना इंग्लंड सोडवे लागले; त्यांना अमेरिकेला जाणे भाग पडले. युरोपातील अनेक देशांत अशा धार्मिक छळाचा तोंडा देवावे लागले.

१६ व्या शतकात युरोपात धर्मसुधारणा चळवळीला गती मिळाली. या सुधारणेतून क्रिश्चन धर्माला प्रोटेस्टंट व कॅथोलिक्स असे दोन पंथ पडले. त्यांच्यातील असहिष्णू प्रवृत्तीतून युरोपात अनेक देशांत संघर्ष घडून आले.

१६ व्या शतकात युरोपातील प्रत्येक सत्ताधीशाने आपल्या प्रभुत्वासाठी व प्रभावासाठी कोकोळले उपाय योजले. त्यातूनच मजकूर म्हणते धार्मिक प्रेम्स व्हावे म्हणून त्यांनी प्रजेला आपल्या पंथात जोडवण्याचा प्रयत्न केला. त्यातून अनेक देशांत धार्मिक संघर्ष घडून आले. हे संघर्ष फक्त देशांतर्गत होते असे नसून देशादेशांतही धार्मिक बोबीकरण पुढे झाले. त्यातील प्रमुख संघर्ष पुढीलप्रमाणे -

* फ्रान्समधील धार्मिक संघर्ष - फ्रान्समध्ये मोठ्या प्रमाणात धार्मिक संघर्ष दाडून आले. दक्षिण फ्रान्समधील अल्बीगेनिस या उपपंथाने कॅथोलिकांच्या दक्षिणेकडून फोर प्रहार केले. त्यामुळे चिडलेल्या पोपने कडव्या कॅथोलिकांना तेथे पाठवले. त्यांनी तेथे मोठ्या प्रमाणावर कत्तल केली. याशिवाय इतर पंथांमधील पाखंडी म्हणूनही हत्या केली. पुढे कॅथोलिक्स या धर्मसुधारकांनी फ्रान्समध्ये रोमन कॅथोलिक्स पंथाकडून संघर्षित मोल्मि काढली. मात्र पोपच्या दडपणामुळे फ्रान्स सोडून जावे लागले. फ्रान्समध्ये दुसरा हेन्री याच्या कारकिर्दीत ह्युजेनॉट पंथाची लोकप्रियता वाढली. त्यांनी धार्मिक सुधारणेबरोबर राजकीयही मागणी केली. त्यामुळे १५६२ ते १६८८ या काळात रोमन कॅथोलिक्स व ह्युजेनॉट यांच्यात यादवी हेन्री ह्युजेनॉटची अमानुष कत्तल केली. यात क्रिसरा हेन्री राजाने कॅथोलिकांची बाजू उचलून हारल्यामुळे ह्युजेनॉट पंथावर बंदी घातली. पुढे चौथ्या हेन्रीने

धर्मसुधारणा चळवळीने ज्ञान, विज्ञान, साहित्य, कला, तंत्रज्ञान इत्यादी क्षेत्रांवर धर्मसंस्थेचे असलेले निबंध दूर झाले. त्यामुळे या क्षेत्रात बुद्धिप्रभावाने प्रयोगशीलता व अनुभव मिडवतेला भरून मिळाले. प्रत्येक क्षेत्रात संशोधनाला चालना मिळाली. मानवी जीवनाचे प्रतिक्रम साहित्य व कलात दिवून घेऊ लागले. धर्मसुधारणा चळवळीने शारिक, राजकीय, शारिक, वैचारिक, साहित्यिक व कलेच्या क्षेत्रात प्रकाशनाच्या युगाचा सुरुवात झाली. त्यामुळेच युरोपच्या इतिहासात धर्मसुधारणा चळवळीना अत्यंत महत्त्वपूर्ण स्थान आहे.

“Dissemination of Education for Knowledge, Science, and Culture”
-Shikshanmaharshi Dr. Bapuji Salunkhe
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon

DEPARTMENT OF BOTANY

B.Sc.III

REPORT OF CIE ACTIVITIES

**(SEMINARS)
2019-20**

**Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon
Department of Botany**

Date: 10/08/2019

To
The Principal
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli.416312

Subject: To seek Permission to conduct “**CIE Activities.**”

Respected sir,

With reference to above mentioned subject, Department of Botany is going to conduct “**CIE Activities.**” For the academic year 2019 - 2020 .We are kindly requesting you to allow us to conduct this activity for our students.

Thanking you,

Yours Faithfully



(Dr. N. A. Kulkarni)

**Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon
Department of Botany**

Date:10/08/2019

Notice

All students of **B. Sc. Part III** hereby informed that we are going to conduct various CIE activities like Seminars, Class Tests and Brain Storming Sessions in 2019 – 2020. It is compulsory to all students to participate in this activity.



(Dr. N. A. Kulkarni)

REPORT

Department of Botany has conducted various CIE activities like Seminars, Class Tests and Brain Storming Sessions in 2019 – 2020. Total 18 Students of B.Sc. part III were participated in these activities.

The entire activity conducted by **Dr. A. P. Inamdr** and Assisted by **Dr. S.K.Khade**. For successful conduction of this activity **Hon. Prin. Dr. Minlind Hujare** guided nicely and Prof. **Dr. N.A. Kulkarni** cooperated well for the proramme.



Prof. **Dr. N.A. Kulkarni**
Head of the Department

Department of Botany

B. Sc. III 2019-20

Schedule of the Seminars from 11/09/2019 to 16/09/2019

Every student should have to submit their synopsis of the topic selected for the seminar before the seminar presentation. Seminars should be completed within 2.00 pm to 05.00 pm every day after lectures.

Sr. No.	Roll No.	Name of the student	Topic of the seminar	Date
1.	6121	Ashtekar Rutuja Rajendra	Sexual Repr.in <i>Albugo</i>	11/09/2019
2.	6122	Deshmukh Amruta Baburao	Haplontic and Diplontic Life cycle	11/09/2019
3.	6123	Jadhav Neha Kisan	Asexual Reprod. in <i>Albugo</i>	11/09/2019
4.	6124	Karade Suvarna Satyawar	<i>Ectocarpus</i> External Morphology	12/09/2019
5.	6125	Khandagale Tanuja Shivaji	<i>Chara</i> Occurrence, Structure	12/09/2019
6.	6126	Khedakar Ashwini Arvind	<i>Funaria</i> Gametophyte	12/09/2019
7.	6127	Mali Prajakta Ashok	<i>Funaria</i> Int. Str. And Vegetative Repro.	13/09/2019
8.	6128	More Tejashree Rangrao	<i>Chara</i> Life cycle	13/09/2019
9.	6129	Patil Gauri Sharad	Veget. Repr. in <i>Chara</i>	13/09/2019
10.	6130	Patil Nehal Vilas	<i>Funaria</i> : Sexual Repr.	14/09/2019
11.	6131	Patil Poonam Appasaheb	<i>Uncinula</i> : Sexual Repr..	14/09/2019
12.	6132	Patil Rohini Sampatrao	Reprod. In Fungi	14/09/2019
13.	6133	Patil Samrudhi Shamrao	Isomorphic and Het. life cycles in Algae	15/09/2019
14.	6134	Pawar Snehal Madhukar	<i>Ectocarpus</i> : Sexual Repr.	15/09/2019
15.	6135	Salunkhe Ankita Satish	Haplobiontic and Diplobiontic life cycles	15/09/2019
16.	6136	Salukhe Snehal Bramhanand	<i>Chara</i> : Sexual Repr.	16/09/2019

17.	6137	Shikalgar Tammama Jamir	<i>Funaria</i> Sporophyte	16/09/2019
18.	6138	Sutar Prajka Pandurang	<i>Ectocarpus</i> : Asexual Rep.	16/09/2019



ALLOTMENT OF THE TOPICS TO THE STUDENTS

"ज्ञान विज्ञान प्राप्ति सुरक्षितान् यान्मती शिक्षणप्रदाय"
शिक्षणमन्त्री डॉ. बापूजी नाईको

Shri Swami VivekanandShikshanSanstha,Kolhapur's

Padmabhushan Dr. VasanttraodadaPatilMahavidyalaya, Tasgaon

Department of Botany, B. Sc. III 2019-20 (Dr. Alka P. Inamdr)

11/09/2019 to 16/09/2019

S. N.	Roll No.	Name of the Student	Title of Seminar	Sign.
1	6121	Ashtekar Rutuja Rajendra	Sexual Repr.in <i>Albugo</i>	<i>Rutuja</i>
2	6122	Deshmukh Anruta Baburao	Haplontic and Diplontic Life cycle	<i>Anruta</i>
3	6123	Jadhav Neha Kisan	Asexual Reprod. in <i>Albugo</i>	<i>Neha</i>
4	6124	Karade Suvarna Satyawani	<i>Ectocarpus</i> External Morphology	<i>Suvarna</i>
5	6125	Khandagle Tanuja Shivaji	<i>Chara</i> Occerrence, Structure	<i>T.S.Khandagale</i>
6	6126	Khedkar Ashwini Arvind	<i>Funaria</i> Gametophyte	<i>Ashwini</i>
7	6127	Mali Prajakta Ashok	<i>Funaria</i> Int. Str. Vegetative Repro.	<i>Prajakta</i>
8	6128	Mali Tejashree Rangrao	<i>Chara</i> Life cycle	<i>Tejashree</i>
9	6129	Patil Gauri Sharad	Vegetative Repr. in <i>Chara</i>	<i>Patil G</i>
10	6130	Patil Nehal Vilas	<i>Funaria</i> : Sexual Repr.	<i>Nehal</i>
11	6131	Patil Poonam Appasaheb	<i>Uncinula</i> : Sexual Repr..	<i>Poonam</i>
12	6132	Patil Rohini Sampatrao	Reprod. In Fungi	<i>Rohini</i>
13	6133	Patil Samrudhi Shamrao	Isomorphic and Het. life cycles in Algae	<i>Samrudhi</i>
14	6134	Pawar Snehal Madhukar	<i>Ectocarpus</i> : Sexual Repr.	<i>Snehal</i>
15	6135	Salunkhe Ankita <i>Satish</i>	Haplobiontic and Diplobiontic life cycles	<i>Salunkhe</i>
16	6136	Salunkhe Snehal Bramhanand	<i>Chara</i> : Sexual Repr.	<i>Snehal</i>
17	6137	Sutar Prajakta Pandurang	<i>Ectocarpus</i> : Asexual Rep.	<i>Prajakta</i>

Alka

Alka P. Inamdr
HEAD
DEPARTMENT OF BOTANY
PADMABHUSHAN DR. VASANTRAO DADA PATIL
MAHAVIDYALAYA, TASGAON, DAST SANGLI

ASSESSMENT OF THE SEMINARS

"ज्ञान विज्ञान आणि सुनोबकार कानाडी शिक्षणप्रसार"
 शिक्षणसंस्था डी. सांगली जिल्हा
 Shri Swami VivekanandShikshanSanstha,Kolhapur's

Padmabhushan Dr. VasantaoDadaPatilMahavidyalaya, Tasgaon
 Department of Botany, B. Sc. III 2019-20 (Dr. Alka P. Inamdr)
 11/09/2019 to 16/09/2019

S. N.	Roll No.	Name of the Student	Title of Seminar	Presenta tion (10)	Subject knowledge and Discusst(10)	Total (20)
1	6121	Ashtekar Rutuja Rajendra	Sexual Repr.in <i>Albugo</i>	08	09	17
2	6122	Deshmukh Amruta Baburao	Haplontic and Diplontic Life cycle	07	06	13
3	6123	Jadhav Neha Kisan	Asexual Reprod. in <i>Albugo</i>	08	09	17
4	6124	Karade Suvarna Satyawar	<i>Ectocarpus</i> External Morphology	07	08	15
5	6125	Khandagle Tanuja Shivaji	<i>Chara</i> Occurrence, Structure	08	08	16
6	6126	Khedkar Ashwini Arvind	<i>Funaria</i> Gametophyte	08	07	15
7	6127	Mali Prajka Ashok	<i>Funaria</i> Int. Str. And Vegetative Repro.	08	08	16
8	6128	Mali Tejashree Rangrao	<i>Chara</i> Life cycle	06	06	12
9	6129	Patil Gauri Sharad	Veget. Repr. in <i>Chara</i>	08	08	16
10	6130	Patil Nehal Vilas	<i>Funaria</i> : Sexual Repr.	08	07	15
11	6131	Patil Poonam Appasaheb	<i>Uncinula</i> : Sexual Repr.	09	08	17
12	6132	Patil Rohini Sampatrao	Reprod. In Fungi	05	07	12
13	6133	Patil Samrudhi Shamrao	Isomorphic and Het. life cycles in Algae	07	06	13
14	6134	Pawar Snehal Madhukar	<i>Ectocarpus</i> : Sexual Repr.	06	06	12
15	6135	Salunkhe Ankita Satish	Haplobiontic and Diplobiontic life cycles	05	05	10
16	6136	Salunkhe Snehal Bramhanand	<i>Chara</i> : Sexual Repr.	08	08	16
17	6137	Sutar Prajakta Pandurang	<i>Ectocarpus</i> : Asexual Rep.	06	06	12
18	6138					

Alka


 HEAD
 DEPARTMENT OF BOTANY
 PADMABHUSHAN DR. VASANTRAO DADA PATIL
 MAHAVIDYALAYA, TASGAON, DIST. SAINGLI

Photos of the Seminar Activity



Dr. Alka Inamdr
Department of Botany
PDVP Mahavidyalaya, Tasgaon

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasanttraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक ०१.१०.२०१९

वी.कॉम.भाग २ मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, शनिवार दिनांक ०५.१०.२०१९ रोजी Tally ERP 9 ची PPT दाखविण्यात येणार आहे याची नोंद घ्यावी.



डॉ. ए. जी. सोनवले

HEAD

Department of Commerce
P.D.V.P.College, Tasgaon.



डॉ. मिलिंद एस. हुजरे

प्राचार्य

पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगाव (जि. सांगली)

Department of Commerce

Tally ERP 9



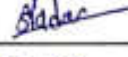
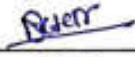
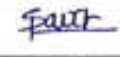

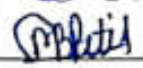
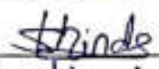
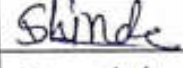
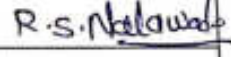



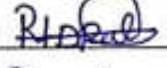
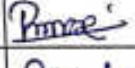
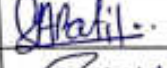
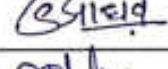
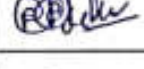


PPT PRESENTATION

Class: B.Com.II

Subject: Corporate Accounting

Date: 05.10.2019

Time : 8:50 to 10:15

Sr.No.	Name	Signature
1)	Pallavi Ramesh Yadav	
2)	Nikita Kumar Yadav	
3)	Sonali Shivaji Yadav	
4)	Namrata Papat sutar	
5)	Swati Subhash Pawar	
6)	Harshada Bhanudas Patil	
7)	Mrunali Balawant Patil	
8	Sandhya Dhanaji Shinde	
9	Shruti Vishwas Shinde	
10.	Rutuja Shahaji Nalawade	
11	Kalase Pallavi Dilip	
12	Chavan Sonam Sambhaji	
13	Patil Shweta Satish	
14	Patil Rutuja Dnyaneshwar	
15	More Pranali Madhukar	
16	Patil Shubhangi Aravind	
17	Jadhav Shital Shashikant	
18	Jadhav Rutuja Rajendra	
19.	Mane Shweta Anil.	
20.	Mane Rutuja Bhagwan.	


21	Nalavade Radhika Vishvas	<u>RL</u>
22	Maske kamal suresh	<u>RM</u>
23	Shinde Pratiksha Adhikar	P. A. Shinde
24	Yalamat Sonali Popat	<u>SPYalamat</u>
25	Jadhav Mamata Mahadev.	<u>Mamata</u>
26	More Madhuri Nandkumar	M. N. More
27	Salunkhe Anital Vinayak.	<u>Asalunkhe.</u>
28	salunkhe Anuradha Dnyande	<u>Asalunkhe.</u>
29	More Yagita Jayjayram	<u>More</u>
30	Gehadake Pooja Shashikanth	<u>Pooja</u>
31	Patil Sima Dinkar	<u>Patil</u>
32	Patil Manisha Balkrishna	<u>Patil</u>
33	Patil Komal Deepak	<u>Patil</u>
34	Suryawanshi Apurva Amal	<u>Amal</u>
35	Yadav Pooja Omkar	<u>Pooja</u>
36.	gutar Nilam suresh	<u>N. S. gutar</u>
37	mali someer Hunamunt	<u>Mali</u>
38	nisale Akshay Tambuwant	<u>Nisale</u>
39	Ingaole Pratik Dipak	<u>P. Ingaole</u>
40	Sulankar Gabil Santosh	<u>Sulankar</u>
41	Sutar Sushant Shivaji	<u>Sutar</u>
42	Aniket kisan Jadhav	<u>Aniket</u>

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

नोटीस

दिनांक १७.०८.२०१९

वाणिज्य विभागातील वी. कॉम. भाग १ मधील सर्व विद्यार्थ्यांना कळविण्यात येते की, शुक्रवार
दिनांक २३.०८.२०१९ रोजी योग्यता चाचणी (Aptitude Test) घेण्यात येणार आहे याची
सर्वानी नोंद घ्यावी.


(डॉ. ए. जी. सोनवले)

HEAD
Department of Commerce
P.D.V.P.College, Tasgaon.


(डॉ. मिलिंद हुजरे)

Principal
Padmabhushan Dr. Vasantraodada Patil
Mahavidyalaya, Tasgaon, (Sangli)

DEPARTMENT OF COMMERCE

SLOW LERNERS & ADVANCED LERNERS LIST

2019-20

Sr. No.	Name	Marks (12 th)	Signature
1)	Kalina Jabangje Mane	57.80	<u>Mane</u>
2)	Akanksha Uttam zambare.	73	<u>Akanksha</u>
3)	Sandhyarani Dhanaji Mane	66.20	S.D. Mane
4)	Pradnya Ramchandra Patil	50.1%	P. Patil
5	Harshada Bandu krode.	61.1%	<u>Harshada</u>
6	Tunali Dagadu Lokhande	60.92%	T. D. Lokhande
7	Monali Laxman Wadkar.	64.1%	M. L. Wadkar
8	Saniya Shermaahmad sayyed	63.0%	<u>Saniya</u>
9	Ghripali Dilip Shendage	63.1%	<u>Shendage</u>
10	Gaikwad Bhagyashri Hanamant.	65.54%	<u>B. Gaikwad</u>
11	Shreyasha Subal Gavali	64.1%	<u>Shreyasha</u>
12	Rutuja Dilip Patil	70.76.1%	<u>Rutuja</u>
13.	Ankita Gajanan Phadatore.	46.70.1%	<u>Ankita</u>
14	shital shankar more	72.92.1%	<u>Shmore</u>
15	Shweta Ashok Gaikwad	59.38	<u>Gaikwad</u>
16	Poonam Vilas Jadhav	60.00	<u>P. Jadhav</u>
17	Siddhi Anil Kadam	62.30	<u>S. Kadam</u>
18	Madhura Nivas Gurav	68.62	<u>M. Gurav</u>

19)	Mali Katal Krushna	52.46%	<u>P.Mali</u>
20)	Mali Nisha Shrikant	63.38%	<u>P.S.Mali</u>
21)	Patil Anuradha Yashwant	69.69%	<u>Patil</u>
22)	Shinde Nikita Balaso	69.52	<u>Shinde</u>
23)	Devkule Akshata Sugriv	63.23%	<u>Puf</u>
24)	Ghotkar Shivani Shivaji	59.99%	<u>S.S.Ghotkar</u>
25)	Mohite Rutuja Nandkumar	54.50%	<u>R.Mohite</u>
26)	Chavan Sujata Mahadev	81.85%	<u>Chav</u>
27)	Shweta Siddheshwar Patil	65.54%	<u>Shweta</u>
28)	More Bhagyashri Sudhakar	72%	<u>B.More</u>
29)	Patil Shivani Rajendra	73.54%	<u>Patil</u>
30)	Patil Ankita Bharat	60.15%	<u>Patil</u>
31)	Patil Shital Sambhaji	66.46%	<u>Patil</u>
32)	Patil Ankita Ramchandra	68.00%	<u>Patil</u>
33)	Patil Shrutika Anandrao	51.08%	<u>Patil</u>
34)	Ghodke Akshada Milind	63.08%	<u>A.Ghodke</u>
35)	Khot Shital Dattatray	65.53%	<u>Khot</u>
36)	Kharmate Swapnali Satish	63.08%	<u>S.S.Kharmate</u>
37)	Pardeshi Prajakta Vdy	64.75%	<u>P.V.Pardeshi</u>
38)	Pardeshi Sanskruti Narayan	60.30%	<u>Pardeshi</u>
39)	Koli Hema Bhagwan	72.92%	<u>H.Koli</u>
40)	Pardeshi Namrata Vinayak	60%	<u>N.V.Pardeshi</u>
41)	Benkar Laxmi Sijaram	76.92%	<u>Benkar</u>

41)	Patil Kernal Babasaheb	68.74%	Kernal
42)	Limbe Dnyu Baban	70.82%	Patil
43)	Athar Tusneem Ilahi	57%	Patil
44)	Rawar Manika sanjay	59.85%	M.S.P
45)	Patil Nikita Baban	67.23%	Patil
46)	Ankita Ashok Kote	51.38%	Patil
47)	Rawar Shrotha Kumar	75.92%	Shrotha
48)	Charan Anuja Amar	72%	Charan
49)	Jadhav Akshita Dilip	49.69%	Jadhav
50)	Mali Piyusha Shankar	69.1%	Mali
51)	Mali Shivani Bapu	70.1%	Mali
52)	Wagmode Megharani Sudam	73.08%	M.S.W.
53)	Manc Ankita Jottram	73.69%	Manc
54)	Satpute Rutuja Sanjay	62.46%	Patil
55)	Jadhav Piyanka Pradip	72.31%	Patil
56)	Patil Prashant Rajendra	78.00%	Patil
57)	Mulla Sahil Suraj	64.60%	Patil
58)	Patil Vaibhav Suresh	76.46%	Patil
59)	Nikan Sahin A. Achan	60.00%	Nikan
60)	Patil Pratik Suresh	42.90%	Patil
61)	Santosh Basappa Malawde	48%	Santosh
62)	Charan Rohan Ramesh	61.32%	Charan
63)	Patil Goraksh Suraj	48.46%	Patil

65	Patil Ajay Suchakar	56%	Aspatil
66	Abhijeet Ashok Mane	69.69%	Amo
67	AJAY VIJAY ALVARE	57.85%	Alvar
68	PRATHNESH PRAASHANT BOCHHE	53.28%	PBP
69	Ganesh Somanath Warde	53%	G.S.Warde
70	Aniket Anil Patil	60.77%	APatil
71	Patil Amit sharmgondal	71.23%	APatil
72	Aniket phanudas Guliy	67.38	A.B.Guliy
73	Rahul Ashok mali	57%	RAM
74	Shubham Vasant Mali	70.15%	S.V.Mali
75	Rohit Komalji Nikam	65.28%	R.K.Nikam
76	Tahid Khalid Manin	40.92%	Tahid
77	Abhishek. Vittal Sawant	64.80%	A.V.Sawant
78	Vighnesh sharaad patil.	54.7%	VPatil
79	omkar datta mali	67%	O.D.Mali
80	Akash sambhaji chendage	65.8%	Achendage
81	shubham dipak & koli	61.23%	Shubham
82	varun machendranath patil	54.80%	Vmpatil
83	Nikhil Suresh Sutar	64%	N.Sutar
84	Mayur Sanjay Mali	62.92%	M.Mali
85	Tingare Indrajeet Revansiddh	65%	T.Revansiddh
86	Dhondiram Tanaji Waghmode	52%	D.Waghmode
87	MANE Shivam Vishwanath	64.15%	SM

"ज्ञान विज्ञान अणि सुसंस्कार वांगडी शिक्षणप्रकार" - शिक्षणनदर्या डॉ. वापूजी सादुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक 0१.१0.२0१९

वी.कॉम.भाग ३ मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, सोमवार
दिनांक 0७.१0.२0१९ रोजी Advanced Accountancy – I या विषयाची सराव परीक्षा
(Preliminary Exam) घेण्यात येणार आहे तरी सर्वांनी याची नोंद घ्यावी.



(डॉ. ए. जी. सोनवले)

HEAD
Department of Commerce
P.D.V.P. College Tasgaon.

CLASS: B.COM.III

TIME: 08:45 TO 10 :45

Subject : Advanced Accountancy

Date : 07.10.2019

Marks : 40

-
- Instruction : 1) All question are compulsory.
2) Figures to the right indicate full marks.
3) Use of calculator is allowed.

Q.1) The following figures are extracted from the books of Nirvighna Bank Ltd. as on 31 March 2019.

Particulars	Rs.
Interest on Loan	3,10,000
Interest on cash credit	4,90,000
Interest on balance with RBI	25,000
Interest on Investment	1,90,000
Interest on Fixed Deposits	2,60,000
Interest on Saving Accounts	1,10,000
Interest on borrowings from RBI	10,000
Profit on sale of Investment	65,000
Loss on sale on Investment	5,000
Profit on Revaluation of Investment	60,000
Loss on Revaluation of Investment	10,000
Salaries to Staff	65,000
Rent & Taxes	8,000
Depreciation on Bank Property	21,000
Sundry Income	18,000
Printing & Stationary	31,000
Advertisement	15,000
Audit Fees	14,000
Postage & Telephone Charges	15,000
Profit on 01.04.2018	1,20,000

Adjustments :-

- i) Write off Rs. 19,000 for bad and doubtful debts.
- ii) Provide 40% for Taxation.
- iii) Rebate on bill discounted is to be provide for Rs. 20,000.

From the above information prepare

- a) Profit and loss account in vertical form 08
- b) Necessary schedule required for profit and loss account 08

Q.2) attempt any two of the following A,B,C

A. M/s G. Dipak Traders is insured under a loss of profit for Rs. 1,26,000. Their books of accounts are Closed on 31st December every year. The fire occurred on 1.9.2015. The records saved disclosed the following information:

Turnover during the year ended 31.08.2015	14,40,000
Turnover during the year ended 31.03.2015	12,00,000
Turnover from 1.9.2015 to 30.11.2015	60,000
Turnover in the corresponding period of 2014	3,60,000
Standing charges for the year ending 31.03.2015	72,000
Net profit during the year ending 31.03.2015	48,000

It has been ascertained that the business has consistently shown an increase of 25% in turnover in the months preceding the fire over corresponding period of the previous year. Calculate the amount of claim under loss of profit policy.

B. A fire occurred in the business premises of M/s Ashok Traders on 15.10.1999. From the following ascertain the claim for loss of stock.

Stock (1.1.1998)	34,000
Purchases (1.1.98 to 31.12.98)	1,22,000
Sales (1.1.98 to 31.12.98)	1,80,000
Stock 31.12.98	30,000
Purchases (1.1.99 to 14.10.1999)	1,47,000
Sales (1.1.99 to 14.10.1999)	1,50,000

Stock salvaged was worth Rs. 18,000. The amount of Policy was Rs. 60,000. There is an average clause in the policy.

C. From the following particulars find out the amount of claim for loss of profit date of fire 30.06.2000

Period of indemnity: six months.

Amount of policy Rs.40,000

Net profit for the accounting year ended 31st March 2000 Rs. 12,500.

Sales for the year ended 30.6.2000 Rs. 2,00,000.

Standing charges for the year ended 31.3.2000 Rs. 28,500.

Turnover for the year ending 31.3.2000 Rs. 1,98,000.

Turnover for the indemnity period from 1.7.2000 to 31.12.2000 Rs. 56,000.

Turnover for the period from 1.7.1999 to 31.12.1999 Rs. 1,10,000.

The turnover for the year 2000-2001 had shown a tendency of increase of 10% over the turnover of the preceding year.

Q.3) Write Short Note (Any Two)

i. Loss of Profit Policy

ii. Loss of Stock Policy

iii. Advantages of management accounting

iv. Higher Purchase System

@@@

Sr.No.	Name	Signature	Marks
1)	श्रीराम राहुल सुधाकर	Sudhakar	27
2)	पार्ली निलम नईद	Nilam	29
3)	माली तृषाली नारायण	Trishali	32
4)	करडि अनानी दिलीप	Anani	18
5)	सांबरे स्वाती मानसिंग	Manasing	34
6)	जाधव प्रतिक्षा प्रकाश	Pratiksha	27
7)	जाधव मयुरी माणिक	Manik	21
8)	माने प्रेरणा उमनिल	Prerna	09
9)	पार्ली सायमी सुभाष	Saimi	22
10)	काकर सुभांगी संभाजी	Subhangi	24
11)	साळुंखे जमता मवनाथ	Jamata	29
12)	मुल्गणी रेश्मा लालासखेव	Rashma	44
13)	गायकवाड शिवानी संपत	Shivani	29
14)	चव्हाण सुजाता विठ्ठल	Chavan	28
15)	जाधव स्वप्नाली चंद्रकांत	Swapnali	17
16)	गायकवाड स्वप्नाली अनिल	Swapnali	19
17)	शिंदे शोभिता वांकर	Shinde	22
18)	माली मयुरी पुकार	Murali	29
19)	कांबीकर अंजली प्रकाश	Kambikar	14
20)	सुतार काजल उमेश	Sutar	19
21)	चव्हाण अनुजा अर्जुन	Chavan	24
22)	कुंभार पूजा राजाराम	Kumbhar	24
23)	जाधव ज्येष्ठा जगन्नाथ	Jadhav	21
24)	माली प्रतिक्षा राजाराम	Pratiksha	19
25)	शिंदे मिलुला बाळासो	Milule	30
26)	कपूर पमिता बाळासो	Kapoor	27
27)	वांडरे मयुरी बाळासो	Vandare	32
28)	चव्हाण मयुरी सनेश	Chavan	09
29)	पवार निकीता सुर्यकांत	Pawar	14
30)	मुल्ला आशेषा बाबू	Mulla	27
31)	माने अपर्णा शिवाजी	Mane	32
32)	कोळी स्वाती अशोक	Koli	34
33)	गावारी रुपाली माणिक	Garari	29
34)	भांडगे राणी रामचंद्र	Bhandge	31
35)	सुतार लक्ष्मी सुरेश	Sutar	32
36)	माने कोमल धोंडीराम	Mane	09

"आम शिक्षण आणि सुरंस्कृत संताठें विवणपना" - विवणपनापी शी. बापुगी मळुंये
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक 0९.0९.२0१९

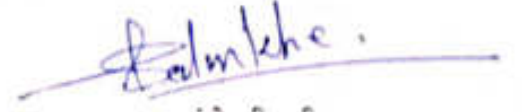
शी. कॉम. भाग १ मधिल सर्व विद्याध्यांना कळविण्यात येते की, शुक्रवार दिनांक
१३.0९.२0१९ रोजी व्यवस्थापनाची तत्वे व उपाययोजना या विषयाची घटक चाचणी (Unit
Test) घेण्यात येणार आहे तरी सर्वानी याची नोंद घ्यावी.



डॉ. ए. जी. सोनवले

HEAD

Department of Commerce
P.D.V.P. College, Tasgaon



पा. साकुंये पी. वी.

2060152

Unit Test.

13/9/2019

Lucky Page No.:
Date: / /

* Student Name.*	Mark 10	Sign
1] Abhishek Sawant	7	Abhishek A.S.
2] Akash Sambhaji chendage	6	Chendage
3] Aniket Anil Patil	6	Patil
4] Ajay Sudhakar Patil	7	Patil
5] shubham Ashok Patil	6	S.A.Patil
6] Patil Amit Shermunda	6	Patil
7] Jadhav Pratik Sungh	6	Patil
8] Mane Abhijeet Ashok	8	Mane
9] mulani mahamad Raju	7	mulani
10] Gunesh Somnath Harde	6	G.S. Harde
11] Mali Mamata Balasa	8	Mamali
12] Waghmode megharani & Suleam	8	M.S. Waghmode
13] mali Shivani Bapu	8	Bapu
14] Devkule Akshata sugriv	7	Devkule
15] kharade nikita dhondiram	6	Nikita
16] Shinde Nikita Balasa	7	Shinde
17] Patil shivani Rajendra	6	Patil
18] More Bhagyashri Suelhakar	7	More
19] Mohite Rutuja Nandkumar	8	R.Mohite
20] Gholkar Shivani Shivaji	6	Gholkar
21] Gaikwad Shweta Ashok	7	Gaikwad
22] Gaikwad Akanksha Mahadev	6	Gaikwad
23] Kadam Siddhi Anil	6	Kadam
24] Gurav Madhura Nivas	6	Gurav
25] Jadhav Poonam Vilay	7	Jadhav
26] khot shital Dattatray	8	khot
27] kharmate swapnali satish	8	kharmate
28] Phadkare Ankita Gajanan	7	Phadkare
29] Sayyad Saniya Shermahmad	6	Sayyad
30] Gaikwad Bhagyashri Hanamant.	7	Gaikwad
31] Yamgar Sonali Sadashiv	7	S.S. Yamgar

		8	Sajash
32	Govali Shreyasha Suhel	9	Patil
33	Patil Vijita Sunil	9	T.D. Lokhande
34	Lokhande Tumali Daggadu	9	M.L. Wadkar
35	Wadkar Manali Laxman		

Total = 35

~~Galunkhe.~~
13/9/2019

(Miss. P.B. Galunkhe)

Shree...

...

...

...

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
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Department of Commerce

सूचना

दिनांक १३.०९.२०१९

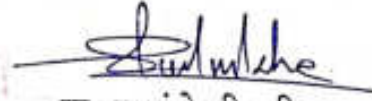
वी.कॉम.भाग २ मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, बुधवार दिनांक १८.०९.२०१९ रोजी उदयोजकतेची मूलतत्वे या विषयाची घटक चाचणी (Unit Test) घेण्यात येणार आहे तरी सर्वांनी याची नोंद घ्यावी.



डॉ. ए. जी. सोनवले

HEAD

Department of Commerce
P.D.V.P.College, Tasgaon.



प्रा. साळुंखे पी. वी. . .

Unit test.

B.Com.II

18/9/2019

Page No.

Date

Roll No	Students Name	Sign	Mark.
4606	Reshma Dhanaaji Sawant		8
4711	Rutuja Aakaram Salunkhe		7
4649	Jadhav Pratiksha Dinkar		9
4570	Patil Amruta Anil		8
4584	Smita Rajaram Patil.		8
4529	Mayuraj Mahadev Hivare		9
4710	Aishwarya Avinash Salunkhe		7
4642	Striket Kisan Jadhav		5
5.	Sutar Sushant Shivaji		7
	Pawar Nilesh Nander		8

Total = 10

18/9/2019

(Miss. P.B. Salunkhe)

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक ११.०९.२०१९

वी.कॉम.भाग १ मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, बुधवार दिनांक १८.०९.२०१९ रोजी व्यवस्थापनाची तत्वे व उपयोजन या विषयाची घटक चाचणी (Unit Test) घेण्यात येणार आहे तरी सर्वांनी याची नोंद घ्यावी.



डॉ. ए. जी. सोनवले
HEAD

Department of Commerce
P.D.V.P.College, Tasgaon.



प्रा. चव्हाण पी. आर.

Sub-Management, Std-B.com-I

	Students Name	Marks	Sign
1	Khot Dipali Dilip	9	<u>Dipali</u>
2	Nyaynirgune Arati Arjun	8	<u>Nyaynirgune</u>
3	Mane Pooja Lalaso	8	<u>Mane</u>
4	Dound Priti Dilip	7	<u>PDD</u>
5	Khayat Shivani Mohan	6	<u>S.M.Khayat</u>
6	Patil Sameuddhi Babusao	10	<u>S.B.Patil</u>
7	Jadhav Pooja Ananda	9	<u>P.A.Jadhav</u>
8	Kamble Asha Shashikant	8	<u>A.Kamble</u>
9	Sakate Shauli Keshav	8	<u>Sakate</u>
10	Godbole Sarjwani Santosh	5	<u>Sarjwani</u>
11	Tawar Samiksha Ravaso	8	<u>Tawar</u>
12	Patil Shital Tanaji	9	<u>S.Patil</u>
13	Patil Tejashree Popat	8	<u>T.P.Patil</u>
14	Babar Rutuja Sanjay	6	<u>R.S.Babar</u>
15	GURAV Swapnali Shrikant	7	<u>Swapnali</u>
16	Autade Madhuri Bharat	9	<u>Autade</u>
17	Patil Pratiksha Manik.	7	<u>Patil</u>
18	Mainkas Archana Anil.	8	<u>AM</u>
19	Mane Yogita Jalindar	7	<u>Mane</u>
20	Pawar Aditi Dilip	9	<u>Pawar</u>

21	Chavan Sayali Anil	8	
22	Jadhav Yogesh Ankush	7	
23	Sutar Nikhil Suresh	7	
24	Jadhav Yogesh Mohan	8	
25	Kumbhar Sourabh Pandurang	6	SPTS
26	Kumbhar Abhishek Balasa	7	ABK
27	Sutar Amit Dhondiram	8	Sutar
28	Patil Shubham Suresh	10	Spatil
29	Mane Nilesh Shivaji	8	Mane
30	Mane Shivam Vishwanath	7	SM
31	Kumbhar Revati Mahadev	8	Rohumbhar
32	Pole Aishwarya Shankar	8	Aspata
33	Jadhav Shilpa Hanmant	9	SH.Jadhav
34	Chavan Manshi Vinayak	8	M.V.Chavan
35	Sapkal Swapnali Bharat	8	Sapkal
36			
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43			
44			

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

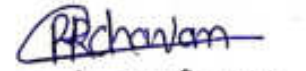
सूचना

दिनांक १६.०८.२०१९

वी.कॉम.भाग २ मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, सोमवार दिनांक २३.०८.२०१९ रोजी उदयोजकतेची मूलतत्वे या विषयाची घटक चाचणी (Unit Test) घेण्यात येणार आहे तरी सर्वांनी याची नोंद घ्यावी.


डॉ. ए. जी. सोनवले
HEAD

Department of Commerce
P.D.V.P.College, Tasgaon.


प्रा. चव्हाण पी.आर.

No	Name	Sign	Marks
1.	Mane Rutuja Bhagwan	<u>RMane</u>	6
2.	Mane Shweta Anil.	<u>Shweta</u>	7
3)	Shinde Shrutti vishwas.	<u>Shinde</u>	6
4)	Guryawanshi APURVA Amar	<u>Apueli</u>	9
5]	Patil Pratibha vishvanath	<u>Patil</u>	6
6)	gutar Nilam suresh	N.S.sutar	9
7]	Bhasme Ankita Sandip	A.S.Bhasme	6
8)	Sagare Arati sudesh	<u>Sagare</u>	9
9]	Hake Rutuja Krishna	<u>Hake</u>	8
10)	NYAYJINGUNE Komal Anil	<u>Jingune</u>	9
11]	Mali Sonali Rajendra	<u>Sonali</u>	8
12]	Patil Nutan Subhash	<u>Patil</u>	9
13)	Rokane Pallavi Ravindra	P.R.Rokane	9
14	Gavali Pratiksha Buresh	<u>Gavali</u>	9
15)	Pawar Poadnya Poadeep	<u>Pawar</u>	8
16]	GhosPade Susheta Madhukar	<u>GhosPade</u>	7
17)	Mali Ashwini Anil	<u>Mali</u>	6
18)	snehal Balasaheb Bhasate	<u>snehal</u>	6
19)	charan Reshma Anil	<u>Charan</u>	6
20)	Mithare Rutuja Kalgonda.	<u>Mithare</u>	6
21)	Shinde Anita Renukaben	<u>Anita</u>	7
22)	Potdare Gitonjali Dayanand	<u>Potdare</u>	7
23)	khoslat snehal Satish	<u>khoslat</u>	6
24]	Bobade Rohan Sanjay	<u>Bobade</u>	8
25)	Pawar Arjun Jambardan	<u>Pawar</u>	6
26]	BansalVade Dattatray Ashok	<u>Bansal</u>	4
27)	shubham sanjay Ghule	<u>S.S.Ghule</u>	6
28]	Patil Pratik Vites Jitendra	<u>P.D.Patil</u>	9
29]	Koti Nilesh Sudam	<u>Koti</u>	7

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. बापूजी साळुंखे
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Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक ०६.०९.२०१९

वी. कॉम. भाग ३मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, शनिवार दिनांक
१४.०९.२०१९ रोजी सहकाराची विकास या विषयाची घटक चाचणी (Unit Test) घेण्यात
येणार आहे तरी सर्वांनी याची नोंद घ्यावी.


डॉ. ए. जी. सोनवले
HEAD

Department of Commerce
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प्रा. पाटील जे. आर.

Students Name

	Signa	Mark
1) Dhandashri Dhondiram Kokate.	<u>Dhandashri.</u>	12
2) Patil Ashwini Arvind	<u>Patil.</u>	10
3) Jadhav Mayuzi Manik	<u>Jadhav.</u>	12
4) Shinde Nituja Balaso	<u>Shinde</u>	12
5) Kadam Pramila Balaso	<u>Kadam</u>	10
6) Shinde Yogita Shankar	<u>Shinde</u>	12
7) Gaikwad Swapnali Anil	<u>Gaikwad</u>	11
8) Patil Rutuja Ankush	<u>Patil</u>	10
9) Patil Ashwini Baban	<u>Patil</u>	10
10) Bhise Rutuja Vijay	<u>Bhise</u>	12
11) Paadeshi Aishwarya Shankar	<u>Paadeshi</u>	13
12) Sayad Muskan Riyaj	<u>Sayad</u>	10
13) Varade Pami Somanath	<u>Varade</u>	10
14) Chavan Komal Bajrang	<u>Chavan</u>	12
15) Manek Taireen Zakir	<u>Manek</u>	12
16) Utpat Sayali Sunil	<u>Utpat</u>	10
17) Sbiyani Sakshi Sanjay	<u>Sbiyani</u>	11
18) Modagekar Pooja Shivaji	<u>Modagekar</u>	10
19) Suryawanshi Bhagyashai Bajrang	<u>Suryawanshi</u>	10
20) Shinde Vashali Anandrao	<u>Shinde</u>	10
21) Jadhav Rohini Kakasaheb	<u>Jadhav</u>	12
22) Jadhav Swapnali Chandrakant	<u>Jadhav</u>	12
23) Nalawade Anjali Sukhadev	<u>Nalawade</u>	12
24) Phalake Sarita Nivas	<u>Phalake</u>	10
25) More Pramila Manohar	<u>More</u>	10
26) Jagtap Rutuja Kisan	<u>Jagtap</u>	10
27) Patil Shweta Sampat Rao	<u>Patil</u>	11
28) Jadhav Sneha Jagannath	<u>Jadhav</u>	13
29) Chavan Sujata Vittal	<u>Chavan</u>	10
30) Pawar Priyanka Bhanudas	<u>Pawar</u>	10

31)	Chavan Rutuja Sanyasa	<u>Chavan</u>	11
32)	Vandare Mayuri Balaso	<u>Vandare</u>	11
33)	Mulani Tamanna Latif	<u>Mulani</u>	11
34)	Patole Sakshi Parashram	<u>SRP</u>	11
35)	Pawar Nikita Sanyakant	<u>Pawar</u>	10
36)	Mulla Aayasha Babu	<u>Mulla</u>	11
37)	Shrudha Bharat Patil	<u>Patil</u>	11
38)	Manasi Krishnadev Chavan	<u>Manasi</u>	12
39)	Payal Vasant Gaikwad	<u>Payal</u>	11
40)	Shivani Sampat Gaikwad	<u>Shivani</u>	12
41)	Anupa Tamaji Wagh	<u>ATWagh</u>	10
42)	Swati Mansing Zambare	<u>Swati</u>	15
43)	Patil Sayali Subhash	<u>SPatil</u>	14
44)	Patil Apurva Chandrakant	<u>APatil</u>	10
45)	Jadhav pratiksha Prakash	<u>Pratiksha</u>	13
46)	Patil Nilam Nivas	<u>N.N.Patil</u>	13
47)	Mane Aishwarya Anil	<u>Mane</u>	12
48)	Raste Kajal Ramchandra	<u>KRaste</u>	12
49)	Velhal Sayali Ramchandra	<u>Velhal</u>	11

Bcom III Co-operation

Student's Name	Sing	Marks
1. शोरान राहुल सुधाकर.	Sweet	12
2. केळी विशाल कुमाव	Shani	12
3. गणेश बाबुनाथ पाटीले	Shakti	11
4. विनीत विठ्ठल	Rohit	11
4. पाटील राहुल रघुनाथ	Shane	11
5. माने कोमल द्योडीराम	Hastur	12
6. सुतार हर्षदा सुरेश	Nishantkhe	11
7. आर्जुन नमता नवनाथ	Gtk	12
8. गौरी हणमंत खराडे	Pratwanj	14
9. मुवाणी रेश्मा बाबासाहेब	Raj	12
10. पाटील निलम नैर्द्र	Josali	12
11. साळी वृषाली नारायण	Shani	10
12. पाटील श्वेता काकासी	Shantika	10
13. बनसोडे मेघा राजाराम	Harsh	10
14. जगताप हर्षदा दादासाहेब	Shane	12
15. माने प्रेरणा अनिल	K-v Nandale	13
16. नवाळे कविता वसंत	Shantika	12
17. साळी मयुरी प्रकाश	Rambhikam	12
18. कांबीकर अंजली प्रकाश.	SM	10
19. श्वेता महादेव बाबु	Ashraf	12
20. आश्लेषा चंद्रकांत पाटीले		

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
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Padmabhushan Dr. Vasanttraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक ०४ . ०८ . २०१९

वी.कॉम.भाग ३ मधिल सर्व विद्यार्थ्यांना कळविण्यात येते की, शनिवार दिनांक
१३ . ०८ . २०१९ रोजी Advanced Accountancy Paper I या विषयाची घटक चाचणी
(Unit Test) घेण्यात येणार आहे तरी सर्वांनी याची नोंद घ्यावी .


डॉ. ए. जी. सोनवले
HEAD
Department of Commerce
P.D.V.P.College, Tasgaon.


डॉ. मिलिंद एस. हुजरे
प्राचार्य,
पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगांव, जि. सांगली. (O.S.)

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon

Department of Commerce

Unit Test

Class: B.Com. III

Date: 13.08.2019

Subject: Advanced Accountancy – I

Total Marks: 24

.....

Q.1 Aniket purchased a motor car on hire purchase system from Sahyadri Motors on 1-04-2004 on paying Rs. 50,000 down. The balance amount (including 10% interest p.a.) was paid at the end of each year as under. 31-03-2005 Rs. 65,000, 31-03-2006 Rs. 60,000, 31-03-2007 Rs. 55,000. Aniket depreciated the car at 20% on WDV. Give Motor Car Account in the books Aniket. **8**

Q.2 Mr. Ravishankar purchased a machine on hire purchase basis from the Telco Machines Ltd. On 1st April 2019. The cash price of the machine was Rs. 100000. Mr. Ravishankar paid Rs. 20000 on signing the agreements and took the delivery of the machine. The balance amount was paid in four equal instalments of Rs. 24000 each at the end of the each year. Depreciation is charged @ 10% p.a. on the WDV basis.

Prepare Machine account & Telco Machine account in the books of Ravishankar. **8**

Q.3 On 1st April 2019, Asin Ltd. Purchased a machine on hire purchase system from Sunshine Ltd. The cash price of the machine was Rs. 74,500/- and the payment was to be made as follows.

Rs. 20,000/- was to be made on signing to agreement and the balance in three annual instalments of Rs. 20,000/- each at the end of each year. 5% interest is charged by Sunshine Ltd. Per annum. Asian Ltd. Was decides to write off 10% depreciation per annum on reducing value of the machine.

Prepare necessary ledger accounts in the books of Asian Ltd. **8**

"ज्ञान विज्ञान आणि सुसंस्कार यांचाच विकास" - विकासकर्ते डॉ. बापूजी काकुळे
 Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
 Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon
 Department of Commerce

Class: B.Com. III

Date: 13.08.2019

Subject: Advanced Accountancy - I

Total Marks: 24

Sr. No.	Name	Signature	Marks
1]	Suraj Pravin Darekar	<u>Suraj</u>	14
2]	Mulani Reshma Lalasaheb	<u>@LMulani</u>	22
3]	Salunkhe Namrata Navnath	<u>@Salunkhe</u>	18
4]	Raste Kajal Ramchandra	<u>KRST</u>	21
5]	Zambare Swati Mansing	<u>Swati</u>	20
6.	Patil Nilam Nivas.	N.N.Patil.	16
7.	Patil Sayali Subhash	<u>SPatil</u>	18
8.	Mali majuzi Prakash	<u>Mali</u>	12
9.	Velhal Sayali Ramchandra	<u>Sayali</u>	18
10	Sutar Haashada Suresh	<u>HSutar</u>	16
11.	mane Komal Dhondiram	K.D.MANE	14
12.	Chavan Manasi Krishnadev	<u>Manasi</u>	18
13.	Shinde Varshali Anandrao	<u>VSInde</u>	14
14.	Jadhav Rohini Kakasaheb	<u>RJadhav</u>	10
15.	Gaikwad Payal Vasant	<u>Payal</u>	14
16.	Jadhav Sneha Jagganath	<u>Sneha</u>	08
17.	Jadhav Swapnali Chandrakant	<u>Swadhav</u>	09
18	Grasavi Rupali Manik	<u>RGrasavi</u>	14
19	Kambhakar Anjali Prakash	<u>Rambhakar</u>	18
20.	Mulla Aayasha Babu	<u>Ayasha</u>	14
21.	Pawar Nikita Suryakant	<u>Nikita</u>	12
22.	Gaikwad Shivani Sampat	<u>Shivani</u>	10
23.	Chavan Suviata	<u>Shivani</u>	12


"ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार" - शिक्षणमहर्षी डॉ. वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasanturadada Patil Mahavidyalaya, Tasgaon
Department of Commerce

सूचना

दिनांक ०४.०८.२०१९

वी.कॉम.भाग १ मधील सर्व विद्यार्थ्यांना कळविण्यात येते की, शुक्रवार दिनांक ०६.०९.२०१९ रोजी विमाशास्त्र विषयांतर्गत 'जीवन विमा : फायदा की तोटा' या घटकावरती गट चर्चा (Group Discussion) होणार आहे तरी सर्वांनी याची नोंद घ्यावी.


डॉ. ए. जी. सोनवले
HEAD
Department of Commerce
P.D.V.P.College, Tasgaon.


डॉ. मिलिंद एस. हुजरे
प्राचार्य,
पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगांव, जि. सांगली. (O.S.)

निष्कर्ष :- जीवन विमा हा आवश्यकच आहे. कारण त्यामुळे लक्षात नियमित होते तसेच आर्थिक सुरक्षता देखील मिळते.

"ज्ञान विज्ञान आणि सुसंस्कार यांचाही विकासप्रकार" - शिक्षणमन्त्री डॉ. कापूरी मारुते
 Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
 Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon
 Department of Commerce

गट चर्चा

(Group Discussion)

विषय: विभाशास्त्र
 वर्ग: बी. कॉम. भाग १

घटका जीवन विभा: योग्य की अयोग्य
 दिनांक: 0९.0९.२०१९

गट अ

अ.क.	नाव	सही
1	वेभव सदाशिव पारील	...
2	साहील सुरज मुक्ती	...
3	प्रेम अजय न्यायनिग	...
4	अभिजित अशोक माने.	...
5	अजय सुधाकर पारील	...
6	अमरजीत मंडकार पेंकर	...
7	राहुल अशोक माकी	...
8	सुनील रामदास कुर्बि	...
9	पुत्रिक सुनील जाधव	...
10	पार्थिव अनिकेत अणिल	...
11	आकाश अंभाजी चेंडगे.	...
12	शुभम विक्रम डोळ	...
13	विरोध ओमनाथ वाड	...
14	अोकार विला माकी	...
15	महमद राजू मुलाषी	...
16	पद्माल अजय वारील	...
17	धनंजय राजेंद्र पवार	...
18	मुस्तकीम झाकीर पठाण	...
19	आरिफ आमीर बारगीर	...
२०		

Dr. A. G. Sonawale
 (Signature)

"ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार" - शिक्षणनहर्षो डॉ. वापूजी साळुंखे
 Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
 Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon
 Department of Commerce

गट चर्चा

(Group Discussion)

विषय: विमाशास्त्र

वर्ग : बी. कॉम. भाग १

घटक: जीवन विमा : अयोग्य

दिनांक : 0६.0९.२0१९

गट व

अ.क.	नाव	सही
1	Khot Dipali Dilip	Dipali
2	Patil Shital Tanaji	SPatil
3	Daund Paiti Dilip	PDD
4]	CHAVAN SWATI POPAT	Swati
5)	Jagtap Amruta Sanjay	A.S.J
6)	Pol Aishwarya Shankar	Pol
7)	Jadhav Shilpa Hanmant	shilpa
8)	Mulani Tabbasum Munir	T.M.Mulani
9]	Mulani Reisa Sikandar	R.S.Mulani
10]	Mandale Jaysbri Dilip	J.D.Mandale
11]	Autade Madhuri Bharat	Autade
12]	Kumbhar Revati Mahadev	Rakumbhar
13]	Patil Tejaswree Popat	T.P.Patil
14)	Babar Rutuja Sanjay	R.S.Babar
15)	Gadbale Sanjivani Santosh	Sanjivani
16)	Pawar Manika Sanjay	M.S.P.
17]	patil Nikita Baban	NPatil
18]	Attar Tasneem Ilahi	TAttar
19)	Kumbhar Priyanka Ramchandra	Rakumbhar
20]	Bhasale Nikita Rajaram	NBhasale
21]	pawar shradha kumar	shradha
22]	more shital shankar	Smore
23)	Kamble Truphi Dipak	TKamble

Dr. A. G. Sonawale
 (In-charge)


- 24) भांडी रामना वाळणो Bhawali
- 25) भांडी शिवानी बापू Bhawali
- 26) भांडीव भांडीव विजोय Bhawali
- 27) वाळणो मेळणो सुभाष W.M.S
- 28) विजोयानी शिवानी शिवनीडे Vihoade
- 29) चव्हाण सुभाषा महादेव Chavan
- 30) शैला विश्वंकर वाहील Shaila
- 31) परेगी लक्ष्मी वित्तायक N.V. Gore
- 32) पाटील श्रुतिका अमनदेराव Patil
- 33) चौहदे अरुदा मिनिंद Chouhade
- 34) पाटील अनुशया यशवंत Patil
- 35) भांडी काळल लुळा Bhawali
- 36) मोरे शाळ्यश्री सुधाकर More
- 37) Kharode Nikita Ghondiram Nikita
- 38) Ghotkat Shivani Shivaji Ghotkat
- 39) शिंदे निकीता लालामो Shinde
- 40) देवकुळे लक्ष्मी युगवि Devkule
- 41) खरभाटे स्वप्नाली शक्तिश Kharmate
- 42) माने अंकिता जेतीराम Mane
- 43) सातपुते श्रुतिका संजय Satpute
- 44) रास्ते रेखा रामचंद्र Raste
- 45) माईनकर अर्चना अनिल Mane
- 46) फाळके प्रतिष्ठा विजोय Falake
- 47) वाडकर मोनाली लक्ष्मी Wadkar
- 48) पाटील समृद्धी लक्ष्मी Patil
- 49) माने पूजा लालामो Mane
- 50) व्याघ्रनिर्गुणे भारती अर्चना Vyaghre
- 51) पटेल शानिया सिखंदर Patel
- 52) कांबळे आरुदा शरिकांत Kamble

“ज्ञान विज्ञान आणि सुसंस्कार चांसाटी शिक्षणप्रकार” - शिक्षणनहयो डॉ. वापूजी साकुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasanttraodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

नोटीस

दिनांक १९.०८.२०१९

वाणिज्य विभागातील वी. कॉम. भाग १ मधील सर्व विद्यार्थ्यांना कळविण्यात येते की, गुरुवार
दिनांक २२.०८.२०१९ रोजी विमाशास्त्र या विषयाची घटक चाचणी (Unit Test) घेण्यात येणार
आहे याची सर्वांनी नोंद घ्यावी.


प्रद्युम्न,
पदाभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगांव, जि. सांगली. (०९)


(डॉ. ए. जी. सोनवले)
HEAD
Department of Commerce
P.D.V.P. College, Tasgaon.

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणमहर्षी डॉ. वापूजी माळुंबे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon
Department of Commerce

प्रश्न क.१ विमा म्हणजे काय ते सांगून वाणिज्यच्या दृष्टिकोनातून विम्याचे महत्त्व स्पष्ट करा. १०

प्रश्न क.२ विम्याची प्राथमिक मूलतत्वे स्पष्ट.

१०

①

“ज्ञान विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” - शिक्षणनहर्षी डॉ. वापूजी साळुंवे
 Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
 Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya, Tasgaon
 Department of Commerce

Class: B.Com. I

Date: 22.08.2019

Subject: Insurance

Total Marks: 20

Sr. No.	Name	Signature	Marks
1]	Sut+raJec+chundrakunh+ ^{Petkar}	S.C. Petkar	12
2]	Rahul Ashok mali	RAM	08
3]	Abhijeet Ashok manre	Amr.	07
4]	Aniket Bhanudas Gulij	A.B. Gulij	02
5]	Prern Ajay Nyanit.	Prern	09
6]	mulla sahil suraj	S.S. Mullig.	08
7]	Sutar Nikhil Suresh	Devar	14.
8]	surve kunal sukhadev	surve	17
9]	Shubham dipak Koli	Shubham	02
10]	Abhishek vittal Sawant	AU. Sawant	18
11)	Akash sambhaji chendage	Chendage	09
12)	Adake Shital Mohan	Adake	07
13)	Patil Aniket Anil	Patil	05
14)	Jadhav Pratik Surah.	Jadhav	10
15]	Patil Amit shumyenda.	APUL.	12
16]	Hingnire Rushikesh Somnath	R.S. Hingnire.	14
17)	Padwar saagar Rajendra	SA Padwar	10

18) Joglekar Aniruddha Dilip

Pratik Surah

Sr. No.	Name	Signature	Marks
1.	विजिता सुनिल पाटील	<u>VBH</u>	17
2.	तेजस्विनी संभाजी साळुंखे	T.S.Salunkhe	10
3	दिपाली दिलीप खोत	<u>Dhat</u>	12
4)	शीतल तानाजी पाटील	<u>Shital</u>	14
5)	अंकिता गजानन फडने	<u>AFD</u>	09
6)	प्रतुजा संजय वावर.	<u>Pratibha</u>	09
7)	मोनिषा संजय पावर	M.S.P.	07
8)	प्रियांका प्रताप जाधव	<u>P.P.J</u>	11
9)	अंकिता अशोक कोरे	<u>A.K.</u>	14
10]	निकीता लज्जत पाटील	<u>N.Patil</u>	12
11]	प्रतुजा संजय सातपुते	<u>Pratibha</u>	14
12)	अमृता संजय जगताप	A.G.J.	14
13]	शिंदे निकीता बाळामो	<u>Shinde</u>	16
14]	देवकुळे अक्षता सुशीव	<u>Raj</u>	12
15]	वाद्यमाते मेघारानी सुवाम	M.S. Vaidya	11
16)	माळी शिवाणी बापू	<u>Malvi</u>	09
17)	जाधव अक्षता दिलीप	<u>Adhav</u>	12
18)	परदेशी संस्कृती नारायण	<u>Paradeshji</u>	07
19]	पेंवतान नेहा अनिल.	<u>Penwatan</u>	10
20]	शाश्वती सुवशाज चौले	<u>Shashwati</u>	08
21)	मिज्बा आरशिद जमादार	<u>Mizba</u>	08
22]	श्वेता अशोक गायकवाड	<u>Shweta</u>	09
23]	शुभांगी सोपान गायकवाड	<u>Shubhangi</u>	07
24)	मधुरा निवास गुरव	<u>Madhura</u>	12
25]	रेवती महादेव कुंभार	<u>Ravati</u>	14
26)	सादनी जितेंद्र अमृतसागर	S.J. Amrutsagar	07
27]	प्रतिक्षा विनोद फाळके	<u>Pratiksha</u>	09
28]	भारती अर्जुन व्यासनिगुणे	<u>Bharitya</u>	10
29]	पूजा लालाराम माने	<u>Pooja</u>	11
30)	शितल संभाजी पाटील	<u>Shital</u>	07
31	अंताडे माधुरी भरत	<u>Antade</u>	08
32	जाधव पल्लवी रामदास	<u>Jadhav</u>	08
33	माने पुनम लज्जत	<u>Mane</u>	04

Sr. No.	Name	Signature	Marks
1	मीनाली लक्ष्मण वाडकर		
2	कुरिना जतांगीर मोरे	M.L.Wadkar	09
3	आकाशा उत्तम आवरे	Akamee .	08
4	धर्षदा बडू कराडे	Dh.	08
5	नसनीम इलाही आतार	Nasim	12
6	समृद्धी बाबुराव पाटील	S.Patil	12
7	अदथा कुमार पवार	A.D.K.	14
8	निकिता राजाराम मोसले	N.Mosale	13
9	आरुधा शारिकान कांबळे	A.R.	12
10	पद्मवी चंद्रकांत गाडे	P.Gade	10
11	अंकिता जेतीराम मोने	A.M.	08
12	प्रातिक्षा माणिक पाटील	P.Patil	12
13	रईसा सिकंदर मुलानी	R.Mulani .	16
14	मुभाणी इलखसुम मुनीर	T.m.mulani	14
15	मंडने जयश्री दिनीप	Mandale .	18
16	मुजावर अविशा आशुब	Mujawar	10
17	वायसी सुरेश कांबळे	V.S.K.	12
18	पुशवर्षा रांकर पोळ	P.P.	14
19	सोहिते त्रधुजा नंदकुमार	S.Hite	08
20	माळी काजल कृष्णा	M.M.	08
21	माळी निशा श्रीकांत	N.S.Mali	07
22	पाटील अनुराधा चरापंत	P.Patil	06
23	मोरे भाव्यश्री सुधाकर	B.Moore	04
24	माईनकर अर्चना अमोल	M.M.	09
25	पवार पूजा शहाजी	P.P.	11
26	वंडार शितल परशुराम	V.P.	10
26	काठ त्रधुजा संतोष	K.K.	07
25	जाधव रवना दत्तात्रय	J.J.	04
26	पाटील अंकित रामचंद्र	P.P.	08
27	लाखंड पुजाली दगडू	L.L.	09
28			

प्र. 1 आधुनिक व्यवस्थापनाची व्याख्या -

- ① मनमोहन प्रसाद - या लेखकाने व्यवस्थापन नामाचा हवाला देऊन असे मत मांडले आहे की, आधुनिक व्यवस्थापन हे अभिजात, आदि, नवअभिजात दुपटीकोनामध्ये भार घालून त्याचे संशोधन करणे म्हणजे व्यवस्थापन प्रगत झाले असून सन 1950 नंतर ते अस्तित्वात झाले.

प्रस्तावना -

औद्योगिक कंपनीपूर्वी काळात म्हणजेच इ.स 1750 पूर्वी व्यवस्थापनाची संकल्पना रचली असली तरी व्यवस्थापनाचे अर्थिक व व्यवसायिक दृष्ट्या महत्व फार कमी होते. वाफेच्या, इंजिनचा शोध लागल्यामुळे इंग्लंड मध्ये औद्योगिक क्रांतीचा प्रारंभ झाला. औद्योगिक क्रांतीनंतर उद्योग वाढीच्या क्षेत्रात आमुलाग्र बदल झाले. मोठ्या प्रमाणावर उत्पादनास प्रारंभ झाला. औद्योगिक आंदोलन, श्रम, संघर्ष इ. उत्पादन साधनामध्ये वाढ करणे हा निवार्थ झाले. त्यामुळे आधुनिक व्यवस्थापनाचे महत्व वाढा गेले ते पुढीलप्रमाणे.

अ) आधुनिक व्यवस्थापनाचे अर्थिक महत्व

व्यवसायाचे अविमूक्त अस्तित्त्व व विकास या दृष्टीकोनातून महत्व पुढीलप्रमाणे -

- ① परिवर्तनाची आव्हाने पेलणे -

जागतिकीकरणामुळे व्यापार उद्योग क्षेत्रामध्ये अनेक नवी आव्हाने निर्माण झाली आहेत. प्रगत विज्ञान व यंत्रज्ञानामुळे व्यवसायक्षेत्रामध्ये अनेक प्रकारचे बदल व परिवर्तन झाली आहेत. अर्थव्यवस्थेचे स्वरूप अधिकधिक खुले होत आहे. या सर्व परिवर्तनाचे आव्हाने पेलण्यासाठी कुशल व धंदेवादी व्यवस्थापनाची गरज भासत आहे. त्यांच्याकडे कुशल धंदेवादी व अनुभवी आधुनिक व्यवस्थापन आहे. त्यांचे अस्तित्त्व मान्य ठरवून आहे.

2) उपलब्ध साधनाचा योग्य व महत्तम उपयोग कोणाच्याही व्यवस्थापनासाठी सुभी, श्रम, सांडवल व संघटन या सर्व साधनांची जोडणी करावी लागते, व साधनाची जोडणी करून चालवून नाही. जर या साधनाचा काढकसरीन व योग्य उपयोग करणवा लागतो. व्यवस्थापनाची उद्दिष्टे. सामग्रीचा उपयोग वापर करणवाचे कोणत्या व्यवस्थापनाकडे असवे.

3) शास्त्रीय शोधाचा व नवीन मंत्रांचा लाभ घेणे व्यवस्थापनाची उद्दिष्टे पूर्ण करणवासाठी कल्पक योजना आखाण्या लागतात. या योजना पूर्ण करणवासाठी नवीन मंत्रांचा विचार आखा करणवा लागतो. नवीन मंत्रांचा वापर करणवा व कुठे करणवा याबाबत निश्चय व्यवस्थापन घेते.

4) बाजारपेठेतील ठाळेकापू व तीव्र स्पर्धा औद्योगिक क्रांतीपूर्व उत्पादन : भारताच्या पुर्णिकरनेच अद्युचित हेतू त्यामुळे बाजारपेठेत कोणातीही स्पर्धा नव्हती. औद्योगिक क्रांतीनंतर उत्पादनात मोठ्या प्रमाणात वाढ आली. बाजारपेठेचा विस्तार राष्ट्रीय व आंतरराष्ट्रीय बाजारीपर्यंत आला.

5) श्रमसमस्याची समाधानकारक सोडवणूक श्रमसमस्या हाताळणे यात आधुनिक व्यवस्थापनाची अरी कसोटी लागते आजचा कामगार किंवा कर्मचारी हा आक्रमक कळकळीने होते होते.

6) गतिमान नेतृत्व करणवाची गरज आधुनिक व्यवस्था

पुन हे आपल्या संघटनेला नेतृत्व देण्याचे कार्य करणे. कर्मचाऱ्यांचा कृतीशील प्रतिसाद व सहकार्य मिळवण्यासाठी कार्यक्षम नेतृत्वाची आवश्यकता असते.

7) उद्देशाची परिणामकारक पूर्तता

आधुनिक व्यवस्थापनाच्या सर्व कार्याचा शोध हा व्यवसायाच्या सर्व उद्देशांचा परिणामकारक पूर्ततेकडील असतो. व्यवसायाच्या शोषणाची कार्यक्षम अंमलबजावणी करून घेतली. मरच उद्देशाची पूर्तता होऊ शकते.

(ब) राष्ट्रच्या हृदयीने अर्थिक व्यवस्थापनाचे महत्त्व

1) राष्ट्रीय साधनसंपत्ती शोध उपयोग व विकास -

कुशल
य कल्पक आधुनिक व्यवस्थापनामुळे राष्ट्राच्या नवनव्या साधनसंपत्तीचा शोध होऊन त्याचा राष्ट्रविकासासाठी उपयोग करता येतो. भारतामध्ये विभिन्न प्रकारची भूगर्भ साधनसंपत्ती असल्याचे सांगितले जाते. प्रगत आधुनिक व्यवस्थापन तंत्राद्वारे त्या साधनसंपत्तीचा शोध घेते.

2) नियोजनाचे यश -

भारतामध्ये अर्थिक नियोजनाचा अंमल लक्ष केला जातो आहे. पंचवार्षिक योजना आखून वेगवेगळ्या क्षेत्रांच्या विकासाच्या योजना आखल्या जातात. या पंचवार्षिक योजनांचे यशस्विलपणे कुशल प्रसन्न प्रगत व्यवस्थापनावर अवलंबून असते.

3) औद्योगिक विकास

देशातील सर्व प्रतारच्या उद्योगधंद्यांचा विकास जर आधुनिक व्यवस्थापनाशी प्रत्यक्ष पथे निगडित आहे. उद्योग स्थापना करणे

5) नियति वाढ -

विदेशी व्यापारामध्ये नियति वाढवण्यासाठी व्यावसायिक कौशल्य व संपैदासकपणा कार महत्वाचा आहे. कोणत्याही देशाच्या वृद्धीने व्यापार महत्वाचा आहे

गुणवत्ता वाढवा
गुणवत्ता

ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार.
शिक्षण महर्षी- डॉ बापुजी साळुंखे

College Name :

Padmabhushan Dr. Vasantraodada Patil
Mahavidyalaya Tasgaon.

Subject Name :

Advanced Accountancy Paper - II.

[Auditing]

Seminar Subject Name :

Meaning of Audit And Explain

// Explain principles of Auditing.

Class : B - Com - III.

Roll No : 4764

Student Name : Pradnya Ghanashyam Kamble.

11/10/2019

Meaning of Audit.

Definition:

"Audit means a systematic and critical examination of accounts and records so as to satisfy the auditor that financial statements are well in order and they reflect a true and correct view of operating results as well as financial position of the business / profession."

Thus audit involves indepth examination of books of accounts and records to judge the truth and correctness of operating results and financial position of business.

That is why, audits a specialized job. In amount all countries in the world, separate institution has been well established to educate the persons for a separate institution viz, The Institute of chartered Accountants of India was established on 1st July 1949 under the act of Parliament. A student who cleared the final Examination of this Institute becomes a chartered Accountant. Thus, for the work of audit, the chartered Accountants are eligible.

Principle of Audit.

In the business world, Audit has become the need of time. It has been gaining tremendous importance because Audit helps to enjoy a no. of benefits.

"Audit refers to a systematic and critical examination of books of accounts and records, so as to satisfy the Auditors that financial statements are well in order and they reflect a true and correct view of operating results and financial position of business."

In order to realize the basic objectives of audit, the Audit Statement - 1 gives following 9 basic principles governing audit.

1) Integrity, Objectivity and Independence :

Auditor should be honest, sincere and unbiased. He should be independent while conducting Audit.

2) Confidentiality :

Auditor gets crystal clear information of business of client. It is must for the auditor to keep such information in confident from others.

3) Skill and Competence :

Audit is specialized job. That is why, for purpose of audit, C.A. Qualification is a must. Even ICAI conducts seminars / Workshops / Conferences etc. to update the knowledge of the chartered Accountants.

4) Work performed by other :

In the process of audit, the Auditor may take help from assistants, articles, experts, and even other C.A. However the ultimate responsibility lies with

the Auditor. It is also necessary to clearly state the help from the others from Audit Report.

5) Documentation:

Auditor has to keep important documents which provide evidence that audit is carried out well in order for example:

- a) Working Papers.
- b) Audit Notebooks

6) Planning:

Auditor is required to chalk out plan in respect of every audit work. Audit becomes effective, meaningful and fruitful if it is well planned. Here, audit program plays a very significant role in success of the audit.

7) Audit Evidence:

Audit basically depends on evidence. Auditor tries to find out whether there is sufficient and reliable evidence of accounting entries recorded in the books of Accounts Auditor, use techniques like inspection, observation, enquiry, confirmation, vouching, verification, etc. in the process of audit.

8) Accounting system and Internal Control:

Auditor examines the accounting system followed by the client so as to evaluate its appropriateness. Moreover, he judges the internal control system. The technique to be followed for Audit depends to a great extent upon the quality of accounting system, internal check system and internal control.

9) Audit Conclusion and Reporting:

On completion of audit process, Auditor expresses his opinion to confirm that:

Good

निव. मंजुरी बाळासो वांडरे

इयत्ता - ७.०००० - ५५५

दिनांक - ३१-०८-२०१९

शेळ नं -

प्र १ सहकार म्हणजे काय ? सांगुन सहकाराची वैशिष्ट्ये सांगा.

→ प्रस्तावना -

सहकार म्हणजे सहकारी तत्वावर व्यवसाय करणारी व्यापारी संघटना किंवा सहकारी व्यवसाय संघटना होय. सहकारी संस्र तत्त्वे म्हणजे सहकारी संस्थेतील कार्य आणि जिवनासंबंधी सांगितलेले काही नियम होय. सर्वप्रथम इंग्लंडमध्ये सहकारी तत्वावर व्यापार करणारी संस्था स्थापना करण्यात आली. १८४४ मध्ये इंग्लंडमधील शॅडेल येथे २५ विनकर कामगारांनी एकत्र येऊन शॅडेल सहकारी सांठारची स्थापना केली. भारतामध्ये सहकारी संस्था स्थापन येवयाची सुमारे १०० वर्षे कालावधी लागली या संस्था सहकारी तत्वावर कार्य करतात. या संस्थांमध्ये अल्प उत्पन्न असणाऱ्या एकत्रित येऊन ही संस्था सुरु करतात. व जीवनावश्यक वस्तु स्वस्त व कमीदरात उपलब्ध करून देतात.

व्याख्या -

① आर्थिकदृष्ट्या दुर्बल घटकातील व्यक्ती स्वइच्छेने एकत्र येऊन संस्था स्थापन करतात त्यास सहकारी संस्था असे म्हणतात.

② स्वइच्छेने आणि समानतेच्या भावनेने व्यक्तींनी एकत्र येऊन केलेली संस्था स्थापन म्हणजे सहकारी संस्था होय.

सहकारी संस्थेची वैशिष्ट्ये -

सहकारी संस्थेची वैशिष्ट्ये पुढीलप्रमाणे सांगता येतील.

① सहकार म्हणजे व्यक्तिये संघटन -

सहकार म्हणजे व्यक्तिये संघटन किंवा अनेक विविध प्रकारच्या व्यक्तिये एकत्रिकरण आहे. आर्थिकदृष्ट्या दुर्बल व्यक्ति स्वतःच्या उन्नतीसाठी सहकारी संस्थेमार्फत एकत्रित येतात. व त्यांच्या आर्थिक विकास मदत करून आर्थिक विकास करूयाचा प्रयत्न करतात.

② सरकारकडे अक्तीयने नोंदणी -

वेगवेगळ्या छुट्ट्यातील कमीत कमी 10 व्यक्ती एकत्रित येऊन सहकारी संस्था स्थापन करतात असे अपेक्षित असते. त्या व्यक्तियेनी आपल्या सहकारी संस्थेची नोंदणी सरकारकडे करणे आवश्यक असते.

③ स्वतंत्र कार्यक्षेत्र अस्तित्व -

सहकारी संस्थेला अन्य कंपनीप्रमाणे स्वतंत्र अस्तित्व असते. तिला स्वतःचे बौध्दिकत्व असते व तिच्या उपबोधा सहीप्रमाणे संस्थेमध्ये केला जातो.

④ सहकार म्हणजे सहकारी तत्वावर कार्य करणारी संस्था -

सहकार म्हणजे सहकारी तत्वावर कार्य करणारी संस्था. त्या सहकारी संस्थेने इतर व्यवसाय संस्थेप्रमाणे कार्य करायची अपेक्षा असते. सहकारी संस्था सहकारी तत्वावर कर्णावस्था कार्य

करीत असतात.

७) सहकार म्हणजे ठेविलेला संघटना -

समाजातील अर्थिकदृष्ट्या दुर्बल व्यक्ति एकत्र येऊन सहकारी संस्था स्थापन करित असतात. या संस्थांमध्ये सर्व व्यक्ती आपल्या मज्जेने स्वयंचेने समासद होतात. व जेव्हा त्यांना वारिले तेव्हा ते सहकारी संस्था सोडून जाऊ शकतात. त्यांच्यावर कोणत्याही प्रकारचे नियंत्रण संस्थेचे असते.

८) नफ्यापेक्षा आढळतावर भर -

सहकारी संस्था ही नोकशाही नत्वा वर कार्य करित असतात. सर्व समासदांना समान अधिकार दिलेला असतो. या संस्था समासदांना नफ्यापेक्षा समासदांच्या अर्थिक हितान्या सर्वप्रथम विचार करतात.

९) समान हित -

आपल्या समासदांचे समान हित करणे हे सहकारी संस्थांचे प्रमुख वैशिष्ट्य आहे. उदा. कर्जपुरवठा करणे.

१०) समान गरजा असणाऱ्यांची संघटना -

सहकारी संस्थेमध्ये समान गरजा असणाऱ्या सर्व व्यक्ति एकत्रित येऊन ही संस्था स्थापन करतात. अशा व्यक्ति सहकारी संस्थेचे समासद होतात.

११) सहकारी संस्थेतील समता -

सहकारी संस्थेतील सर्व समासद

समान आतेत. सह संस्थेमध्ये सर्व या तत्वावर कार्य करतात.

⑩ सामाजिक जबाबदारी -

सहकारी संस्थेमध्ये सर्व समासदांची सामाजिक जबाबदारी समान असते व सर्व समासद सामाजिक दृष्ट्या कार्य करतात. सामाजिक जबाबदारी असणारे व्यक्ति समासद असतात. अशाप्रकारे सहकाराची वैशिष्ट्ये आहेत.

Siddha

From ~~11~~
 Lot: 32.43
 NAME: NARAI VISHWANATH DHANISHA

Books of Laxmi Bank Ltd

Particulars	S.No.	Rs. P.S.	Rs. P.S.
(A) Capital and liabilities			
Shares Capital	1	50,00,000	
Reserve and surplus	2	26,37,000	
Depositors	3	29,50,000	
Borrowing	4	Nil	
Other liabilities and provision	5	Nil	
Total (Rs.)			
(B) Assets and Properties			
Cash in hand with other	6	2,73,000	
Bank	7	Nil	
Balance with other banks	8	1,50,000	
Money at call and short notices	9	33,95,000	
Government	10	4,50,000	
Advances	11	Nil	
Fixed Assets			
Other Assets			
Total (Rs.)			
(C) Provisions contingencies (liabilities)	12		
Bills for collection			
Total (Rs.)			

Profit and Loss Account

	S. No	Rs	Rs
(I) Income			
Interest earned	13	12,42,000	
Other Income	14	2,50,000	

TOTAL (RS) 14,92,000

(II) Expenditure			
Interest expended	15	2,20,000	
Operating expenses	16	10,75,000	
Provision and contingencies		50,000	

TOTAL (RS) 13,45,000

(III) Profit and Loss A/c			
Open Net Profit during the Year (I-II)		1,47,000	
Opening Net Profit		160,000	

TOTAL (RS) 3,09,000

(IV) Appropriation			
Transfer to Statutory Reserve (20% Net Profit)		29,400	
Transfer to dividend		1,70,000	
Transfer to Balance sheet Reserve and Salaries		1,09,600	

TOTAL (RS) 3,09,000

details of schedule

Schedule No-1

Share Capital

Particulars		RS.
equity share capital of Rs: 100 each	Rs: 50000 shares	50,00,000
TOTAL (RS)		<u>50,00,000</u>

Schedule No-2

Reserve and surplus

Particulars		RS
Statutory Reserve	25,00,000	25,00,000
20% Net Profit	29,400	29,400
Balance c/d Reserve and surplus		1,07,600

Add -

		<u>26,37,000</u>
--	--	------------------

Schedule No-3

deposits

Particulars		RS.
current deposits	10,00,000	10,00,000
fixed deposits	12,50,000	12,50,000
saving deposits	50,000	50,000
recurrent deposits	2,00,000	2,00,000
TOTAL (RS)		<u>29,50,000</u>

Schedule No-4
Borrowings

Particulars		RS
Nil		Nil
TOTAL (RS)		<u>Nil</u>

Schedule No-5
Other liabilities and provision

Particulars		RS
TOTAL (RS)		

Schedule No-6

Particulars		RS
Cash in hand and with balance with R.B.F.		2,73,000
TOTAL (RS)		<u>2,73,000</u>

Schedule No-7
Balance with other banks and money at call short notice

Particulars		RS
Nil		Nil
TOTAL (RS)		<u>Nil</u>

Schedule No - 8

Investment

Particulars	RS.
Investment	4,50,000

TOTAL (RS)	<u>4,50,000</u>
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Schedule No-9

Advances

Particulars	RS.
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Long cash credit and overdrafts and overdrafts	28,50,000
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Bad debts	50,000	28,00,000
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Bills purchased and discount		5,45,000
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TOTAL (RS)	<u>33,45,000</u>
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Schedule No-10

Fixed Assets

Particulars	RS.
-------------	-----

Premises	5,00,000	5,75,000
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S.I. on dep'n	<u>2,50,000</u>	47,50,000
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TOTAL (RS)	<u>47,50,000</u>
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Schedule No-11

Other Assets

Particulars	RS
-------------	----

Nil	Nil
-----	-----

TOTAL (RS)	<u>Nil</u>
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Schedule No-12
Contingencies - liabilities and
Bills for collection

Particulars

Nil

TOTAL (₹)

₹

Nil

Nil

Schedule No-13
Interest earned

Particulars

Interest and discount

on unpaid discount

12,80,000

88,000

₹.

12,42,000

TOTAL (₹)

12,42,000

Schedule No-14
Other Income.

Particulars

commission and exchange

₹.

2,50,000

TOTAL (₹)

2,50,000

Schedule No-15
Interest expended

Particulars

Interest and deposits

₹.

2,20,000

TOTAL (₹)

2,20,000

Schedule No-10
 Operating Expenses

Particulars	RS
Salaries	2,18,000
Director's fees	2,71,000
Other operating expenses	2,74,000
Dep'n	2,50,000
TOTAL (RS)	10,75,000

prepare format in
 Accounting form

(Signature)

Name :- Revati Dattatry Jadhav

Q. 1 Ans. विमा म्हणजे काय ?
 एकच प्रकारच्या शोक्याची शक्यता असणाऱ्या अनेक व्यक्तींकडून विमा हातल्या स्वभावात जमा केल्या जाणाऱ्या निधीकडून ज्या व्यक्तीचे नुकसान होईल त्याची भरपाई करवणेच एक सामाजिक व्यवसाय म्हणजे विमा होय.

विमा ही एक अशी व्यवस्था आहे की ज्यामध्ये विमदाराला सोसाव्या लागणाऱ्या आर्थिक नुकसानीची भरपाई विशिष्ट मोबदल्यात करवणेच विमकरी केली.

Q. 2 विम्याचे वाणिज्याच्या दृष्टीकोनातून महत्त्व स्पष्ट करा !

Q. 3 विम्याची मूलतत्वे लिहा.

- 1) विमाहित
- 2) नुकसान भरपाई तत्व
- 3) पस्पट विश्वास तत्व
- 4) सहकार्येचे तत्व
- 5) संभाव्यतेचे तत्व

विमाहित : विमदाराला भविष्यात निर्माण होणाऱ्या शोक्यातून होणारे नुकसान भरपाई करून देवणेच संरक्षण दिले जाते. परंतु असे संरक्षण देत असताना विमाहितचा विचार केला जातो. विमाहित म्हणजे विमदाराने विविध वस्तूत आर्थिक संबंध असणे होय. आर्थिक हितसंबंध नसतील तर विमा संरक्षण संरक्षण घेता येत नाही. हे तत्व सर्व विमा प्रकाराला लावू पडेल.

1) विमाहित हे भौतिक अस्तित्त्व असणाऱ्या वस्तू व व्यक्तींमध्ये असावे पाहिजे.

2) विमातेत वस्तूला भविष्यकाळात नुकसान, विम पोहचवण्याची शक्यता असावी लागते.

3) विमातेत वस्तू व विमदार यांच्यात कथ्येने आर्थिक हितसंबंध

असावे .

उदा. पती - पत्नी, भागीदार - मालक, घनको - त्रुणको इ.

२) नुकसान भरपाईचे तत्व :- विमा कंपनी व विमोदारात लिबिल स्वतंत्रता झालेल्या करारात विमा असू शकते. या करारानुसार विमोदाराचे झालेले नुकसान भरपाई केल्याचे मान्य केले जाते. त्याच विम्याचे भाषेत नुकसान भरपाई म्हणले जाते. या तत्वानुसार विमीत मालमत्तेचे जेवढे नुकसान झाले आहे तेवढीच भरपाई केली जाते. कोणत्याही परिस्थितीत नुकसानीची रक्कम दिली जात नाही. त्यामुळे हे तत्व आर्थिकविश्वास जास्त कायदा होऊ नये यासाठी आहे. त्यासाठी विमा कंपनीकडून विशिष्ट नुकसान भरपाईच्या पध्दतीच्या विमोदाराचे जेवढे आर्थिक नुकसान झाले असते.

३) वस्तूची दुहस्ती :- यापद्धतीमध्ये विशेषतः मोटार, स्कूटर, इ. वाहनांचा समावेश केला जातो. यामध्ये नुकसान भरपाई करीत असतात. विविध वस्तू पूर्ण पदावृत्त येण्यासाठी ती दुहस्त करून दिली जाते. दुहस्तीसाठी थोडो सर्व खर्च विमा कंपनी करते.
क) वस्तू बदलून देणे :- या पद्धतीमध्ये विमोदाराच्या मागणी केल्यास विमा कंपनीच्या नियमानुसार वस्तूचा बदलामध्ये पसून देवत येते.

वरील तीन पद्धतीमध्ये कोणत्यातरी एक पद्धतीत विमा निवाड करता येते. कमीत - कमी आर्थिक असवारी पद्धती निवडल्या येते.

४) पसपर विश्वास तत्व :- विमोदार व विमा कंपनी विमा करार करीत असतात. कागदापत्रे व दिलेली माहिती पसपर विश्वासावरती स्वीकारली जाते. त्यामुळे दोन्ही घटकांत दिलेली माहिती सत्य व अचूक दिली जाते. कारण करार मान्यता द्यायची नाही. पसपर विश्वासाच्या तत्वावर जास्त जास्त जबाबदारी ही विमोदारावती सोपवली जाते. त्यामुळे विमा संरक्षण घेत असताना वास्तव स्वतंत्रता असते.

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PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASAGAON
DEPARTMENT OF CHEMISTRY
(Academic Year 2019-20)

Notice

Date: - 27/08/2019

All the students of B. Sc. I are here by informed that, they should submit their Home assignment on or before 30/08/2019. Questions for Home assignments are as follows:-

A] Inorganic Chemistry

Q. 1. Define Ionic bond & explain in detail energetic of ionic bonding

Q.2 Discuss in details born haber cycle.

B] Organic Chemistry

Q.1. Discuss the mechanism of sulphonation & nitration of benzene.

Q.2 What are aromatic compound & non aromatic compound & explain it with example.

Q.3 Explain molecular orbital picture of benzene.



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DEPARTMENT OF CHEMISTRY
P.D.V.P. COLLEGE, TASAGAON

18/25

Name - Mohite Pranali Adhikrao
Roll No - 5007
Date - 30-08-19

Q.1 Define Ionic bond & explain in detail Energetics of Ionic bonding.

i) Ionic bond - A chemical bond formed by transfer of certain number of electrons from valence shell of [electropositive] one atom to valence shell of another atom [electronegative] so both acquire stable noble gas configuration.

OR

It is the interionic electrostatic force of attraction developed due to transfer of electrons from one atom to another.

OR

It is the electrostatic force of attraction betⁿ oppositely charged ions. It is called ionic bond.

ii) Energetics of Ionic bonding formation -

The ionic bond depends upon -

- 1) The ease of formation of cation from electropositive ion atom i.e. ionisation potential^{or} energy of atom of element.
- 2) The ease of formation of an anion from electronegative atoms i.e. electron affinity of an atom of element.
- 3) The ease of combination of appropriate number of oppositely charged ions i.e. lattice energy of the resulting ionic solid.

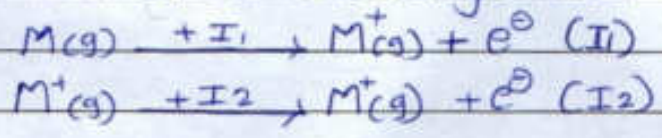
Name - chavan Pooja P.
Roll No - 5160
subject - chemistry
Date - 30-8-2019

i] Define ionic bonding and explain in detail energetic of ionic bonding.

→ Ionic bond - A chemical bond formed by transfer of certain number of electrons from valence shell of one atom to valence shell of another atom so both acquire stable noble gas configuration or it is the interionic electrostatic force of attraction developed due to transfer of electrons from one atom to another or it is the electrostatic force of attraction betⁿ oppositely charged ions. it is called ionic bond

Energetics of ionic bond - i] The ease of formation of cation from electropositive atom. i.e ionisation potential or energy of an atom of element.
2] The ease of formation of an atom of element anion from electronegative atom.
i.e electron affinity of an atom of element
3] The ease of combination of appropriate no. of oppositely charged ions. i.e lattice energy of the resulting ionic solid.

i] Ionisation Potential (I) - The amount of energy required to remove an electron completely from an isolated gaseous atom is known as ionisation potential it is always +ve.

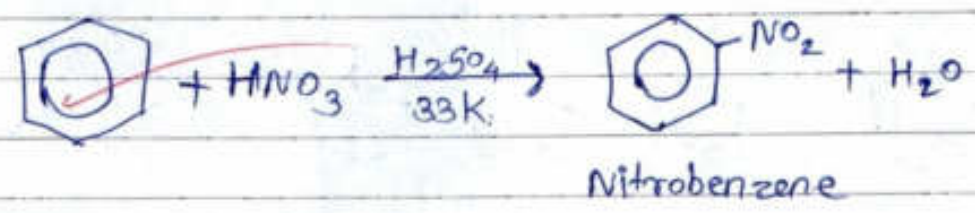


Q1 Discuss the mechanism of Nitration & sulphonation of benzen.



Nitration:

The substitution of hydrogen by Nitro group is called Nitration. In presence of concentrated sulphuric acid, benzene reacts with nitric acid to form nitrobenzene.



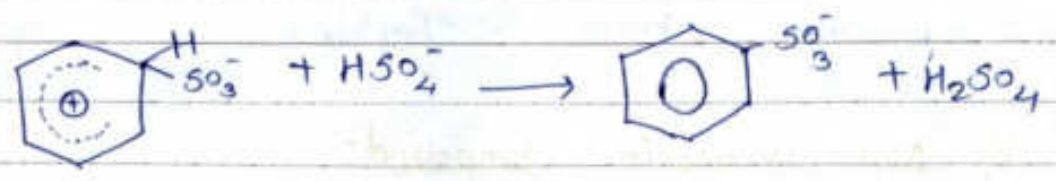
Mechanism:

It proceeds through following steps.

04

i) Formation of electrophile Nitronium ion (NO_2^+) from concentrated HNO_3 sulphuric acid acts as a catalyst.

ii) Loss of proton by σ -Complex to HSO_4^-



benzene sulphonic acid.

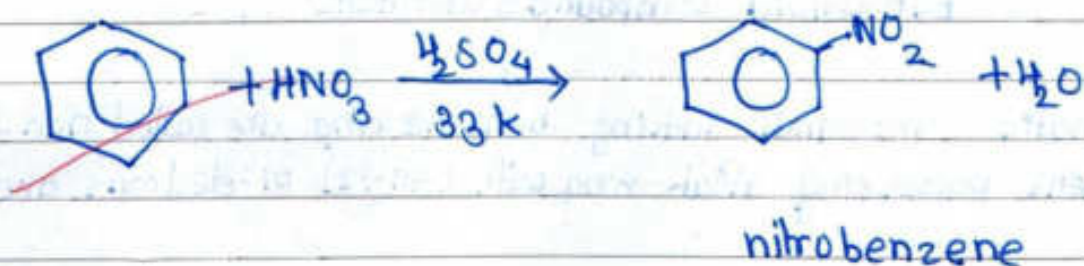
iii) ~~protonation~~ of the product formed in step (ii)



Q. 1 Discuss the mechanism of Nitration and sulphonation of benzene

Nitration:

The substitution of hydrogen by nitro group is called nitration. In presence of concentrated sulphuric acid, benzene reacts with nitric acid to form nitrobenzene.

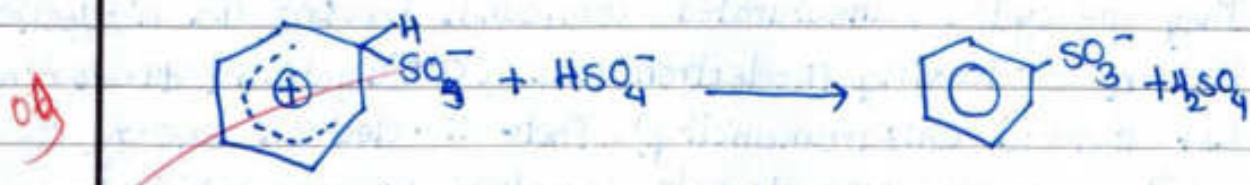


Mechanism:

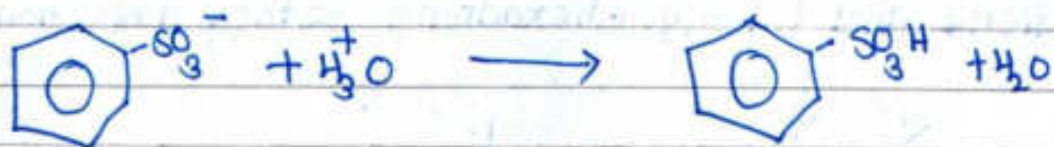
It proceeds through following steps.

i) Formation of electrophile nitronium ion (NO_2^+) from concentrated HNO_3 sulphuric acid acts as a catalyst.

ii) Loss of proton by σ -complex to HSO_4^-



iii) Protonation of the product formed in step (ii)



benzene sulphonic acid

Q. 2 What are aromatic compound and Non-aromatic compound and explain with examples?

a) Formation of compound:

They are unusually stable, cyclic, planar

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PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASAGAON

DEPARTMENT OF CHEMISTRY

(Academic Year 2019-20)

Notice

Date: - 16/09/2019

All the students of B.Sc I (Chemistry) here by informed that, there class test is scheduled on 24/09 /2019 at 01:35 pm. The details of the topic for the unit test are as follows.

1. Inorganic chemistry (Paper-I): Chemical bonding and molecular structure VBT
2. Organic Chemisry (Paper-II):Cycloalkanes ,Cycloalkenes and Alkadienes

It is compulsory to all the students to attend the test as per the college scheduled.



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DEPARTMENT OF CHEMISTRY

UNIT TEST

Class :- B.Sc.I

Sub.:- Organic & Inorganic

Marks :- 20

20
20

Date :- 24-9-2019

Time :- 1.35 to 2.20 pm

1. Butadiene on polymerization forms _____ rubber.
 - a) Polyethene
 - b) benzene
 - c) bunna
 - d) none of these
2. The carbon - carbon bond length in 1,3 butadiene is _____ A⁰.
 - a) 1.54
 - b) 1.46
 - c) 1.33
 - d) 1.35
3. Dehydration of cyclohexanol forms _____.
 - a) Cyclohexan
 - b) cyclobutane
 - c) methyl cyclopentane
 - d) cyclohexene
4. Propadiene is an example of _____ diene.
 - a) isolated
 - b) cumulated
 - c) conjugated
 - d) none of these
5. Alkadiene are represented by general formula _____.
 - a) C_nH_{2n+2}
 - b) C_nH_{2n-2}
 - c) C_nH_{n-2}
 - d) C_nH_{2n}
6. Cyclohexene undergo allylic halogenation when react with _____.
 - a) Cr₂O₃
 - b) Conc H₂SO₄
 - c) NBS
 - d) Zn-Cu
7. Catalytic bromination of cycloalkane gives _____.
 - a) 1,3 dibromo cycloalkane
 - b) 1,3 dibromo cycloalkene
 - c) 1,3 dibromo alkane
 - d) 1,3 dibromo alkene
8. In Diels-Alder reaction diene react with _____ at 473K to fom cycloalkene
 - a) Nicrome
 - b) dienophile
 - c) acetylene
 - d) allene
9. Equimolar mixture of bromine and 1,3 butadiene gives _____.
 - a) tetrabromobutane
 - b) 1,4 dibromo-2 butene
 - c) 1,4 dibromo-2 butane
 - d) dibromobutane

Padmabhushan Dr Vasantodada Patil Mahavidhyalaya, Tasgaon

DEPARTMENT OF CHEMISTRY

Roll No. - 5239

Div - B

UNIT TEST

Class : -B.Sc.I

Sub.: Organic & Inorganic

Marks :- 0

18

20

Date :- 24 - 9 - 2019

Time :- 1.35 to 2.20 pm

1. Butadiene on polymerization forms _____ rubber.
a) Polyethene b) benzene c) binna d) none of these
2. The carbon - carbon bond length in 1,3 butadiene is _____ Å.
a) 1.54 b) 1.46 c) 1.33 d) 1.35
3. Dehydration of cyclohexanol forms _____
a) Cyclohexan b) cyclobutane c) methyl cyclopentane d) cyclohexene
4. Propodiene is an example of _____ diene.
a) isolated b) cumulated c) conjugated d) none of these
5. Alkadiene are represented by general formula _____
a) C_nH_{2n+2} b) C_nH_{2n-2} c) C_nH_{n-2} d) C_nH_{2n}
6. Cyclohexene undergo allylic halogenation when react with _____
a) Cr_2O_3 b) Conc H_2SO_4 c) NBS d) Zn-Cu
7. Catalytic halogenation of cycloalkane gives _____
a) 1,3 dibromo cycloalkane b) 1,3 dibromo cycloalkene
c) 1,3 dibromo alkane d) 1,3 dibromo alkene
8. In Diels-Alder reaction diene react with _____ at 473K to form cycloalkane
a) Nicrome b) dienophile c) acetylene d) allene
9. Equimolar mixture of bromine and 1,3 butadiene gives _____
a) tetrabromobutane b) 1,4 dibromo-2 butene

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PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASAGAON

DEPARTMENT OF CHEMISTRY

(Academic Year 2019-20)

Notice

Date: - 16/12/2019

All the students of B.Sc I (Chemistry) hereby informed that, there class test is scheduled on 21 /12 /2019 at 12:05 pm. The details of the topic for the unit test are as follows.

1. Physical chemistry (Paper-I) : Thermodynamics
2. Analytical Chemisry (Paper-II):Analysis of Fertilizers.

Notice

It is compulsory to all the students to attend the test as per the college scheduled.



Head,

Dept. of Chemistry

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DEPARTMENT OF CHEMISTRY

P.D.V.P. COLLEGE, TASGAON

16/20

Gita

Roll No :- 5265.

Date :- 21-12-2019

1] What is Thermochemistry.

→ Thermochemistry is a branch of Physical chemistry deals with thermal or heat changes during chemical reaction.

2] Define Exothermic Reaction.

→

oo

3] Give the Statement of third law of thermodynamics.

→ The entropy of all crystalline solid may be taken as zero at the absolute zero temperature.

4] Define Adiabatic Process.

→ The process in which no heat can flow into or out of the system called as Adiabatic Process.

5] Define order of Reaction.

→ The sum of power of the concentration of the reactant in the rate of eqn of that particular

6] Define fertilizer.

→ Fertilizer is a substance that provides one or more chemical elements necessary for plant growth & development.

7] What is mean by sampling

→ The sampling is the process of extracting a small particle for the large quantity

Q1 → Thermochemistry - Thermochemistry is branch of physical chemistry which deals with thermal or heat changes during the chemical reaction.

Q2 Exothermic Reaction - The reaction in which the heat is evolved are called exothermic reaction.

Q3 Third law of thermodynamics -

The entropy of all crystalline solids may be taken as zero at the absolute zero temperature.

Q4 Adiabatic process -

The process in which no heat can flow into or out of the system is called as adiabatic process.

Q5 Order of reaction -

The sum of the power of the concentration of the reactant is called as order of reaction.

Q6 Fertilizers -

Fertilizers may be defined as any organic or inorganic material of natural or synthetic origin added to the soil to supply one or more plants nutrients.

Q7 Sampling -

Sampling is a process of extracting small portion of material from its large quantity which represents the composition of whole material.

Q.1) What is first order reaction? Derive an expression for rate constant of first order reaction?

First order reaction -

In a first order reaction, a single atom or molecule of a reactant undergoes a change and gives the product. In this case, the rate of a reaction is directly proportional to the concentration of the reacting substance.

1. Derivation of equation for rate constant or velocity constant of first order reaction.

Consider a general first order reaction as,
 $A \longrightarrow \text{products}$

If C_A = concentration of the reactant A at any time 't', then the rate of reaction is given by,

$$-\frac{dC_A}{dt} \propto C_A \quad \text{or} \quad -\frac{dC_A}{dt} = kC_A$$

If a = Initial concentration of the reactant A and x = decrease in concentration in time 't', then the concentration remaining at time,

$$'t' = a - x \quad \text{i.e. } C_A = (a - x) \quad \text{at time } t.$$

$$\therefore -\frac{d(a-x)}{dt} = k(a-x)$$

$$\therefore \frac{dx}{dt} = k(a-x) \quad \because \left(\frac{da}{dt} = 0, \text{ as 'a' is constant}\right)$$

$$\therefore \frac{dx}{a-x} = k \cdot dt$$

Integrating the above equation we get,

$$\int \frac{dx}{a-x} = \int k \cdot dt$$

$$\text{i.e. } -\log(a-x) = kt + c \quad (\text{where } c = \text{Constant of integration})$$

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PADMABHUSHAN Dr. VASANTRAODADA PATIL MAHAVIDYALAYA, TASAGAON
DEPARTMENT OF CHEMISTRY
(Academic Year 2019-20)

Notice

Date: - 18/01/2020

All the students of B. Sc. I are here by informed that, they should submit their Home assignment on or before 27/01/2020. Questions for Home assignments are as follows:-

A] Physical Chemistry

Q. 1 What is first order of reaction? Derive an expression for rate constant for first order reaction?

B] Organic Chemistry

Q.1 What is mean by error? Explain the types of error.



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A single atom or molecule of a reactant undergoes a change and gives the product.

Expression for 1st order reaction:-

Consider a general first order reaction as $A \rightarrow \text{products}$.

If C_A = Concentration of the reactant A at any time 't', then the rate of reaction is given by

$$-\frac{dC_A}{dt} \propto C_A \quad \text{or} \quad -\frac{dC_A}{dt} = kC_A$$

If a = initial concentration of the reactant A and x = decrease in concentration in time 't', then the concentration remaining at time,

't' = $a - x$ i.e. $C_A = (a - x)$ at time t

$$\therefore \frac{-d(a-x)}{dt} = k(a-x)$$

$$\therefore \frac{dx}{dt} = k(a-x) \quad (\because \frac{da}{dt} = 0 \text{ as 'a' is constant})$$

$$\therefore \frac{dx}{a-x} = k \cdot dt$$

Integrating the above eqⁿ. we get,

$$\int \frac{dx}{a-x} = \int k \cdot dt$$

i.e. $-\log_e(a-x) = kt + c$ (where, c = constant of integration)

when $t = 0$, $x = 0$ of above eqⁿ becomes,

$$-\log_e a = c$$

Putting this value of 'c' in above eqⁿ, we get

$$-\log_e(a-x) = kt - \log_e a$$

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DEPARTMENT OF CHEMISTRY

(Academic Year 2019-20)

Notice

Date: - 03/02/2020

All the students of B. Sc. I are here by informed that, they should submit their Home assignment on or before 12/02/2020. Questions for Home assignments are as follows:-

A] Physical Chemistry

Q. 1 Give the difference between spontaneous & non spontaneous process

Q.2 Explain in details Carnot's cycle.

B] Analytical Chemistry

Q.1.Explain sampling method inn fertilizers analysis.

Q.2.How is nitrogen estimated by kjedahl's method.



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Q1 Explain sampling method in fertilizer analysis.

→ Sampling:-

The procedure of collecting a sample from whole lot of material is called sampling. A sample is a small quantity of representative material picked out from a bulk quantity for testing / analysis.

Then the only results of sample analysis are meaningful and truly reflect the whole lot, different sampling methods are given below.

- i) methods in which whole material is examined, this method is impracticable.
- ii) casual sampling on an adhoc basis. this method is totally unscientific.
- iii) methods in which portions of the material are selected on the basis of statistical probabilities. this is the only reliable method and it is mathematical one.

Sampling of gases, liquids and solids are carried out by different techniques. Sampling devices for gases are usually made from glass & are fitted with stop-cocks at both the end.

the gases may be collected by expansion into an evacuated container, fusing & displacement of a liquid technique.

In the sampling of solids, e.g a truck load of an ore containing lumps of vastly different sizes and variable composition. the particle size is taken into consideration, since composition of particles of different sizes may vary. large size sample are converted into small & suitable size for analysis by different method.

07

CHEMISTRY - HOME ASSIGNMENT

Q. write difference between Spontaneous and non-spontaneous process.

→ Spontaneous process. (Irreversible)	Non-spontaneous. (Reversible).
① It is a natural process.	① It is a hypothetical process.
② opposing force is much less than driving force.	② driving and opposing forces differ by very small amount.
③ Equilibrium is attained only when the process is complete.	③ There is an equilibrium at every stage.
④ AS opposing force is much less than a lot of energy is dissipated and maximum work cannot be obtained.	④ AS driving and opposing forces are nearly equal, maximum work can be obtained.
⑤ Process may be fast or slow.	⑤ process is slow.
⑥ e.g:- flow of heat from hot body to a cold body.	⑥ e.g:- Reversible expansion of an ideal gas.

10

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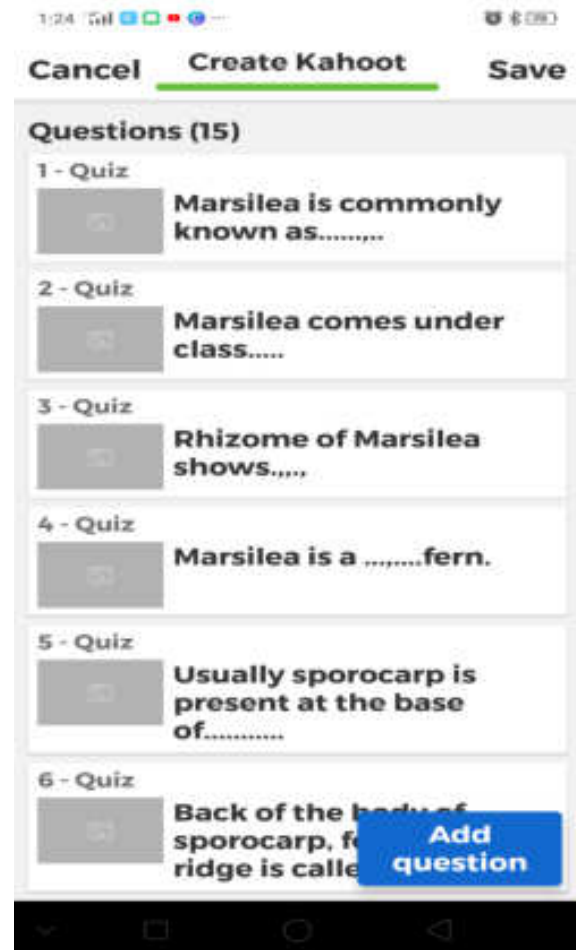
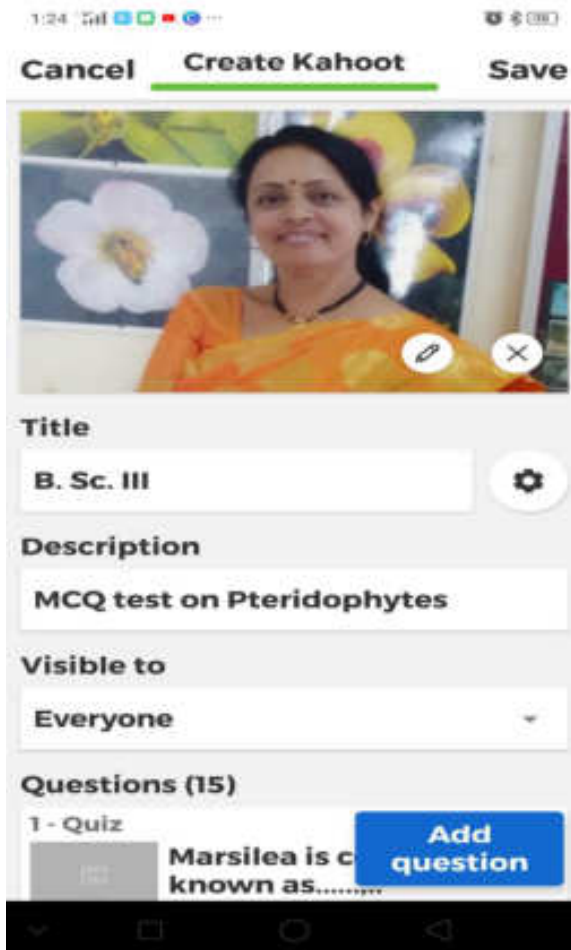
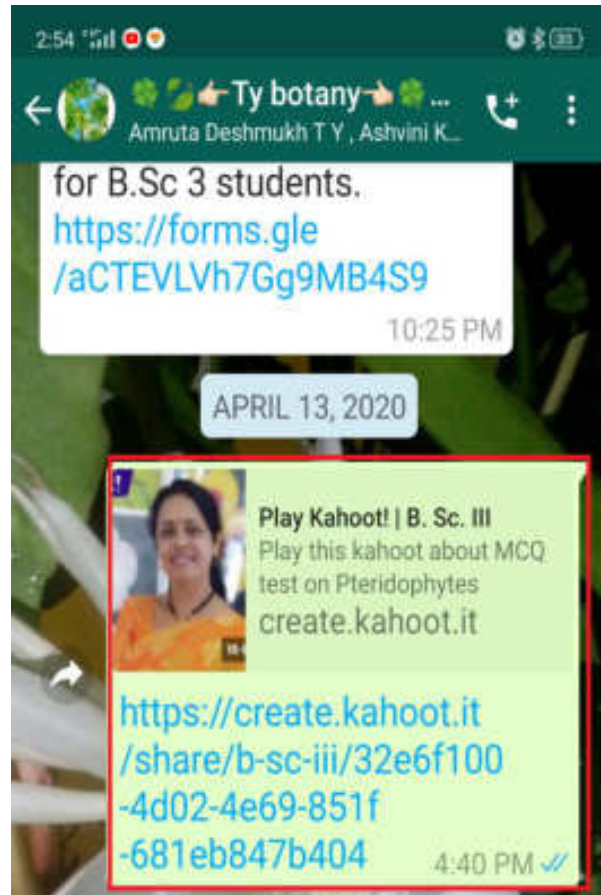
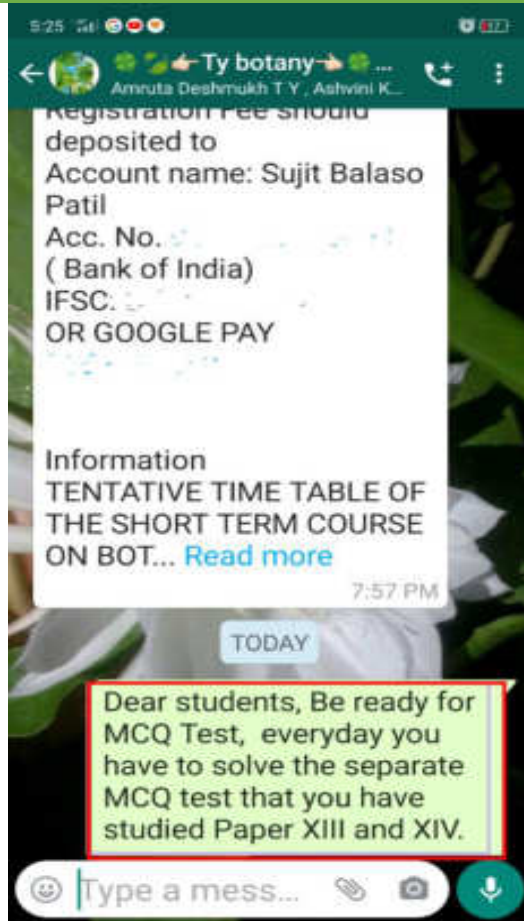
--Shikshanmaharshi Dr. Bapuji Salunkhe

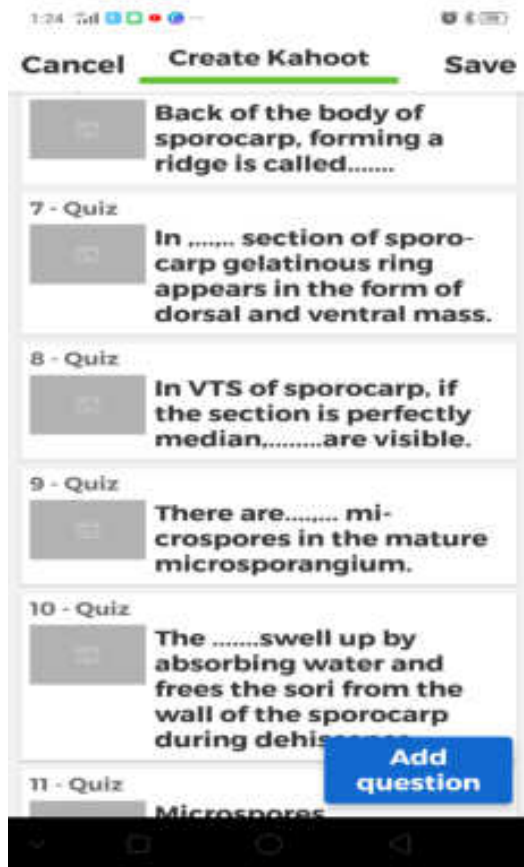
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Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon

DEPARTMENT OF BOTANY

B. Sc. III Paper XIII **Online Test 2** on ***Marsilea***

- ❖ The students were informed about the test periodically during this lockdown period.
- ❖ On 7th April 2020 the students were notified about the test.
- ❖ Online objective test was conducted for B. Sc. III students on 13 April 2020 on **Kahoot!**
- ❖ Two days prior information was communicated to the student on WhatsApp group of TY B. Sc. III about the test on learnt topics of the paper XIII of Semester VI.
- ❖ The students were suggested to study this material as well as the notes supplied in the classroom.
- ❖ The students' response was good for the test. Total 17 students have successfully completed and submitted the test.
- ❖ Students viewed their scores immediately after submission, which can be useful for their further studies.





Dr. Alka Inamdr
Department of Botany
PDVP Mahavidyalaya, Tasgaon

“Dissemination of Education through Knowledge, Science and Culture”.

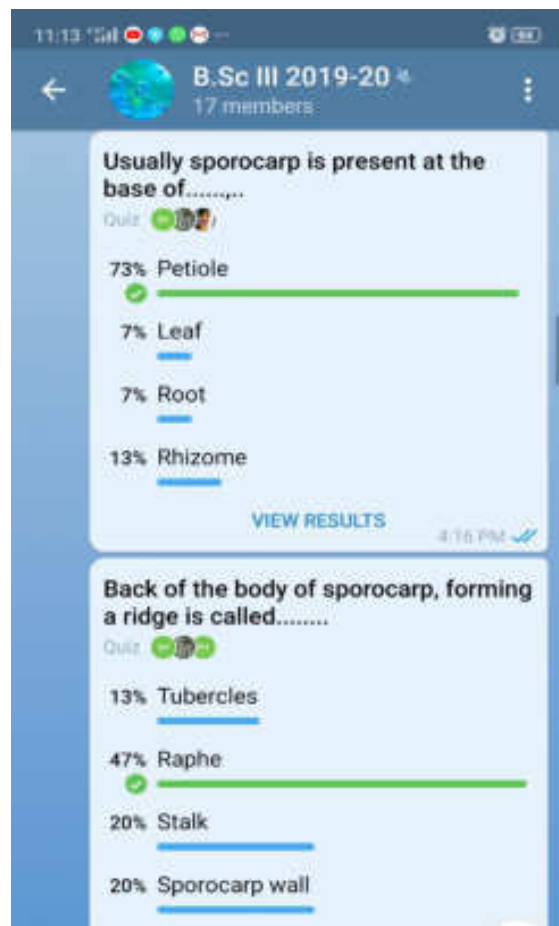
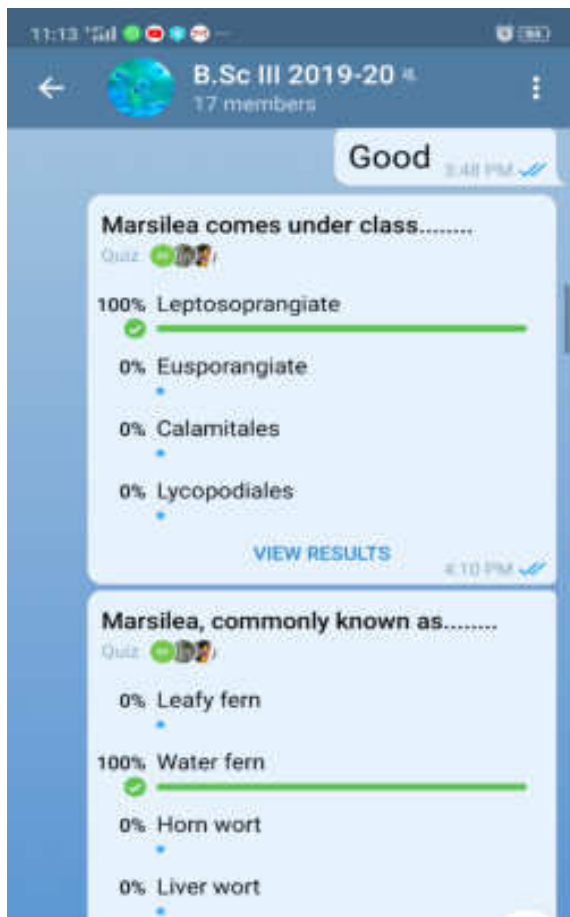
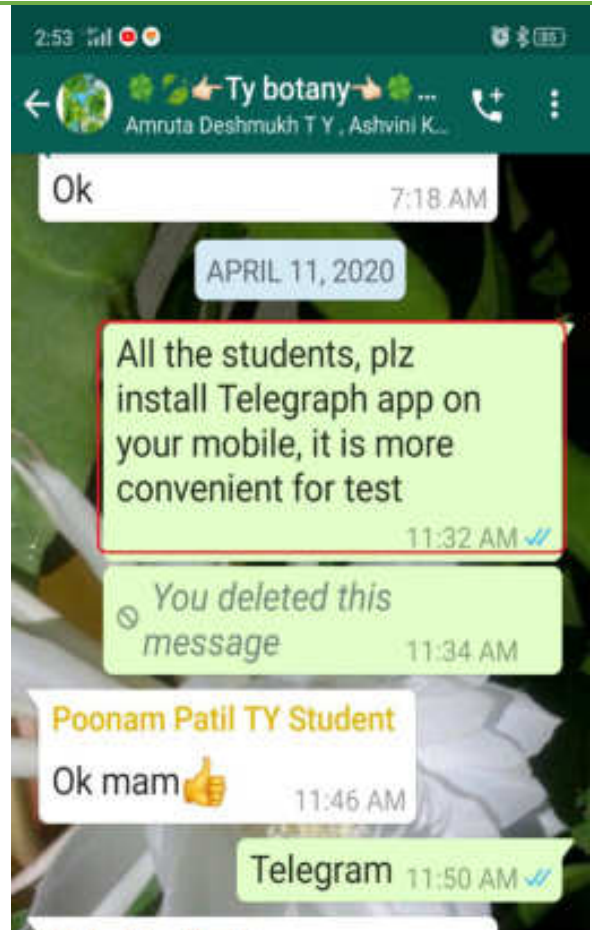
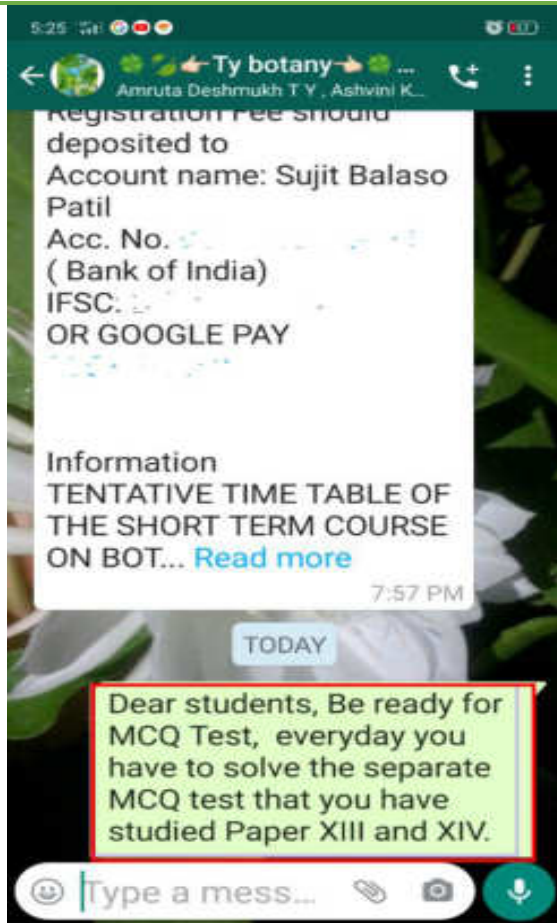
--Shikshanmaharshi Dr. Bapuji Salunkhe

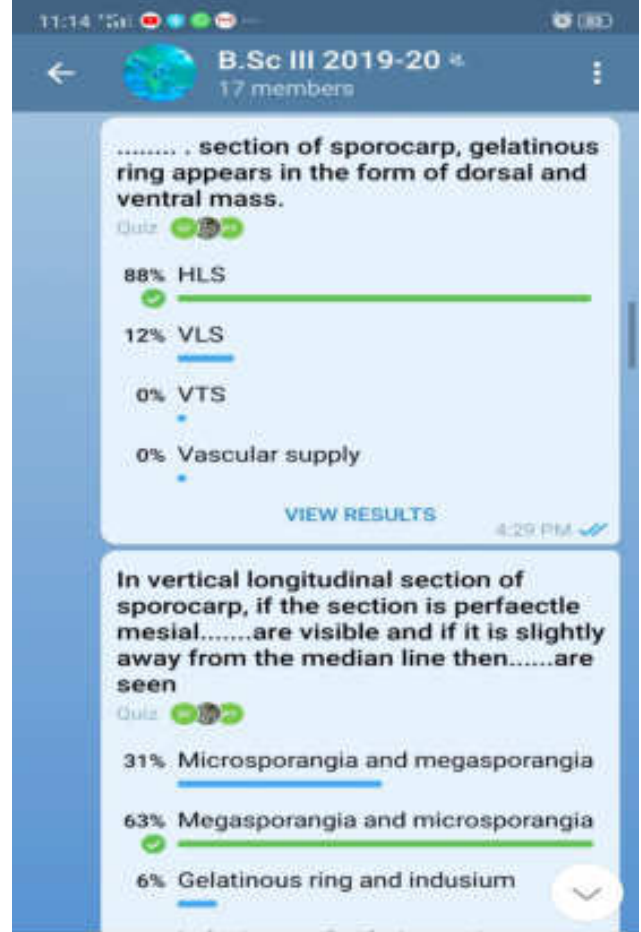
Shri. Swami Vivekananda Shikshan Sanstha's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon

DEPARTMENT OF BOTANY

B. Sc. III Paper XIII Online Test 3 on *Marsilea*

- ❖ The students were informed about the test periodically during this lockdown period.
- ❖ On 11th April 2020 the students were notified about the installation of Telegram app on mobile.
- ❖ Online objective test was conducted for B. Sc. III students on 15th April 2020 on **Telegram**.
- ❖ Two days prior information was communicated to the student on WhatsApp group of TY B. Sc. III about the test on learnt topics of the paper XIII of Semester VI.
- ❖ The students were suggested to study this material as well as the notes supplied in the classroom.
- ❖ The students' response was good for the test. Total 17 students have successfully completed and submitted the test. T
- ❖ Students viewed their scores immediately after submission.
- ❖ The students received their responses immediately, which can be useful for their further studies.





Dr. Alka Inamdr
 Department of Botany
 PDVP Mahavidyalaya, Tasgaon

“Dissemination of Education through Knowledge, Science and Culture”.

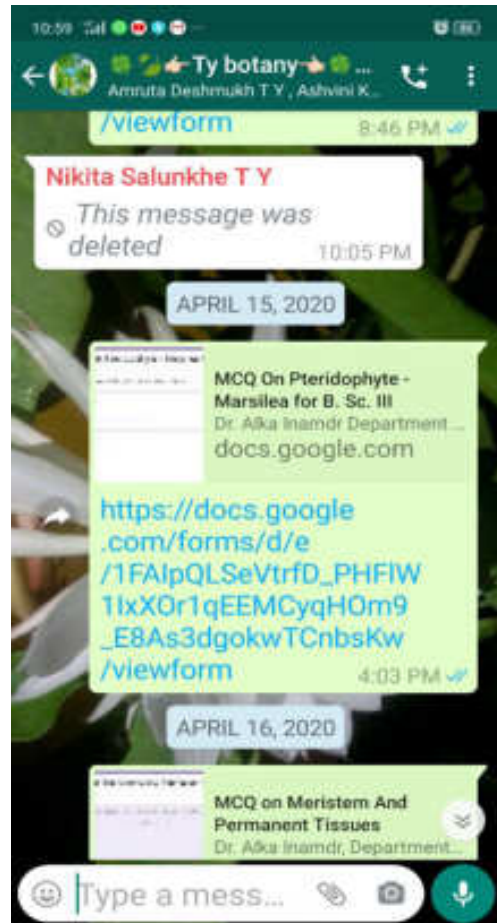
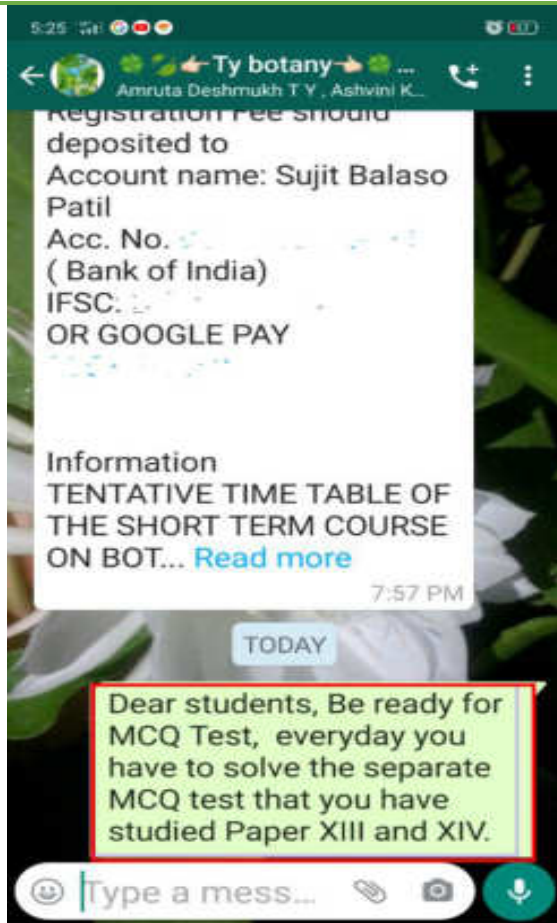
--Shikshanmaharshi Dr. Bapuji Salunkhe

Shri. Swami Vivekananda Shikshan Sanstha's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon

DEPARTMENT OF BOTANY

B. Sc. III Paper XIII **Online Test 4** on
Pteridophytes and *Marsilea*

- ❖ The students were informed about the test periodically during this lockdown period.
- ❖ On 11th April 2020 the students were notified about the installation of Telegram app on mobile.
- ❖ Online objective test was conducted for B. Sc. III students on 16th April 2020 on **Google Form**.
- ❖ Two days prior information was communicated to the student on WhatsApp group of TY B. Sc. III about the test on learnt topics of the paper XIII of Semester VI.
- ❖ The students were suggested to study this material as well as the notes supplied in the classroom.



MCQ On Pteridophyte - Marsilea for B. Sc. III

Dr. Aka Inamdi Department of Botany P. D. V. P. College, Talgaon
 * Required

1. Email address *

2. Marsilea, commonly known as.....
 Mark only one oval.

Horn wort
 Beerwort
 Leaf fern
 Water fern

3. Marsilea comes under class.....
 Mark only one oval.

Leptosporangiate
 Eusporangiate
 Calamitales
 Lycopodiales

6323520 MCQ On Pteridophyte - Marsilea for B. Sc. III

4. Rhizome of Marsilea shows

Mark only one oval.

Protostele

Ductosteles

Amphitelic Siphonostele

Simple Stele

5. Marsilea is a _____ fern

Mark only one oval.

Homosporous

Heterosporous

Microsporous

Megasporous

6. usually Sporocarp is present at the base of

Mark only one oval.

Petiole

leaf

Axil

root

7. back of the body of sporocarp, forming a ridge is called

Mark only one oval.

Tubercles

Raphe

stalk

sporocarp wall

<https://docs.google.com/forms/d/1wFYyqzH5Kj8B9m2vWwDMMn0zGMA4eLWYkz4E/> 25

6323520 MCQ On Pteridophyte - Marsilea for B. Sc. III

8. In _____ section of sporocarp, gelatinous ring appears in the form of dorsal and ventral mass

Mark only one oval.

vascular supply

VTS

VLS

HLS

9. In VLS of sporocarp, if section is perfectly median, _____ are visible and if slightly away from median line, _____ are seen

Mark only one oval.

Microsporangiate and Megasporangiate

Megasporangiate and Microsporangiate

gelatinous ring and Indusium

indusium and gelatinous ring

10. There are _____ microspores in a mature Microsporangium

Mark only one oval.

8-16

10-15

32-64

65-70

<https://docs.google.com/forms/d/1wFYyqzH5Kj8B9m2vWwDMMn0zGMA4eLWYkz4E/> 26

6/23/2020 MCQ On Pteridophyte - Marsilea for B. Sc. III

11. The _____ swell up by absorbing water and frees the sori from wall of the sporocarp during dehiscence

Mark only one oval.

- indusium
- sporocarp wall
- gelatinous ring
- sori

12. Microspore develops _____ and _____

Mark only one oval.

- Microgametophyte and archegonium
- microgametophyte and antherozoid
- megagametophyte and antherozoid
- megagametophyte and archegonium

13. In Marsilea, antherozoids are _____

Mark only one oval.

- coiled and uniflagellate
- coiled and multiflagellate
- straight and uniflagellate
- straight and multiflagellate

14. A fern differs from a moss in having _____

Mark only one oval.

- swimming antherozoid
- independent gametophyte
- vascular tissue
- distinct alternation of generations

<https://docs.google.com/forms/d/1ar7yqz8Hkxj08yewW0r0u0DkMk0t9MAA4eLW7H0u0d0/> 4/5

6/23/2020 MCQ On Pteridophyte - Marsilea for B. Sc. III

15. A stele without central pith is called _____

Mark only one oval.

- protosteles
- dictyostele
- siphonostele
- solenostele

16. The plant body of Marsilea is a _____

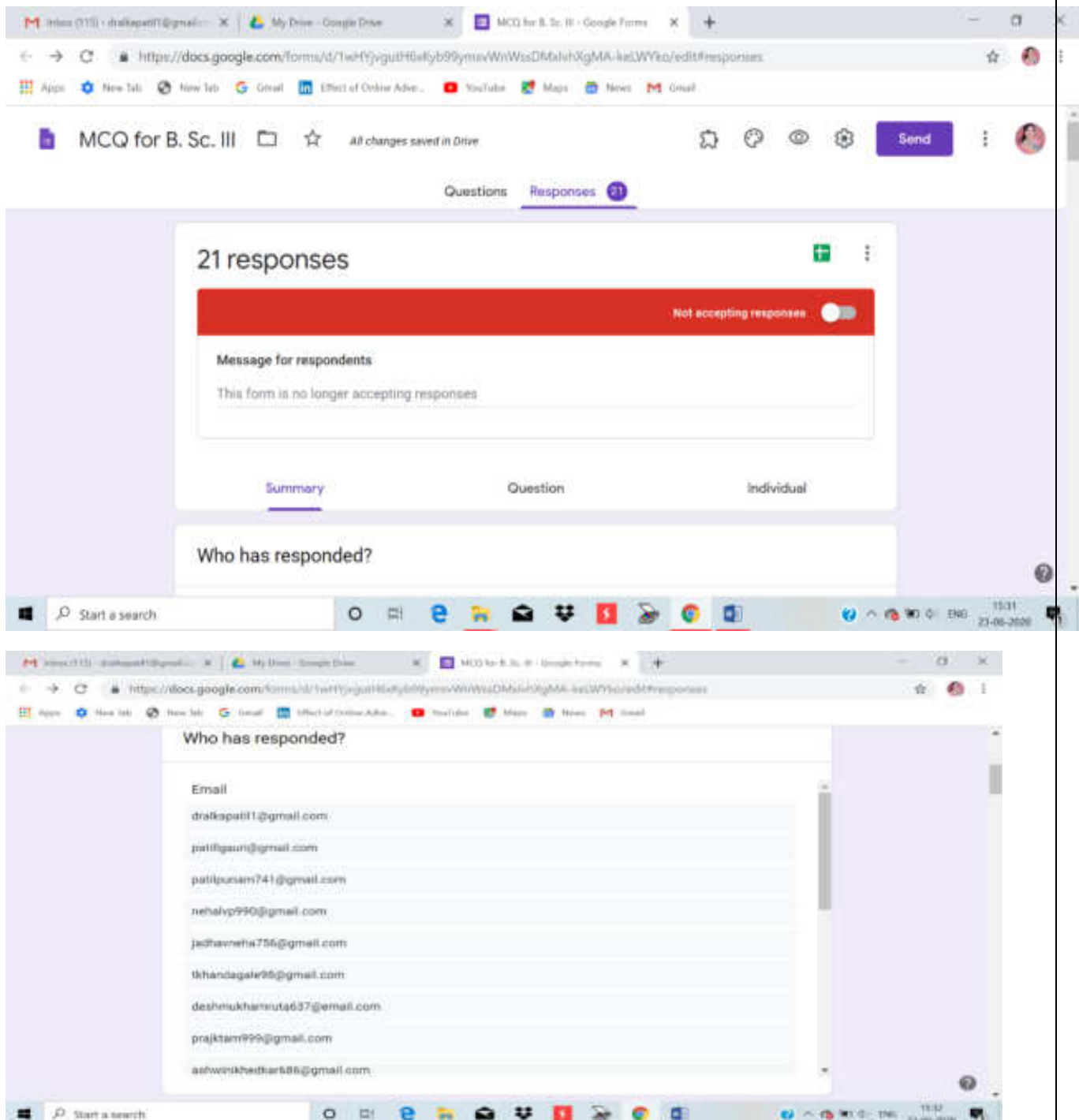
Mark only one oval.

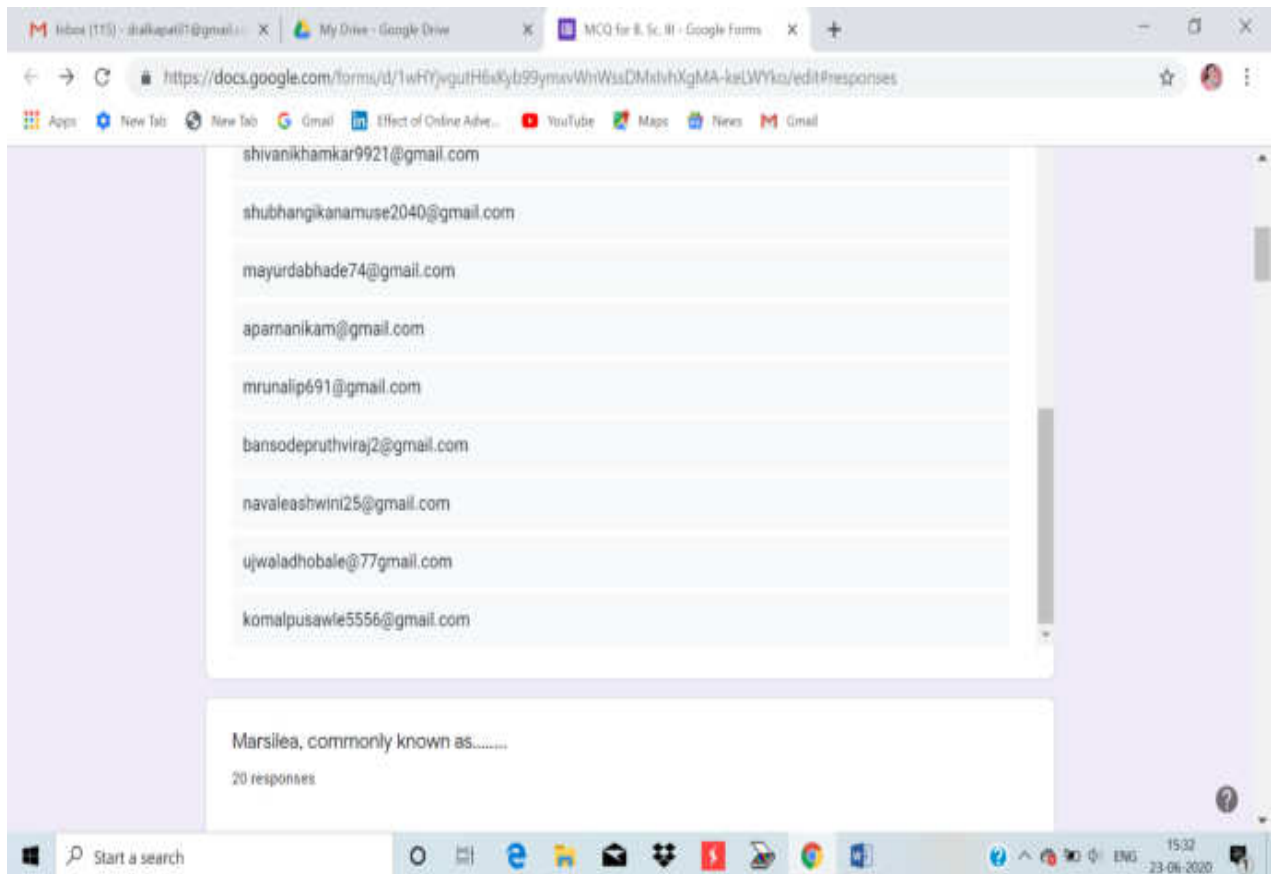
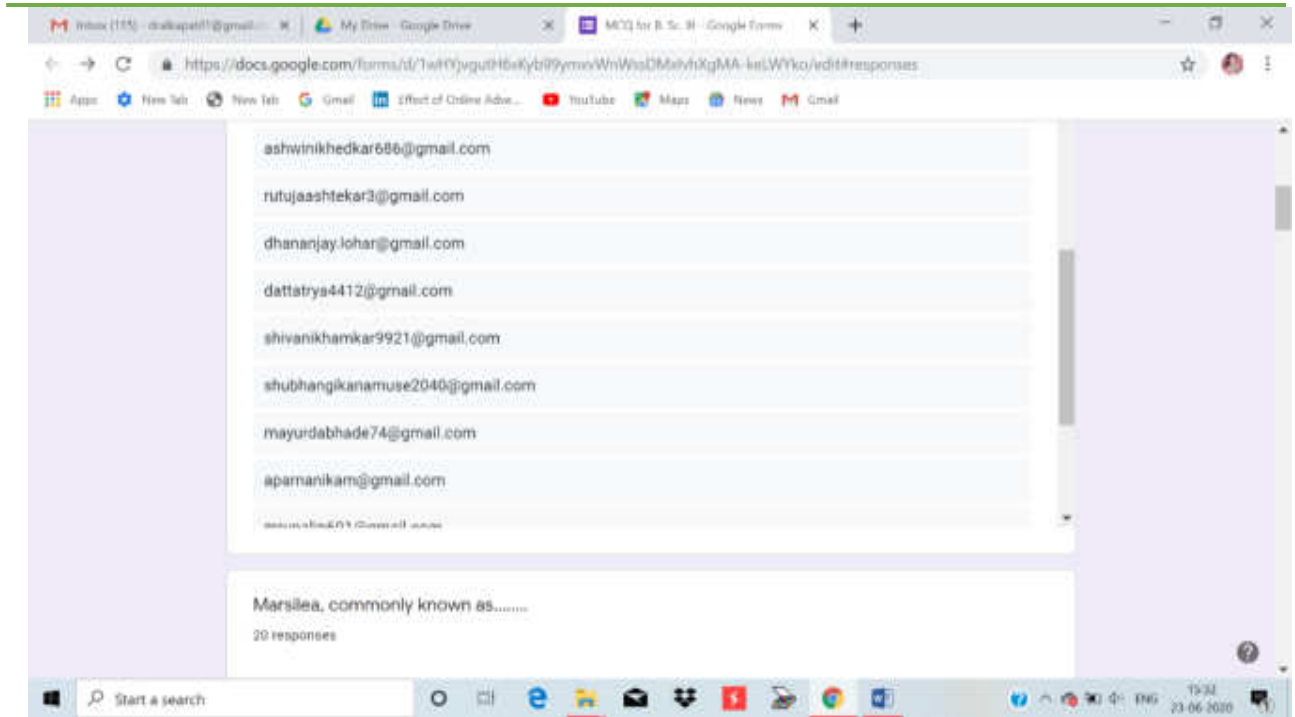
- gametophyte
- sporophyte
- creeper
- climber

This content is neither created nor endorsed by Google

Google Forms

- ❖ The students' response was good for the test. Total 17 students have successfully completed and submitted the test.
- ❖ Students viewed their scores immediately after submission.
- ❖ The students received their responses immediately, which can be useful for their further studies.





Dr. Alka Inamdr
Department of Botany
PDVP Mahavidyalaya, Tasgaon

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--Shikshanmaharshi Dr. Bapuji Salunkhe

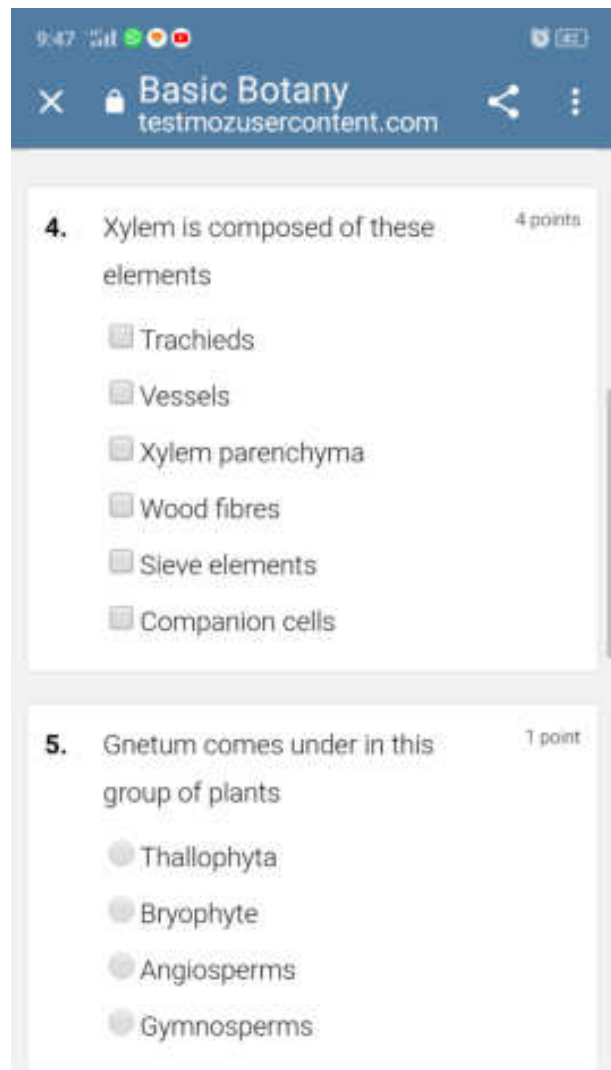
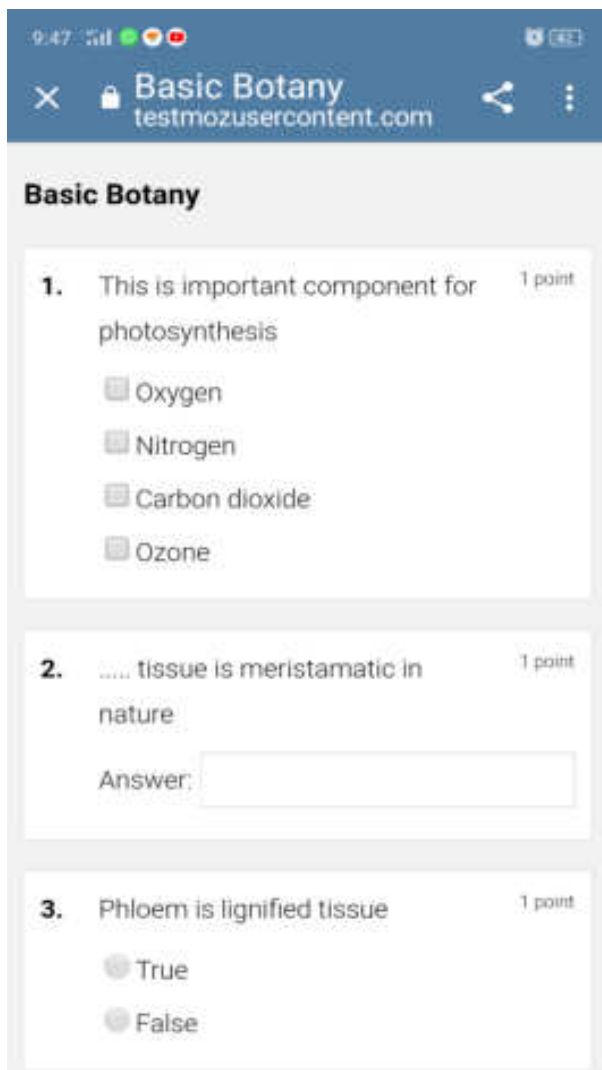
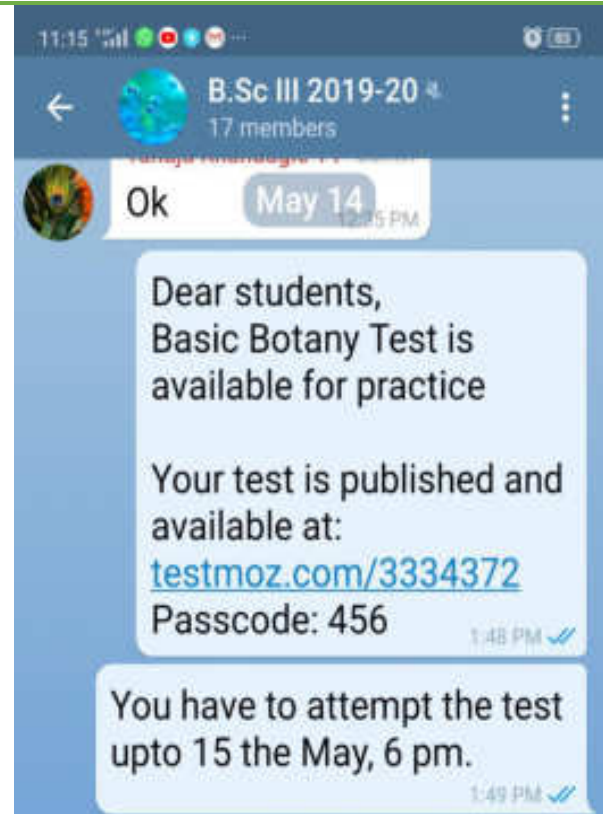
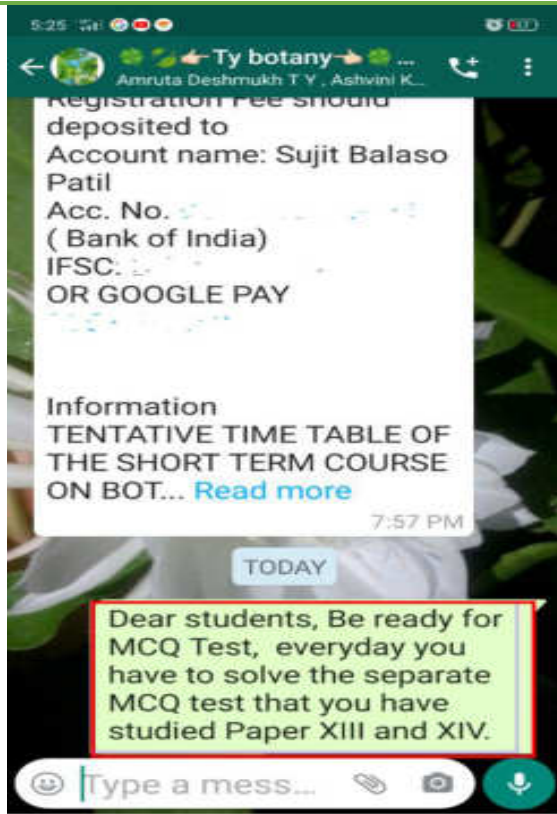
Shri. Swami Vivekananda Shikshan Sanstha's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon

DEPARTMENT OF BOTANY

B. Sc. III Paper XIII **Online Test 7** on

Basic Botany

- ❖ The students were informed about the test periodically during this lockdown period.
- ❖ On 11th April 2020 the students were notified about the installation of Telegram app on mobile.
- ❖ Online objective test was conducted for B. Sc. III students like M. Sc. Entrance Exam on 14th May 2020 on **Testmoz**.
- ❖ Two days prior information was communicated to the student on WhatsApp group of TY B. Sc. III about the test on learnt topics of the paper XIII of Semester VI.
- ❖ The students were suggested to study this material as well as the notes supplied in the classroom.



9:47 5G 100% 87%

6. These are important mechanisms in plants. 3 points

- Photosynthesis
- Transportation
- Respiration
- Automization
- Brownion movement

7. The plants are living organisms. 1 point

True

False

8. Function and types of Chlorophyll in plants 1 point

9. The plant part...is doing that absorption of water and minerals. 1 point






Answer:





10. Plant part and it's functions 4 points

<input type="checkbox"/> Root	1. Cell division
<input type="checkbox"/> Xylem	2. Water
<input type="checkbox"/> Phloem	3. Water
<input type="checkbox"/> Leaf	4. Food
	5. Photosynthesis

Submit

- ❖ The students' response was good for the test. Total 17 students have successfully completed and submitted the test. T
- ❖ Students viewed their scores immediately after submission.
- ❖ The students received their responses immediately, which can be useful for their further studies.

9:43      42

  **Basic Botany**
Lite | testmoz.com  

Menu

Basic Botany Results


[Export to CSV: Point Grid | Response Grid](#)
[Copy to Clipboard](#)
[View Audit Logs](#)

Average Score
70%
0% to 100%

Average Time
0:27:59
0:03:57 to 3:37:45

Responses
14

Score Histogram



Name	Score	Started On	Finished On
456	82% (14/17)	2020-05-15 11:48 a.m.	2020-05-15 12:00 p.m.
amruta	69% (11.67/17)	2020-05-14 3:43 p.m.	2020-05-14 4:13 p.m.
Gauri	59% (10/17)	2020-05-14 8:33 p.m.	2020-05-14 8:43 p.m.
Neha	65% (11/17)	2020-05-14 2:34 p.m.	2020-05-14 2:42 p.m.
Nehal Vilas patil	67% (11.33/17)	2020-05-14 3:24 p.m.	2020-05-14 3:34 p.m.
Nikita	96% (16.33/17)	2020-05-14 6:02 p.m.	2020-05-14 6:07 p.m.
Poonam	88% (15/17)	2020-05-14 2:15 p.m.	2020-05-14 2:29 p.m.
Prekta Ashok Mali	41% (7/17)	2020-05-15 10:56 a.m.	2020-05-15 11:09 a.m.
Rutuja Ashtekar	80% (13.67/17)	2020-05-14 3:09 p.m.	2020-05-14 3:21 p.m.
Sarnudhi shamrao patil	88% (15/17)	2020-05-14 3:49 p.m.	2020-05-14 4:25 p.m.
Snehal	0% (0/17)	2020-05-14 2:23 p.m.	2020-05-14 6:00 p.m.
Snehal salunkhe	100% (17/17)	2020-05-14 8:10 p.m.	2020-05-14 8:14 p.m.
Tanuja khandagale	94% (16/17)	2020-05-14 7:10 p.m.	2020-05-14 7:14 p.m.
Tanuja Shivaji khandagale	53% (9/17)	2020-05-14 6:26 p.m.	2020-05-14 6:41 p.m.

Dr. Alka Inamdr

Department of Botany
PDVP Mahavidyalaya, Tasgaon

“Dissemination of Education through Knowledge, Science and Culture”

Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur

**PADMABHUSHAN DR.VASANTRAODADA PATIL
MAHAVIDYALAYA, TASGAON**

Department of Compter Science

B. Sc. III Advancd linux application

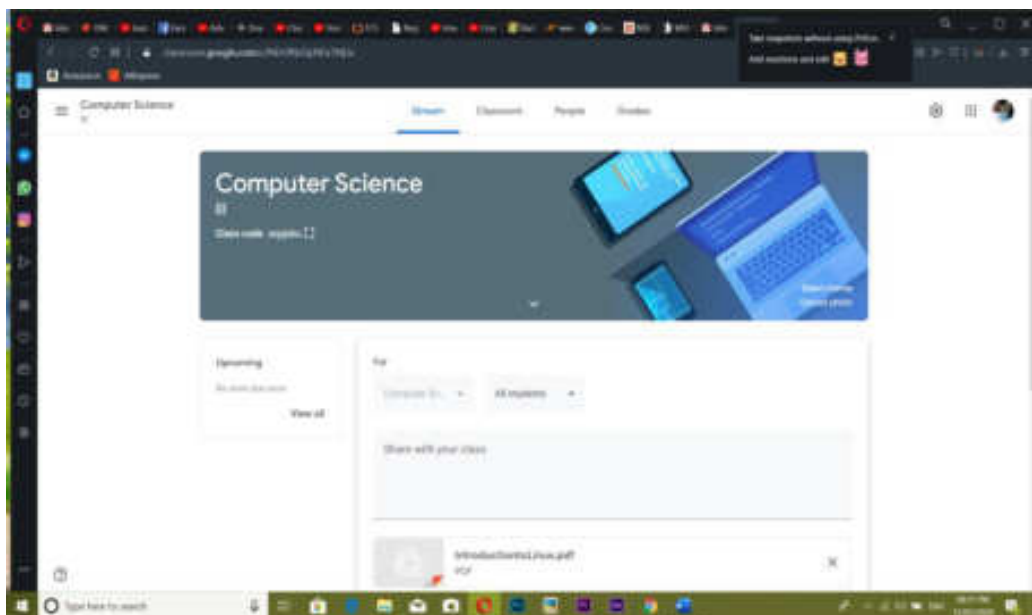
(Paper -XV)

Online Test No. 1

1. B. Sc. III student’s online objective test was conducted on 10TH May 2020.

❖ One week prior to the test a study material was shared with the students on Google classroom. The students were suggested to study this material as well as the notes supplied in the classroom.

❖ Shared study material on Google classroom:



- ❖ The students were informed about the test by WhatsApp

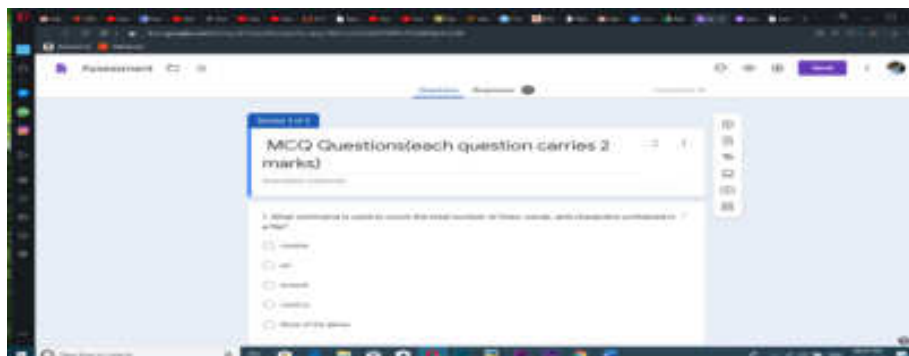


On 10TH May 2020 the test was provided on Google form to the students.



- ❖ The link of corresponding test was given below:
[https://docs.google.com/forms/d/e/1FAIpQLSep45WRH3-yG3htdbFfiHsMO_TiEV16EiMs7qQZCrcMKqkFPQ/viewform?usp=sf link](https://docs.google.com/forms/d/e/1FAIpQLSep45WRH3-yG3htdbFfiHsMO_TiEV16EiMs7qQZCrcMKqkFPQ/viewform?usp=sf_link)

❖ And the questions of the Online tests are as follows:



❖ The students' response was good for the test. Total 23 students have successfully completed and submitted the test.

❖ Students viewed their scores immediately after submission. The students received their responses by an email, which can be useful for their further studies.

❖ The students' responses were recorded as follows:

Timestamp	Score	Roll No	Name	Email
5/10/2020 14:33:14	20 / 20	6236	Gayatri laxman joshi	gayatrijoshi83082@gmail.com
5/10/2020 14:49:40	20 / 20	6242	Sanket Ekanath Mali	sanketmali856@gmail.com
5/10/2020 14:55:31	20 / 20	6242	Mali Shubhangi Ashok	malishibhangi@gmail.com
5/10/2020 15:07:43	14 / 20	6246	Mulla Sana Dilawar	sanamulla015@gmail.com
5/10/2020 15:20:43	20 / 20	6250	Yogesh Patil	Patilyogesh0172@gmail.com
5/10/2020 15:22:08	20 / 20	6252	Pravin Pratap Pawar	pravinpawar200.pp72@gmail.com
5/10/2020 15:23:20	20 / 20	6230	Shubham Keshav Bobade	shubham191099@gmail.com
5/10/2020 15:25:07	20 / 20	6232	Pranav Piraji Hankare	prahan1999@gmail.com

5/10/2020 15:50:21	20 / 20	6235	JamdadeVishvajeet Vasantrao	vishvajeetjamdade@gmail.com
5/10/2020 16:14:51	20 / 20	6251	Pawar Pooja Balaso	pbp4432@gmail.com
5/10/2020 16:57:53	20 / 20	6240	Pooja Rajaram Mali	malipooja487@gmail.com
5/10/2020 17:14:11	20 / 20	6228	Avinash Raghu Bhosale	bhosaleavinash05@gmail.com
5/10/2020 17:34:10	16 / 20	6244	Vijay Prakash Mane	vijaymane400@gmail.com
5/10/2020 20:16:17	20 / 20	6255	Shubham Sanjay Rajput	shubham.prajval@gmail.com
5/10/2020 20:48:39	20 / 20	6247	Patil Ashwini Sarjerao.	patilashu2698@gmail.com
5/11/2020 8:37:36	18 / 20	6260	Wakade Shweta Sunil	shwetawakade6943@gmail.com
5/11/2020 15:02:21	20 / 20	6231	Manoj shamrao Gunjal	Manojgunjal555@gmil.com
5/11/2020 15:18:26	20 / 20	6259	Ajay babaso thorat	ajayat8055@gmail.com
5/11/2020 19:50:57	20 / 20	6237	Tanaji Namdev kengare	tanajikengare97@gmail.com
5/12/2020 11:45:50	20 / 20	6245	Mulani masum rajekhan	Mulanimasum@gmail.com
5/12/2020 11:53:39	20 / 20	6257	Nikita Suresh Shendage	nikitashendage2371@gmail.com
5/12/2020 13:37:51	10 / 20	6227	Amrutsagar pranali rajendra	pranaliamrutsagarpanu@gmail.com

➤ Statistical Analysis

❖ Marks Analysis:

Marks	No. Of Students
Below 10	0
10 – 15	2
16 – 20	21
Total	23

- ❖ The test was conducted by using **Google forms**.
- ❖ Further tests or quizzes are based on other topics of this paper.
- ❖ The most important objectives regarding to take tests or quizzes are as follows:
 - To evaluate how much students study during the lockdown periods.
 - To check performance of the student individually.

Miss.Kumbhar N.V.
Assistant Professor,
Department of computer science

ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार - शिक्षणमहर्षी डॉ. बापूजी साठुखे

Shri Swami Vivekanand Shikshan Sanstha , Kolhapur

PADMABHUSHAN DR. VASANTRAODADA PATIL MAHAVIDYALAYA,

TASGAON, DIST SANGLI

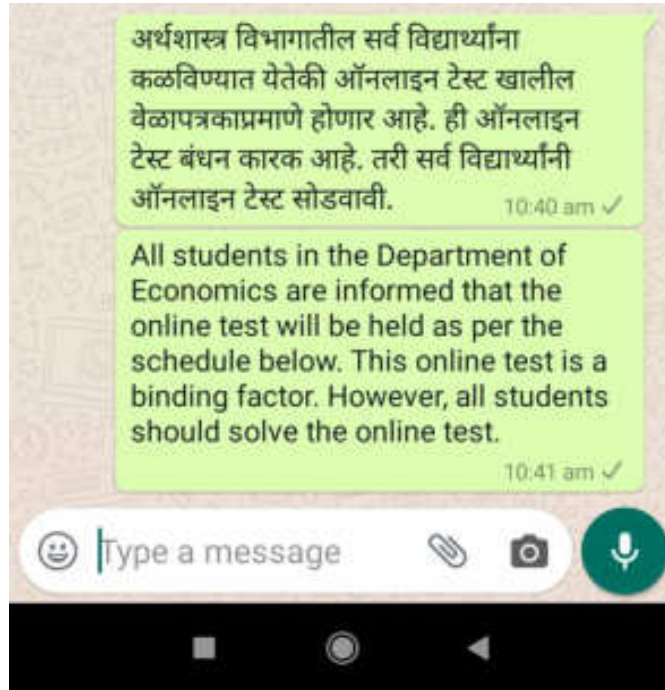
DEPARTMENT OF ECONOMICS

२०१९-२०

M.A.Part-II

ADVANCED BANKING

- B.A. Part-III students' online test was conducted on 28th April 2020
- On 26th April 2020 the students were notified about the test.



- On 28th April 2020 the link for the test was provided to student.



- The Students response was good for the online test. Total 37students have successfully completed and submitted the test
- Students viewed their scores immediately after submission. The students received their responses by email, which can be useful for their further students.
- The Students responses were recorded.



Advanced Banking (M.A.,-II) Results

View by table / partition by tag / group by name Search

Export to CSV | Print Grid | Response Grid
Copy to Clipboard
View Audit Logs

Average Score
72%
35% to 100%

Average Time
0:08:00
0:00:20 to 0:48:33

Responses
37

Score Histogram

Names	Score	Started On	Finished On	Time
<input type="checkbox"/> Suman, Karolina, Ramon	100% (40/40)	2020-04-28 11:37 a.m.	2020-04-28 11:39 a.m.	0:02:01
<input type="checkbox"/> Julia, sven	60% (24/40)	2020-04-28 5:56 p.m.	2020-04-28 5:59 p.m.	0:02:21
<input type="checkbox"/> Karolina, Ramon, Suman	60% (24/40)	2020-04-27 4:26 p.m.	2020-04-27 4:47 p.m.	0:20:34
<input type="checkbox"/> Sat, niyasha, balaji	70% (28/40)	2020-04-25 12:42 p.m.	2020-04-25 1:17 p.m.	0:35:11
<input type="checkbox"/> Karan, Anand, Palli	90% (36/40)	2020-04-27 6:05 p.m.	2020-04-27 6:05 p.m.	0:00:04
<input type="checkbox"/> Karan, Anand, stouk	60% (24/40)	2020-04-27 8:42 p.m.	2020-04-27 8:46 p.m.	0:04:01
<input type="checkbox"/> MAMATHA	60% (24/40)	2020-04-25 12:42 p.m.	2020-04-25 12:45 p.m.	0:02:34
<input type="checkbox"/> Mahesh, sany	85% (34/40)	2020-04-25 12:48 p.m.	2020-04-25 12:49 p.m.	0:01:22
<input type="checkbox"/> Manoj, Ramesh, Deepansh	85% (34/40)	2020-04-25 10:32 a.m.	2020-04-25 10:45 a.m.	0:13:21
<input type="checkbox"/> Manoj, Ramesh, Deepansh	100% (40/40)	2020-04-25 12:19 p.m.	2020-04-25 12:21 p.m.	0:01:48
<input type="checkbox"/> Meeta	80% (32/40)	2020-04-27 5:25 p.m.	2020-04-27 5:37 p.m.	0:11:11
<input type="checkbox"/> Meeta	60% (24/40)	2020-04-27 6:33 p.m.	2020-04-27 6:35 p.m.	0:01:51
<input type="checkbox"/> Muzen, Rajesh, Anil	90% (36/40)	2020-04-27 5:46 p.m.	2020-04-27 5:47 p.m.	0:01:19
<input type="checkbox"/> Muzen, Rajesh, Anil	70% (28/40)	2020-04-30 5:35 p.m.	2020-04-30 5:38 p.m.	0:02:02
<input type="checkbox"/> Muzen, Rajesh, Anil	85% (34/40)	2020-04-27 5:27 p.m.	2020-04-27 5:32 p.m.	0:04:29
<input type="checkbox"/> Muzen, Rajesh, Anil	40% (16/40)	2020-04-27 5:39 p.m.	2020-04-27 5:41 p.m.	0:02:21
<input type="checkbox"/> Muzen, Rajesh, Anil	70% (28/40)	2020-04-27 5:43 p.m.	2020-04-27 5:45 p.m.	0:01:24
<input type="checkbox"/> Praveen, sany, Mohan	60% (24/40)	2020-04-27 4:51 p.m.	2020-04-27 5:17 p.m.	0:25:24
<input type="checkbox"/> Praveen, sany	60% (24/40)	2020-04-25 12:57 p.m.	2020-04-25 1:09 p.m.	0:12:25
<input type="checkbox"/> Pratiksha	70% (28/40)	2020-04-26 7:23 p.m.	2020-04-26 8:12 p.m.	0:48:31
<input type="checkbox"/> Pratiksha, Anand, Pallavi	90% (36/40)	2020-04-27 12:09 p.m.	2020-04-27 12:23 p.m.	0:14:01
<input type="checkbox"/> Pratiksha, Anand, Pallavi	60% (24/40)	2020-04-27 8:29 p.m.	2020-04-27 8:36 p.m.	0:07:05



- The test was conducted by using <https://testmoz.com/q/>

Dr. Bandu J Kadam

Assistant Professor

Department of Economics



“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार”- शिक्षणमहर्षी डॉ.वापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit

Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312
DEPARTMENT OF HISTORY
S.Y.B.A. and T. Y. B. A.

ONLINE REMEDIAL COACHING CLASSES

05/07/2021 to 16/07/2021.

Submitted to
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya
Tasgaon, Dist- Sangli, Pin-416312

Organizer
Department of History
Tasgaon,

Convener & Head
Dr. H. D. Nadaf
Department of History
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon,

2020-2021

ONLINE REMEDIAL COACHING CLASSES

PROGRAMME REPORT

Title of Programme	: Online Remedial Coaching Classes
Organizing Department	: History
Date	: 05/07/2021 to 16/07/2021.
Venue	: Online Platform
No. of Participants	: 45 (Male: 35, Female: 10)
Convener	: Dr. Haji Nadaf
President	: Prin. Dr. Dr. Milind Hujare

The Department of History conducted '**Online Remedial Coaching Classes** activity on **S.Y.B.A. & T. Y. B. A.**, from **05/07/2021 to 16/07/2021**. A total of 45 students have participated in this same activity. All students have presented in the classes. The syllabuses of these classes were the basis on a base of curriculum. The following papers are billow, Modern Indian History, History Writings, History of Modern Maharashtra, History of Medieval India, History of Ancient India and Modern Maharashtra.

All Students of B.A.II and III participated in this activity. Students have responded to the good dialogue. The faculty has successfully tried to appreciate the student. Students are given their reflection about the classes. A total of **74%** of students have presented this activity. This entire activity conducted by **Dr. H. D. Nadaf** for this activity. **Hon. Prin. Dr. Milind Hujare** has guided and cooperated well with the program.


HEAD
Department of History
P.D.V.P.College,Tasgaon

Date 01/07/2021

To
The Principal
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli.416312

Subject: Application for the Permission to conduct 'Online Remedial Coaching Classes' activity.

Respected sir,

Our department of History is going to conduct 'Remedial Coaching Classes' activity on **S. Y. B. A.** and **T. Y. B. A.** from **05/07/2021** to **16/07/2021**. We are kindly requesting you to allow us to conduct this activity. Students of B.A. Part- I. & II, are going to participate in the same activity.

Thanking you

Yours Faithfully


Principal
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon. (तासगाव)


HEAD
Department of History
P.D.V.P.College, Tasgaon

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” -शिक्षणमहर्षी डॉ.बापूजी साळुंखे
M Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY
(2020-21)

Notice

Date: 01/07/2021

All admitted history students of S. Y. B. A. and T. Y. B. A. hereby indicate that we will be conducting Online remedial coaching classes' activities on "Curriculum" from 05/07/2021 to 16/07/2021. for slow learning students. The topics will be related to the given topics and all students will be listed with this instruction. All enrolled students are required to attend Classes.


HEAD
Department of History
P.D.V.P.College,Tasgaon



Dissemination of Education for Knowledge, Science and Culture”
Shikshanmaharashi Dr. Bapuji Salunkhe



Shri Swami Vivekanand Shikshan Sanstha Kolhapur's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya
Tasgaon, Dist- Sangli 416312
DEPARTMENT OF HISTORY
Bachelor of Arts, Part: III
(Academic Year 2020-2021)

Online Remedial Coaching Classes

(B.A.II., and B. A. III.)

INAUGURAL SESSION

05th July 2021 Morning 10.00 am

Inaugurator

Prin. Dr. Milind S. Hujare

Convener and Head

Dr. Haji Nadaf

05/07/2021 to 16/07/2021

Lecturer:

Dr. Haji Nadaf

Mr. Ranjit Kumbhar, Dr V. D. Chorge

Time: 10.00 am. / Venue-Room No.26. (Date: 05/07/2021 to 16/07/2021)

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” -शिक्षणमहर्षी डॉ.बापूजी साळुंखे
M Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY
(2020-21)

Online Remedial Coaching Class:

Remedial coaching class facility is available in the department for the slow learners and all categories can also join these classes. All the three classes are organized by taking the components of the curriculum which the students of history find difficult. Usually, these classes run in both semesters each year. Lectures on each topic are given in each term. After the lecture, the students are assessed by taking tests and guided accordingly. Students are also given study materials and reference books for improvement. The aim of coaching classes is to get good results in university exams.

The department has started coaching for students who have not been able to pass certain courses as well as those who want to improve their performance in certain courses. The aim is to improve the performance and quality of students and enhance their abilities. Counseling for remedial coaching was established in the college to enable students who need therapeutic coaching to proceed to a higher level of efficiency and to reduce their failures and reduce the dropout rate to the required level. Coaching at the undergraduate level. The college strives to develop students' specific abilities that can contribute to their academic success. It is very important to guide the students in this regard. Professors try to guide students informally. Teachers help students understand the undergraduate program guidelines and explain various aspects of the curriculum. They constantly encourage students to come forward with their ideas at all stages of development. Faculty members can get in touch and help advance academic goals by sharing guidance, experience and skills. Students come from increasingly diverse backgrounds, so teachers communicate personally and try to help them adapt to change in a beautiful way. It also lowers stress levels and boosts students' confidence.

Objectives:

1. Identify slow students based on previous exams as well as tests taken in classes.
2. Selecting the study area of their difficulties and arranging hours to overcome these difficulties.
3. Department of History to conduct classes on selected subjects.
4. To reduce the drop-out ratio.
5. To strive for the holistic development of the students.
6. To meet their educational requirements.
7. Focus on students in rural and hilly areas.
8. To ascertain how beneficial therapeutic coaching classes have been.

Highlights of Coaching Classes:

1. The attraction about difficult subjects.
2. Increase of knowledge level.
3. The positive impact of coaching.
4. Planning of Curriculum and implementation as per schedule.
5. Empowering the slow learner to keep pace with academic challenges.
6. To bring the slow learners together with good learners.
7. To provide the remedy to the academic problems of the slow learners.
8. To give special attention to the problems of slow learners.
9. To improve the academic performance of the slow learners
10. To Increase students' confidence and competence level.
11. To improve the academic performance of the slow learners.

Following Subjects and concerned Faculties are involved in Remedial Coaching**Classes:**

Sr. No.	Name of the Faculty	Educational Qualification
1.	DR. HAJI NADAF	M. A., M.Phil, Ph. D.
2.	DR. V. D. CHORGE	M. A., Ph. D.SET
3.	MR. R. S KUMBHAR	M. A., SET/NET

Outcome:

Through this effort of the faculty, the slow learners are highly benefitted and are able to understand their lessons in a better way. Remedial teaching is a kind of follow up and it is done after an initial phase of teaching and evaluation. Remedial teaching as a part of curriculum is being done for the sake of slow learners with the aim of improving their understanding ability through easy and simple language exercises. Study materials are specially prepared for them for easier understanding. There was marked improvement in the level of achievement after the rigorous remedial teaching at regular intervals.

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Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY
(2020-21)

Online Remedial Coaching Class

Lecture Scheduled

T.Y. B. A.						
Day	Monday	Tuesday	Wednesday	Thursday	Friday	Saturday
Time	10.00 am - 11.00am	10.00 am - 11.00am	10.00 am - 11.00am	10.00 am - 11.00am	10.00 am - -11.00am	10.00 am - 11.00am
Teacher	Dr. Haji Nadaf	Dr. V. D. Chorge	Mr. R. S Kumbhar	Dr. Haji Nadaf	Dr. V. D. Chorge	Mr. R. S Kumbhar

S.Y. B. A.						
Day	Monday	Tuesday	Wednesday	Thursday	Friday	Saturday
Time	11.00 am - 12.00am	11.00 am - 12.00am	11.00 am - 12.00am	11.00 am - 12.00am	11.00 am - -12.00am	11.00 am - 12.00am
Teacher	Mr. R. S Kumbhar	Dr. Haji Nadaf	Dr. V. D. Chorge	Mr. R. S Kumbhar	Dr. Haji Nadaf	Dr. V. D. Chorge

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Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin- 416312
DEPARTMENT OF HISTORY
(2020-21)
Participant Student

Sr. No	PRN. NO.	GENDER	NAME OF THE STUDENTS	Class
1.	2019053264	M	BHALDARE SURAJ SUDAN	B.A. II.
2.	2019053248	M	BHANDARE SURAJ SHAILESH	B.A. II.
3.	2019053308	M	CHAVAN ANIKET PRUTHVIRAJ	B.A. II.
4.	2019053344	M	CHAVAN RAVINDRA GANESH	B.A. II.
5.	2019053304	F	*CHOUGULE RUTUJA JAYAVANT	B.A. II.
6.	2019053302	M	CHOUGULE SUSHANT RAJENDRA	B.A. II.
7.	2018038435	M	DEVKULE ASHITOSH NISHIKANT	B.A. II.
8.	2019053410	M	DEVKULE SHALMON SHAMUVEL	B.A. II.
9.	2019053478	F	*GHATAGE TEJAL ANIL	B.A. II.
10.	2019056587	M	JADHAV ABHISHEK ASHOK	B.A. II.
11.	2019054043	M	JADHAV MAYUR MANSING	B.A. II.
12.	2019054507	M	KAMBLE AKSHAY SHESHRAO	B.A. II.
13.	2019054511	M	KAMBLE ASHUTOSH RAJENDRA	B.A. II.
14.	2019054402	F	*KAMBLE NIVEDITA NANDKUMAR	B.A. II.
15.	2019053227	F	*KAMBLE SONALI SANGAPPA	B.A. II.
16.	2019054548	M	LANDAGE ASHISH PANDIT	B.A. II.
17.	2019054588	F	*LANDAGE SAYALI KASHINATH	B.A. II.
18.	2019054554	F	*LANDAGE SAYALI VIKAS	B.A. II.
19.	2019056238	M	MANE SAURABH CHANDRAKANT	B.A. II.
20.	2019055964	M	MANDLE SUMIT PRAKASH	B.A. II.
21.	2019056343	M	MORE RAJKIRAN HANMANT	B.A. II.
22.	2019055520	M	NYAYNIRGUNE ABHISHEK RAVINDRA	B.A. II.
23.	2019056483	M	PATIL HARSHVARDHAN SANJAY	B.A. II.
24.	2019055618	M	PATIL NIKHIL BHARAT	B.A. II.
25.	2019056820	M	PATIL OMKAR POPAT	B.A. II.
26.	2019055657	M	PATIL PRATIK PRAKASH	B.A. II.

27.	2019055626	M	SALUNKHE MANISH SUNIL	B.A. II.
28.	2019056606	F	*PATIL SARIKA BALASAHEB	B.A. II.
29.	2019055503	F	*SHETE RADHIKA RAVASO	B.A. II.
30.	2019055648	M	SHINDE PRATHMESH SUNIL	B.A. II.
31.	2019058611	M	SHINDE ABHIJIT MANIK	B.A. II.
32.	2019055647	M	SHINDE DIGVIJAY PRITHVIRAJ	B.A. II.
33.	2019055651	F	*SHINDE RUTUJA SAMPAT	B.A. II.
34.	2019055572	M	SHINDE VAIBHAV RAMCHANDRA	B.A. II.
35.	2019056790	M	TIKKE OMKAR ANIL	B.A. II.
36.	2018038439	M	FAKIR ALFAJ IQBAL	B. A. III
37.	2018038456	M	GURAV SHUBHAM SHIVAJI	B. A. III
38.	2018038475	M	HAJARE AKSHAY KISAN	B. A. III
39.	2018038610	M	KOKATE PRAVIN VAMAN	B. A. III
40.	2018038626	M	KUMBHAR TUSHAR DIPAK	B. A. III
41.	2018043690	M	MALI OMKAR BALASAHEB	B. A. III
42.	2018042594	M	MANE VISHWJEET DIPAK	B. A. III
43.	2018043704	M	AMIT SUKHADEV PATIL	B. A. III
44.	2018038935	F	*PATIL DIPALI JAGANNATH	B. A. III
45.	2018043703	M	PATIL ADESH TANAJI	B. A. III


HEAD
Department of History
P.D.V.P.College, Tasgaon

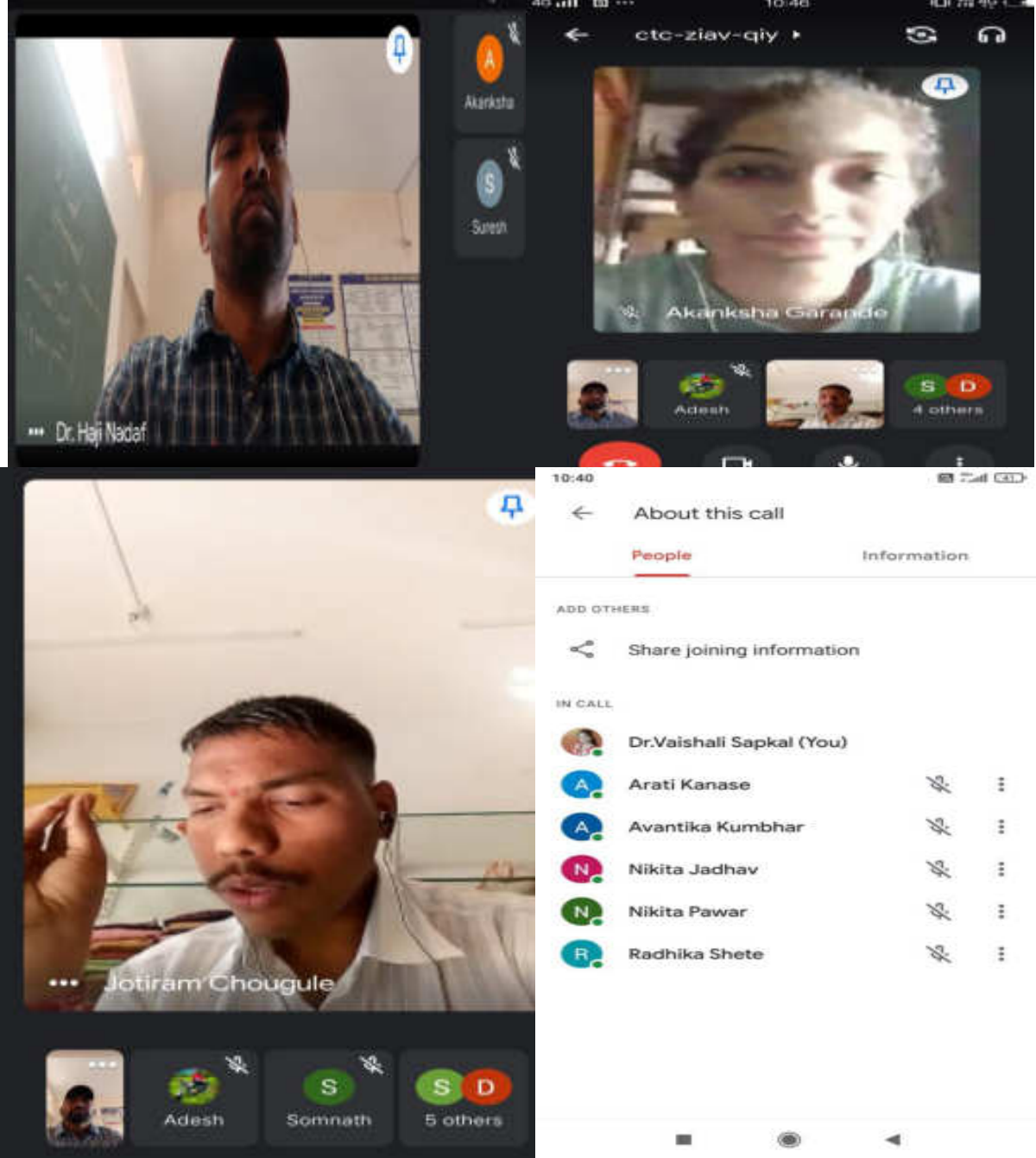
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Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin- 416312
DEPARTMENT OF HISTORY
(2020-21)
Student Online Registration

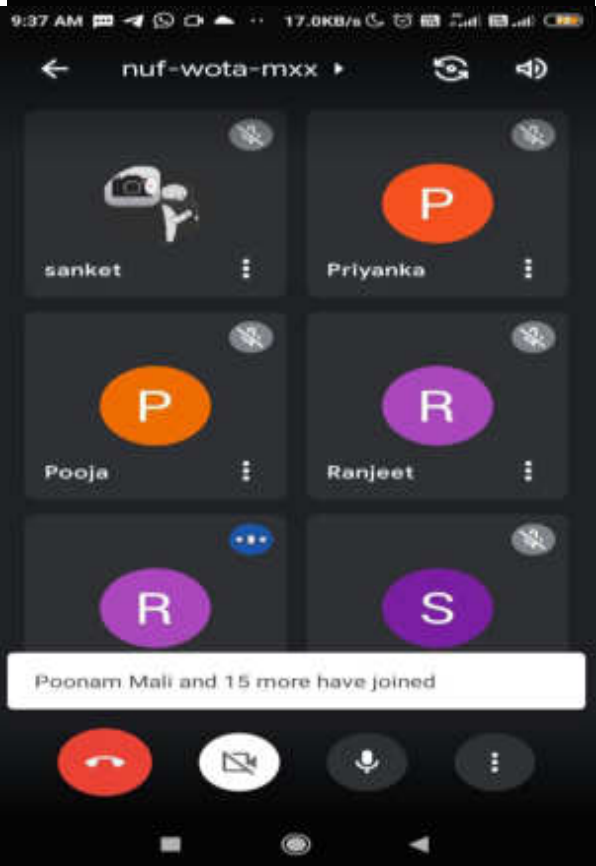
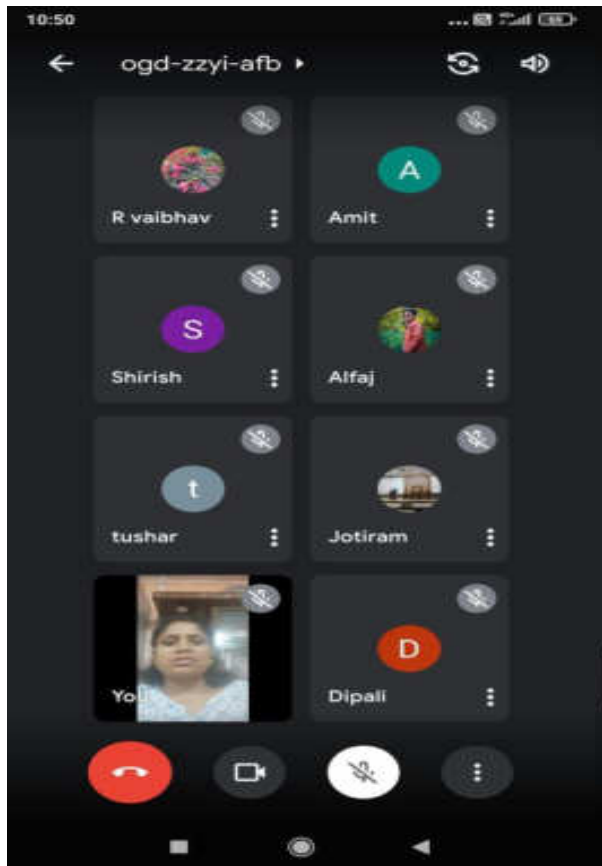
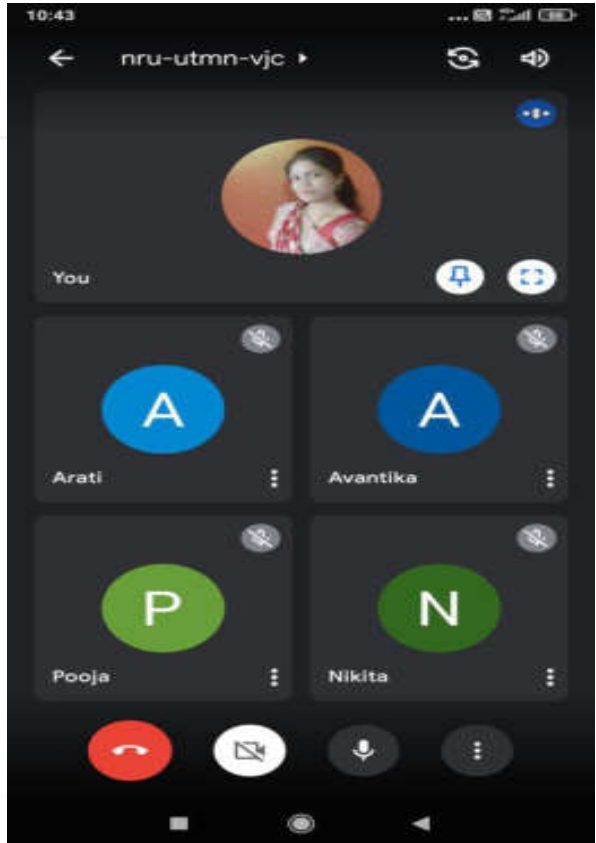
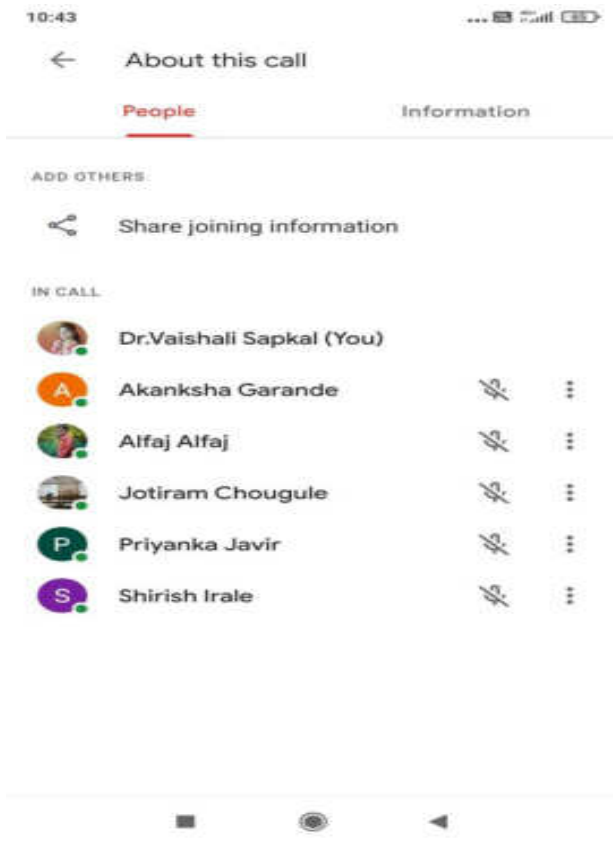
Timestamp	Name of the Student	Roll No./PRN	Email ID	Mobile No.	Subject
5/7/2021	Rutik mandale	3110	rutikmandale61@gmail.com	9730231815	History
5/7/2021	Suraj Sudan bhalbare	0	Surajbhalbare131@gmail.com	9860622332	History
5/7/2021	Vaibhav shinde	0	vaibhavshinde5338@gmail.com	9284135110	History
5/7/2021	Manish sunil salunkhe	3192	manishsalunkhe16@gmail.com	9975607276	History
5/7/2021	Onkar popat patil	3158	Omkupatilbrand406@gmail.com	9970861860	History
5/7/2021	Ghatage Tejal Anil	3250	tejalghatage067@gmail.com	7058684147	History
5/7/2021	Landage sayali kashinath	3090	Saylilandage60@gmail.com	7558315405	History
5/7/2021	Rutuja sampat shinde	10032	rushikeshs702@gmail.com	9130365258	History
5/7/2021	Radhika raosaheb shete	3205	radhikashete18@gmail.com	7620424238	History
5/7/2021	Patil sarika balasaheb	9827	Sarikspatil676@Gmail.com	9022755461	History
5/7/2021	Kamble nivedita Nandkumar	3070	Knivedita987@gmail.com	9579714816	History
5/7/2021	Kamble akshay shashrao	3063	akshaykamble5246@gmail.com	8010595967	History
5/7/2021	Jadhav mayur mansing	2019054043	Mayurjadhav3588@gmail.com	9309967439	History
5/7/2021	Tikke Omkar Anil	3286	omkaromi1137@gmail.com	9309818814	History
5/7/2021	Pramod ravso zende	65166	Pramodzende7499@gmail.com	7218286705	History
5/7/2021	Aniket Pruthviraj Chavan	3012	aniketchavan3483@gmil.com	7447461766	History
5/7/2021	Ashish Suresh Gaikwad	9898	gaikwadashish449@gmail.com	8380889480	History
5/7/2021	Suraj Erandole	65426	surajse1704@gmail.com	9284794245	History
5/7/2021	Pravin Vaman Kokate	2018038610	pravinkokate1011@gmail.com	7620003628	History
5/7/2021	Tushar Dipak Kumbhar	2018038626	tusharkumbhar4414@gmail.com	7820854468	History
5/7/2021	Alfaj Iqbal fakir	2018038439	alfajmakandar0000@gmail.com	07304616172	History
5/7/2021	Omkar Balasaheb Mail	2018043690	omkarmali832@gmail.com	9146117268	History
5/7/2021	Dipali jagannath Patil	2018038935	pdipali561@gmail.com	8180960517	History
5/7/2021	Mane vishwjeet Dipak	2018042594	vishumane345@gmail.com	9112854846	History

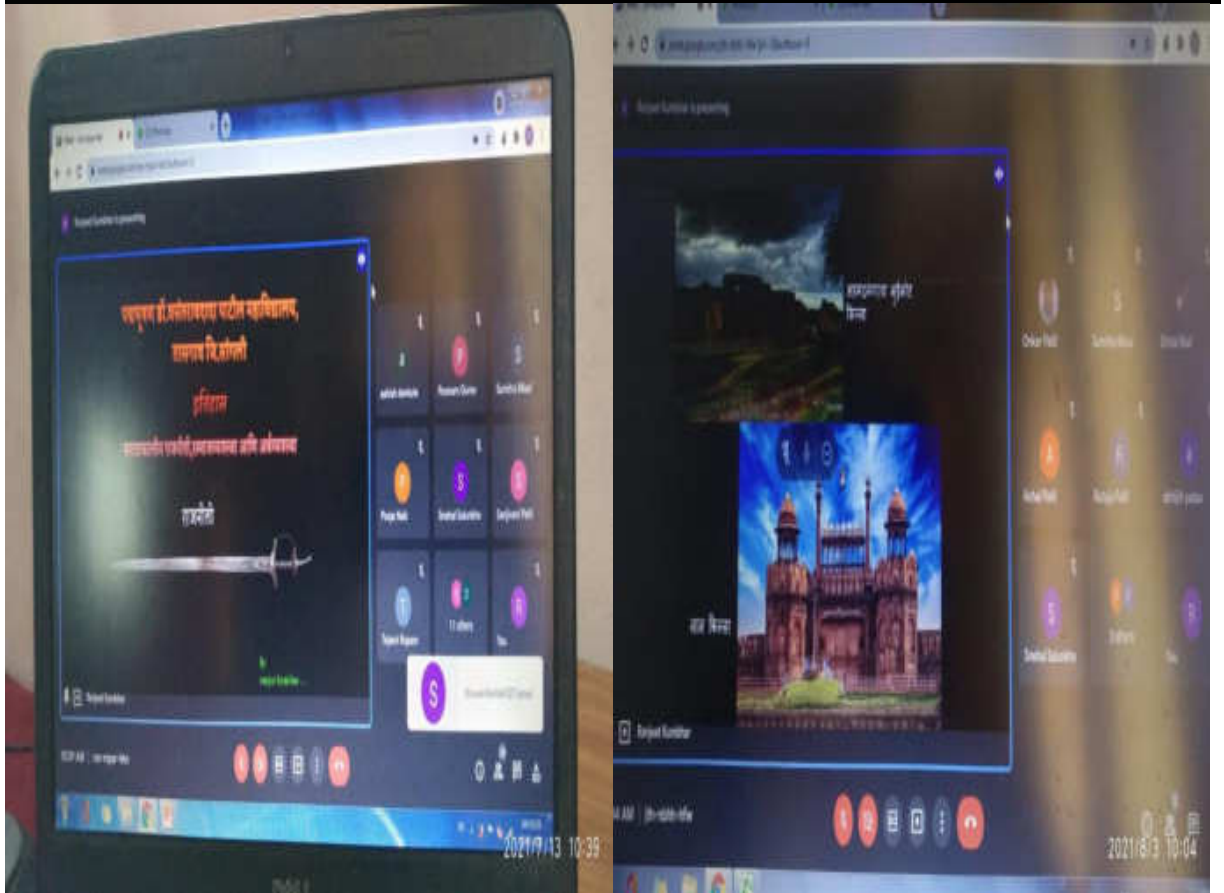
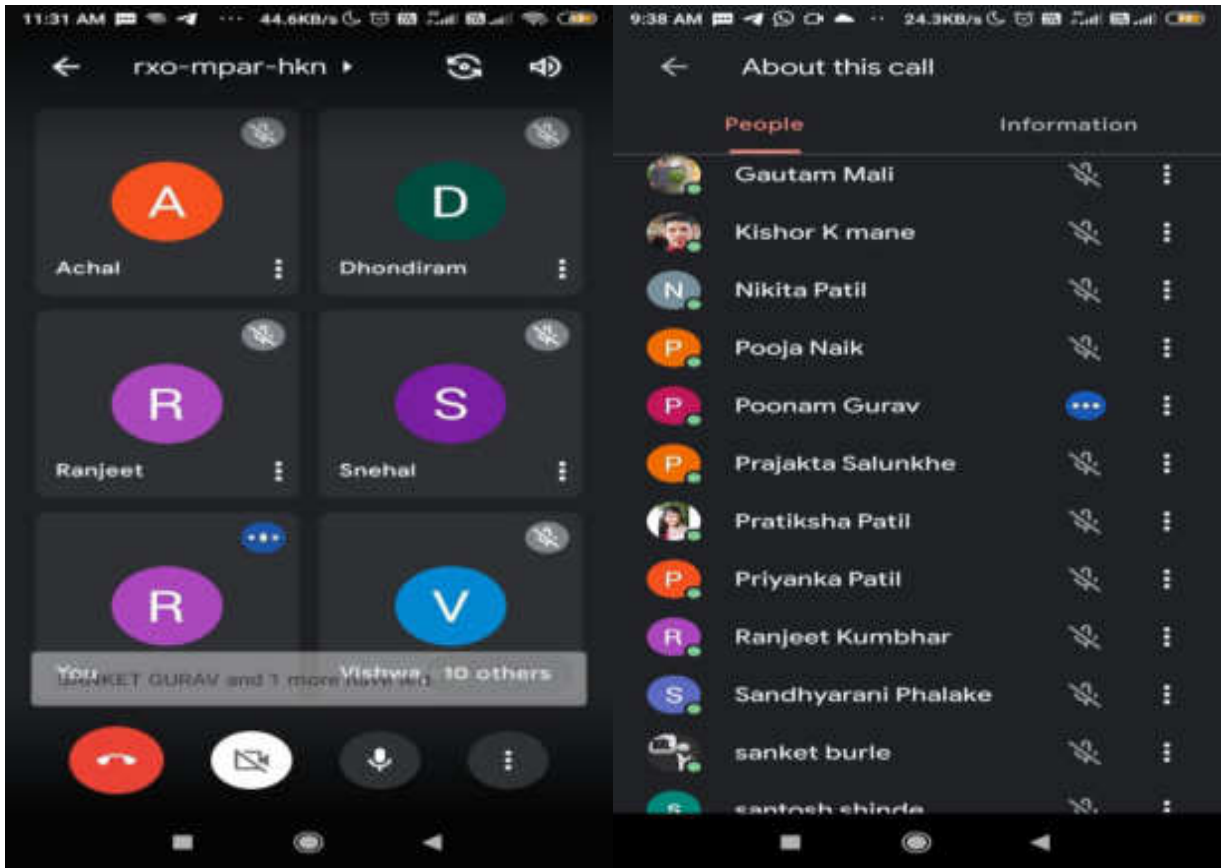
5/7/2021	Adesh Ptil	2018043703	ap539515@gmail.com	7057926689	History
5/7/2021	Amit sukhadev patil	2018043704	dattapatil4577@gmail.com	9373418494	History
5/7/2021	shubham gurav	2018038456	shubhamgurav053@gmail.com	9356053048	History

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” - शिक्षणमहर्षी डॉ.बापूजी साळुंखे
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Tasgaon, Dist- Sangli, Pin- 416312
DEPARTMENT OF HISTORY
(2020-21)

ONLINE REMEDIAL COACHING CLASSES







‘Remedial Coaching’

*Slow Learner and
Advance Learner*

Date: 16/07/2018

**A Report Submitted to
IQAC, Internal Quality Assurance Cell
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312**

**Organized by
Department of History**

**Course Coordinator
Dr. Haji Nadaf**

**Head
Dr.H. D. Nadaf
Department of History
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon,**

2018-2019

Date 13/07/2018

To
The Principal
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli.416312

Subject: Application for the Permission to conduct 'Slow and
- Advance Learner Test'.

Respected sir,

Our department of History is going to conduct 'Slow and Advance
Learner' test on F. Y. B. A. on Monday, 16/07/2018. We are kindly requesting
you to allow us to conduct this activity. Students of B.A. Part- I., are going to
participate in the same activity.

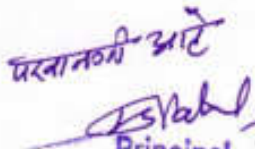
Thanking you

Yours Faithfully


(Dr. H. D. Nadaf)

HEAD
Department of History
P.D.V.P.College, Tasgaon.

Organizer: Dr. H. D. Nadaf


Principal
Padmabhushan Dr. Vasantrodada Patil
Mahavidyalaya, Tasgaon, (Sangli)

Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2018-19)

'Remedial Coaching'

Notice

Date: 13/07/2018

All students of F.Y. B.A. History hereby informed that we are going to conduct '**Remedial Coaching**' Slow and Advance Learner' Class Test on Tuesday, 16/07/2018. The test will be related to History subjects and this test is based on 20 multiple choice questions for 20 marks. It is compulsory to all students to participate with preparation. All students of F.Y. B.A. History hereby informed that we are going to conduct


HEAD
Department of History
P.D.V.P.College, Tasgaon.

(Dr. H. D. Nadaf)

ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण
-शिक्षणमहर्षी डॉ.बापूजी साळुंके
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312
DEPARTMENT OF HISTORY
2018-2019

'Slow and Advance Learner Test'

Day: Tuesday,
Date: 10 July 2018.

B.A.I
(Each question for 02 Marks)

Signature: D.N. Chaudhari
Roll No: 2059

1. सांगली येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोसले क. डफळे ड. निबाळकर
2.हे मुंबई प्रांताचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील ब. यशवंतराव चव्हाण क. बाळासाहेब खेर ड. वसंतराव नाईक
3. डॉ. कोटणीस यांचा जन्म.....जिल्ह्यात झाला.
अ. सातारा ब. सोलापूर क. सांगली ड. कोल्हापूर
4. अजंठा व वेरूळ हे ठिकाणया जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी
5. जागतिक किर्तीचे ताजमहाल शहरात आहे.
अ. गुलबर्गा ब. अग्रा क. हैद्राबाद ड. सोलापूर
6. अकबरसाम्राज्याचा सत्ताधीश होता.
अ. नंद ब. मुघल क. गुप्त ड. सातवाहन
7. छ. शिवाजी महाराजांचे राज्याभिषेक.....या किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रायगड
8. सोलापूर जिल्हा पिकासाठी प्रसिद्ध आहे.
अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. ज्वारी
9. बहमनी साम्राज्याची राजधानीयेथे होती.
अ. पट्टकल्ल ब. हम्पी क. गुलबर्गा ड. बदामी
10.या साली गोवा राज्याची निर्मिती झाली.
अ. १ मे १९६० ब. ३० मे १९८७ क. १ एप्रिल १९६० ड. १ जून १९६०

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

B. A. I

2018-2019

Slow Learner and Advance Learner

Date: 16/07/2018

SR.NO.	ROLL CALL	NAME OF THE STUDENT	SIGNATURE
1.	2560	Garande Atanksha Ashok	Garande
2.	2570	Koli Ajay Adasa	(Koli)
3.	2501	Madane Aniket Sanjay	Madane
4.	2530	Mane Sanket Sanjay	Mane
5.	2690	Hajare Akshay Kisan	Hajare
6.	2686	Chougule Jodram Chanan.	Chougule
7.	2684	Yedake Anol Kisan	A. K. Yedake
8.	2679	Sirsath Akshay Dilip	Sirsath
9.	2671	Shende Vishal Janardhan	Shende
10.	2669	Shaikh Samir Jamal	Shaikh
11.	2657	Patil Shashikant Prakash	Patil
12.	2652	Patil Shankar Himmatrao	Patil
13.	2641	Patil Mayur Arvind	Patil
14.	2635	Patil Amit Suktdev	Patil
15.	2532	Patil Adesh Ranaji	Patil
16.	2018	Bhosale Guresh Bhagwan	Bhosale
17.	2025	Bobade Pratik Dilip.	P. D. Bobade
18.	2026	Bhitorale Parvati	Bhitorale
19.	2034	Charan Ganesh Prathiraj	Charan
20.	2044	Charan Sandesh Suresh	Charan

21.	2047	Chavan Vaibhav Harmanant	Chavan
22.	2049	Chavan Vijay Prakash	D. Prakash
23.	2051	Chavan Vishal Vikhal	Chavan
24.	2052	Choudhari Dadasideri N.	D. Choudhari
25.	2075	Potir Alfaj Iqbal.	Potir
26.	2085	Gujale Vandana Shripati	Gujale
27.	2087	Guar Shankam Shivaji	Guar
28.	2093	Jhar Anandh Netaji	Jhar
29.	2090	Jhar Megha Raju	Jhar
30.	2105	Jadhav Yogesh Parindra	Jadhav
31.	2112	Jawale Rohit Ramch	Jawale
32.	2113	Joshi Abharv Anant.	Joshi
33.	2122	Kamble Maagal Suresh	Kamble
34.	2136	Kharade Rushikesh Keraba	Kharade
35.	2139	Khot Chetana Suryakant.	Khot
36.	2144	Kokate Dhanraj Varman	Kokate
37.	2155	Kumbhar Tuskar Dipak.	Kumbhar
38.	2160	Lokhande Manish Sanjay	Lokhande
39.	2168	Mali Mayuraj Jaykumar.	Mali
40.	2181	Mandke Mahari Dhondiram.	Mandke
41.	2193	Mane Ujjwala Anjum	Mane
42.	2208	Mankari Dhanraj Balekaran	Mankari
43.	2210	Mulani Jubay Jahangir.	Mulani
44.	2216	Mulla Ibrahim Mahammadani	Mulla
45.	2233	Patil Ananda Prakash.	Patil
46.	2236	Patil Chaitanya Prakash.	Patil
47.	2218	Mane Vishwajeet Dipak.	Mane
48.	2210	Manale Abhijeet Rajkumar.	Manale
49.	2218	Mane Vishwajeet Dipak.	Mane
50.	2218	Manale Abhijeet Rajkumar.	Manale

51.	2405	Mali Omkar Balisakeb.	Mali
52.	2408	Mandale Abhijeet Rajkumar.	Mandale
53.	2371	Kamble Pratikraj Suresh	Kamble
54.	2387	Vethar Suhag Uttam.	Vethar
55.	2382	Jrale Shresh Prachant.	Jrale
56.	2373	Derkule Ashish Nishikant.	Derkule
57.	2370	Deshmukh Rajit Rajaram.	Deshmukh
58.	2357	Wagh Swapnil Uttam.	Wagh
59.	2309	Shinde Kshat Prakash.	Shinde
60.	2324	Shinde Ganesh Dny	Shinde
61.	2316	Shende Dipali Chankant.	D.C. Shende
62.	2305	Salunke Shiraj Yashkant.	Salunke
63.	2294	Pawan somnath Shrikaji	Pawan
64.	2287	Patil Vaibhav Vishrant.	Patil
65.	2283	PATIL TEJAS NAVNATH	Patil
66.	2278	Patil Sonali Ganesh	S. G. Patil
67.	2259	Patil Pranati Suresh.	Patil
68.	2254	Patil Pranati Anant	P. A. Patil
69.	2237	Patil Purnod Papat.	Patil
70.	2241	Dipali Jagannath Patil.	D. J. Patil
71.			
72.			
73.			
74.			
75.			
76.			
77.			
78.			

"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार" -शिक्षणमहर्षी डॉ.बापूजी साळुंखे Shri
Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit

Padmabhushan Dr. Vasant Rao Dada Patil Mahavidyalaya

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

Slow Learner and Advance Learner

Date: 16/07/2018

No	Roll No.	Gender M/F	Name of the Student	Mark
1	2018	M	Bhosale Gauresh Bhagwan	08
2	2025	M	Bobade Pratik Dilip	08
3	2026	M	Budhavale Pavan Anil	06
4	2034	M	Chavan Ganesh Pruthviraj	08
5	2044	M	Chavan Sourabh Suresh	10
6	2047	M	Chavan Vaibhav Hanmant	14
7	2049	M	Chavan Vijay Prakash	18
8	2051	M	Chavan Vishal Vitthal	14
9	2052	F	Choudhari Dadmidevi Narsirma	12
10	2075	M	Fakir Alfaj Iqbal	10
11	2085	F	Gujale Archana Shripati	06
12	2087	M	Gurav Shubham Shivaji	08
13	2093	M	Jadhav Avinash Netaji	08
14	2097	F	Jadhav Megha Raju	12
15	2105	M	Jadhav Yogesh Ravindra	08
16	2112	M	Jawale Rohit Ramesh	18
17	2113	M	Joshi Atharv Anant	14
18	2122	F	Kamble Mangal Suresh	12
19	2136	M	Kharapude Rushikesh Keraba	14
20	2139	F	Khot Chetana Suryakant	18
21	2144	M	Kokate Pravin Vaman	16
22	2155	M	Kumbhar Tushar Dipak	12
23	2160	M	Lokhande Monish Sanjay	10
24	2168	F	Mali Mayuri Jaykumar	12
25	2181	M	Mandle Malhari Dhondiram	06
26	2193	F	Mane Ujwala Arjun	08
27	2209	M	Mulani Arbaj Balekhan	10
28	2210	M	Mulani Juber Jahangir	14
29	2216	M	Mulla Ibrahim Mahmadhanif	10
30	2233	M	Patil Ananda Prakash	18
31	2238	M	Patil Chaitnya Babaso	10

32	2241	F	Patil Dipali Jagannath	08
33	2253	M	Patil Pramod Popat	06
34	2254	F	Patil Pranali Amrut	10
35	2257	F	Patil Pranoti Suresh	12
36	2278	F	Patil Sonali Ganesh	18
37	2284	M	Patil Tejas Navanath	14
38	2287	M	Patil Vaibhav Vishrant	12
39	2294	M	Pawar Somnath Shahaji	16
40	2304	M	Salunkhe Dhiraj Yashavant	04
41	2318	F	Shendge Dipali Chandrakant	04
42	2321	M	Shinde Ganesh Dilip	06
43	2329	M	Shinde Vishal Bhaskar	08
44	2353	M	Wagh Swapnil Uttam	10
45	2370	M	Deshmukh Ranjit Rajaram	08
46	2373	M	Devkule Ashitosh Nishikant	12
47	2382	M	Irale Shirish Ramchandra	14
48	2387	M	Jadhav Suhas Uttam	18
49	2391	M	Kamble Pratikraj Surendra	16
50	2405	M	Mali Omkar Balasaheb	14
51	2408	M	Mandale Abhijeet Rajkumar	10
52	2418	M	Mane Vishwjeet Dipak	14
53	2432	M	Patil Adesh Tanaji	18
54	2435	M	Patil Amit Sukhadev	16
55	2441	M	Patil Mayur Arvind	06
56	2452	M	Patil Shankar Himmatrao	08
57	2453	M	Patil Shashikant Prakash	12
58	2469	M	Shaikh Samir Jamal	04
59	2471	M	Shendage Vishal Janardhan	08
60	2479	M	Sirsath Akshay Dilip	12
61	2484	M	Yedake Amol Kisan	18
62	2486	M	Chougule Jotiram Gajanan	16
63	2490	M	Hajare Akshay Kisan	12
64	2500	M	Koli Ajay Appaso	16
65	2501	M	Madane Aniket Lalaso	12
66	2530	M	Mane Sanket Sanjay	04
68	2540	M	Garande Akanksha Ashok	06

Sheela
HEAD

Department of History
P.D.V.P.College, Tasgaon.

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” - शिक्षणमहर्षी डॉ.बापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2018-2019)

‘Remedial Coaching Report’

Title of Programme	: Slow and Advance Learner Test
Organizing Department	: History
Date	: 15th Jan. 2019
Venue	: Lecture Room. No.26
No. of Participants	: 68 (Male:56, Female:12)
Coordinator	: Dr. Haji Nadaf
President	: Prin. Dr. R.R. Kumbhar

The Department of History conducted remedial coaching-‘Slow and Advance Learner’ activity on F.Y.B.A., on **10/07/2018**. Total 68 Students were participated in this same activity. The test was basis on Indian history and Culture.

The Students of B.A.I were participated in this activity and they have responded to well presence. Students are given their knowledge by aptitude exam. In this test 85 percent students have passed. We have divided into two group as slow learner group and advance learner group by present students. Inserts are in the forefront for both of these groups to be independently guided in the future. This entire activity conducted by Dr. H. D. Nadaf for this activity. Prin. Dr. R.R. Kumbhar has guided and cooperated well for the programme.


HEAD
Department of History
P.D.V.P.College, Tasgaon.

(Dr. H. D. Nadaf)



‘Dissemination of Education through Knowledge, Science and Culture’-Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur’s
PADMABHUSHAN Dr. VASANTRAODADA PATIL
MAHAVIDYALAYA, TASGAON
Tal. Tasgaon, Dist.: Sangli.

DEPARTMENT OF COMMERCE

2020-21

Report of
ONLINE REMEDIAL COACHING
ON
FINANCIAL ACCOUNTING



Title of Programme	REMEDIAL COACHING
Organizing Department	COMMERCE
Date	08 th August To 13 th August 2021.
Venue	Online
No. of Participants	91: Male 43 Female: 48

The department of commerce organized online one week remedial coaching classes from 08th August to 13th August 2021. These special classes are conducted outside the regular timetable of class hour, Students belonging to B. Com. I class benefit of these coaching classes. Remedial coaching is to be organized at undergraduate level with a view to improving academic skills and linguistic proficiency of the students in the subject of financial accounting.

“ ज्ञान, विज्ञान आणि सुरतकार यांसाठी शिक्षणप्रसार ” - शिक्षणमहर्षी डॉ. बापूजी साळुंखे

श्री स्वामी विवेकानंद शिक्षण संस्था, कोल्हापूर संघलित

पद्मभूषण डॉ. वसंतरावदादा पाटील महाविद्यालय

तासगांव, जि. सांगली, पिन - ४१६ ३१२ ✆ STD : ०२३४६ - २४० ६६५, २४० ५४५ FAX : २४०५४५

• शिवाजी विद्यापीठ, कोल्हापूर संलग्न •

ई-मेल : san.pdvpm.tas@gmail.com वेबसाईट : www.pdvpm.tasgaon.edu.in

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SEACH X

जि.: C-8

नेटक मूल्यांकन वेगळी "A" (2.75)

शिक्षणमहर्षी डॉ. बापूजी साळुंखे
जी.डी.डी.डी.डी.डी.

मा. चंद्रकांत (दादा) पाटील
अध्यक्ष जी.डी.डी.

प्राचार्य अश्वत्थकृष्ण साळुंखे
ए.ए.टी.

प्राचार्या सौ. शुभांगी गावडे
ए.ए.टी.डी.डी.डी.

डॉ. मिलिंद एस. हजरे
ए.ए.टी.डी.डी.डी.

जावक क्र. : पी.डी.व्ही.पी.एम.टी./

दिनांक :

सूचना

दि. ०६.०८.२०२१

वी.कॉम.भाग एक माधिल विद्यार्थ्यांना कळविण्यात येते, वाणिज्य विभागाकडून **Financial Accounting Paper- I** या विषयासाठी **Remedial Coaching** दिनांक ०८.०८.२०२१ ते ११.०१.२०२१ या दरम्यान घेतले जाणार आहे. तरी ज्या विद्यार्थ्यांना सदर कोर्ससाठी सहभाग घ्यावयाचा आहे त्या विद्यार्थ्यांनी **Whatsapp Group** मध्ये सहभागी व्हावे.


डॉ. यशवंतराव ए.जी.
(वाणिज्य विभाग प्रमुख)


डॉ. मिलिंद हजरे
प्राचार्य
पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगांव (जि. सांगली.)

WHATSAPP GROUP LINK

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REMEDIAL COACHING B.COM.I

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8/8/2021

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+91 79722 98866 joined via an invite link

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B. Com. (CBCS Pattern)
Part-I (Semester I)
Financial Accounting Paper I
Core Course

Introduced from June-2018-19

Credits : 4

Course Inputs :

Unit I : Basic Accounting Concepts and Conventions, Accounting Process, Accounting Standards-Need & Procedure, Concept of IFRS (15 Hours)

Unit II - Amalgamation of Partnership Firms (15 Hours)

Unit III : Consignment Accounts-Important Terms and Accounting in the books of Consignor and Consignee (15 Hours)

Unit IV - Accounts of Professionals –preparation of Receipts and Expenditure Account and Balance sheet of Medical Practitioners and Professional Accountants. (15 Hours)

PADMABHUSHAN DR VASANTRAODADA PATIL

MAHAVIDYALAYA, TASGAON

DEPARTMENT OF COMMERCE

REMEDIAL COACHING 2020-21


B.COM.I


FINANCIAL ACCOUNTING – I

TIME TABLE 2020-21

Time	Name of Teacher	Date					
		8.08.2021	9.08.2021	10.08.2021	11.08.2021	12.08.2021	13.08.2021
9:30am to 11:00	Dr. Sonawale A.G.	Basic Accounting Concepts		Amalgamation of Partnership Firm		Consignment Accounts	
	Mr. Patil G.R.		Accounting Process & Standards		Amalgamation of Partnership Firm		Consignment Accounts

- Practice Exam (14.08.2021)


Dr. Amol Sonawale
HEAD
Department of Commerce
P.D.V.P.College, Tasgaon.


Dr. Milind Hujare
Principal
Padmabhushan Dr. Vasantrodada Patil
Mahavidyalaya, Tasgaon. (Sangli)

STUDY MATERIAL

PADMABHUSHAN DR VASANTRAODADA PATIL MAHAVIDYALAYA,

TASGAON

DEPARTMENT OF COMMERCE

REMEDIAL COACHING 2020-21

B.COM.I

FINANCIAL ACCOUNTING – I

STUDY MATERIAL

- 1) Anticipate no profit but provide for all possible losses while recording business transactions.
- A) Conservatism
B) Full Disclosure
C) Consistency
D) Materiality Convention
- 2) IASC means
- A) International Accounting Standards Committee
B) Internal Accounting Standards Committee
C) Intentional Accounting Standards Committee
D) None of the above
- 3) The determination of expenses for an accounting period is based on the principles of –
- A) Objectivity
B) Materiality
C) Matching
D) Periodicity
- 4) A business is considered to be having an indefinite life according to
- A) Accounting period concept
B) Cost concept
C) Going concern concept
D) Dual aspect concept
- 5) IFRS means
- A) International Financial Reporting Standards
B) Internal Financial Reporting Standards
C) Intentional Accounting Financial Reporting Standards
D) None of the above
- 6) Depreciation accounting is a
- A) AS – 4
B) AS - 5
C) AS – 6
D) AS - 7

7) Retirement benefits is a

- A) AS - 14
- C) AS - 16

- B) AS - 15
- D) AS - 17

8) ASB established in

- A) 1978
- C) 1973

- B) 1977
- D) 1974

9) International Accounting Standards Committee came into existences

- A) 1978
- C) 1973

- B) 1977
- D) 1974

10) Amalgamation is a

- A) AS - 14
- C) AS - 17

- B) AS - 15
- D) AS - 18

11) In amalgamation -

- A) At least two companies are involved
- C) Only two companies are involved

- B) At least one companies are involved
- D) None of the above

12) If goodwill account is not to be maintained in the books of new firm, the amount of goodwill is debited to all partners' capital account in their -

- A) Old profit & loss ratio
- C) Old partners capital ratio

- B) New profit & loss ratio
- D) New partners capital ratio

13) If any assets not taken over by new firm, it is distributed to both partners in their

- A) Old profit & loss ratio
- C) Old partners capital ratio

- B) New profit & loss ratio
- D) New partners' capital ratio

14) If goodwill account is written off in the books of new firm then total amount of goodwill is debited to -

- A) Revaluation A/c
- C) Cash / bank account

- B) Combined partners capital Account
- D) New firm account

15) Reduction in any liabilities

- a. Del credere commission
C) Normal commission

- B) Special commission
D) Gross commission

80.- Goods sent on consignment account is -

- a. Nominal account
C) Personal account

- B) Real account
D) Representative personal account

81.- If the loss of discount is treated as a financial loss -

- a. It is not charged to consignment account
b. It is charged to consignment account
c. It is charged goods sent to consignment account
d. None of the above

82. If Del -credere commission is given to the consignee, regarding bad debts -

- a. No entry is to be passed by the consignor
b. Entry is to be passed by the consignor
c. Entry is to be passed to the goods sent to consignment
d. No Entry is to be passed to the goods sent to consignment

83. - Non-recurring expenses of consignee is considered in calculation of abnormal loss -

- A) When goods are under the custody of consignee
B) When goods are under the custody of consignor
C) When goods are not under the custody of consignee
D) None of the above

84. In order to introduce a new product in the market an additional commission is given to the consignee.


- a. Del credere commission
C) Gross commission


- B) Special commission
D) Over-riding commission

85. - When abnormal loss of goods, if goods are insured and insurance company paid the amount of claim -

- A) Cash /bank account is credited
C) Consignment account is debited

- B) Insurance claim is credited
D) Abnormal loss account is debited


Dr. Alka Inamdar
IQAC & NAAC Coordinator
IQAC Co-Ordinator,
P.D.V.P. Mahavidyalaya,
Tasgaon.


Dr. Amol Sonawale
HOD, Commerce Department
HEAD
Department of Commerce
P.D.V.P.College, Tasgaon.


Dr. Milind Hujare
Principal
Padmabhushan Dr. Vasantraodada Patil
Mahavidyalaya, Tasgaon. (Sangli)

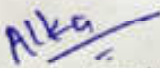


Practice Examination Result

Sr. No.	FULL NAME	Marks Out of 50
1	Sahil Anil Patil	38
2	Sarade Nikita Ajit	40
3	Rutuja Sachin Jamadade	38
4	Kale Tejaswini Dhanaji	34
5	Pranav Pradip Patil	42
6	Gurav Samiksha Sandip	28
7	Sayali Subhash Patil	36
8	Kamble Prathmesh Arvind	38
9	Prajwal Ajay Kadam	40
10	Patil Sanket Adhikrao	38
11	Dattatray Tukaram Sapkal	36
12	Sakshi Satish Kshirsagar	38
13	Swapnil Somshekhar Padsalmani	34
14	Suraj Hari Patil	40
15	Patil Radhika Girish	38
16	Rushikesh Vijay Jamdade	38
17	Amol Ananda Salunkhe	40
18	Savali Sudam Mohite	32
19	Mohite Prajakta Dilip	32
20	Pratik Dadaso Rajmane	42
21	Ghodake Sanket Sanjay	44
22	Aishwarya Maruti Chavan	38
23	Keshav Dipak Mane	40
24	Nikita Vijay Patil	36
25	Suraj Jagannath Bhosale	38
26	Ankita Balu Kamble	38
27	Prajwal Vinayak Kumbhar	40
28	Patil Priyanka Ganesh	36
29	Aditi Sachin Patil	40
30	Sandip Bhagavan Salunkhe	40
31	Vaishali Anil Mali	40
32	Pawar Suraj Nivas	36
33	Shinde Snehal Vijay	40
34	Jagatap Tanuja Kisan	36
35	Anisha Tanaji Dubal	40
36	Sourabh Jagannath Patil	32
37	Pranali Ananda Mali	36
38	Bhagyshri Shivaji Nikam	38
39	Adarsh Anandrao Patil	44
40	Neha Mahadv Koli	40

41	Patil Amruta Balaso	40
42	Patil Dipali Sanjay	38
43	Sakshi Anil Shinde	38
44	Sahil Ismail Mulani	38
45	Tejashri Raghunath Budhavale	40
46	Pradnya Suresh Kamble	34
47	Kale Vaibhav Dilip	34
48	Akanksha Kuber Patil	36
49	Sandhyarani Ramachandra Kumbhar.	34
50	Mulani Madina Hamid	36
51	Nikita Shankar Thite	44
52	Madhugandha Pradip More	42
53	Pratiksha Arjun Sutar	34
54	Patil Amruta Adhikrao	36
55	Yogesh Shivaji Shinde	40
56	Rajan Kiran Adake	40
57	Sagar Balasaheb Wagh	32
58	Vanita Sushant Thokale	38
59	More Pratiksha Pandurang	42
60	Shirote Pooja Rajendra	36
61	Shridhar Dattatray Patil	40
62	Sakshi Kashinath Kumbhar	40
63	Ajay Ramesh Shinde	24
64	Chavan Sujata Ashok	36
65	Chavan Swapnil Sambhaji	40
66	Gavali Saurabh Santosh	36

REPORT

The online remedial coaching class conducted by the Commerce Department from 8th August 2021 to 13th August 2021 was successfully completed. Out of 91 students who participated in this, 66 students appeared for the practice exam. This class provided the students with in-depth information in the field of Financial Accounting. This will definitely benefit them for the university exams. Hon. Principal, Dr. Milind Hujare's guidance helped to complete this remedial coaching class. Mr. Gorknath Patil guided the students for this course.

 Dr. Alka Inamdar IQAC & NAAC Coordinator IQAC Co-Ordinator, P.D.V.P. Mahavidyalaya, Tasgaon.	 Dr. Amol Sonawale HOD, Commerce Department HEAD Department of Commerce P.D.V.P.College, Tasgaon.	 Dr. Milind Hujare Principal Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon. (Sangli)
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Special Programs for
Advanced Learners and Slow Learners

Date: 11/07/2019

A Report Submitted to
IQAC, Internal Quality Assurance Cell
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

Organized by
Department of History

Head
Dr.H. D. Nadaf
Department of History
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon,

2019-202

Date 09/07/2019

To
The Principal
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli.416312

Subject: Application for the Permission to conduct '**Slow and Advance
Learner Test**'.

Respected sir,

Our department of History is going to conduct '**Slow and Advance
Learner**' test on **F. Y. B. A. on Thursday, 11/07/2019**. We are kindly requesting you to allow us to conduct this activity. Students of B.A. Part- I., are going to participate in the same activity.

Thanking you

Yours Faithfully


Dr. Milind S. Hujare
Principal


HEAD
Department of History
P.D.V.P.College, Tasgaon.

“ज्ञान, विज्ञान आणि गुणस्कार यांसाठी शिक्षण प्रसार”-शिक्षणमहर्षी डॉ.बापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY
(2019-20)

Notice

Date: 10/07/2019

All students of F.Y. B.A. History hereby informed that we are going to conduct 'Slow and Advance Learner' Class Test on Thursday, 11/07/2019. The test will be related to History subjects and this test is based on 20 multiple choice questions for 20 marks. It is compulsory to all students to participate with preparation.


(Dr. H. D. Nadaf)

HEAD
Department of History
P.D.V.P.College, Tasgaon.

ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण -शिक्षणमहर्षी डॉ.बापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2019-2020)

'Slow and Advance Learner Test'

Day: Wednesday,

Date: 10 July 2019.

(Each question for 02 Marks)

Signature: _____

Roll No: _____

खेना नायकर (उ.)

2209

5/11

16/20
5/11

1. जत येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोसले क. डफळे ड. निंबाळकर
2.हे महाराष्ट्राचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील ब. यशवंतराव चव्हाण क. बाबासाहेब भोसले ड. वसंतराव नाईक
3. आण्णाभाऊ साठे यांचा जन्म जिल्ह्यात झाला.
अ. सातारा ब. सीलापूर क. सांगली ड. कोल्हापूर
4. अजंठा व वेरुळ हे ठिकाण या जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी
5. जागतिक कितीचे गोलधुमट शहरात आहे.
अ. गुलबर्गा ब. बिजापूर क. हैद्राबाद ड. सोलापूर
6. सम्राट अशोक साम्राज्याचा सत्ताधीश होता.
अ. नंद ब. मौर्य क. गुप्त ड. सातवाहन
7. छ. शिवाजी महाराजांचे राज्याभिषेक या किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रायगड
8. सांगली जिल्हा व पिकासाठी प्रसिद्ध आहे.
अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. हळद व ऊस
9. विजयनगर साम्राज्याची राजधानी येथे होती.
अ. पट्टकल्ल ब. हम्पी क. एहोळे ड. बदामी
10. या साली महाराष्ट्र राज्याची निर्मिती झाली.
अ. १ मे १९६० ब. 15 ऑगस्ट १९४७ क. १ एप्रिल १९६० ड. १ जून १९६०

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण -शिक्षणमहर्षी डॉ.बापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312


DEPARTMENT OF HISTORY

(2019-2020)

‘Slow and Advance Learner Test’

Day: Wednesday,

Date: 10 July 2019.

Signature: 

Roll No: 2303

(Each question for 02 Marks)

१. जत येथेयांचे संस्थान होते. (20 Marks)

अ. पटवर्धन ब. भोसले क. डफळे ड. निंबाळकर

२.हे महाराष्ट्राचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.

अ. वसंतदादा पाटील ब. यशवंतराव चव्हाण क. बाबासाहेब भोसले ड. वसंतराव नाईक

३. आण्णाभाऊ साठे यांचा जन्म.....जिल्ह्यात झाला.

अ. सातारा ब. सोलापूर क. सांगली ड. कोल्हापूर

४. अजंठा व वेरूळ हे ठिकाणया जिल्ह्यात आहेत.

अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी

५. जागतिक किर्तीचे गोलघुमट शहरात आहे.

अ. गुलबर्गा ब. विजापूर क. हैद्राबाद ड. सोलापूर

६. सम्राट अशोकसाम्राज्याचा सत्ताधीश होता.

अ. नंद ब. मौर्य क. गुप्त ड. सातवाहन

७. छ. शिवाजी महाराजांचे राज्याभिषेक.....या किल्ल्यावर झाले.

अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रायगड

८. सांगली जिल्हाव पिकासाठी प्रसिद्ध आहे.

अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. हळद व ऊस

९. विजयनगर साम्राज्याची राजधानीयेथे होती.

अ. पट्टदकल्ल ब. हम्पी क. एहोळे ड. बदामी

१०.या साली महाराष्ट्र राज्याची निर्मिती झाली.


अ. १ मे १९६० ब. 15 ऑगस्ट १९४७ क. १ एप्रिल १९६० ड. १ जून १९६०

'Slow and Advance Learner Test'

Day: Wednesday,

Date: 10 July 2019.

(Each question for 02 Marks)

Signature: 

Roll No: 22151

18
OK

१. जत येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोसले क. डफळे ड. निंबाळकर
२.हे महाराष्ट्राचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील ब. यशवंतराव चव्हाण क. बाबासाहेब भोसले ड. वसंतराव नाईक
३. आण्णाभाऊ साठे यांचा जन्मजिल्ह्यात झाला.
अ. सातारा ब. बीजापूर क. सांगली ड. कोल्हापूर
४. अजंठा व वेरूळ हे ठिकाणया जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी
५. जागतिक किर्तीचे गोलघुमटशहरात आहे.
अ. गुलबर्गा ब. बिजापूर क. हैद्राबाद ड. सोलापूर
६. सम्राट अशोकसाम्राज्याचा सत्ताधीश होता.
अ. नंद ब. मौर्य क. गुप्त ड. सातवाहन
७. छ. शिवाजी महाराजांचे राज्याभिषेकया किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रायगड
८. सांगली जिल्हावपिकासाठी प्रसिद्ध आहे.
अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. हळद व ऊस
९. विजयनगर साम्राज्याची राजधानीयेथे होती.
अ. पट्टकल्ल ब. हम्पी क. एहोळे ड. बदामी
१०.या साली महाराष्ट्र राज्याची निर्मिती झाली.
अ. १ मे १९६० ब. १५ ऑगस्ट १९४७ क. १ एप्रिल १९६० ड. १ जून १९६०

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

2019-2020

'Slow and Advance Learner Test'

प्रश्नपत्राची टीकासमोने

Day: Wednesday,

Date: 10 July 2019.

(Each question for 02 Marks)

Signature: P. R. Patil

Roll No: _____

1. जत येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोंसले क. डफळे ड. निवाळकर
2.हे महाराष्ट्राचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील ब. यशवंतराव चव्हाण क. बाबासाहेब भोंसले ड. वसंतराव नाईक
3. आण्णाभाऊ साठे यांचा जन्मजिल्ह्यात झाला.
अ. सातारा ब. सोलापूर क. सांगली ड. कोल्हापूर
4. अजंठा व वेरूळ हे ठिकाणया जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी
5. जागतिक किर्तीचे गोलधुमट शहरात आहे.
अ. गुलबर्गा ब. विजापूर क. हैद्राबाद ड. सोलापूर
6. सम्राट अशोकसाम्राज्याचा सत्ताधीश होता.
अ. नंद ब. मौर्य क. गुप्त ड. सातवाहन
7. छ. शिवाजी महाराजांचे राज्याभिषेकया किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रायगड
8. सांगली जिल्हावपिकासाठी प्रसिद्ध आहे.
 अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. हळद व ऊस
9. विजयनगर साम्राज्याची राजधानीयेथे होती.
अ. पट्टदकल्ल ब. हम्पी क. एहोळे ड. बदामी
10.या साली महाराष्ट्र राज्याची निर्मिती झाली.
अ. १ मे १९६० ब. 15 ऑगस्ट १९४७ क. १ एप्रिल १९६० ड. १ जून १९६०

-शिक्षणमहर्षी डॉ.बापूजी साठुंबे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312
DEPARTMENT OF HISTORY
2019-2020

'Slow and Advance Learner Test'

Day: Wednesday,

Date: 10 July 2019.

(Each question for 02 Marks)

Signature: *Amrane*

Roll No: 2162

- (10)
1. जत येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोसले डफळे निंबाळकर
2.हे महाराष्ट्राचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील विश्वंतराव चव्हाण क. बाबासाहेब भोसले ड. वसंतराव नाईक
3. आण्णाभाऊ साठे यांचा जन्मजिल्ह्यात झाला.
अ. सातारा सीतापूर क. सांगली ड. कोल्हापूर
4. अजंठा व बेरुळ हे ठिकाणया जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर औरंगाबाद ड. परभणी
5. जागतिक कित्तीचे गोलघुमट शहरात आहे.
अ. गुलबर्गा विजापूर क. हैद्राबाद ड. सोलापूर
6. सम्राट अशोकसाम्राज्याचा सत्तापीठ होता.
अ. नंद भीर्य क. गुप्त ड. सातवाहन
7. छ. शिवाजी महाराजांचे राज्याभिषेकया किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा रायगड
8. सांगली जिल्हावपिकासाठी प्रसिद्ध आहे.
 ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. हळद व ऊस
9. विजयनगर साम्राज्याची राजधानीयेथे होती.
अ. पट्टदकल्ल ब. हम्पी एहोळे ड. बदामी
10.या साली महाराष्ट्र राज्याची निर्मिती झाली.
अ. १ मे १९६० १५ ऑगस्ट १९४७ क. १ एप्रिल १९६० ड. १ जून १९६०

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

2019-2020

'Slow and Advance Learner Test'

Day: Wednesday,

Date: 10 July 2019:

(Each question for 02 Marks)

Signature: [Signature]

Roll No: 2260

1. जत येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोसले क. डफळे ड. निंबाळकर
2.हे महाराष्ट्राचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील क. वर्यशवंतराव चव्हाण क. बाबासाहेब भोसले ड. वसंतराव नाईक
3. आण्णाभाऊ साठे यांचा जन्म.....जिल्ह्यात झाला.
अ. सातारा क. सोलापूर क. सांगली ड. कोल्हापूर
4. अजंठा व वेरूळ हे ठिकाणया जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी
5. जागतिक किर्तीचे गोलघुमट शहरात आहे.
अ. गुलबर्गा क. बिजापूर क. हैद्राबाद ड. सोलापूर
6. सम्राट अशोकसाम्राज्याचा सत्ताधीश होता.
अ. नंद क. मौर्य क. गुप्त ड. सतिवाहन
7. छ. शिवाजी महाराजांचे राज्याभिषेक.....या किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रायगड
8. सांगली जिल्हावपिकासाठी प्रसिद्ध आहे.
अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. हळद व ऊस
9. विजयनगर साम्राज्याची राजधानीयेथे होती.
अ. पट्टदकल्ल ब. हम्पी क. एहोळे ड. बंदापी
10.या साली महाराष्ट्र राज्याची निर्मिती झाली

R.A.-I. History

Sr. No.	Name	Mark	Mobile No.	Signature
1.	Jinal Teekha Shah	37	9922011282	Jinal
2.	Avantika somnath kumbhar	29	9403229110	Akumbhar
3)	Shehac Harichindrakumar	27	9404054905	Shehac
4)	Pallavi Tanaji Dhound	23	7378770554	Pallavi
5)	Anisha Pandurang Patil	23	9765219481	Patil
6)	saniya Jahangir Mulla	35	9890964099	S.Mulla
7)	shinde madhuri manohar		9503883199	Shinde
8)	shinde Rutuja Sampat		7823054180	Shinde
9)	Pandit Shradha Sanjay	33	7249721053	Pandit
10)	Radhika Rovabe shete		7620424238	Rshete
11)	Landage sayali kashinath		9765315052	S.K.Landage
12)	Mujawar Anjum Harun	39	9420680686	Mujawar
13)	Jadhav Tanuja Pradip	29	8956807498	Jadhav
14)	Jadhav shubhangi outkarnay		9890542672	shubh?
15)	Kamble Vaishali Namder	26	9665223431	V.N.Kamble
16)	Kamble Rupali Namdev		9011526157	R.Kamble
17)	Pawar Poojanka TejKumar	34	7620445250	P.T.Pawar
177	Mahadik Gayatri Rajaram	38	8080423292	Gemalati D.
18)	Desai Nikita Appaso		7840985606	Desai
19)	Patil shweta Chandrakani		9561809688	Patil
20)	Mehite madhuri Pandurang		7620467255	Mehite
21)	Dhanavade Anjali Vishnu		9373785502	A.V.Dhanavade
22)	Patil Dhanashri Dhanaji		9823618253	Patil
23)	Mane Pooja Vijay		7028559100	Mane
24)	Mane Akhal Dattatatsay	30	9730177080	A.D.Mane
25)	Mane Amruta Bhimrao	31	9172845301	Mane
26)	Patil shital Kashinath	31	8320173126	Patil
27)	Patil Aramitha Somkittav	33	7507171954	A.S.Patil
28)	Rendalkar Deepthi Vijay			

B.A. I. History.

Date: 31-2-2019

Sr. No.	Name	M.H. No.	Roll No.	Signature
1	shinde madhuri manohar	9503883199	2308	<u>shinde.</u>
2)	rodka sruhat hazichanum	9404054905		<u>S. Rodka</u>
3)	Mandale Priyanka kumar	9881342820	2150	<u>Priyankam</u>
4)	Mulla Sanija Jahangir	9890964099	2195	<u>S. Mulla</u>
5)	Patil Anisha Pandurang	9765219431	2220	<u>Patil</u>
6)	Dhond Pallavi Tanaji	7378770554	2042	<u>Dhond</u>
7)	Satpute Priyanka Dipak	7498605833	2293	<u>Satpute</u>
8)	Patil Megha Mahipati	9766496772	2229	<u>Patil</u>
359)	Mane Anati Jalindar	9503379438	2161	<u>Anane</u>
3)	Mal. Priyanka Tukaram	3561931422	2143	<u>Patil.</u>
11)	Jadhav Tanuka Pradip	8956807498	2079	<u>Jadhav</u>
12)	Mujawar Anjum Horun	9420680686	2438	<u>Mujawar</u>
13)	Jadhav shabhani Sattatray	9970542692	2588	<u>Jadhav</u>
14)	Mane Pranali Dayanand	3130046972	2168	<u>Mane</u>
15)	Patil Sarika Balasahab	9022755461	2244	<u>Patil</u>
16)	Pandit Araddha Sanjay	7249721053	2214	<u>Pandit</u>
17)	Gauti Bhavrat Malavade	9579391363	2202	<u>G.B. Malavade</u>
18)	Patil Pratiksha Haridas	9096921574		<u>P.H. Patil</u>
19)	salunthe. sarita. vishay	7498988045	2287	<u>S.V. salunthe</u>
20)	Rawar Priyanka Tejkumar	7620445250	2268	<u>P.T. Rawar</u>
21)	Radhika Ravgo Shete	7620424238	2303	<u>Radhika</u>
22)	Ghatage Tejal Anil	7058684147	2051	<u>Ghatage</u>
23)	Mahadik Gayatri Rajaram	8080123292	2134	<u>Gemalati R.</u>
24)	shinde Rutuja Sampat	78230584180	2504	<u>Shinde.</u>
25)	Patil Shital Kashirath	8080123136	2245	<u>Patil.</u>
26)	Mane Archana Vinayak	7758911860	2162	<u>Mane</u>
27)	Mane Ananta Bhimaram	9172845901	2160	<u>Mane</u>
28)	Kumbhar Anantika Samrath	9403229110	2121	<u>Kumbhar</u>
29)	Jinal Teekar Shah	9922011282	2229	<u>Jinal</u>

B.A. I. (2019-2020)

Sr.No.	Name	Roll No.	Sign.
①	Shweta Vishwas Nyayningune	2209	<u>Snyayningune</u>
②	Nikita Rajendra Patil	2231	<u>NPatil</u>
③	Avantika Somnath Kumbhar	2121	<u>AKumbhar</u>
④	Jinal Tejvan Shah	2298	<u>JShah</u>
⑤	Poonam Pankesh Jadhav	2073	P.P. Jadhav
⑥	Sayli Dhiraj Sontakke	2923	S.Sontakke
⑦	Nikita Ravindra Khode	2118	<u>NKhode</u>
⑧	Komal Vishwasrao Tamkhade	2336	K.V. Tamkhade
⑨	Shivani Nandkumar Shendage	2494	S.N. Shendage
⑩	Snehal Harichandra Todkar		S.H.Todkar
11)	Radhika Ranbo Shete	2303	<u>RShete</u>
12	Pallavi Tanaji Dhound	2042	<u>P.Dhound</u>
13	Anisha Pandurang Patil	2320	A.Patil
14)	Tanuja Pradip Jadhav	2089	T.P. Jadhav
15)	Archana Vinayak Mane	2162	<u>AVMane</u>
16)	Mal. Priyanka Tukaram	2143	P.T. Mal.
17)	Mane Anadi Jalindar	2161	<u>Mane</u>
18)	Ramija Amire Naldat	2198	<u>RAmire</u>
19)	Deeksha Pragati Shinde	2036	<u>DShinde</u>
20)	Rutuja Sanjay Shivankar	2321	<u>RShivankar</u>

B.A. I. History.

Sr. No.	Name	mark	mobile No.	Signature
1)	Kamble Ashutosh Vijay			
2)	रुमित कुमार शेंडे		7666366218	<u>R. Kumar</u>
3)	Vinayak Maruti Babar		8766482166	<u>Shankar</u>
4)	Tushar Girish Potdar		9049469572	v.m. Babar
5)	Ambar Kumar Kamble		9860458827	<u>Ambar</u>
6)	सुरज शैलेश अंडारे		766938708	<u>Suraj</u>
7)	आदित्य विक्रम काटील		7249504996	<u>Aditya</u>
8)	गणेश संभाजी गावडे		9960643284	<u>Ganesh</u>
9)	आकाश अत मोसा		7620242357	<u>Sub.</u>
10)	एसुला रमजान मुल्का		7620222357	<u>Asula</u>
11)	निलेश बाबुराव सुनीलराव		8459846552	<u>Pranav</u>
			9022695528	<u>NBS</u>

Sl. No.	Name	Mob. No.	Sign.
1	Jadhav Sahas Uttam	8698161274	<u>Jadhav</u>
2	Jangam Akshay Shanudas	7219276331	<u>AK</u>
3	Ushal Vilas Jadhav	7410115113	<u>Ushal</u>
4)	Kherade Rakesh Jitesh	9420883832	<u>Rakesh</u>
5)	Musawar Jafar Hassan		<u>Musawar</u>
6)	Sangawanshi Aniket Rajnarain	9530104246	<u>Aniket</u>
7)	Musawar Sahil Salim	7558588061	<u>Musawar</u>
8)	Pirane Rutik Rajbhosale	9766431757	<u>Pirane</u>

Date
12/09/2019

B. A. J.

Advanced learners & slow learners

Sl. No.	Name	Roll No.	Sign.
1.	Shalman shamuel Deshpande	2373	<u>Shalman</u>
2.	Mitesh Baburao Suryavanshi	2328	<u>NBS</u>
3.	Rasul Ramjan Mulla	2133	<u>Rasul</u>
4)	Aditya vikas Patil	2216	<u>A.</u>
5)	Vikram shripati Adgate	2001	<u>Vikram</u>
6)	Vinayak Rajaram Sadakale	2282	<u>Vinayak</u>
7)	Amar Kumar Kambale	2094	<u>Amr</u>
8)	-vinayak Mazuti Babar	2004	<u>v.m.Babar</u>
9)	<u>Mikes</u> Rajaram Suryavanshi	2327	<u>Mikes</u>
10)	Caanesh sathuram Subhakar	2324	<u>Caanesh</u>
11)	Ritik Rameshram Mandate	2401	<u>Ritik</u>
12)	Suresh Santati Yalwar	2351	<u>Suresh</u>
13)	<u>Mall</u> Anand Santati	2138	<u>Mall</u>
14)	Patil Pradip Jalindar	2326	<u>Patil</u>
15	Tushar Dhanraj Potdar	2483	<u>Tushar</u>
16)	Mayur mansing Jadhav	2071	<u>Mayur</u>
17	Gumeet kumar shendage	2495	<u>Gumeet</u>
18)	Patil Suraj Sanjay	2253	<u>Patil</u>
19)	Nygnirgune Abhishek Ravindra	2210	<u>Nygnirgune</u>
20)	Tharadat Suyog Vijay	2343	<u>Tharadat</u>
21)	Akash love Morya	2520	<u>Akash</u>
22)	maresh suttaruy	2452	<u>maresh</u>
23)	Glover Kiron Howale		<u>Glover</u>
24)	Pravall Ramchandra thite	2339	<u>Pravall</u>

B.A. I. History

S. NO.	Name	Roll No.	Sign.
1	कु. कण्ठे आरता विजय	2401	
2	कु. बेडगे प्रिती उमोल	2007	
3	कु. धनवडे अंजली विपव	2040	A.v. Dhankar
4	कु. प्रिभांका कनिका मधुले	2132	Pradane
5	कु. आरती सुवेश लुंग्माव	2120	A.S. Kumbhara
6	कु. पुजा शिम्भराव जाटेळे	2263	P.B. Patole
7	कु. गायत्री राजाराम मलहिक	2134	Ganabudik
8	कु. तेजस्य अनिल धारगे	2051	Tatage
9	कु. अंजली चंद्रकांत जाहील	2252	Jatil
10	कु. शोभना भागिनी शिंदे	2313	Shinde
11	कु. विडिता आशाशी देसाई	2081	Desai

'Remedial Coaching'

*Slow Learner and
Advance Learner*

Date: 16/07/2018

**A Report Submitted to
IQAC, Internal Quality Assurance Cell
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312**

**Organized by
Department of History**

**Course Coordinator
Dr. Haji Nadaf**

**Head
Dr.H. D. Nadaf
Department of History
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon,**

2018-2019

Date 13/07/2018

To
The Principal
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli.416312

Subject: Application for the Permission to conduct 'Slow and
- Advance Learner Test'.

Respected sir,

Our department of History is going to conduct 'Slow and Advance
Learner' test on F. Y. B. A. on Monday, 16/07/2018. We are kindly requesting
you to allow us to conduct this activity. Students of B.A. Part- I., are going to
participate in the same activity.

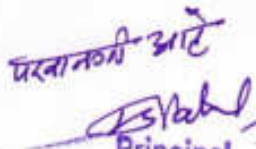
Thanking you

Yours Faithfully


(Dr. H. D. Nadaf)

HEAD
Department of History
P.D.V.P.College, Tasgaon.

Organizer: Dr. H. D. Nadaf


Principal
Padmabhushan Dr. Vasantrodada Patil
Mahavidyalaya, Tasgaon, (Sangli)

Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2018-19)

'Remedial Coaching'

Notice

Date: 13/07/2018

All students of F.Y. B.A. History hereby informed that we are going to conduct **'Remedial Coaching' Slow and Advance Learner** Class Test on Tuesday, 16/07/2018. The test will be related to History subjects and this test is based on 20 multiple choice questions for 20 marks. It is compulsory to all students to participate with preparation. All students of F.Y. B.A. History hereby informed that we are going to conduct


HEAD
Department of History
P.D.V.P.College, Tasgaon.

(Dr. H. D. Nadaf)

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी विभाग
-शिक्षणमहर्षी डॉ.बापूजी साळुंके
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312
DEPARTMENT OF HISTORY
2018-2019

‘Slow and Advance Learner Test’

Day: Tuesday,
Date: 10 July 2018.

B.A.I
(Each question for 02 Marks)

Signature: D.N. Chankar
Roll No: 2059

1. सांगली येथेयांचे संस्थान होते. (20 Marks)
अ. पटवर्धन ब. भोसले क. डफळे ड. निंबाळकर
2.हे मुंबई प्रांताचे पहिले मुख्यमंत्री म्हणून ओळखले जातात.
अ. वसंतदादा पाटील ब. यशवंतराव चव्हाण क. बाळासाहेब खेर ड. वसंतराव नाईक
3. डॉ. कोटणीस यांचा जन्म.....जिल्ह्यात झाला.
अ. सातारा ब. सोलापूर क. सांगली ड. कोल्हापूर
4. अजंठा व वेरूळ हे ठिकाणया जिल्ह्यात आहेत.
अ. नाशिक ब. अहमदनगर क. औरंगाबाद ड. परभणी
5. जागतिक किर्तीचे ताजमहाल शहरात आहे.
अ. गुलबर्गा ब. आग्रा क. हैद्राबाद ड. सोलापूर
6. अकबरसाम्राज्याचा सत्ताधीश होता.
अ. नंद ब. मुघल क. गुप्त ड. सातवाहन
7. छ. शिवाजी महाराजांचे राज्याभिषेक.....या किल्ल्यावर झाले.
अ. प्रतापगड ब. तोरण क. पन्हाळा ड. रावगड
8. सोलापूर जिल्हा पिकासाठी प्रसिद्ध आहे.
अ. ऊस व द्राक्षे ब. बेदाणा व द्राक्षे क. द्राक्षे व हळद ड. ज्वारी
9. बहमनी साम्राज्याची राजधानी येथे होती.
अ. पट्टकल्ल ब. हम्पी क. गुलबर्गा ड. बदामी
10.या साली गोवा राज्याची निर्मिती झाली.
अ. १ मे १९६० ब. ३० मे १९८७ क. १ एप्रिल १९६० ड. १ जून १९६०

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

B. A. I

2018-2019

Slow Learner and Advance Learner

Date: 16/07/2018

SR.NO.	ROLL CALL	NAME OF THE STUDENT	SIGNATURE
1.	2560	Garande Atantara Ashok	Garande
2.	2570	Koli Ajay Aoraso	(Koli)
3.	2501	Madane Aniket Sanjay	Madane
4.	2530	Mane Sanket Sanjay	S. S. Mane
5.	2690	Hejare Akshay Kisan	Hejare
6.	2486	Chougule Jodiram Gayanath.	Chougule
7.	2684	Yedake Anol Kisan	A. P. Yedake
8.	2679	Sirsath Akshay Dilip	Sirsath
9.	2671	Shendge Vishal Janardhan	V. S. Shendge
10.	2469	Shaikh Samir Jamal	Shaikh
11.	2457	Patil Shashikant Prakash	Patil
12.	2452	Patil Shankar Himmatrao	Patil
13.	2641	Patil Mayur Arvind	Patil
14.	2435	Patil Amit Suktdev	A. Patil
15.	2532	Patil Adesh Ranaji	Patil
16.	2018	Bhosale Gurusesh Bhagwan	Bhosale
17.	2025	Bobade Pratik Dilip.	P. D. Bobade
18.	2026	Bhitorale Parat Anil	Bhitorale
19.	2034	Charan Ganesh Pruthviraj	Charan
20.	2044	Charan Suresh Suresh	Charan

21.	2047	Chavan Vaibhav Harmanant	Chavan
22.	2049	Chavan Vijay Prakash	Chavan
23.	2051	Chavan Vishal Vikhal	Chavan
24.	2052	Choudhari Dadasidevi N.	Choudhari
25.	2075	Patil Alfaj Iqbal.	Patil
26.	2085	Gujale Vandana Shripati	Gujale
27.	2087	Guan Sankam Shiraji	Guan
28.	2093	Jhar Anandh Netaji	Jhar
29.	2090	Jhar Megha Raju	Jhar
30.	2105	Jadhav Yogesh Parindra	Jadhav
31.	2112	Jawale Rohit Ramch	Jawale
32.	2113	Joshi Abhar Anant.	Joshi
33.	2122	Kamble Mangal Suresh	Kamble
34.	2136	Kharade Rushikesh Keraba	Kharade
35.	2139	Khot Chetana Suryakant.	Khot
36.	2145	Kotate Dhanraj Varman	Kotate
37.	2155	Kumbhar Tuskar Dipak.	Kumbhar
38.	2160	Lokhande Monish Sanjay	Lokhande
39.	2168	Mali Mayuri Jaykumar.	M. J. Mali
40.	2181	Mandale Mahari Dhondiram.	Mandale
41.	2193	Mane Ujwala Anim	Mane
42.	2208	Mankari Anand Balekaran	Mankari
43.	2210	Mulani Jubey Jahangir.	Mulani
44.	2216	Mulla Ibrahim Mahomedhanij	Mulla
45.	2233	Patil Ananda Akash	Patil
46.	2236	Patil Chaitanya Babasa.	Patil
47.	2218	Mane Vishwajeet Dipak.	Mane
48.	2210	Manale Abhijeet Rajkumar.	A. R. Manale
49.	2418	Mane Vishwajeet Dipak.	Mane
50.	2408	Manale Abhijeet Rajkumar	Manale

51.	2405	Mali Omkar Balisakeb.	Mali
52.	2408	Mondale Abhijeet Rajkumar.	Mondale
53.	2371	Kamble Pratikraj Sureshwar	Kamble
54.	2387	Vethar Suhag Uttam.	Vethar
55.	2382	Jrale Shrish Prachandra.	Jrale
56.	2373	Derkule Ashish Nishikant.	Derkule
57.	2370	Deshmukh Rajit Rajaram.	Deshmukh
58.	2357	Wagh Swapnil Uttam.	Wagh
59.	2309	Shinde Kshatir Bhaskar.	Shinde
60.	2324	Shinde Gunesb Anoop	Shinde
61.	2316	Shende Dipali Chantakant.	D.C. Shende
62.	2305	Salunke Shisaj Yashkant.	Salunke
63.	2294	Pawan somnath Shrikaji	Pawan
64.	2287	Patil Vaibhav Vishrant.	Patil
65.	2284	PATIL TEJAS NAVANATH	Patil
66.	2278	Patil Sodali Gunesb	C. G. Patil
67.	2252	Patil Pranati Suresh.	Patil
68.	2253	Patil Pranati Anant	P. A. Patil
69.	2237	Patil Anant Popat.	Patil
70.	2241	Dipali Jagannath Patil.	D. J. Patil
71.			
72.			
73.			
74.			
75.			
76.			
77.			
78.			

"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार" -शिक्षणमहर्षी डॉ.बापूजी साकुंखे Shri
Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit

Padmabhushan Dr. Vasant Rao Dada Patil Mahavidyalaya

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

Slow Learner and Advance Learner

Date: 16/07/2018

No	Roll No.	Gender M/F	Name of the Student	Mark
1	2018	M	Bhosale Gauresh Bhagwan	08
2	2025	M	Bobade Pratik Dilip	08
3	2026	M	Budhavale Pavan Anil	06
4	2034	M	Chavan Ganesh Pruthviraj	09
5	2044	M	Chavan Sourabh Suresh	10
6	2047	M	Chavan Vaibhav Hanmant	14
7	2049	M	Chavan Vijay Prakash	18
8	2051	M	Chavan Vishal Vitthal	14
9	2052	F	Choudhari Dadmidevi Narsirma	12
10	2075	M	Fakir Alfaj Iqbal	10
11	2085	F	Gujale Archana Shripati	06
12	2087	M	Gurav Shubham Shivaji	08
13	2093	M	Jadhav Avinash Netaji	08
14	2097	F	Jadhav Megha Raju	12
15	2105	M	Jadhav Yogesh Ravindra	08
16	2112	M	Jawale Rohit Ramesh	18
17	2113	M	Joshi Atharv Anant	14
18	2122	F	Kamble Mangal Suresh	12
19	2136	M	Kharapude Rushikesh Keraba	14
20	2139	F	Khot Chetana Suryakant	18
21	2144	M	Kokate Pravin Vaman	16
22	2155	M	Kumbhar Tushar Dipak	12
23	2160	M	Lokhande Monish Sanjay	10
24	2168	F	Mali Mayuri Jaykumar	12
25	2181	M	Mandle Malhari Dhondiram	06
26	2193	F	Mane Ujwala Arjun	08
27	2209	M	Mulani Arbaj Balekhan	10
28	2210	M	Mulani Juber Jahangir	14
29	2216	M	Mulla Ibrahim Mahmadhanif	10
30	2233	M	Patil Ananda Prakash	18
31	2238	M	Patil Chaitnya Babaso	10

32	2241	F	Patil Dipali Jagannath	08
33	2253	M	Patil Pramod Popat	06
34	2254	F	Patil Pranali Amrut	10
35	2257	F	Patil Pranoti Suresh	12
36	2278	F	Patil Sonali Ganesh	18
37	2284	M	Patil Tejas Navanath	14
38	2287	M	Patil Vaibhav Vishrant	12
39	2294	M	Pawar Somnath Shahaji	16
40	2304	M	Salunkhe Dhiraj Yashavant	04
41	2318	F	Shendge Dipali Chandrakant	04
42	2321	M	Shinde Ganesh Dilip	06
43	2329	M	Shinde Vishal Bhaskar	08
44	2353	M	Wagh Swapnil Uttam	10
45	2370	M	Deshmukh Ranjit Rajaram	08
46	2373	M	Devkule Ashitosh Nishikant	12
47	2382	M	Irale Shirish Ramchandra	14
48	2387	M	Jadhav Suhas Uttam	18
49	2391	M	Kamble Pratikraj Surendra	16
50	2405	M	Mali Omkar Balasaheb	14
51	2408	M	Mandale Abhijeet Rajkumar	10
52	2418	M	Mane Vishwjeet Dipak	14
53	2432	M	Patil Adesh Tanaji	18
54	2435	M	Patil Amit Sukhadev	16
55	2441	M	Patil Mayur Arvind	06
56	2452	M	Patil Shankar Himmatrao	09
57	2453	M	Patil Shashikant Prakash	12
58	2469	M	Shaikh Samir Jamal	04
59	2471	M	Shendage Vishal Janardhan	08
60	2479	M	Sirsath Akshay Dilip	12
61	2484	M	Yedake Amol Kisan	18
62	2486	M	Chougule Jotiram Gajanan	14
63	2490	M	Hajare Akshay Kisan	12
64	2500	M	Koli Ajay Appaso	16
65	2501	M	Madane Aniket Lalaso	12
66	2530	M	Mane Sanket Sanjay	04
68	2540	M	Garande Akanksha Ashok	06

Sheela
HEAD

Department of History
P.D.V.P.College, Tasgaon.

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” -शिक्षणमहर्षी डॉ.बापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya


Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

Advance Learner

Sr. No	Roll No.	Gender M/F	Name of the Student	Mark
1	2044	M	Chavan Sourabh Suresh	10
2	2047	M	Chavan Vaibhav Hanmant	14
3	2049	M	Chavan Vijay Prakash	18
4	2051	M	Chavan Vishal Vitthal	14
5	2052	F	Choudhari Dadmidevi Narsirma	12
6	2075	M	Fakir Alfaj Iqbal	16
7	2097	F	Jadhav Megha Raju	12
8	2112	M	Jawale Rohit Ramesh	18
9	2113	M	Joshi Atharv Anant	14
10	2122	F	Kamble Mangal Suresh	12
11	2136	M	Kharapude Rushikesh Keraba	14
12	2139	F	Khot Chetana Suryakant	18
13	2144	M	Kokate Pravin Vaman	16
14	2155	M	Kumbhar Tushar Dipak	12
15	2160	M	Lokhande Monish Sanjay	10
16	2168	F	Mali Mayuri Jaykumar	12
17	2209	M	Mulani Arbaj Balekhan	10
18	2210	M	Mulani Juber Jahangir	14
19	2216	M	Mulla Ibrahim Mahmadhanif	16
20	2233	M	Patil Ananda Prakash	18
21	2238	M	Patil Chaitnya Babaso	10
22	2254	F	Patil Pranali Amrut	10
23	2257	F	Patil Pranoti Suresh	12
24	2278	F	Patil Sonali Ganesh	18
25	2284	M	Patil Tejas Navanath	14
26	2287	M	Patil Vaibhav Vishrant	12
27	2294	M	Pawar Somnath Shahaji	16

28	2353	M	Wagh Swapnil Uttam	10
29	2373	M	Devkule Ashitosh Nishikant	12
30	2382	M	Irale Shirish Ramchandra	14
31	2387	M	Jadhav Suhas Uttam	18
32	2391	M	Kamble Pratikraj Surendra	16
33	2405	M	Mali Omkar Balasaheb	14
34	2408	M	Mandale Abhijeet Rajkumar	10
35	2418	M	Mane Vishwjeet Dipak	14
36	2432	M	Patil Adesh Tanaji	18
37	2435	M	Patil Amit Sukhadev	16
38	2453	M	Patil Shashikant Prakash	12
39	2479	M	Sirsath Akshay Dilip	12
40	2484	M	Yedake Amol Kisan	18
41	2486	M	Chougule Jotiram Gajanan	14
42	2490	M	Hajare Akshay Kisan	12
43	2500	M	Koli Ajay Appaso	16
44	2501	M	Madane Aniket Lalaso	12


HEAD
Department of History
P.D.V.P.College, Tasgaon.
(Dr. H. D. Nadaf)

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” -शिक्षणमहर्षी डॉ.बापूजी साळुंखे


Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

Slow Learner

Sr. No	Roll No.	Gender M/F	Name of the Student	Marks
1	2018	M	Bhosale Gauresh Bhagwan	8
2	2025	M	Bobade Pratik Dilip	8
3	2026	M	Budhavale Pavan Anil	6
4	2034	M	Chavan Ganesh Pruthviraj	8
5	2085	F	Gujale Archana Shripati	6
6	2087	M	Gurav Shubham Shivaji	8
7	2093	M	Jadhav Avinash Netaji	8
8	2105	M	Jadhav Yogesh Ravindra	8
9	2181	M	Mandle Malhari Dhondiram	6
10	2193	F	Mane Ujwala Arjun	8
11	2241	F	Patil Dipali Jagannath	8
12	2253	M	Patil Pramod Popat	6
13	2304	M	Salunkhe Dhiraj Yashavant	4
14	2318	F	Shendge Dipali Chandrakant	4
15	2321	M	Shinde Ganesh Dilip	6
16	2329	M	Shinde Vishal Bhaskar	8
17	2370	M	Deshmukh Ranjit Rajaram	8
18	2441	M	Patil Mayur Arvind	6
19	2452	M	Patil Shankar Himmatrao	8
20	2469	M	Shaikh Samir Jamal	4
21	2471	M	Shendage Vishal Janardhan	8
22	2530	M	Mane Sanket Sanjay	4
23	2540	M	Garande Akanksha Ashok	6


HEAD
Department of History
P.D.V.P.College, Tasgaon.
(Dr. H. D. Nadaf)

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार” -शिक्षणमहर्षी डॉ.बापूजी साळुंखे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2018-2019)

‘Remedial Coaching Report’

Title of Programme	: Slow and Advance Learner Test
Organizing Department	: History
Date	: 15th Jan. 2019
Venue	: Lecture Room. No.26
No. of Participants	: 68 (Male:56, Female:12)
Coordinator	: Dr. Haji Nadaf
President	: Prin. Dr. R.R. Kumbhar

The Department of History conducted remedial coaching-‘Slow and Advance Learner’ activity on **F.Y.B.A.**, on **10/07/2018**. Total 68 Students were participated in this same activity. The test was basis on Indian history and Culture.

The Students of B.A.I were participated in this activity and they have responded to well presence. Students are given their knowledge by aptitude exam. In this test 85 percent students have passed. After the test we have divided into two group as ***slow learner group and advance learner group*** by present students. Inserts are in the forefront for both of these groups to be independently guided in the future. This entire activity conducted by Dr. H. D. Nadaf for this activity. Prin. Dr. R.R. Kumbhar has guided and cooperated well for the programme.


HEAD
Department of History
P.D.V.P.College, Tasgaon.

(Dr. H. D. Nadaf)

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” -शिक्षणमहर्षी डॉ.बापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit

Padmabhushan Dr. Vasantodada Patil Mahavidyalaya

Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2019-2020)

T.Y.B.A.

History Study Fourm

दि.06/08/2019

प्रति,

मा.प्राचार्य,

पद्मभूषण डॉ. वसंतरावदादा पाटील महाविद्यालय.

तसगाव

विषय :महाविद्यालयातील इतिहास विभागात विद्यार्थ्यांसाठी इतिहास अभ्यास
मंडळ स्थापन करण्यास परवानगी मिळणेबाबत

महोदय,

उपरोक्त विषयास अनुसरून अर्ज करतो की ,महाविद्यालयातील वी.ए. भाग २,
व ३ या विभागातील इतिहास विषयाच्या विद्यार्थ्यांचा शैक्षणिक गुणवत्ता विकास
होण्याच्या दृष्टीने विद्यार्थ्यांचा इतिहास अभ्यास मंडळस्थापन करण्याचे योजले आहे.
या मंडळातील विद्यार्थ्यांचे नाव व पद खालीलप्रमाणे ठरविणेत आले आहे. तरी
विद्यार्थ्यांचा इतिहास अभ्यास मंडळस्थापन करण्यास परवानगी मिळणेस सहकार्य
व्हावे ही विनंती.

कळावे,

Permission granted
[Signature]
प्राचार्य

पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तसगाव (जि. सांगली) ;

आ.वि.

[Signature]
डॉ. एन. डी. नदाफ
HEAD

Department of History
P.D.V.P.College, Tasgaon

"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार"
-शिक्षणमहर्षी डॉ. बापूजी साठुंबे
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY
(2019-2020)
T.Y.B.A.

दिनांक : ०४ / ०९ / २०१९

नोटीस
इतिहास विभाग

इतिहास विभागातील बी.ए.भाग ३ या वर्गातील सर्व विद्यार्थ्यांना सूचित करण्यात येते की, शैक्षणिक वर्ष २०१९-२०२० मध्ये सालावादप्रमाणे विभागात 'इतिहास अभ्यास मंडळ' उद्घाटन समारंभ उद्या दि. ०५ सप्टेंबर २०१९ रोजी सकाळी ठीक १०.०० वाजता प्रमुख पाहुणे मा. महावीर मुळे व महाविद्यालयाचे प्राचार्य डॉ. मिलिंद हुजरे यांच्या हस्ते संपन्न होणार आहे. तरी सर्व विद्यार्थ्यांनी उपस्थित राहावयाचे आहे याची सर्वांनी नोंद घ्यावी.

विभागप्रमुख

HEAD

Department of History
P.D.V.P.College, Tasgaon.

१) K.K. Kulkarni. ←
२) Dr. M.P. Nandaf. - Shri.

"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार"

-शिक्षणमहर्षी डॉ.बापूजी साळुंखे

श्री स्वामी विवेकानंद शिक्षण संस्था, कोल्हापूर संचलित

पद्मभूषण डॉ.वसंतरावदादा पाटील महाविद्यालय

तासगाव, जि-सांगली

'इतिहास अभ्यास मंडळ उदघाटन समारंभ'

• कार्यक्रम पत्रिका •

-
- प्रतिमापूजन :
 - प्रास्ताविक व प्रमुख पाहुण्यांची ओळख : डॉ. एच.डी. नदाफ
 - स्वागत :
 - प्रमुख पाहुण्यांचे मनोगत : मा. महावीर मुळे
 - अध्यक्षीय मनोगत : मा.उपप्राचार्य. के.एस. पाटील
 - प्रमुख उपस्थिती : डॉ. अमोल सोनावले
रणजीत कुंभार, वैशाली चोरगे
 - आभार : सौ. किर्ती कोलप
 - सूत्रसंचालन : कु. पूजा जाधव

दिनांक: 05 सप्टेंबर २०१९

स्थळ: इतिहास विभाग

Shri Swami Vivekanand Shikshan Sanstha Kolhapur Sanchlit,
Padmabhushan Dr.Vasandraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312


DEPARTMENT OF HISTORY
(Academic Year-2019-2020)

HISTORY STUDENT FORUM

Sr.No.	Name of the Member	Class	Designation
1.	Miss. Pooja Jadhav	B.A.III	President
2.	Miss. Shubhangi Mali	B.A.III	Vice- President
3.	Mr. Sanket Awale	B.A.III	Acting President
4.	Mr. Satyam Kharade	B.A.III	Secretary
5.	Mr. Sagar Shingade	B.A.III	Member
6.	Miss. Jyoti Kumare	B.A.III	Member
7.	Miss. Nikita Koli	B.A.III	Member
8.	Miss. Poonam Pol	B.A.III	Member
9.	Mr. Pravin Kokate	B. A. III	Member

Dr. Haji Nadaf
Head

Dr. Milind S. Hujare
Principal


प्रिचारीय .
पद्मभूषण डॉ. वसंतरावदादा पाटील
हाबिद्यालय, तासगांव (जि. सांगली)


HEAD
Department of History
P.D.M.P. College, Tasgaon.

Student Participation

Name	Roll No.	Sign.
Pooja Jayram Jadhav.	3643	<u>Pooja</u>
Arati Sunil Babar.	3638	<u>A.S. Babar</u>
Ashwini Pratapsao Kistakar	3767	<u>A.P. Kistakar</u>
Rutuja Shivaji Mane	3653	<u>R. Mane</u>
Mulani Namira Inayatulla	3656	<u>Mulani</u>
Pol Poonam Poudip	3667	<u>Pol</u>
Mau Shubhangi Ashok	3649	<u>Mau</u>
Patil Saurabh Sanjay	3663	<u>Patil</u>
Mohamedpust Anand Anaso	3651	<u>M.</u>
Chougule Pravin Suresh	3640	<u>Chougule</u>
Suryawanshi Pravin Mohan	3670	<u>Suryawanshi</u>
Khosale Satyam Sanjay	3644	<u>Khosale</u>
Arti Prakash Patil		<u>Arti Patil</u>
Jyoti Sadashiv Kurnare		<u>Jyoti</u>
Pooja Popatrao Bhoskar		<u>Pooja</u>
Koli Nikita Vasudev		<u>Koli</u>
Mujawar Ashraf Ayub		<u>Mujawar</u>
Mare Rakul Damodar		<u>Mare</u>







Inauguration of Forum , Chief Guest Prof. Mahaveer Mule President: IC Prin. Prof K. S. Patil and Chief Presence Dr. A. G. Sonawale

Historical Exhibition
On
**'Old Coin, Postal Ticket,
Medieval Manuscript and
Wooden Art'**

Date: 25/01/2019

Submitted to
IQAC, Internal Quality Assurance Cell
Padmabhushan Dr.Vasandraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

Organized by
Department of History

Course Coordinator
Mr. Ranjeet Kumbhar

Head
Dr.H. D. Nadaf
Department of History
**Padmabhushan Dr.Vasandraodada Patil Mahavidyalaya,
Tasgaon**

2018-2019

दि. २३/०१/२०१९

प्रति,
मा. प्राचार्य
पी. डी. व्ही. पी. महाविद्यालय, तासगाव,
जि. सांगली.

विषय:- ऐतिहासिक जुनी नाणी-नोटा, चलन, पोस्टल स्टॅम्प व माती कला (पॉटरी) यांचे प्रदर्शन
भरवण्यास परवानगी मिळणेबाबत...

महोदय,

आपल्या महाविद्यालयात इतिहास विभागातर्फे जुनी नाणी, परकीय चलन, पोस्टल स्टॅम्प व माती
कला (पॉटरी) या विषयावर शुक्रवार, दि. २५ जानेवारी २०१९ रोजी प्रदर्शन भरवण्याचे आयोजित केले
आहे. तरी त्यासंदर्भात आपली परवानगी मिळावी ही विनंती.

कळावे,

Allowed
PR
23/1/2019

HEAD
Department of History
P.D.V.P.College, Tasgaon.

आपला विश्वासू,

प्रा. रणजीत सर्जेराव कुंभार
इतिहास विभाग

श्रीस्वामी विवेकानंद शिक्षण संस्था कोल्हापूर संघलित.

पद्मभूषण डॉ. वसंतरावदादा पाटील महाविद्यालय
तासगाव जि. सांगली
इतिहास विभाग

सुचना

महाविद्यालयातील सर्व विद्यार्थ्यांना सूचित करण्यात येते की, उद्या शुक्रवार दि. २५ जानेवारी २०१९ रोजी सकाळी १०.०० वाजता आपल्या महाविद्यालयात इतिहास विभागातर्फे जुनी नाणी-नोटा, पोस्टल स्टॅम्प व मातीकला यांचे प्रदर्शन आयोजित केले आहे. तरी सर्व विद्यार्थ्यांनी उपस्थित राहावे.

ठिकाण:- खोली क्र. 13 (भूगोलविभाग)

प्राचार्य,



डॉ. आर. आर. कुंभार
प्राचार्य,

पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगाव, जि. सांगली. (O.S.)



‘‘ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार’’ - शिक्षणमहर्षी डॉ. बापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

PADMABHUSHAN DR. VASANTRAODADA PATIL MAHAVIDYALAYA

TASGAON, Dist. Sangli, Pin- 416 312 ☎ - STD : 02346-250665, 250575 FAX : 250575

● Affiliated to Shivaji University, Kolhapur ●

E-mail : san.pdvpm.tas@gmail.com Website : www.pdvpmtasgaon.edu.in

NAAC Reaccredited 'B' (2.78) Established Year : June 1962 P. B. No. : 14 Jr. College No. : J22-10-001 Sr. College Code No. : SIACH X Jr.: C-8



Shikshanmaharshi
Dr. Bapuji Salunkhe
B.A., B.T., D.Lit.
FOUNDER

Hon. Chandrakant (Dada) Patil
PRESIDENT
B.Com.
Minister of Revenue, Public Works
Govt. of Maharashtra

Prin. Abhaykumar Salunkhe
M.A.
CHAIRMAN

Prin. Mrs. Shubhangi Gawade
M.Sc. B.Ed.
SECRETARY

Dr. R. R. Kumbhar
M.Sc. M.Phil., Ph.D.
PRINCIPAL

Ref.No. : PDVPMT / 1665 / 2018-19


Date : 25/11/2019

प्रति,
मा. गजानन पाटील,
पलूस.

महोदय,

आपल्या महाविद्यालयामध्ये शिक्षणमहर्षी डॉ.बापुजी साळुंखे यांच्या जन्मशताब्दी निमित्त शुक्रवार दि.२५/०१/२०१९ रोजी इतिहास विभाग आयोजित ऐतिहासिक वस्तू व कला प्रदर्शनात आपलेकडील काष्ठ शिल्पकला प्रदर्शन व काष्ठ शिल्प कला संग्रह प्रदर्शित केला. आपला संग्रह विद्यार्थी-विद्यार्थिनीस तसेच पुढील पिढीस निश्चितच प्रेरणादायी, माहितीपूर्ण व ज्ञानवर्धक आहे. संग्रह प्रदर्शित केल्याबद्दल आम्ही आपले आभारी आहोत.

धन्यवाद !


(डॉ.आर.आर.कुंभार)
प्राचार्य,

परामूर्ण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगांव, जि. सांगली. (OS)



'ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार' - शिक्षणमहर्षी डॉ. बापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

PADMABHUSHAN DR. VASANTRAODADA PATIL MAHAVIDYALAYA

TASGAON, Dist. Sangli, Pin- 416 312 ☎ - STD : 02346-250665, 250575 FAX : 250575

• Affiliated to Shivaji University, Kolhapur •

E-mail : san.pdvpm.tas@gmail.com Website : www.pdvpmtasgaon.edu.in

NAAC Reaccredited 'B' (2.78) Established Year : June 1962 P. B. No. : 14 Jr. College No. : J22-10-001 Sr. College Code No. : ^{SWAC/4}X Jr.: C-8



Shikshanmaharshi
Dr. Bapuji Salunkhe
B.A., B.T., D.L.S.
FOUNDER

Hon. Chandrakant (Dada) Patil
PRESIDENT B.Com.
Minister of Revenue, Public Works
Govt. of Maharashtra

Prin. Abhaykumar Salunkhe
CHAIRMAN M.A.

Prin. Mrs. Shubhangi Gawade
SECRETARY M.Sc. B.Ed.

Dr. R. R. Kumbhar
PRINCIPAL M.Sc. M.Phil., Ph.D.

Ref.No. : PDVPMT / 1666/2018-19

Date : 25/01/2019

प्रति,
श्रीमती सुजाता रेणके,
जयसिंगपूर.

महोदय,

आपल्या महाविद्यालयामध्ये शिक्षणमहर्षी डॉ.बापुजी साळुंखे यांच्या जन्मशताब्दी निमित्त शुक्रवार दि.२५/०१/२०१९ रोजी इतिहास विभाग आयोजित ऐतिहासिक वस्तू व कला प्रदर्शनात आपलेकडील चित्रांचा संग्रह प्रदर्शित केला. आपला संग्रह विद्यार्थी-विद्यार्थिनीस तसेच पुढील पिढीस निश्चितच प्रेरणादायी, माहितीपूर्ण व ज्ञानवर्धक आहे. संग्रह प्रदर्शित केल्याबद्दल आम्ही आपले आभारी आहोत.

धन्यवाद !

(डॉ.आर.आर.कुंभार)
प्राचार्य,

संपूर्ण डॉ. वसंतरावदादा पाटील
विद्यालय, तासगांव, जि. सांगली. (05.)



"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार" - शिक्षणमहर्षी डॉ. बापूजी साळुंखे

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Established Year : June 1962 • P. B. No. : 14 • Jr. College No. : J22-10-001 • Sr. College Code No. : ^{SHVACIA}X Jr.- C-8



Shikshanmaharshi
Dr. Bapuji Salunkhe
B.A., B.T., D.Lit.
FOUNDER

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Prin. Abhaykumar Salunkhe
M.A.
CHAIRMAN

Prin. Mrs. Shubhangi Gawade
M.Sc. B.Ed.
SECRETARY

Dr. R. R. Kumbhar
M.Sc. M.Phil., Ph.D.
PRINCIPAL

Ref.No. : PDVPMT / 1664/2018-19

Date : 25/01/2019

प्रति,
मा.प्रशांत पाटील
कवठे एकंद.

महोदय,

आपल्या महाविद्यालयामध्ये शिक्षणमहर्षी डॉ.बापुजी साळुंखे यांच्या जन्मशताब्दी निमित्त शुक्रवार दि.२५/०१/२०१९ रोजी इतिहास विभाग आयोजित ऐतिहासिक वस्तू व कला प्रदर्शनात आपलेकडील माती कला प्रदर्शन व मातीचे भांडी यांचा संग्रह प्रदर्शित केला. आपला संग्रह विद्यार्थी-विद्यार्थिनीस तसेच पुढील पिढीस निश्चितच प्रेरणादायी, माहितीपूर्ण व ज्ञानवर्धक आहे. संग्रह प्रदर्शित केल्याबद्दल आम्ही आपले आभारी आहोत.

धन्यवाद !

pu

(डॉ.आर.आर.कुंभार)
प्राचार्य,

पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तारगांव, जि. सांगली. (O.S.)



"ज्ञान, विज्ञान आणि सुरंस्कार यांसाठी शिक्षणप्रसार" - शिक्षणमहर्षी डॉ. बापूजी साळुंखे

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Shikshanmaharshi
Dr. Bapuji Salunkhe
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FOUNDER

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SECRETARY

Dr. R. R. Kumbhar
M.Sc. M.Phil., Ph.D.
PRINCIPAL

Ref.No. : PDVPMT / 1664 / 2018-19


Date : 23/01/2019

प्रति,
मा.अशोक सडोलीकर
कोल्हापूर.

महोदय,

आपल्या महाविद्यालया मध्ये शिक्षणमहर्षी डॉ.बापुजी साळुंखे यांच्या जन्मशताब्दी निमित्त शुक्रवार दि.२५/०१/२०१९ रोजी इतिहास विभाग आयोजित ऐतिहासिक वस्तू व कला प्रदर्शनात आपलेकडील जुनी नाणी, नोटा, देश विदेशातील नाणी व चलनी नोटांचा संग्रह, टपाल तिकिटांचा संग्रह प्रदर्शित केला. आपला संग्रह विद्यार्थी-विद्यार्थिनीस तसेच पुढील पिढीस निश्चितच प्रेरणादायी, माहितीपूर्ण व ज्ञानवर्धक आहे. संग्रह प्रदर्शित केल्याबद्दल आम्ही आपले आभारी आहोत.

धन्यवाद !


(डॉ.आर.आर.कुंभार)
प्राचार्य,

पद्मभूषण डॉ. वसंतरावदादा पाटील
महाविद्यालय, तासगांव, जि. सांगली. (OS)



“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार” शिक्षणमहर्षी डॉ.बापूजी साळुंखे
स्वामी विवेकानंद शिक्षण संस्था, कोल्हापूर संचलित

पद्मभूषण डॉ.वसंतरावदादा पाटील महाविद्यालय तासगाव, जि- सांगली

शिक्षणमहर्षी डॉ.बापूजी साळुंखे यांच्या जन्मशताब्दी वर्षानिमित्त
इतिहास विभाग आयोजित

‘ऐतिहासिक वस्तू व कला प्रदर्शन’

शुक्रवार दि. २५ जानेवारी, २०१९

☆ कार्यक्रमपत्रिका ☆

सकाळी: ९. ३०. वा.

प्रदर्शन उदघाटन समारंभ

प्रास्ताविक व स्वागत

: प्रा. रणजीत कुंभार

उदघाटक व प्रमुख पाहुणे

: प्रा. डॉ.सी.जी.पाटील माजी विभागप्रमुख, पदवी व
पदव्युत्तर इतिहास विभाग, पी. डी. व्ही. पी. महाविद्यालय,
तासगाव

प्रदर्शनकर्त्यांचा सत्कार समारंभ

: मा.अशोक सडोलीकर- जुनी नाणी, नोटा आणि पोस्ट
तिकिटे यांचे संग्रहकर्ते

मा. प्रशांत पाटील- मातीकला कलावंत

श्रीमती. सुजाता रेणके- चित्रकार

श्री. गजानन पाटील- काष्ठ शिल्पकार

अध्यक्षीय मनोगत

: मा. प्राचार्य. डॉ. आर. आर. कुंभार

प्रमुख उपस्थिती

: प्रा. जी. के. पाटील ज्येष्ठ वक्ते व विचारवंत

डॉ. बी. टी. कणसे सदस्य, सी.डी.सी

डॉ. हाजी नदाफ विभागप्रमुख, इतिहास विभाग

आभार

: प्रा. कीर्ती कोलप

सूत्रसंचलन

: कु. वैजयंती गेळे

स्थळ: मुख्य इमारत, वर्ग खोली क्रमांक- १३

वेळ : दि. २५ जानेवारी, २०१९ रोजी सकाळी १०.०० ते सायंकाळी ५.०० वाजेपर्यंत.

Shri Swami Vivekanand Shikshan Sanstha Kolhapur Sanchlit
 Padmabhushan Dr. Vasantgadada Patil Mahavidyalaya,
 Tasgaon, Dist- Sangli, Pin-416312
 DEPARTMENT OF HISTORY
 PRESENT SHEET (2018-2019)

Title of the Programme: Historical exhibition Class: _____ Date: 25/01/20

Sr.No.	Name of the Student		
3647	19	Irasi Mallesh Umakant	530839969 <u>Mhambur</u>
		Jadhav Swapnil Sanjay	9112606975 <u>Sure</u>
		SUTAR ANURAG DATTATRAY	7420004292 <u>31/12/2018</u>
		Githwar Kodaj Pravin Kumar	9930944025 <u>Prkajade</u>
		NIYAT AMIT NADAT	9960254012 <u>shaleh</u>
		Lokhande Nilesh BHASKAR	9130066585 <u>Ab</u>
		name Ajay Dalascheb	9657728707 <u>Ab</u>
		omkar Shivaji Babar	7409891407 <u>Ab</u>
		Mulani Reham Rajin	860531130 <u>Plur j</u>
		NIYAT AMIN NADAT	9960254012 <u>shaleh</u>
		Jadhav Ajun Nagesh	7218855749 <u>Abhadhar</u>
		Patole Ajay Krushna	8000522210 <u>Hi up</u>
		PATIL NILESH Kumar	9579051101 <u>Shetal</u>
		Deshmukh Saurabh Shashikant	9970300373 <u>Satukh</u>
		Hogale Vikas Vishwanath	7385919170 <u>VHogale</u>
		Dobe Rohit Ramchandra	9130515828 <u>Pran</u>
		Sukhadev. Amit Bajarang	779653197 <u>ABShah</u>
		santosh Tanaji Suryawshi	7057918302 <u>Suryawshi</u>
		somnath Shivaji Ashtekar	77448485 <u>Somnath</u>
		Sajid Jaiuddin mulani	2410532337 <u>Sajid</u>
		Janabade Amit subhash	8390138052 <u>Janabade</u>
		Chanchakar Sampat more	9075523958 <u>Sampat</u>
		Bharat Baburab chipate	9284274171 <u>Bharat</u>
		Mohite Nilesh Eknath	9168626012 <u>Nilesh</u>
		Kharode Akshay Pandurang	9860300202 <u>Ak</u>
		Sandip Mathurao Ingawale	8928344628 <u>Sandip</u>
		more Rushikosh Putang	976320809 <u>more</u>
		wade nitinkumar Ashok	8303462968 <u>Wade</u>
		Chorade Vikas Nandkumar	830576287 <u>Chorade</u>
		sunny Rajesh Salunkhe	7083977810 <u>sunny</u>

Historical Exhibition

Title of Programme	: ‘Historical Exhibition on wooden art, old Manuscripts and Coins’
Organizing Department	: History
Date	: 25th January 2019.
Venue	: Main Building Room. No. 14
No. of Participants	: 350, Male: 150, Female: 200
Chief Guest	: Prof. Dr. Chandrakant Patil, Tasgaon
Coordinator	: Mr. Ranjeet Kumbhar
President	: Prin. Dr. R.R. Kumbhar

Our Department organized ‘Historical Exhibition on wooden art, old Manuscripts and Coins’ on 25th January 2019. An exhibition, in the most general sense, is an organised presentation and display of a selection of items. In practice, exhibitions usually occur within a cultural or educational setting such as a museum, art gallery, park, library, exhibition hall. Exhibitions can include many things such as art in major museums and smaller galleries, interpretive exhibitions, natural history museums and history museums, and also varieties such as more commercially focused exhibitions and trade fairs.

Exhibit Objectives:

- To foster a better understanding of the importance of historical artefacts in our live.
- To provide an overview of collecting methods and rationales and help visitors better understand the role of history museums in preserving the past.
- To help visitors learn from historical artefacts and inspire them to learn more about the process.
- To entertain and amuse visitors with the variety and scope of the museum's collection.



Speaker Prof. Dr. Chandrakant Patil stated that, an art exhibition is traditionally the space in which art object meet an audience. The exhibit is universally understood to be for some temporary period unless, as is rarely true, it is stated to be a permanent exhibition. Such expositions may present pictures, drawings, video, sound, installation, performance, interactive art, new media art or sculptures by individual artists, groups of artists or collections of a specific form of art.





It was a comprehensive and searchable listing of exhibitions held at department of history on 29th January 2019. It includes exhibitions planned by departmental staff devoted primarily to educational purposes. Total **350** participants were participated in this same activity. All participants seen following views related to this subject-

- Martaha Period Manuscripts in Modi Script.
- Old Coins of Ancient to Modern Period.
- Postal ticket of Colonial Period.
- Wooden Art of daily items.

This entire activity completed by Mr. Ranjeet Kumbhar and Prin. Dr. R.R. Kumbhar guided and cooperated for the proramme.

Mr. Ranjeet Kumbhar
Coordinator

Dr. Haji Nadaf

“ज्ञान विज्ञान आणि गुंतकार यासाठी शिक्षणप्रकार” शिक्षणदर्शन डॉ. बापूजी माळुंजे

श्री स्वामी विवेकानंद शिक्षण संस्था कोल्हापूर

पदपुस्तक डॉ. वसंतदासदास पाटील भद्रविद्यालय, तामनागव

वाणिज्य विभाग (२०१८-१९)

सुचना

दिनांक २५/०२/२०१९

श्री. कौम. पाग ३ सर्व विद्यार्थ्यांना कळविण्यात येते की, गुंतकार दिनांक २८ / ०२ / २०१९ रोजी
Commerce Talent Search ही Competition देण्यात येणार आहे याची तयारी नीट घ्यावी.

(विषय शिक्षक)

(सौ. जयशंकर ए. शि.)

श्री. एम. डी. पाटील

(विभाग प्रमुख)

HEAD

Department of Commerce
P.D.V.P.College, Tasgaon.

**SHIVAJI UNIVERSITY COMMERCE AND MANAGEMENT TEACHERS ASSOCIATION
(SUCOMATA), KOLHAPUR
COMMERCE TALENT SEARCH (DISTRICT / UNIVERSITY LEVEL) COMPETITION-2018
(CTSC-2019) 28th Feb., 2019**

Attendance Report

Sr.	Name of the Student (Beginning with surname)	Male/Female	Seat No.	Sign. of Student
1)	Nimbalkar Komal Namdev	F	01	
2)	Jagtap Aparna Narayan.	F	02	
3)	Lohar Meghana Balaso	F	03	
4)	Chavan Pushpanjali Balaso	F	04	
5)	Patil Archana Vikas	F	05	
6)	Patil Nikita Dilip	F	06	
7)	Pawar Sobal Sunil	F	07	
8)	Sutar Vaishali Uttam	F	08	
9)	Patil Brnali Raghunath.	F	09	
10)	Khatal Sujata Hemant	F	10	
11)	Edake Snehal Ajit	F	11	
12)	Mali Radhika Balaso.	F	12	
13)	Chavan Kajal Namdev.	F	13	
14)	Rajmane Taxmi Ramesh.	F	14	
15)	Kopase Supriya Ramchandra	F	15	
16)	Tadhar Meghazani Mahadev	F	16	
17)	Perwat Ashwini Dhanraj	F	17	
18)	Gavali Gourzi Sanjay	F	18	
19)	Guzey Kizan Jagannath	BM	19	
20)	Patil Shubham Sadashiv	M	20	
21)	Karvanje Roshan Gangaram	M	21	
22)	Mane Avinash Yaman.	M	22	
23)	Chavan Rahul Kishor	M	23	
24)	Kopase Mahesh Vijay	M	24	
25)	Savane Kajal Malhari	F	25	
Total present			31	

Date-28/02/2019

Name of the College- P.D.V.P. Maha. Talgaon


Sign of the Supervisor-

Checked and Verified by: (Name) -

Signature-

- Note** - 1. Write 'Absent' in the column of sign. of the student if the student is absent.
2. Take as many copies of this report as you required.
3. Valid identity card must be with the student.

Seat No.		Sign of Student		Sign Supervisor	
----------	--	-----------------	--	-----------------	--

SHIVAJI UNIVERSITY COMMERCE AND MANAGEMENT TEACHERS ASSOCIATION (SUCOMATA), KOLHAPUR

COMMERCE TALENT SEARCH COMPETITION – 2019 (CTSC – 2019)

Day & Date: - Thursday, 28/02/2019

Time: 10.00 a.m. to 11.00 a.m.

Total Marks: 60

INSTRUCTIONS:

1. Write alphabet (a,b,c or d) of correct answer in the box provide to right side of each question.
2. Change in answer or over writing will not be considered.
3. Although Marathi translation is provided wherever necessary, question in English version is the base.
4. All questions are compulsory.
5. There are sixty questions carrying one mark each.
6. Use only blue ink ballpoint pen. Answer in other colors will not be assessed and no marks will be issued.

सूचना

1. प्रश्नांच्या उजवीकडील चौकोनात बरोबर उत्तराचे अनुक्रम दर्शविणारे (a,b,c or d) इंग्रजी अक्षरच लिहावे.
2. उत्तरातील बदल अथवा खाऱ्याखोऱ्या विचारत घेतले जाणार नाहीत व त्याचे पुन दिले जाणार नाहीत.
3. आवश्यक तेथे प्रश्नाचे मराठी भाषांतर दिलेले आहे. तथापि इंग्रजी भाषेतील प्रश्नच प्रमाण मानण्यात येईल.
4. सर्व प्रश्न आवश्यक आहेत.
5. प्रत्येकी एक मार्काचे 60 प्रश्न दिलेले आहेत.
6. फक्त निळ्या शाईचा बॉलपेन वापरावा. अन्य कोणत्याही शाईने लिहिलेली उत्तरे तपासली जाणार नाहीत व त्याचे पुन दिले जाणार नाहीत.

1. insurance can be taken for maximum years
a) Fire b) Life c) Marine d) None of these
.....हा विमा जास्तीत जास्त कितीही वर्षांसाठी घेता येतो.
a) अग्नि विमा b) आयुर्विमा c) सागरी विमा d) यापैकी नाही
2. Due to insurance, foreign business is developed.
a) Marine b) Fire c) Life d) None of these
.....विम्याच्या अस्तित्वामुळे विदेशी व्यापाराचा विकास झालेला आहे.
a) सागरी विमा b) अग्नि विमा c) आयुर्विमा d) यापैकी नाही
3. Third person liability (compensation) includes in.....
a) Crop Insurance b) Livestock insurance c) Accidental insurance d) None of these
त्रयस्थ व्यक्तीच्या नुकसानभरपाईचा समावेश..... विम्यात होतो.
a) पिक विमा b) पशुधन विमा c) अपघात विमा d) यापैकी नाही
4. principle is not related to life insurance.
a) Utmost good faith b) Compensating c) Transfer of ownership d) None of these
..... आयुर्विम्यास तत्त्व लागू होत नाही
a) परस्पर विश्वास तत्त्व b) नुकसान भरपाई तत्त्व c) मालकी हक्क हस्तांतर तत्त्व d) यापैकी नाही
5. Indian Life Insurance Corporation established in.....
a) 1952 b) 1969 c) 1956 d) 1979
भारतीय आयुर्विमा महामंडळाची स्थापना झाली.
a) 1952 b) 1969 c) 1956 d) 1979
6. Which of the following is not a measure of Central tendency.....
a) Mean b) Mode c) Median d) Standard Deviation
7. A batsman has scored following runs in last 5 innings 40, 25, 35, 60, 75. The mean score is.....
a) 47 b) 57 c) 35 d) 40
8. Concept of management by objective was developed by
a) FW Taylor b) Peter Drucker c) Henry Fayol d) George Terry
9. In 1980 Six Sigma was first introduced by
a) एफ. डब्ल्यू टेलर b) पीटर ड्रुकर c) हेनरी फेयॉल d) जॉर्ज टॅरी

10. a) Henry Fayal 1980 मध्ये सर्वात पहिल्यांदा 'सिक्स सिगमा' ही संकल्पना..... यांनी मांडली.
 a) हेची फेचॉल b) रोबिन्सन c) Emails Ross d) Bill Smith
 a) Planning & Organizing b) Staffing & control c) Planning & Control d) विल स्मिथ
 of the following management function are closely related.
 a) Planning & Organizing b) Staffing & control c) Planning & Control d) Planning & decision making
11. ही व्यवस्थापनाची कार्ये परस्पर संबंधी आहेत.
 a) नियोजन आणि संपटन b) भरती आणि नियोजन c) नियोजन आणि नियंत्रण d) नियोजन आणि निर्णय घेणे
 Theory X and Theory Y of motivation is developed by.....
 a) Douglas Mc Gregor b) Herzberg c) Williams Ouchi d) A.H.Maslo
12. यांनी कार्य प्रेरणेचा सिद्धांत मांडला.
 a) डग्लस माग्गोरो b) हचवर्ग c) विल्यम औपी d) अब्राहम मास्लो
 was studied in Hawthorn Experiments.
 a) Human relation b) Workers c) Technology d) None of these
 चा अभ्यास करण्यासाठी होथोर्न प्रयोग केला गेला .
 a) मानवी संबंध b) कामगार c) तंत्रज्ञान d) यापैकी नाही
13. Total Quality Management (TQM) focused on.....
 a) Employee b) Customer c) Both a and b d) None of these
 संपूर्ण गुणवत्ता व्यवस्थापनामध्ये (TQM) यावर भर दिला आहे.
 a) कामगार b) ग्राहक c) दोन्ही अ आणि ब d) वरीलपैकी नाही
14. entrepreneur was related with 'Apple' company?
 a) Steve Jobs b) Mark Zukerberg c) Ratan Tata d) Waren Bufet
 ऑपल या कंपनीशी संबंधित उद्योजक होता.
 a) स्टीव्ह जॉब्स b) मार्क झुकेरबर्ग c) रतन टाटा d) वारेन बफेट
15. A is the documentary evidence in support of a transaction recorded in the books of account.
 a) Cash memo b) Voucher c) Audit d) Receipt
16. In auditing is known as a watch dog.
 a) Businessman b) Creditor c) Auditor d) Assesses
17. is tax on income.
 a) Gift Tax b) GST c) Wealth tax d) Income tax
18. means the activities for earning livelihood which requires intellectual skill and Manual skill.
 a) business b) Profession c) Trading d) Vocation
19. GST stands for.....
 a) Goods and Service Trading b) Goods and Service Trade c) Goods and Service Tax d) None of these
20. Life Insurance Premium is paid by an assessee deducted U/c
 a) 80 D b) 80 CCC c) 80 G d) 80 C
21. is a critical examination of the books and accounts with a specific objectives.
 a) Investigation b) Audit c) Book-keeping d) Accounting
22. Income from is tax free income.
 a) Agriculture b) Rent c) Salary d) Interest
23. audit is compulsory by law.
 a) Statutory b) Internal c) Special d) Test
24. There are types of Accounts.
 a) Two b) Three c) Four d) Several
25. Salary paid to an employee is..... expenditure.
 a) Revenue b) Capital c) Personal d) Waste
26. Carriage on purchase of machinery is..... expenditure.
 a) Capital b) Revenue c) Office d) Marketing
27. When the total of debit side of profit and loss account is more than the total of credit side, it results in
 a) Net Profit b) Net Loss c) Net Sales d) Purchase
28. Canara Bank A/c, Tata Ltd, Manoj capital A/c are the..... accounts.
 a) Nominal b) Real c) Personal d) None of these
29. The sum of direct material, direct wages and direct expenses is termed as.....
 a) Factory cost b) Prime Cost c) Work Cost d) Total Cost.
30. The business entity concept is applicable to.....
 a) Sole Proprietorship b) Partnership c) Corporate firm d) All of the above
31. At break- Even Point there is.....
 a) Profit b) Loss c) No profit- No loss d) None of these.
32. Deduction U/c 80C for the assessment year 2018-19.....
 a) 2,50,000 b) 1,50,000 c) 1,80,000 d) 5,00,000
33. of the following Balance sheet equations is wrong.
 a) Assets + Liabilities = Capital b) Assets - Liabilities = Capital c) Assets - Capital = Liabilities d) Liabilities + capital = Assets
34. Every member of co-operative society has right to give..... Vote.
 a) Two b) One c) Three d) All

सहकारी संस्थेच्या सभासदास मत देण्याचा अधिकार आहे.

35. SEBI is
 a) Body Corporate
 सेबी आहे.
 a) संयुक्त प्रमंडळ
 The Imperial Bank of India was established in
 a) 1931
 इम्पेरियल बँक ऑफ इंडिया ची स्थापना साली करण्यात आली.
 a) 1931
 is the present governor of RBI.
 a) Urjit Patel
 रिजर्व बँकेचे विद्यमान गव्हर्नर कोण आहेत.
 a) उर्जित पटेल
 Hon. Prime Minister Narendra Modi has taken the decision of demonetization on.....
 a) 8th Nov ,2016
 भारताचे पंतप्रधान मा. नरेंद्र मोदी यांनी नोटा बंदीचा निर्णय रोजी घेतला.
 a) 8 नोवेंबर, 2016
 'Mudra Scheme' was started in India in the year.....
 a) 2011
 भारतात 'मुद्रा योजना'साली सुरु करण्यात आली.
 a) 2011
 Banking ombudsman was introduced in India on
 a) 14th June, 1995
 भारतात बँकिंग लोकपाल योजना रोजी सुरु करण्यात आली.
 a) 14 जून, 1995
 'NIESBUD' has its head office at.....
 a) Hyderabab
 राष्ट्रीय उद्योजकता व लघुउद्योग विकास संस्थेचे (NIESBUD) मुख्य कार्यालय आहे.
 a) हैदराबाद
 ' Mahila Gruh Udyog' Limited was established in
 a) 1959
 महिला गृह उद्योग लि. या संस्थेची स्थापना साली झाली.
 a) 1959
 According to David mc-Clelland entrepreneur should have.....
 a) Need for Achievement
 डेविड मॅकलेलेंड यांच्या मते उद्योजकांच्या अंगी हा गुण असावा.
 a) सिद्धी प्रेरणा
 Marketing is and delivery of a standard of living to society.
 a) Co-ordination
 विपणन म्हणजे समाजासाठी राहणीमानाची व प्रदान होय.
 a) समन्वय
 Modern marketing means
 a) All activity of Production
 आधुनिक विपणन म्हणजे
 a) उत्पादनाच्या सर्व क्रिया
 is a component of promotion mix.
 a) Publicity & Public Relation
 हा वृद्धी मिश्रणाचा घटक आहे.
 a) जनसंपर्क व प्रसिद्धी
 Advertisement is.....
 a) Announcement
 जाहिरात म्हणजे.....
 a) उद्घोषणा
 of the following is Marketing concept.
 a) Product Concept
 विपणन संकल्पना आहे.
 a) वस्तू संकल्पना
 is known as father of co-operative movement.
 a) H. Calvert
- b) एक
 b) Co-operative Institute
 सहकारी संस्था
 b) 1941
 साली करण्यात आली.
 b) 1941
 Shaktikant Das
 शक्तीकांत दास
 b) 8th Dec, 2016
 8 डिसेंबर, 2016
 b) 2015
 2015
 b) 1st April, 1935
 1 एप्रिल, 1935
 b) Ahmedabad
 अहमदाबाद
 b) 1949
 1949
 b) Self- Motivation
 स्वयंप्रेरणा
 b) Creativity
 सृजनता
 b) Customer Satisfaction
 ग्राहक समाधान
 b) Customer Satisfaction
 ग्राहक समाधान
 b) Selling Concept
 विक्री
 b) Selling Concept
 विक्री संकल्पना
 b) Robert owen
 रॉबर्ट ओव्हन
- c) तीन
 Charitable Trust
 धर्मादाय संस्था
 1921
 1921
 Bipin Patel
 बिपिन पटेल
 31st March, 2017
 31 मार्च, 2017
 2017
 2017
 19th July, 1964
 19 जुलै, 1964
 New Delhi
 नवी दिल्ली
 1969
 1969
 Motivation
 अग्नीप्रेरणा
 Co-operation
 सहकार्य
 Producer Satisfaction
 उत्पादक समाधान
 Physical Distribution
 प्रत्यक्ष वितरण
 Impersonal Communication
 अद्वैयक्तिक संदेशावहन
 Holistic Marketing
 सर्वव्यापी विपणन
 Watkin
- d) सर्व
 Share Market
 भाग बाजार
 1935
 1935
 Arun Jetley
 अरुण जेटली
 17th Oct, 2017.
 17 ऑक्टोबर, 2017.
 2009
 2009
 11th June, 1995
 11 जून, 1995
 Pune
 पुणे
 1979
 1979
 Intellectual Motivation
 बौद्धिक प्रेरणा
 Creation
 निर्मिती
 None of these
 यापैकी नाही
 Consumer Behavior
 उपभोक्ता वर्तणूक
 Expenditure
 खर्च
 All of the above
 वरील सर्व
 C.R. Fay

Group Discussion
On
Democracy and Youth

Date: 13/08/2018

Submitted to
IQAC, Internal Quality Assurance Cell
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

Organized by
Department of History

Course Coordinator
Dr. Haji Nadaf

Head
Dr.H. D. Nadaf
Department of History
Padmabhushan Dr.Vasatraodada Patil Mahavidyalaya,
Tasgaon

2018-2019

Group Discussion:
PDVPMHISTORY

Date 13 August 2018

To
The Principal
Padmabhushan Dr. Vasantodada Patil
Mahavidyalaya,
Tasgaon, Dist- Sangli. 416312

Subject: Application for the Permission to conduct 'Group Discussion' activity.

Respected sir,

Our department of History is going to conduct Group Discussion activity on the topic entitled 'Democracy and Youth' on 13 August 2018. We are kindly requesting you to allow us to conduct this activity. Students of B.A. part III are going to participate in the same activity.

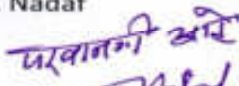
Thanking you

Yours Faithfully


(Dr. H. D. Nadaf)

HEAD
Department of History
P.D.V.P. College, Tasgaon.

Organizer: 1) Dr. H. D. Nadaf


Principal
Padmabhushan Dr. Vasantodada Patil
Mahavidyalaya, Tasgaon, (Sangli)

“ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षणप्रसार”

-शिक्षणमहर्षी डॉ.बापूजी साकुंभे


Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit
Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY
(2018-19)

Notice

Date: 13 August 2018.

All students of T.Y. B.A. History hereby informed that we are going to conduct Group Discussion activity on ‘**Democracy and Youth**’, **Monday, 13 August 2018**. Topics will be related to givens subjects and allotted to each topic, per eight students are enlisted along with this notice. It is compulsory to all students to participate with preparation.


Dr. H. D. Nadaf
HEAD
Department of History
P.D.V.P. College, Tasgaon.

"ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार"

-शिक्षणमहर्षी डॉ.बापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur Sanchlit

Padmabhushan Dr.Vasandraodada Patil Mahavidyalaya,
Tasgaon, Dist- Sangli, Pin-416312

DEPARTMENT OF HISTORY

(2018-2019)

B.A.III

13 August 2018

Group Discussion on 'Democracy and Youth'

Sr.No.	Name of The Student	Signature
1.	Ashtekar Santosh Shivaji	S.S. Ashlekar
2.	Babar Omkar Shivaji	Babar
3.	Bandgar Anil Abasaheb	Anil
4.	Chipate Bharat Baburao	Bharat
5.	Dabe Rohit Ramchandra	R.P. Dabe
6.	Deshmukh Saurabh Shashikant	Shashikant
7.	Ghorpade Vikas Nandkumar	Vikas
8.	Miss.Hatekar Pallavi Shivaji	Hatekar
9.	Hogale Vikas Vishvanath	Vikas
10.	IngawaleSandipMadhukar	S.M. Ingale
11.	Jadhav Akshaykumar Dhanaji	Akshay
12.	Jadhav Arjun Nadesh	Arjun
13.	Jadhav Swapnil Sanjay	Swapnil
14.	Jamadade Amit Subhash	Amit
15.	Kadam Akshay Shivaji	Akshay
16.	Kharade Akshay Pandurang	Akshay
17.	Kodag Pravin Kumar	Pravin
18.	Koli Malleshee Umakant	Malleshee
19.	Lokhande Nilesh Bhaskar	Nilesh
20.	Mali Sujit Ishwar	S.Mali

21.	Mane Ajay Balasaheb	<u>Mane</u>
22.	Mane Ranjeet Ramesh	<u>M.R. Ball</u>
23.	Mane Rushikesh Patang	<u>Rushikesh</u>
24.	Mohite Nilesh Eknath	<u>Nilesh</u>
25.	More Chandrakant Sampat	<u>Chandrakant</u>
26.	Mulani Rehan Raju	<u>Rehan</u>
27.	Mulani Sajid Jainuddin	<u>Sajid</u>
28.	Nadaf Firoj Sikandar	<u>Firoj</u>
29.	Nadaf Niyaj Amin	<u>Niyaj</u>
30.	Miss. Patil Aishwarya Rajendra	<u>A.R. Patil</u>
31.	Patil Nilesh Kumar	<u>N.K. Patil</u>
32.	Patole Ajay Krushna	<u>Ajay</u>
33.	Sadakale Sunil Deepak	<u>S.S. Sadakale</u>
34.	Salunkhe Sunny Rajesh	<u>S.R. Salunkhe</u>
35.	Sukhadev Amit Bajarang	<u>A.S.B</u>
36.	Sutar Akshay Dattatray	<u>Akshay</u>
37.	Waghmode Shivdas Balasaheb	<u>Shivdas</u>
38.	Ware Nitinkumar Ashok	<u>N.A. Ware</u>

Shukh
HEAD
 Department of History
 P.D.V.P. College, Tasgaon.

Group Discussion Report

Title of Programme	: ‘Democracy and Youth: 21st Century’
Organizing Department	: History
Date	: 13th August 2018.
Venue	: Main Building Room. No. 26
No. of Participants	: 38, M-36, Female-02
Coordinator	: Dr. Haji Nadaf
President	: Prin. Dr. R.R. Kumbhar

Our Department organized group discussion on Democracy and Youth’ on 13th August 2018. Democracy is a form of government in which the people have the authority to choose their governing legislation. Democracy is a system of processing conflicts in which outcomes depend on what participants do, but no single force controls what occurs and its outcomes.



The uncertainty of outcomes is inherent in democracy. Democracy makes all forces struggle repeatedly to realize their interests and devolves power from groups of people to sets of rules.



Here comes the responsibility of the future of the nation. There is a large population of youth in India. More than 50 percentages of Indian people belong to the age below 35 years. This is a great hope for the nation. They are looking for a better India, and a new world order. Significant social changes can come through them. They are for socio-economic development. India is a very large country full of diversities – linguistically, culturally and religiously. The level of education of youth is a key to both the successful functioning of democracy and socio-economic development of the country. And perhaps, more importantly, it is an essential condition for human dignity. Right education makes them conscious of the principles of liberty, equality, secular and democratic. Recently, the Right to Education is provided as a fundamental right. We hope that this will help the cause of educating the children universally. Lack of education is the root cause of all kinds of deprivations and inequalities.

All groups have stated about democracy on following points related to this subject

- Sex Ratio, according to 2011 Census, in India is only 914 female children per 1000 male children. This is lower than the 2001 Census which reported child sex ratio of 927 female children per 1000 male children.
- Inequality in the society in the name of gender is a curse on India today.
- Casteism: The caste system which presumably originated in the division of labour in the ancient society has become a more or less rigid group classification, based on birth.
- Communalism: The biggest threat to Indian Democracy and freedom is Communalism and religious fundamentalism.
- Religious Fundamentalism also reinforces communalists in exploiting both religion and politics.
- In order to integrate all these into one we need to have a wider outlook and therefore, regionalism and cultural jingoism cannot be accepted.
- Negative propaganda: Freedom with responsibility is the need of the hour. Citizens must be taught positive thinking. Media and the social agencies must spread positivism to all. Remember positive reinforcement brings positive outcome. Negativity breeds negativity.
- Corruption: We need to uproot the widespread corruption at all levels through imparting values of honesty, commitment, compassion, responsibility and sincerity.
- Exploitation: In a true democratic free country no one should be a victim of exploitation.
- Drugs and violence: Youth must stand against drug addiction and violence of all sorts.

The groups have responded to well discussion. Eventually all participants agree with it should be the aware and attention of good governance. This entire activity completed by Dr. H.D. Nadaf and Prin. Dr. R.R. Kumbhar guided and cooperated for the programme.


HEAD
Department of History
P.D.V.P. College, Tasgaon.

Coordinator and Head

Group Discussion:
PDVPMHISTORY

"Dissemination of Education through Knowledge, Science and Culture".

--Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

Padmabhushan Dr.Vasandraodada Patil

Mahavidyalaya, Tasgaon

Tal. Tasgaon. Dist. Sangli-416 312 (Maharashtra)

(Affiliated to Shivaji University, Kolhapur)

**DEPARTMENT OF
COMPUTER-SCIENCE
And
Computer Application
(2020-21)**

A Guest lecture On

Data Science



"Dissemination of Education through Knowledge, Science and Culture".

--Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr.Vasatraodada Patil

Mahavidyalaya, Tasgaon

Tal. Tasgaon. Dist. Sangli-416 312 (Maharashtra)

(Affiliated to Shivaji University, Kolhapur)

Title of the Guest Lecture

" Data Science"

Organizer:Department Of Computer Science&Appication
and
Internal Quality Assurance cell (IQAC) PDVP College,
Tasgaon

Total No Of Student Participated:22

Total no of faculty conducted : 02

Brochure of Guest Lecture



Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon, Sangli

Department of Computer Science and IQAC

Organizes
Guest Lecture on

“Introduction to Data Science”

Date: Saturday, 10th July 2021

Time: 1:00 pm onwards

Resource Person

Prof. Mrs. Mariyam Maniyar

Experience 13 Years

Assistant Professor, K. K. Wagh Institute of Engineering Education & Research, Nashik

Session will be Online in Google meet platform

Google Meet link: <https://meet.google.com/ruu-sfvw-ndc>

Prof. Vijay Kumbhar
Head, Computer Science Department

Dr. Milind S. Hujare
Principal, P.D.V.P College

Permition letter

दि. 5/7/2021

प्रति,
मा. प्राचार्य,
पी. डी. व्ही. पी. महाविद्यालय,
तामसाव

विषय :- Guest lecture साठी परवानगी
मिळवावासात.

मा. महोदय,

आपला महाविद्यालयीय B.Sc. -III / B.C.A -III

(संगणकशास्त्र / संगणक ॲप्लीकेशन) या वर्गातील
विद्यार्थ्यांसाठी Data Science या Current विषयावरील.

प्रा. मरीचम मनिवार (नाबिकु इन्जिनियरिंग कॉलेज), नाशिक
लांचे lecture देवण्याची परवानगी मिळवावासात.

कुलदे.

संजु
()

शाखा विरगमु,



(प्रा. कुमार व्ही. पी.)

**“Dissemination of Education for Knowledge, Science and Culture” -Shikshanmaharashi Dr. Babuji Salunkhe
Shri Swami VivekanandShikshanSanstha, Kolhapur’s
PadmabhushanDr.vasatraodadaPatilMahavidyalaya, Tasgaon
Department of Computer Science And Application
A Guest Lecture in Data Science
Academic year 2020-21**

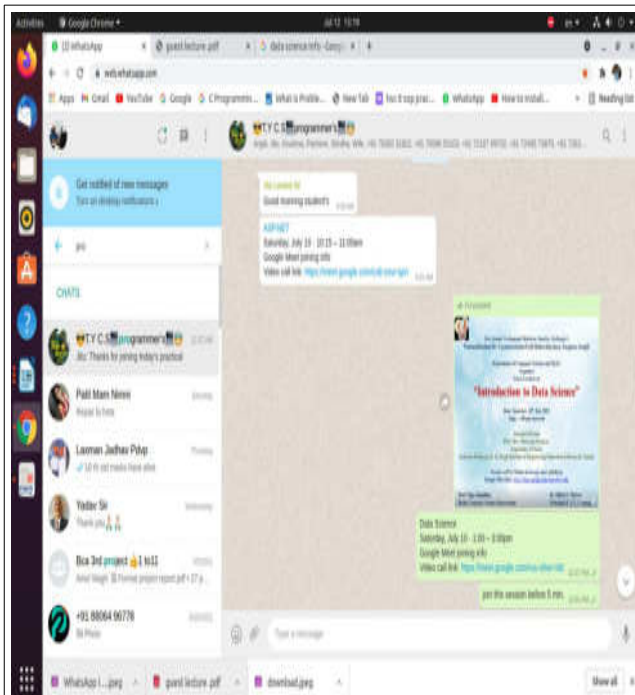
NOTICE

6 July 2021

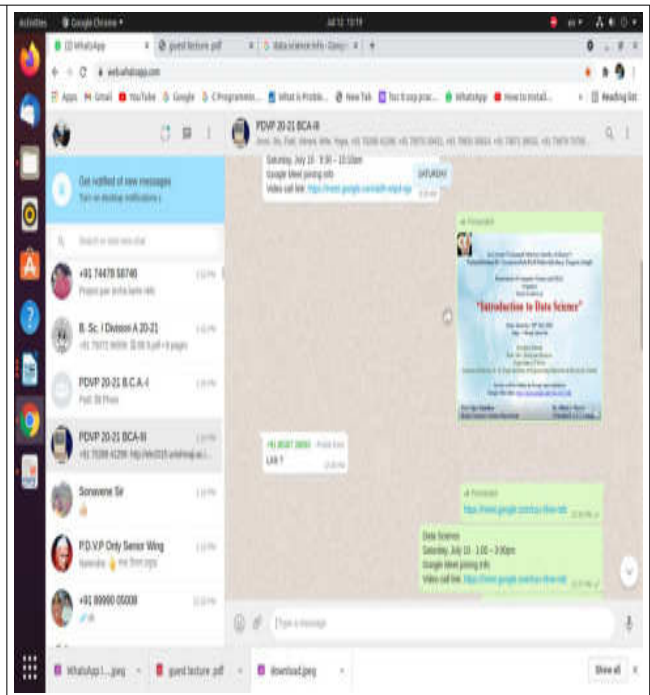
All student’s of B.Sc. PART-III (Computer students)) and BCA Part - 3 are hereby informed that, our college is going to organise guest lecture on “Data Science” for this current academic year .All students should participate this online session at saturday 10-7-2021 before 5 min .link is provided in class whatsapp group.



Communicate Student's through Whatsapp

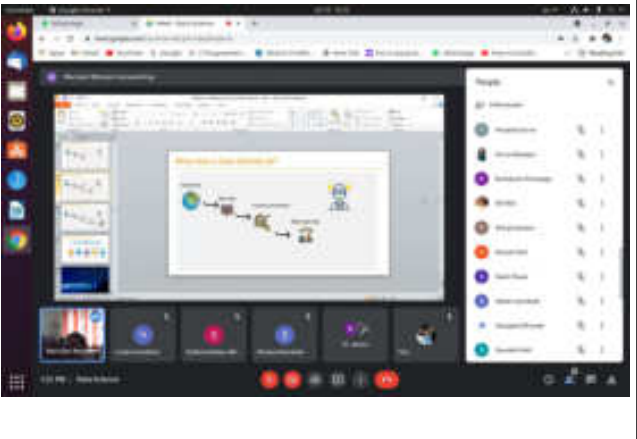
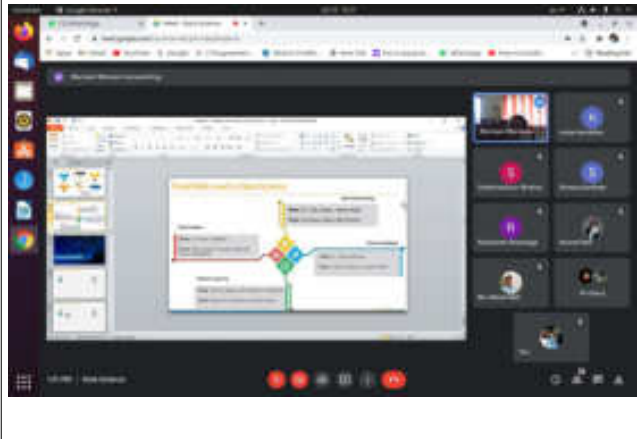
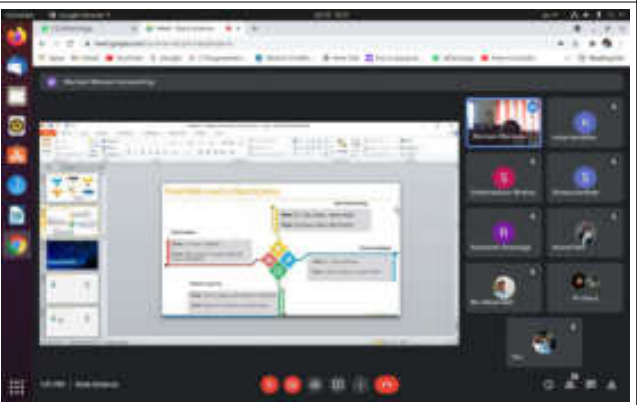
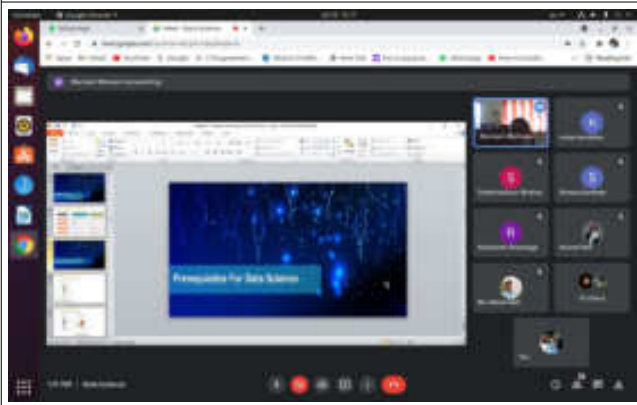
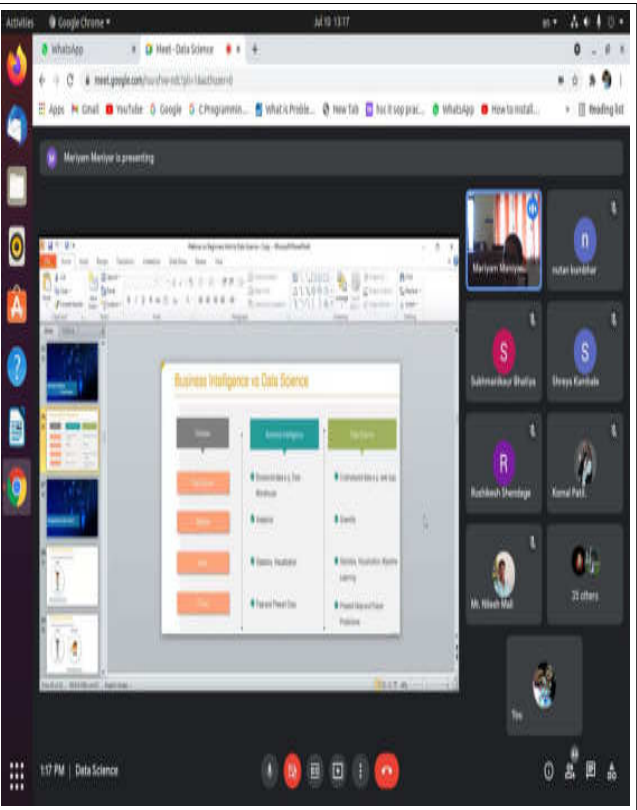
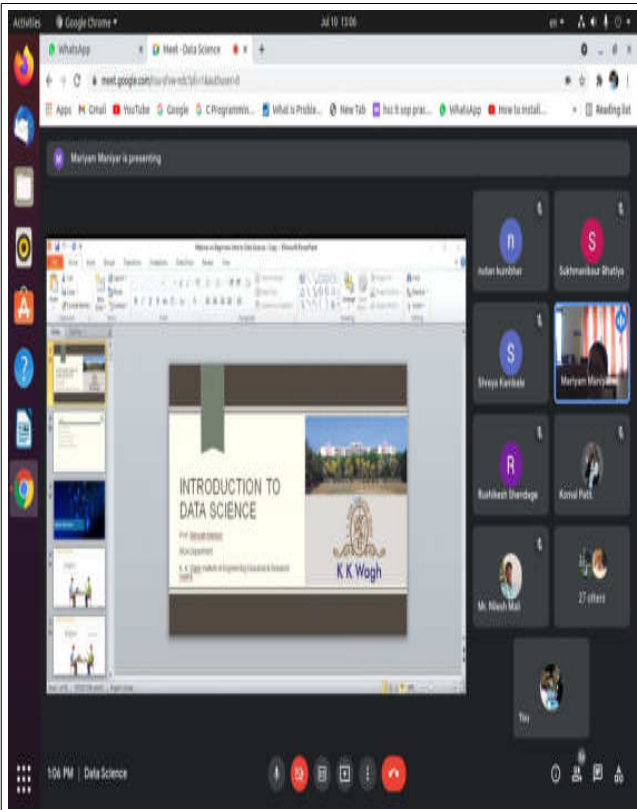


B.Sc.-3 whatsapp group

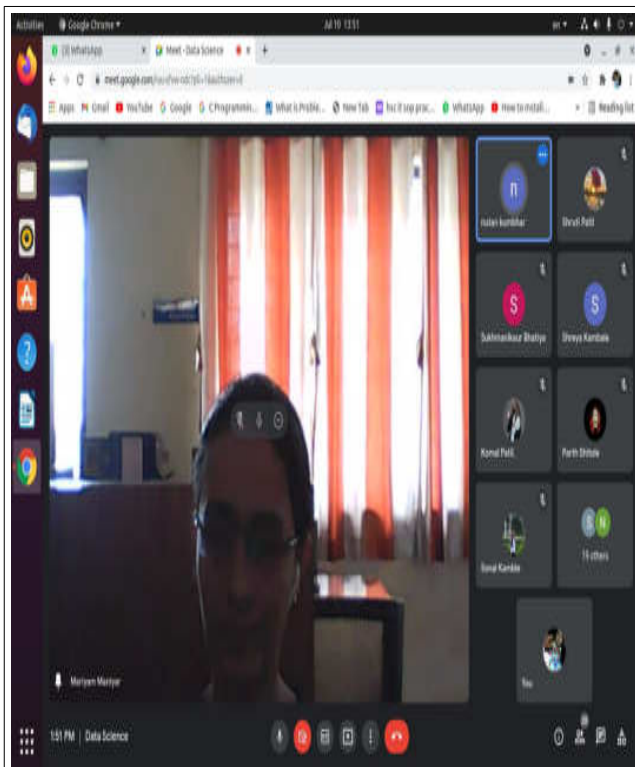


BCA-3 whatsapp group

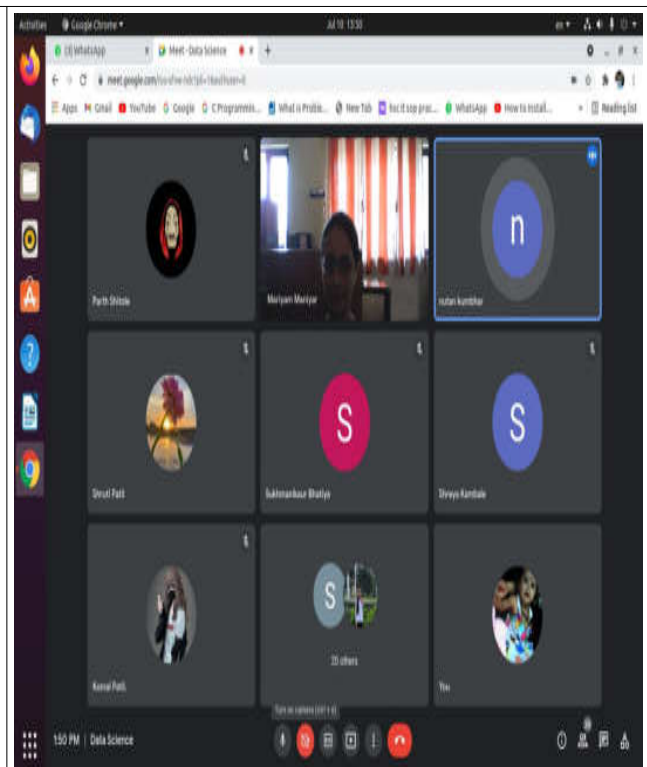
Slide Presentation



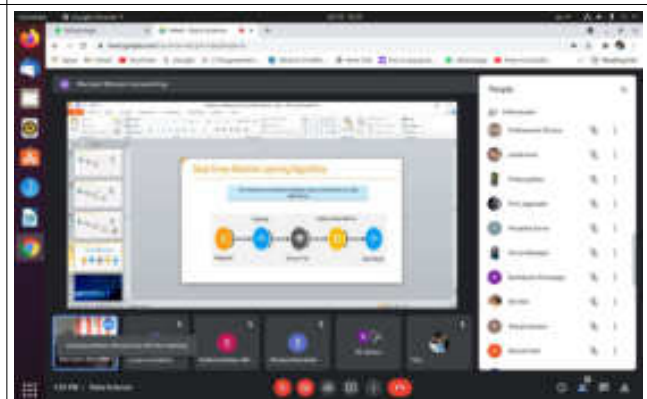
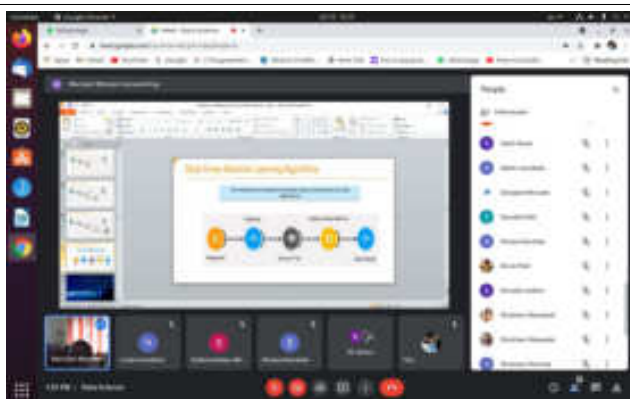
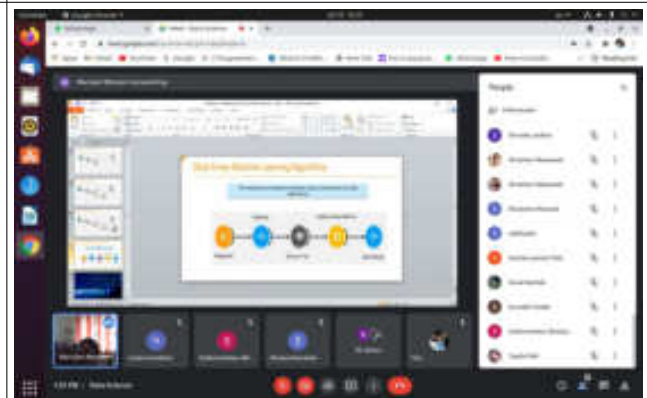
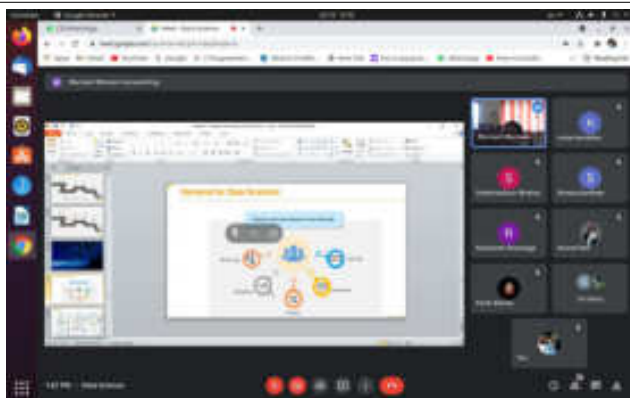
Slide Presentation



Guest



Student Partitipation



Presenty

- * Meet - ruu-sfw-ndc
 - * Meeting code: ruu-sfw-ndc
 - * Created on 2021-07-10 01:00:10
- sub: Data Science

Full Name	First Seen	Time in Call
Anjali sapkal	2021-07-10 01:00:13	00:47:34
Nisha Gaming	2021-07-10 01:00:14	00:48:54
Pooja Jangam	2021-07-10 01:00:15	00:21:06
Pooja Patil	2021-07-10 01:00:16	00:49:06
Pratiksha Patil	2021-07-10 01:00:17	00:50:08
Priyanka Surve	2021-07-10 01:00:18	00:49:27
Purva Mahajan	2021-07-10 01:00:19	00:49:55
Sakshi Thorbole	2021-07-10 01:00:20	00:49:56
Snehal Laxman Patil	2021-07-10 01:00:21	00:39:14
Snehal Sudam Patil	2021-07-10 01:00:22	00:00:51
Swaleha Attar	2021-07-10 01:00:10	00:50:33
Vijay Kumbhar	2021-07-10 01:00:10	00:50:33
Marium maniyar	2021-07-10 01:00:11	00:50:34
Mahesh Chavan	2021-07-10 01:00:12	00:50:35
Nikhil Patil	2021-07-10 01:00:13	00:50:36
Yogita Patil	2021-07-10 01:00:14	00:50:37
Komal Pawar	2021-07-10 01:00:15	00:50:38


Introduction

Data science is related to data mining, machine learning and big data. Data science is a "concept to unify statistics, data analysis, informatics, and their related methods" in order to "understand and analyze actual phenomena" with data. ... However, data science is different from **computer science and information science**.

About session

This session is conducted by **Prof. Marium Maniyar**, **nashik Enggearing College, Nashik Department of MCA**. This session handed by **Prof. Kumbhar V.T.** and vote of thanks taken by **Prof. Kumbhar N.V.**


Alka Inamdar
IQAC Co-Ordinator,
P.D.V.P. Mahavidyalaya,
Tasgaon.


Prin. Dr. Milind S. Hujare
Principal
Padmabhushan Dr. Vasantodada Patil
Mahavidyalaya, Tasgaon (Sangli).

COURSE ON COMMUNICATION SKILLS



**Padmabhushan Dr. Vasantodada Patil Mahavidyalaya
Tasgaon**

"Dissemination of Education for Knowledge, Science and Culture" -Shikshanmaharshi Dr. Bapuji Salunkhe
Shri Swami Vivekanand Shikshan Sanstha Kolhapur's
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon, Dist. Sangli-416312, Maharashtra

Report of certificate course on **"Communication skills"**

❖ **Event: Certificate course on "Communication Skills"**

❖ **Duration: 5th June 2021 to 5th August 2021**

❖ **Venue: online platform (Zoom, Google Classroom, You tube, What's App)**

❖ **Participants:**

Total Student	Male	Female
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157	25	132
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Prepared by Dr M. U. Patil; Department of Chemistry

Contents

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Objectives of course:

The objectives of this course are:

- ✍ To provide an overview of Prerequisites to Communication skills to students.
- ✍ To put in use the basic mechanics of Grammar.
- ✍ To provide an outline to effective Communication.
- ✍ To impart the correct practices of the strategies of Effective writing.

Course Outcomes:

On completion of this course, the students will be able to.....

CO1. To be familiar with the complete course outline/Course Objectives/Learning Outcomes/ Evaluation Pattern & Assignments

CO2. To participate in an online learning environment successfully by developing the implication-based understanding of discussion boards & Referencing Styles.

CO3. To demonstrate his/her ability to write error free while making an optimum use of correct Vocabulary & Grammar.

CO4. To stimulate their Critical thinking by designing and developing clean and lucid writing skills.

CO5. To demonstrate his verbal and non-verbal communication ability through presentations

Emphasis:

This curriculum was developed by the College for students to develop their communication and presentation skill which help to them to present them during interviews.

Effective communication is an integral part of life. Communication is a process of exchanging ideas, messages, information etc. through verbal or nonverbal communication. In this course, the focus had been on improving LSRW skills, i.e. listening, speaking, reading and writing. Students learned how to communicate effectively. Classroom assignments/activities specifically designed to encourage students to play an active role for enhancing their knowledge and developing learning strategies.

Blended learning - online and traditional lectures with other active teaching methodologies, such as group discussions, discussions and assignments enable students towards understanding various aspects of effective communication.

Class participation is a fundamental aspect of this course student are encouraged to actively take part in all group activities and

to give an oral group presentation. Students noticed a huge improvement; they had taken a lot of fun in this learning process.

Broacher of course:

"Dissemination of Education through Knowledge, Science and Culture" -Shikshanamakar:hi Dr. Bopuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha Kolhapur's
Padmabhushan Dr. Vasantrodada Patil
Mahavidyalaya, Tasgaon,
 Sangli-416312, Maharashtra, India
 (Affiliated to Shivaji University Kolhapur)

Online Two Months Course on
Communication Skills
5th June to 5th August 2021
 Course Platform: Google Classroom

Objectives of Course

Part 1

1. Self Introduction
2. Lectures on personality development
3. Do write and have write
4. Articles
5. Semos
6. Auxiliary notes
7. Wt questions
8. Letter writing

Part 2

1. Interview skills
2. Curriculum vitae
3. Dressing sense
4. Doe and Doeth
5. Thanks letter
6. Poster and posters

Part 3

1. Presentation skill
2. Speeches
3. Manners and Etiquettes
4. Anchoring
5. Group Discussion
6. Debate

Application procedure

Join the Course up to 30th May 2021

1. Registrar yourself through this link : <http://bit.ly/3communicationskill>
2. Join Google Classroom for details of course
3. Course material included on Google Classroom
4. Join WhatsApp Group for further Communication
5. Links of Google Classroom and WhatsApp Group are included in registration form

Eligibility condition

- The course is open for Arts, Commerce, Science graduation and post graduation students.
- E-certificate will be issued to the participants after successful completion of course
- No Registration fee

Organising Committee

Dr. Milind S. Hujare	Principal
Dr. Megha U. Patil	Course Coordinator
Dr. Abha P. Inamdr	Dr. Jeevan S. Ghodake
IQAC Coordinator	IQAC Co-coordinator
Mr. Dattatray Y. Sakhare	Dr. Sachinkumar K. Shinde
Course Convenor	Technical coordinator
Invited Talk	
Mr. Ajit S. Pachore	

Course content:

Dissemination of education for knowledge, Science and culture-Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shiksha Sanshta, Kolhapur's

Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon

Course on Communication Skill

Syllabus

2020-21

Part 1

1. Self Introduction
2. Lectures on personality development
3. Be verbs and have verbs
4. Articles
5. Tenses
6. Auxiliary Verbs
7. Wh questions
8. Letter writing

Part 2

1. Interview skills:
2. Curriculum vitae
3. Dressing sense
4. Dos and Don'ts
5. Thanks letter
6. Poster and gestures

Part 3

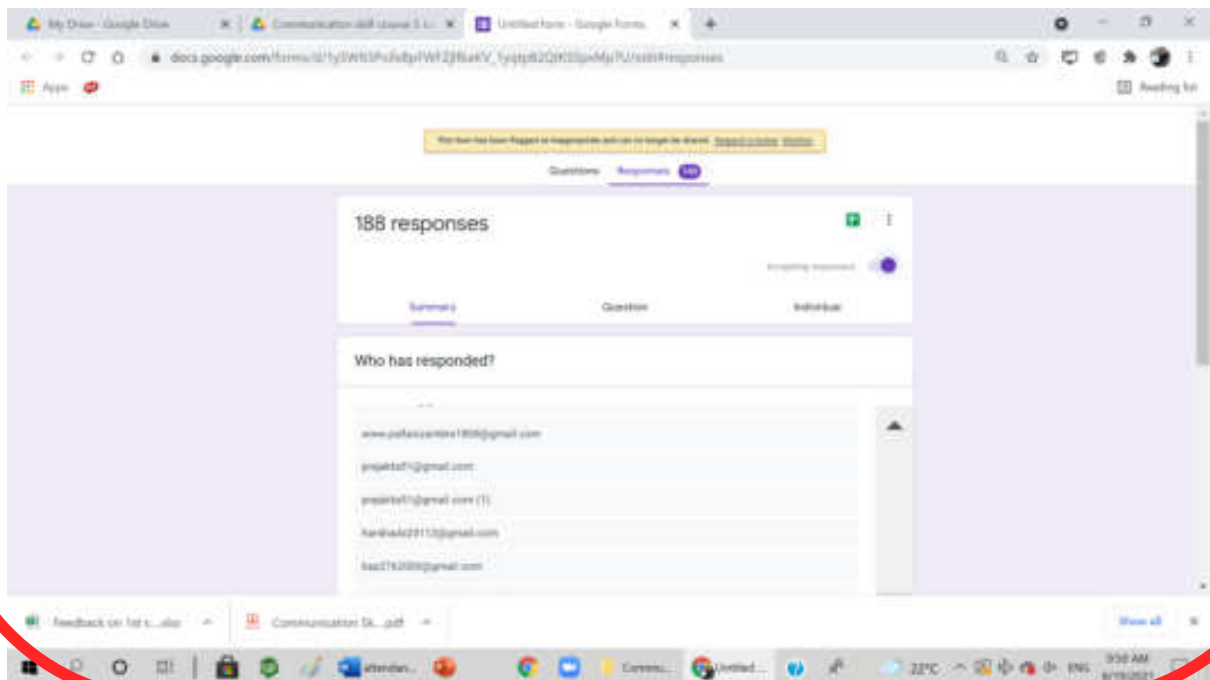
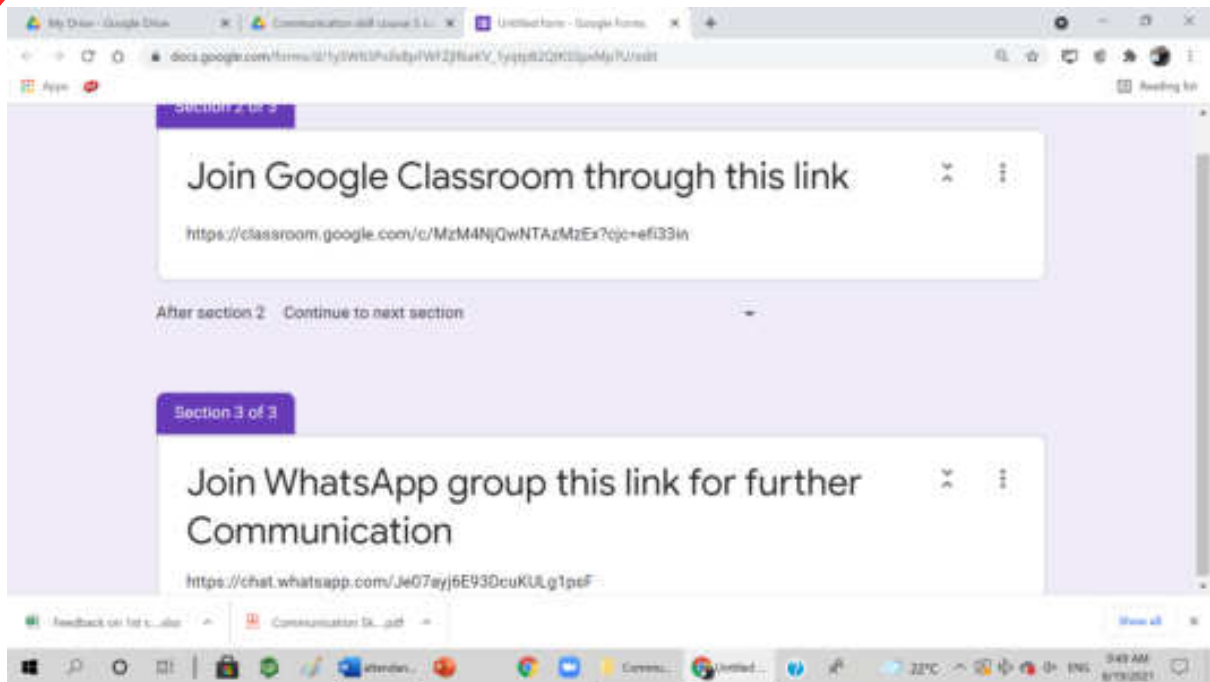
1. Presentation skill
2. Speeches
3. Manners and Etiquettes
4. Anchoring
5. Group Discussion
6. Debate

Teacher incharge
Dr. Megha U. Patil

Notice to students:

The screenshot shows a Google Form titled "Online Course on Communication Skills" from Padmabhushan Dr. Vasantrodada Patil Mahavidyalaya, Tasgaon. The form has two visible sections: "Email" with a "Valid email" error message and a "Change settings" link, and "Name of Student" with a "Short answer text" input field. The browser address bar shows the form URL: docs.google.com/forms/d/1y5Wt3hUvbyfW123fkaKv...yqpp820K03pMjPU/widt. The Windows taskbar at the bottom shows the time as 9:43 AM on 6/19/2021.

The screenshot shows the continuation of the Google Form. It features three sections: "Class" with a "Short answer text" input field, "Contact Number" with a "Short answer text" input field, and "College Name" with a radio button option for "PDVP College, Tasgaon". The browser address bar and Windows taskbar are identical to the previous screenshot, showing the time as 9:43 AM on 6/19/2021.



Participant's list:

Suraj Hanmant Khedkar	Bsc3	9112810094	Pdvp College, Tasgaon
Patil Kirti Suresh	Bsc-3	9561683208	Pdvp College, Tasgaon
Aishwarya Nilkanth Dombe	Bsc-3	9359178576	Pdvp College, Tasgaon
Gaikawad Poonam Shankar	Bsc3rd	7219139115	Pdvp College, Tasgaon
Patil Hanmant Dnyaneshwar	Bsc-Ii	7391859646	Pdvp College, Tasgaon
Rushikesh Ashok Chandavale	Bsc-Ii	9960924584	Pdvp College, Tasgaon
Jadhav Sima Anandrao.	Bsc-Ll	9022348304	Pdvp College, Tasgaon
Pradnya Dadaso Chandanshive	F. Y. B. Sc.	8668320954	Pdvp College, Tasgaon
Pratiksha Sanjay Yadav	Fy Bcom	9503848957	Other
Yadav Sukanta Sambhaji	Fy Bsc	8767192185	Other
Parvati Prakash Mali	M Sc 2	7066972943	Pdvp College, Tasgaon
Snehali Raghunath Mali	M. Sc	8788909868	Pdvp College, Tasgaon
Bhosale Purva Appaso	M. Sc I	7058727373	Pdvp College, Tasgaon
Salunkhe Anjali Gajanan	M. Sc I	7350954416	Pdvp College, Tasgaon
Sakshi Ganapati Bhatale	M.Sc - 1(Statistics)	8421191005	Pdvp College, Tasgaon
Sonam Patil	M.Sc (Statistics)	9130351782	Pdvp College, Tasgaon
Komal Keshav Mali	M.Sc 2	7843074935	Pdvp College, Tasgaon
Mane Monika Popat	M.Sc. Ll	7758000349	Pdvp College, Tasgaon
Snehal Maruti Patil	M.Sc.1st (Statistics)	9850931360	Pdvp College, Tasgaon
Uma Dattatray Patil	M.Sc=1 (Statistics)	9284643805	Pdvp College, Tasgaon
Supriya Mahavir Patil	M.Sc-1	9860490565	Pdvp College, Tasgaon
Avadan Amol Ramchandra	M.Sc-1	9130329976	Pdvp College, Tasgaon
Patil Shrutika Bharat	Msc 1	9130497131	Pdvp College, Tasgaon
Khilari Kiran Dhondiram	Msc 1st Year	9503305449	Pdvp College, Tasgaon
Simran Ajjij Tamboli	Msc 2 (Statistic)	7262855992	Pdvp College, Tasgaon
Rani Siddharth Waghmode	Msc1	8830505677	Pdvp College, Tasgaon
Snehal Gajanan Gidde	Msc-1 (Stat)	9503288650	Pdvp College, Tasgaon
Samruddhi Sanjay Mane	Msc-2 Year	8669860288	Pdvp College, Tasgaon
Mulani Mijaba Firoj	S.Y Bsc	9172016563	Pdvp College, Tasgaon
Saurabh Subhash Kore	Third Year	7448231415	Other
Raturaj Kharat	Ty Bsc	9307541981	Pdvp College, Tasgaon
Rupali Mali	Ty Bsc	7588366033	Pdvp College, Tasgaon
Irfan Harun Mujawar	Ty Bsc	9503978052	Pdvp College, Tasgaon

COURSE ON COMMUNICATION SKILLS

Rutuja Vilas Nalawade	Bsc I (A)	9356688098	Pdvp College, Tasgaon
Nikhil Kishor Gore	Bsc Ii	7666892950	Other
Snehal Sanjay Patil	Bsc Ii	9175719106	Pdvp College, Tasgaon
Patil Ratnadeep Pradeep.	Bsc Iii	9518364719	Pdvp College, Tasgaon
Swati Ravindra Waghmare	Bsc L	8010595709	Pdvp College, Tasgaon
Rutuja Mohan Mane	Bsc L	9067559504	Pdvp College, Tasgaon
Kajal Baburao Rasale	Bsc L	9284589404	Pdvp College, Tasgaon
Akshada Sheshrao Kamble	Bsc Ll	8623008812	Pdvp College, Tasgaon
Pawar Kajal Vishnu	Bsc Ll	9356548058	Pdvp College, Tasgaon
Rina Rajendra Jadhav	Bsc Ll	9529879196	Pdvp College, Tasgaon
Shinde Prachi Ankush	Bsc Ll	7219623436	Pdvp College, Tasgaon
Prasad Ravindra Aradhya	Bsc Ll	7387154845	Pdvp College, Tasgaon
Manisha Sanjay Kharade	Bsc Ll B	9529490679	Pdvp College, Tasgaon
Ajay Netaji Zambre	Bsc Lll	9860818347	Pdvp College, Tasgaon
Chavan Nilam Jalindar	Bsc Lll	9370401987	Pdvp College, Tasgaon
Chavan Omkar Dilip	Bsc Part 2	9552638358	Pdvp College, Tasgaon
Bhosale Sayali Shankar	Bsc Second Year	7264873662	Pdvp College, Tasgaon
Khatal Jyoti Kundlik	Bsc Second Year	9766084342	Pdvp College, Tasgaon
Sujeet Suhas Madvanna	Bsc Sy	7249334808	Other
Sonali Baban Jadhav	Bsc Ty	9765050092	Pdvp College, Tasgaon
Susmitha Wakrekar	Bsc(Statistics)	9356063017	Pdvp College, Tasgaon
Jadhav Harshada Mahadev	Bsc.1	7499556878	Pdvp College, Tasgaon
Kiran Anil Patil	Bsc:-3	9373188511	Pdvp College, Tasgaon
Nishant Maruti Bhosale	Bsc 1	8080245299	Pdvp College, Tasgaon
Sushama Sahebarav Jadhav	Bsc1	9325364545	Pdvp College, Tasgaon
Radhika Tanaji Bhosale	Bsc1	9322361517	Pdvp College, Tasgaon
Anjali Ramesh More	Bsc-1	8421879554	Pdvp College, Tasgaon
Shinde Sakshee Sushilkumar	Bsc-1	8999789971	Pdvp College, Tasgaon
Rajkanya Mohan Jadhav	Bsc-1	9322965979	Pdvp College, Tasgaon
Rohan	Bsc1 St	8421395472	Pdvp College, Tasgaon
Sudiksha Dinesh Jadhav	Bsc2	9022391232	Pdvp College, Tasgaon
Omkar Jadhav	Bsc2	8459430917	Pdvp College, Tasgaon
Yadav Mayuri Sunil	Bsc2	7028777466	Pdvp College, Tasgaon
Rohan Rajendra Chavan	Bsc2	9022183042	Pdvp College, Tasgaon
Patil Sakshi Pandit.	Bsc-2	9970061716	Pdvp College, Tasgaon
Mali Nivedita Gajanan	Bsc-2	9503047032	Pdvp College, Tasgaon
Kamble Tejaswini Sultan	Bsc-2	7410703215	Pdvp College, Tasgaon
Mane Poojashri Balaso	Bsc2nd	9021208571	Pdvp College, Tasgaon
Pawar Vijay Sukhadev	Bsc3	7219427787	Pdvp College, Tasgaon
Jyoti Ashok Patil	Bsc3	8956014931	Pdvp College, Tasgaon
Arati Anil Shinde	Bsc3	9021670516	Pdvp College, Tasgaon

COURSE ON COMMUNICATION SKILLS

Karan Krishnarao Ghotkar	Bsc 1	8080915791	Pdvp College, Tasgaon
Prathmesh Suresh Shinde	Bsc 1	7620441499	Pdvp College, Tasgaon
Ratnadip Sanjay Jadhav	Bsc 1	9860291161	Pdvp College, Tasgaon
Ankita Sanjay Mali	Bsc 1	7262828107	Pdvp College, Tasgaon
Sakshi Patil	Bsc 1	7083072575	Pdvp College, Tasgaon
Suvarna Shankar Shelke	Bsc 1	7709793605	Pdvp College, Tasgaon
Ashutosh Jalindar Rakshe	Bsc 1	8530731431	Pdvp College, Tasgaon
Jadhav Snehal Sadanand.	Bsc 1	9021504191	Pdvp College, Tasgaon
Nalawade Rutuja Santoshkumar	Bsc 1 Year	8308430986	Pdvp College, Tasgaon
Dnyaneshvari Jitendra Hingmire	Bsc 1 Year	8263029958	Pdvp College, Tasgaon
Mane Ritesh Hanmant	Bsc 1st	8767006385	Pdvp College, Tasgaon
Malame Rushikesh Sidram	Bsc- 1st	7414967820	Pdvp College, Tasgaon
Nikita Minil Edake	Bsc 1st Year	7558548244	Pdvp College, Tasgaon
Patil Rohan Manik	Bsc 2	9284891116	Pdvp College, Tasgaon
Shinde Sourabh Ganesh	Bsc 2	8600868560	Pdvp College, Tasgaon
Chougule Rushikesh Sukhadeo	Bsc 2	9370789689	Pdvp College, Tasgaon
Shelke Anuja Anil	Bsc 2	8605508780	Pdvp College, Tasgaon
Rutuja Adhikrao Patil	Bsc 2	8530507798	Pdvp College, Tasgaon
Pawar Priti Kishor	Bsc 2	9766921336	Pdvp College, Tasgaon
Parit Poonam Rajaram	Bsc 2 Year	9172399659	Pdvp College, Tasgaon
Mohabbat Shikalgar	Bsc 2nd	9307078276	Pdvp College, Tasgaon
Gautami Shrikant Shintre	Bsc 2nd	9607180880	Pdvp College, Tasgaon
Kognole Shreya Shital	Bsc 2nd Year	7775874507	Pdvp College, Tasgaon
Smita Ashok Patil	Bsc 3	9373689078	Pdvp College, Tasgaon
Patil Komal Shashikant	Bsc 3	9326733622	Pdvp College, Tasgaon
Jaydip Yuvaraj Patil	Bsc 3	9359020649	Pdvp College, Tasgaon
Shinde Rutuja Ashok	Bsc 3	9359621747	Pdvp College, Tasgaon
Shranita Kamble	Bsc 3	9604927362	Other
Snehali Jambu Mainkar	Bsc 3	7709850424	Pdvp College, Tasgaon
Mohite Shweta Machindra	Bsc 3	8698390769	Pdvp College, Tasgaon
Amruta Dinkar Patil	Bsc 3	8623071024	Pdvp College, Tasgaon
Mane Ankita Adikrao	Bsc 3 Botany	9422530804	Pdvp College, Tasgaon
Pratiksha Tanaji Gaikwad	Bsc 3 Chemistry	9172502049	Pdvp College, Tasgaon
Pooja Tukaram Edake	Bsc First Year	8767992986	Pdvp College, Tasgaon
Prajakta Mahadev Patil	Bsc First Year	9130558644	Pdvp College, Tasgaon
Jadhav Aparna Ramdas	Bsc Fy	8600325569	Pdvp College, Tasgaon
Ankita Anandrao Ghodake	Bsc Fy	9022313848	Pdvp College, Tasgaon
Shahista Maheebul Mulani	Bsc I	9503596454	Pdvp College, Tasgaon
Nitali Ashok Mane	Bsc I	7249112870	Pdvp College, Tasgaon
Shinde Samruddhi Rajaram	Bsc I	7822073857	Pdvp College, Tasgaon
Aditi Avinash Vitekar	Bsc I	8767180136	Pdvp College, Tasgaon

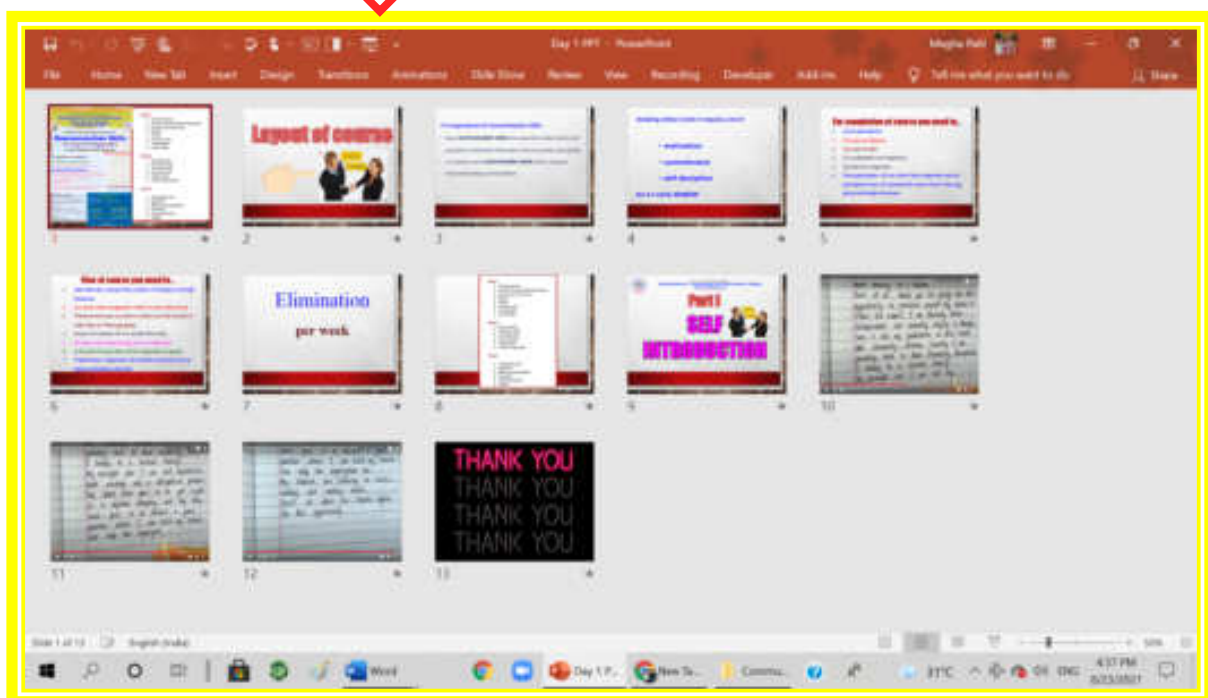
COURSE ON COMMUNICATION SKILLS

Name Of Student	Class	Contact Number	College Name
Megha Sanjay Mane	13	9579383017	Pdvp College, Tasgaon
Amuradha Ramesh Patil	B Sc 1st	7499686184	Pdvp College, Tasgaon
Prathamesh Chandrakant Pawar	B. Sc - Ii	9503873256	Pdvp College, Tasgaon
Pallavi Narayan Mali	B. Sc 1	8482986190	Pdvp College, Tasgaon
Salunkhe Shivani Shivaji	B. Sc 2nd Year	8830886953	Pdvp College, Tasgaon
Shinde Dipali Shamrao	B. Sc 3rd	7620682139	Pdvp College, Tasgaon
Prajakta More	B. Sc L	7558677107	Pdvp College, Tasgaon
Patil Swati Kisan	B. Sc. 1	8010497771	Pdvp College, Tasgaon
Dipti Shahaji Salunkhe	B. Sc-1	8605389916	Pdvp College, Tasgaon
Salunkhe Divya Shahaji	B. Sc-2	8605389916	Pdvp College, Tasgaon
Mohite Amruta Manik	B. Sc3	9021531108	Pdvp College, Tasgaon
Pranali Pawar	B. Tech 2nd Year	9370075049	Other
Nalawade Komal Sharad	B.Sc 1st Year	9730592156	Pdvp College, Tasgaon
Apurva Thorat	B.Sc 2 Yr	7020439172	Pdvp College, Tasgaon
Patil Rutuja Bapusaheb	B.Sc 3	9665604441	Pdvp College, Tasgaon
Shradha Pawar	B.Sc 3	9673641856	Pdvp College, Tasgaon
Komal Suryawanshi	B.Sc 3	9764485232	Pdvp College, Tasgaon
Mali Ashwini Shivaji	B.Sc I	8552047648	Pdvp College, Tasgaon
Pawar Anuradha Amrut	B.Sc I Year	9860376122	Pdvp College, Tasgaon
More Pooja Vishnu	B.Sc Ii	7378673772	Pdvp College, Tasgaon
Namrata Kumar Sutar	B.Sc Ii	7498072804	Pdvp College, Tasgaon
Shinde Omkar Vilas	B.Sc Iii	9096151611	Pdvp College, Tasgaon
Patil Punam Chandrakant	B.Sc L	7798933872	Pdvp College, Tasgaon
Pathan Swaliya Jamirkhan	B.Sc LI Year	8766462085	Pdvp College, Tasgaon
Mali Sujata Sadashiv	B.Sc.	8806069770	Pdvp College, Tasgaon
Shraddha Bajirao Koshti	B.Sc. 1	9049575622	Pdvp College, Tasgaon
Snehal Dinkar More	B.Sc.1 St	9028801974	Pdvp College, Tasgaon
Rutuja Badal Kamble	B.Sc.2	9356383510	Pdvp College, Tasgaon
Patil Akshada Sunil	B.Sc.2	9960465202	Pdvp College, Tasgaon
Abhijit Sanjay Nimbalkar	B.Sc.Ii	9860532119	Pdvp College, Tasgaon
Shweta Chandrakant Ghodake	B.Sc.LIi	9518999034	Pdvp College, Tasgaon
Tasnim Riyaj Sayyad	B.Sc-1	8010743345	Pdvp College, Tasgaon
Adsul Sushant Namdev	B.Sc Iyear	9172354118	Pdvp College, Tasgaon
Chougule Nikita Namadev	Ba C 1	7499830853	Pdvp College, Tasgaon
Patil Divya Arjun	Bas3 Botany	9307418183	Pdvp College, Tasgaon
Duryodhan Ravasaheb Jadhav	Bsc	7620861435	Pdvp College, Tasgaon
Wagh Akanksha Sanjay	Bsc - 3	7499959832	Pdvp College, Tasgaon
Gaikwad Prajakta Nagesh	Bsc . Third Year	9307785918	Pdvp College, Tasgaon
Dhanshree Dattatray Mainkar	Bsc 1	7385002654	Pdvp College, Tasgaon
Sumit Yadav	Bsc 1	7397886747	Pdvp College, Tasgaon

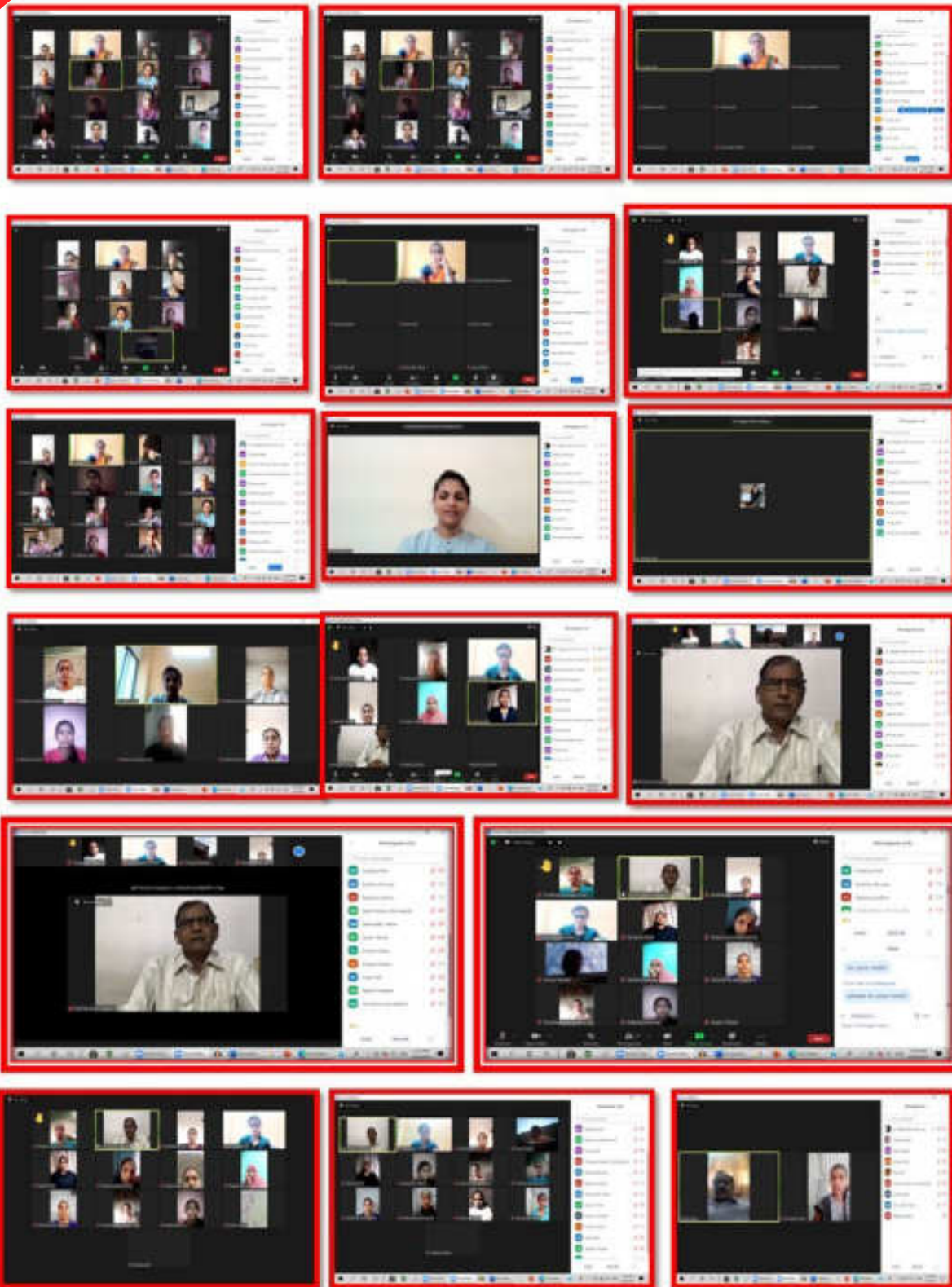
This course is divided into three lessons to ensure that each topic is covered fully and that each student has all the information that they need to learn. Each lesson in this course is followed by a lesson exercise and assignment.

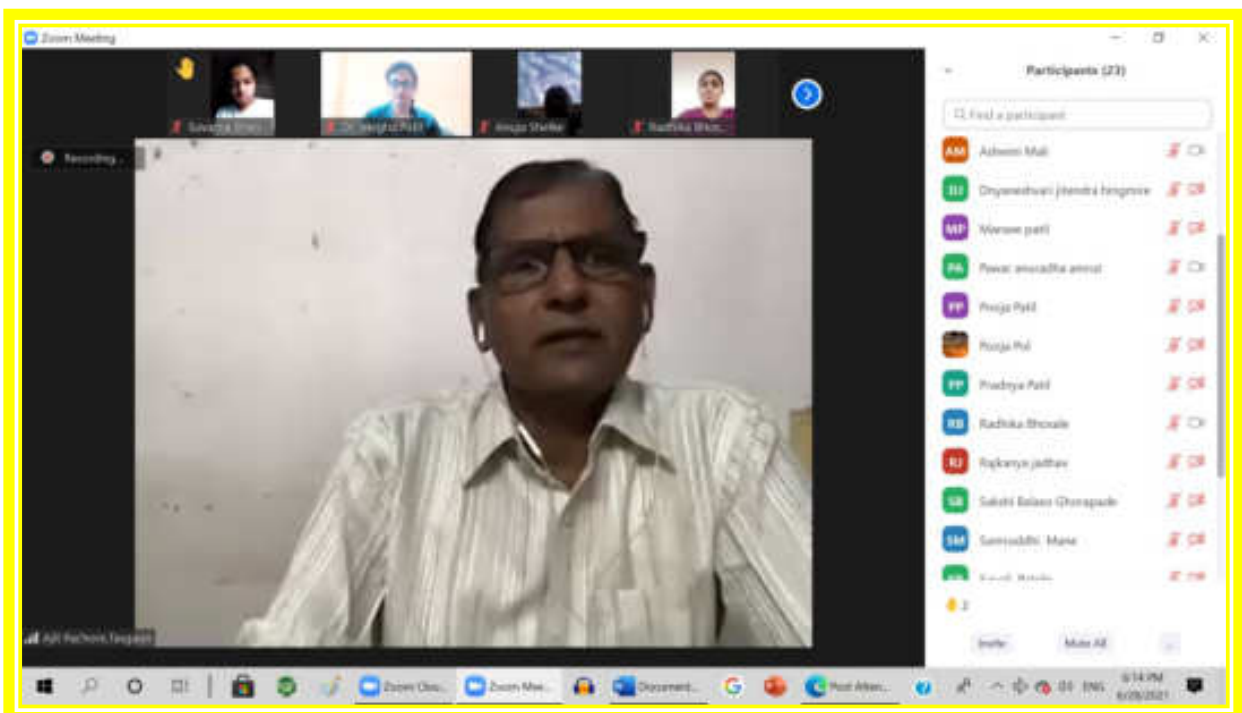
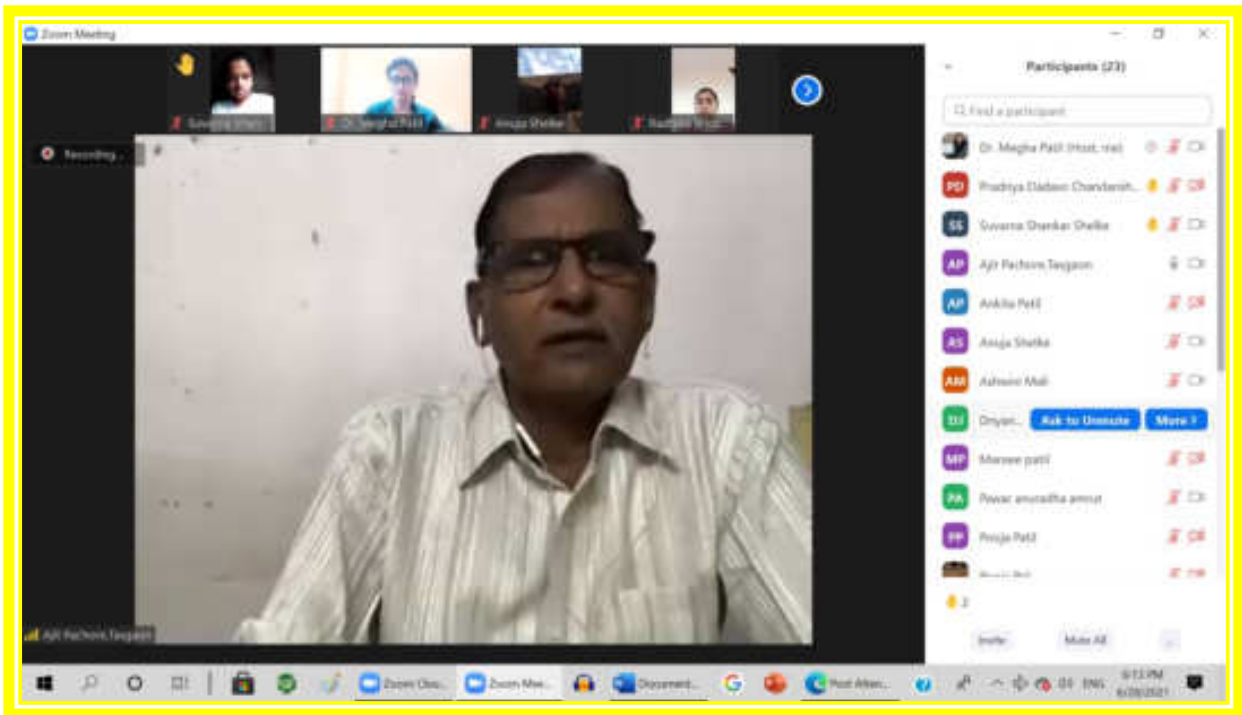
The lesson exercises are mandatory for all students and will determine whether or not a student has satisfactorily completed the course. Lesson assignments are optional. These was not be graded and only given as a tool that provides for a deeper learning and understanding of the material presented.

Use of Zoom platform

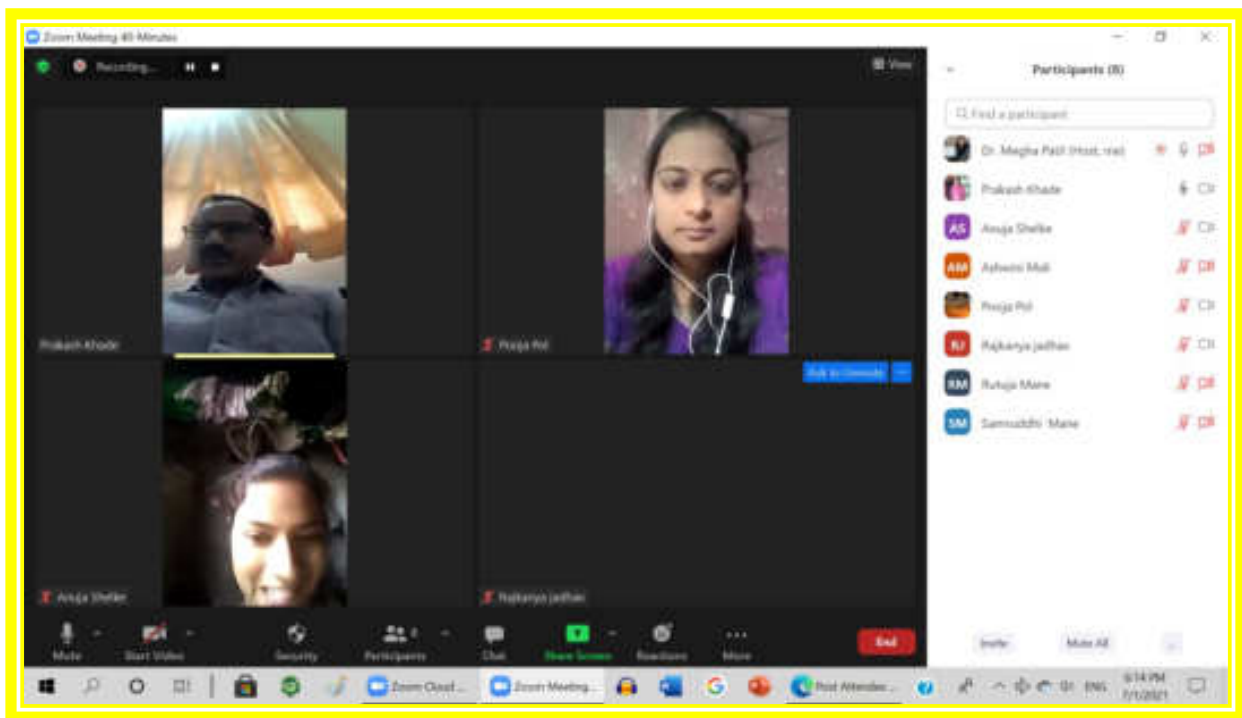


We also invited guest Mr. Ajit Pachore and Mr. Prakash Khade from Department of English from home college during course for delivering lecture on group discussion and interview skill.





Mr. Ajit Pachore delivering lecture during course



Mr. Prakash khade delivering lecture during course

We also provided videos on You tube to get more knowledge.

Use of You tube to share knowledge

❖ Communication skill

Part 1 <https://youtu.be/bg7a1nYXLYs>

Part 2 <https://youtu.be/av2LKvy5iGs>

❖ Improve reading and writing skill

<https://youtu.be/Ov8EoT7EhLM>

❖ Introduce yourself:

<https://youtu.be/TwZ7LgrPwR0>

Personality development

<https://youtu.be/v2euEwGnas8>

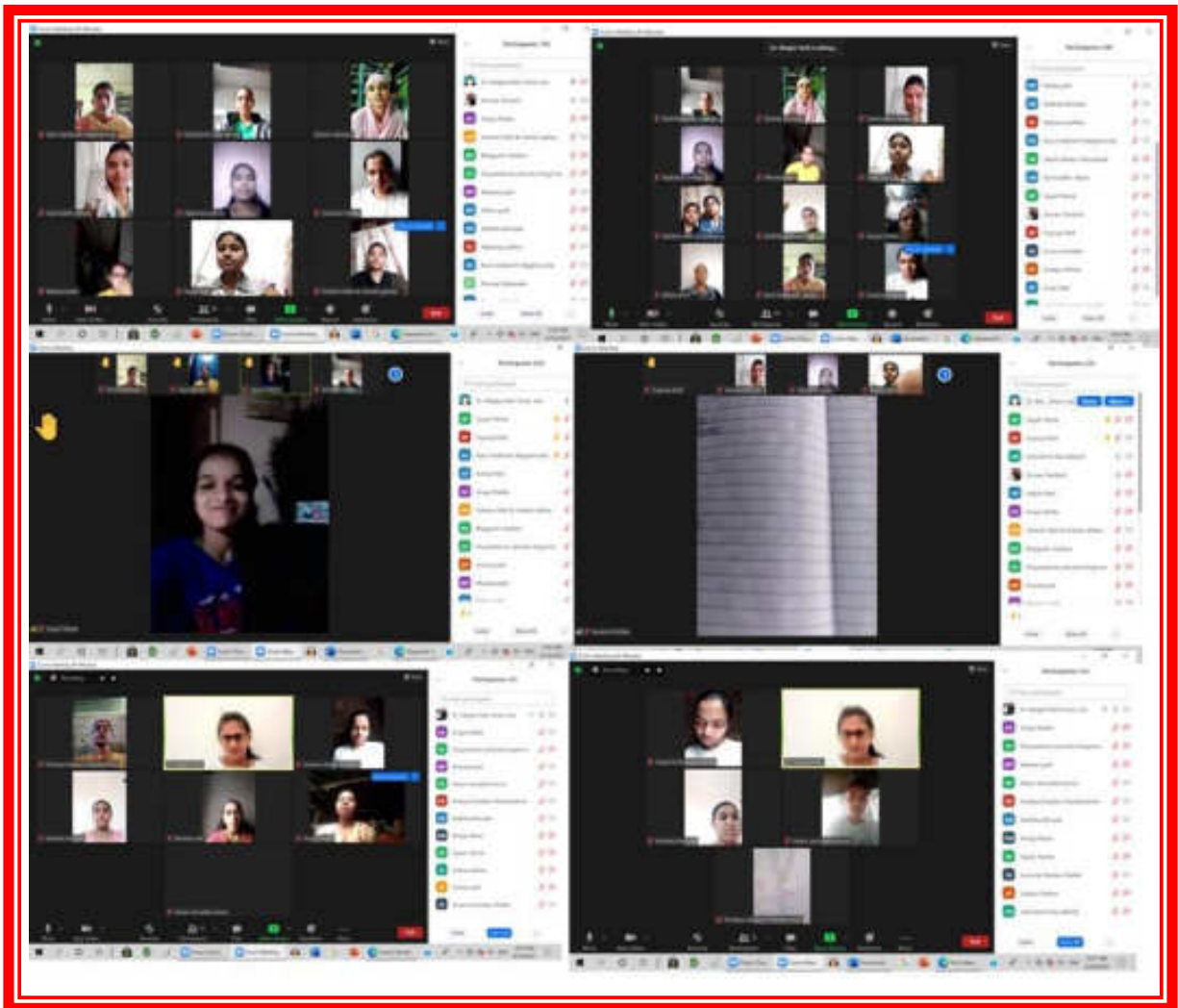
❖ Job interview

https://youtu.be/ZZOvzL9_MjA

❖ Advanced English words

<https://youtu.be/qY0X95EBJDw>

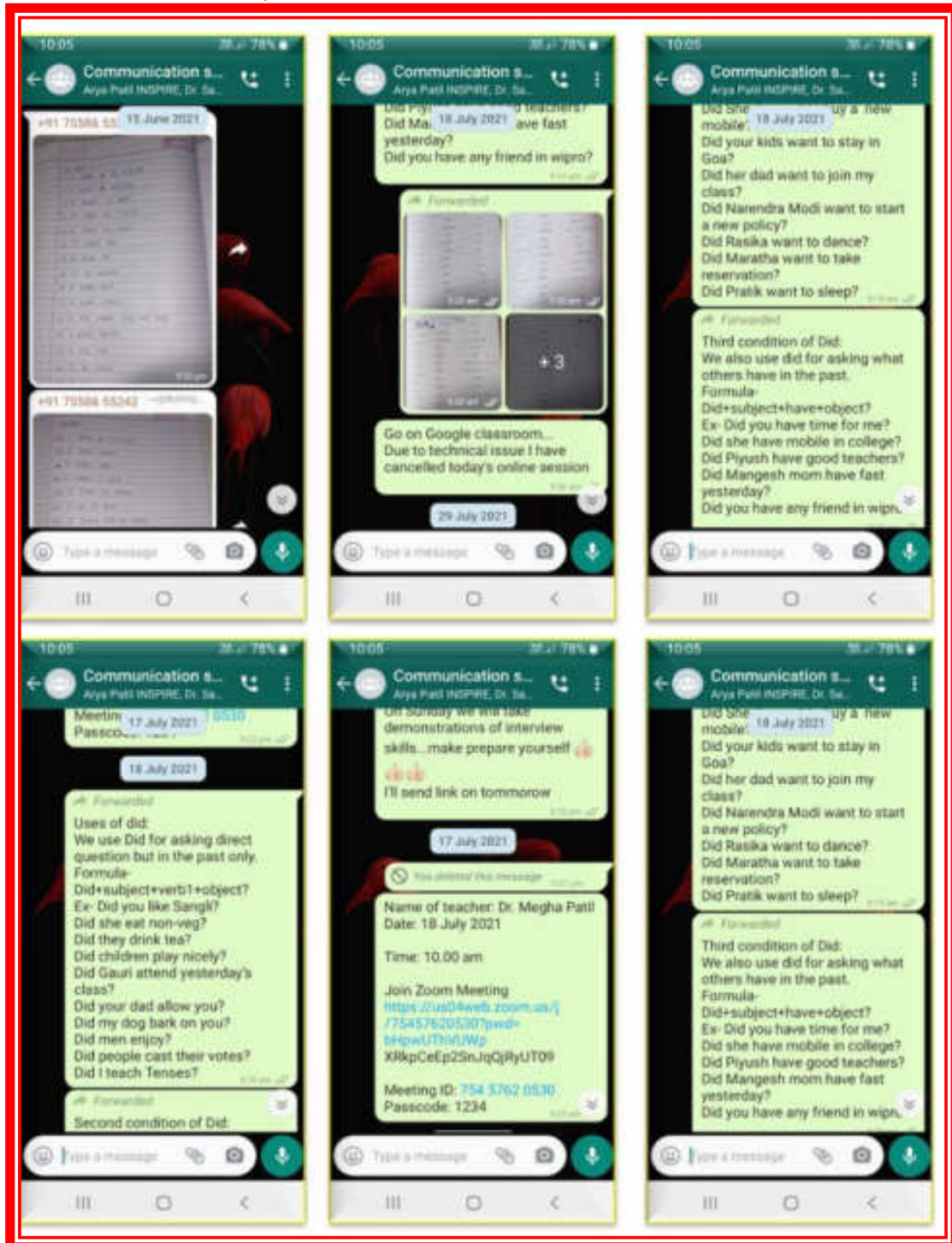
<https://youtu.be/1INwAOp0cfk>

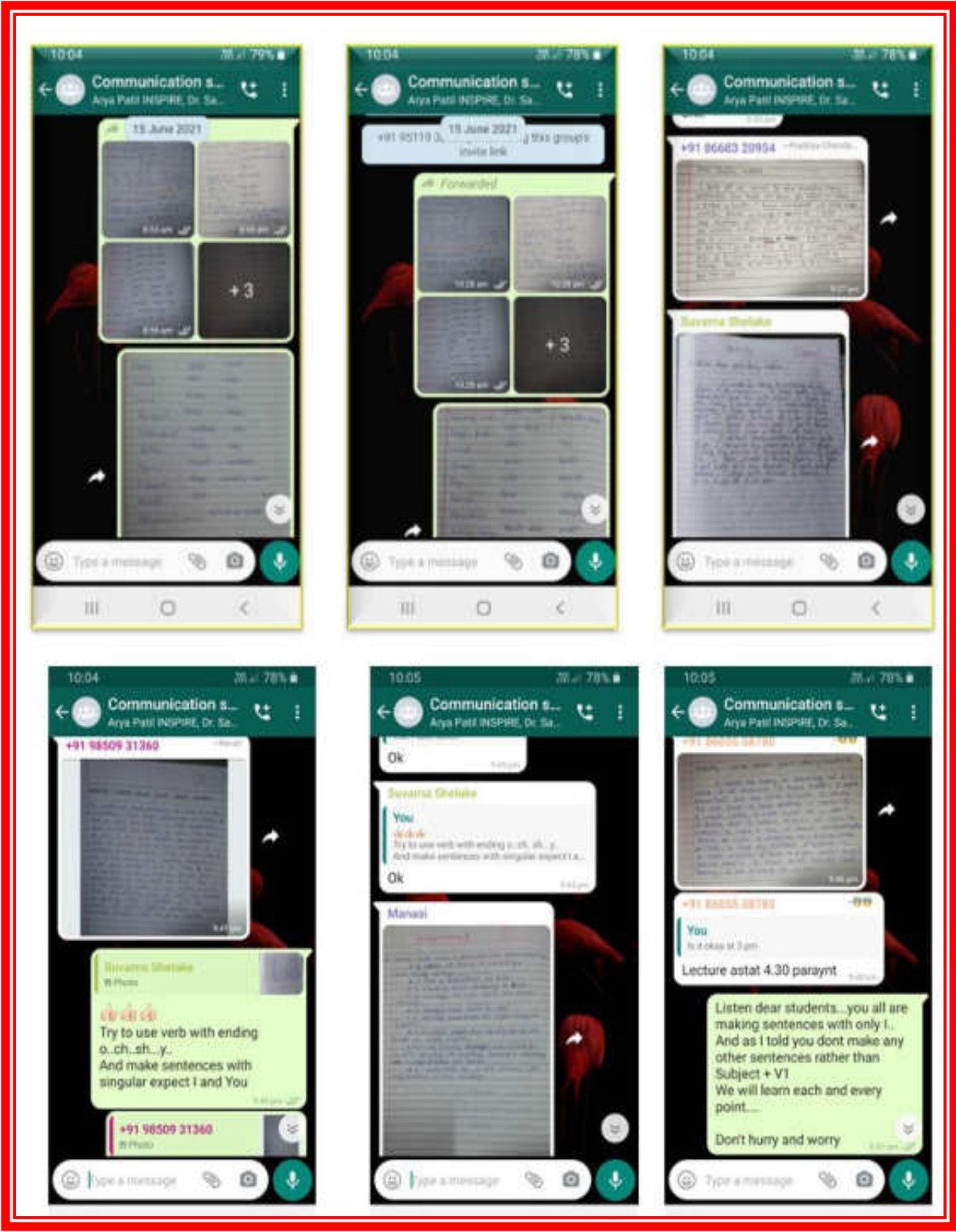


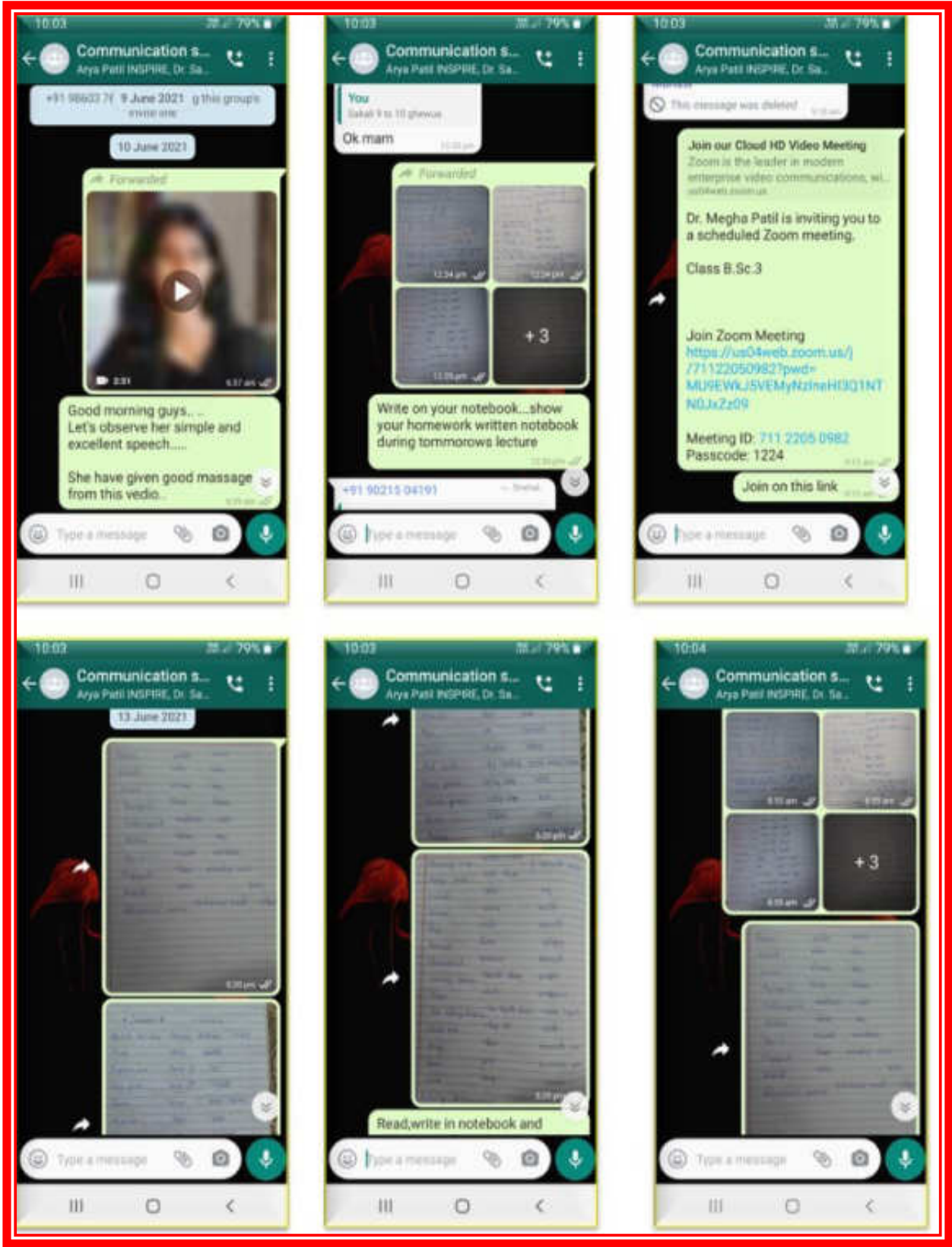
Discussion on you tube videos on zoom platform

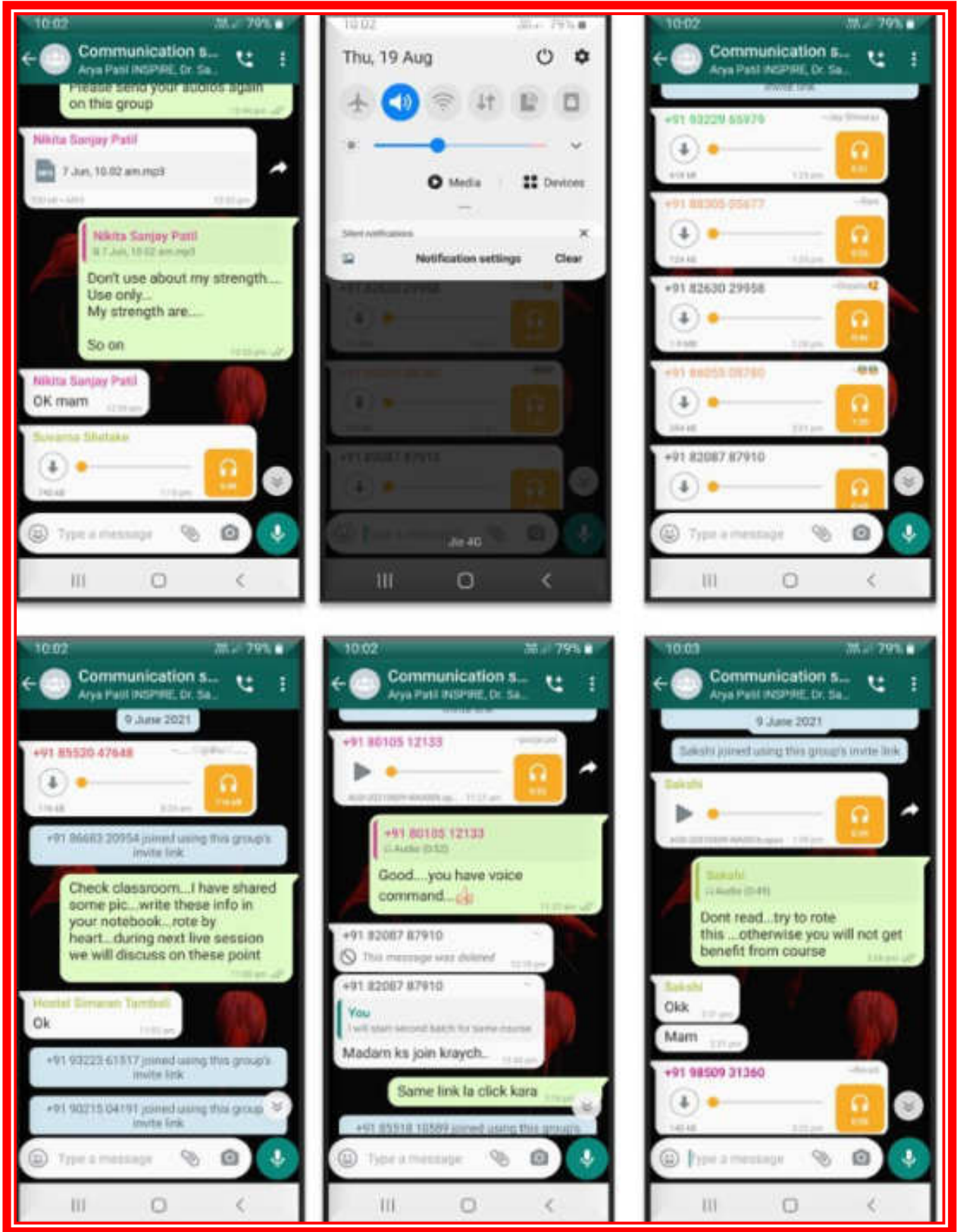
Some useful information also provided using What's App Group. The underlying purpose of WhatsApp is to facilitate communication, and at its most basic level, education is nothing but communication. WhatsApp can provide a channel through which teachers can achieve faster and more seamless communication with their students. It can also increase the level of communication between students and create another venue for learning

Use of What's App Group

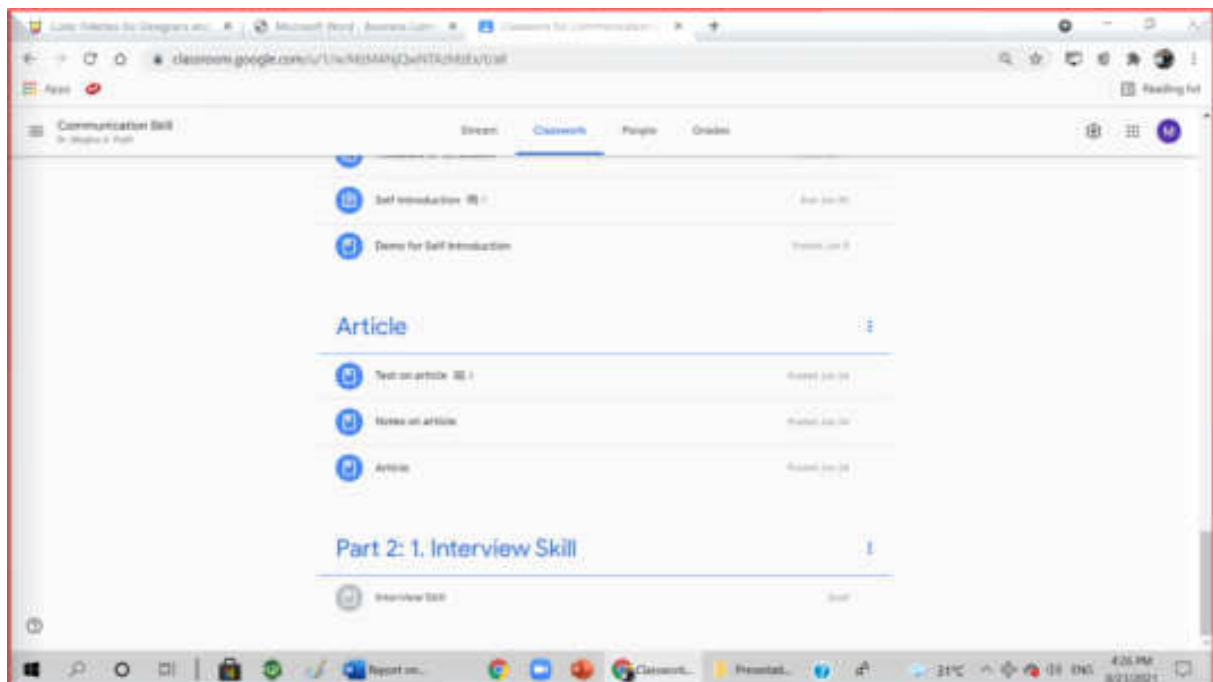
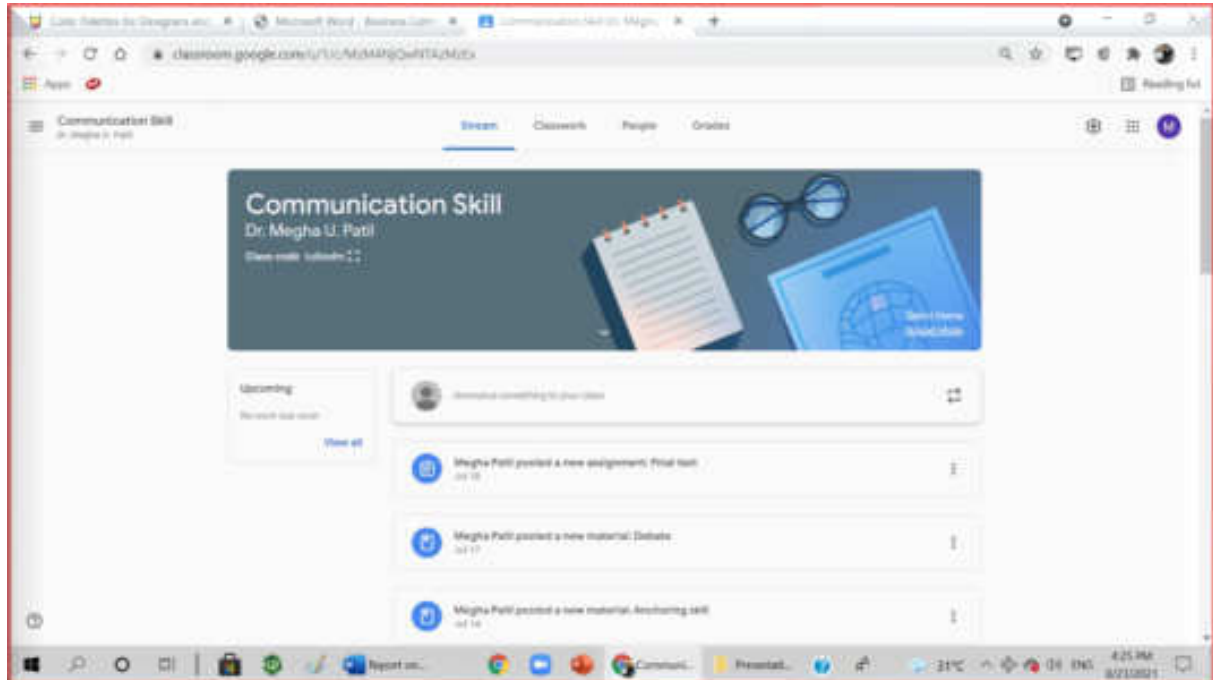


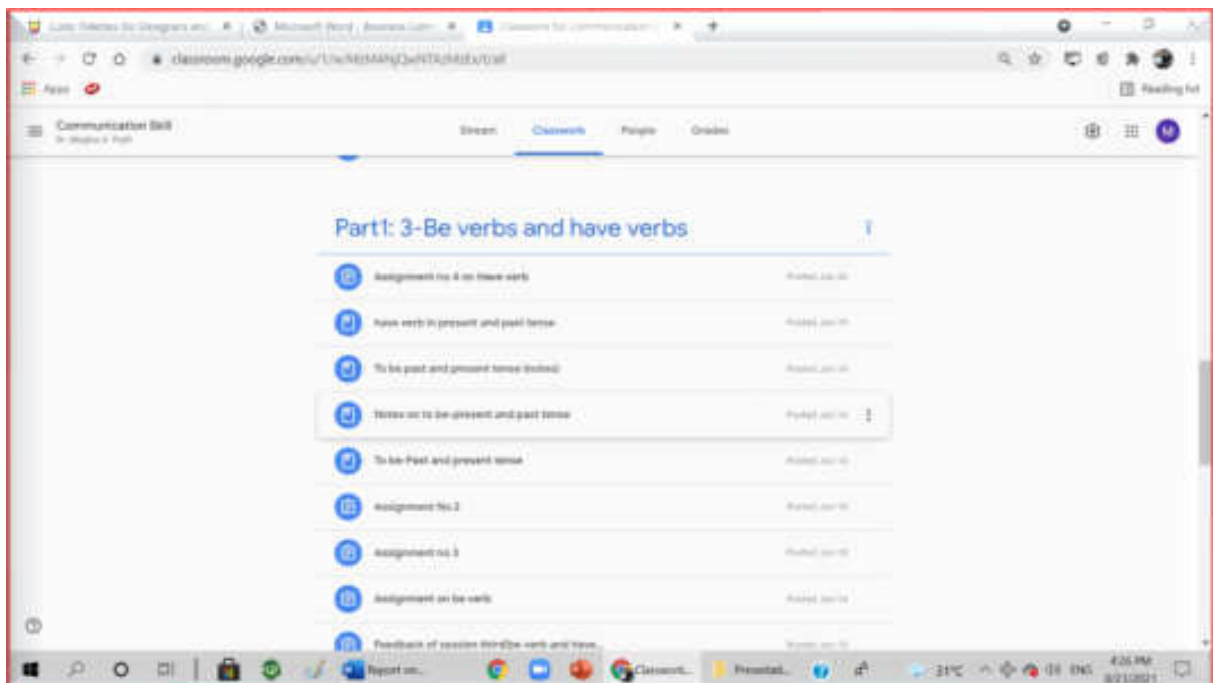
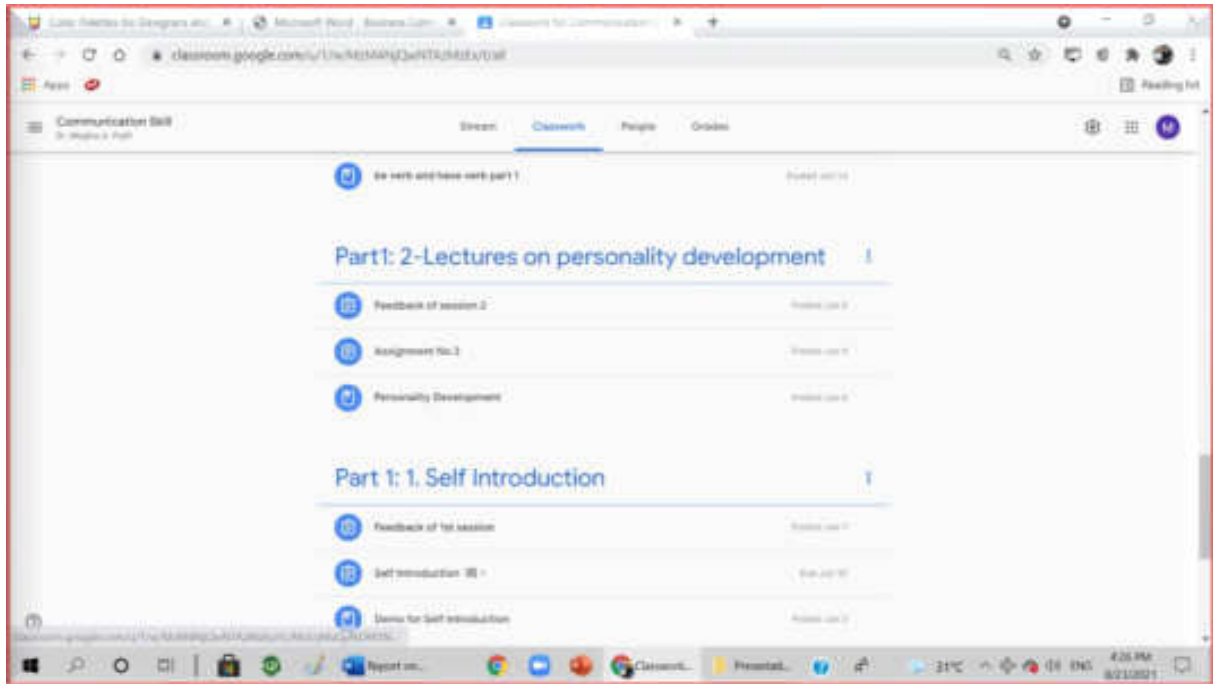


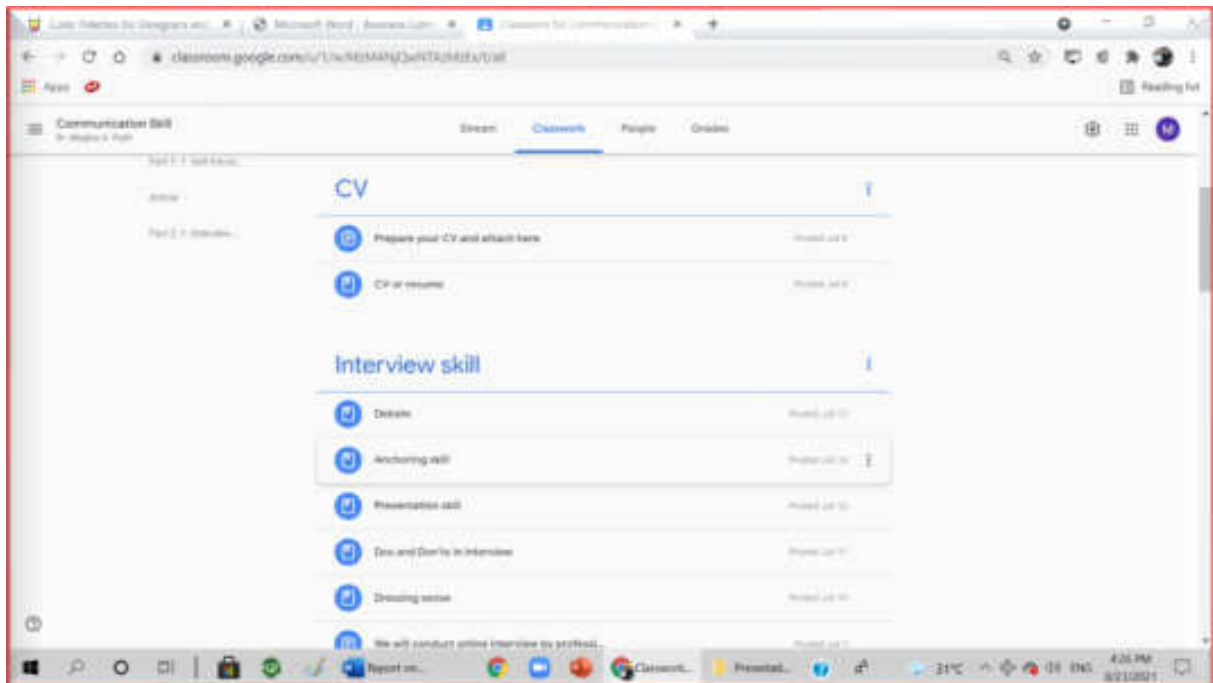
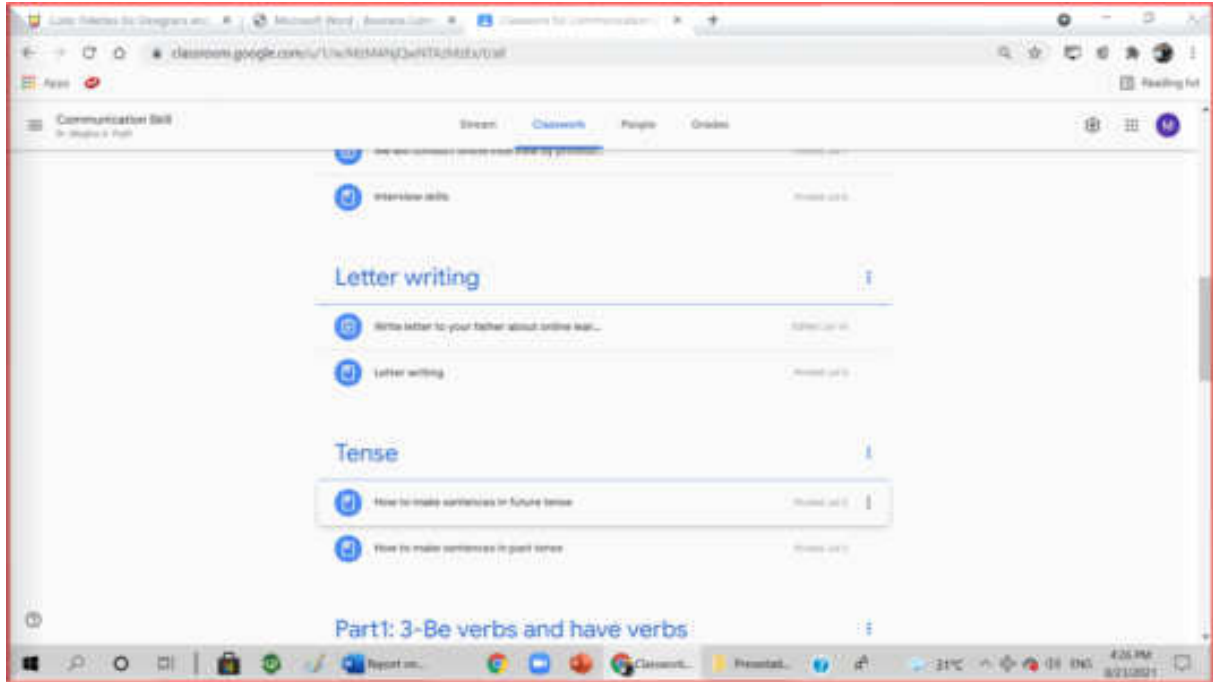


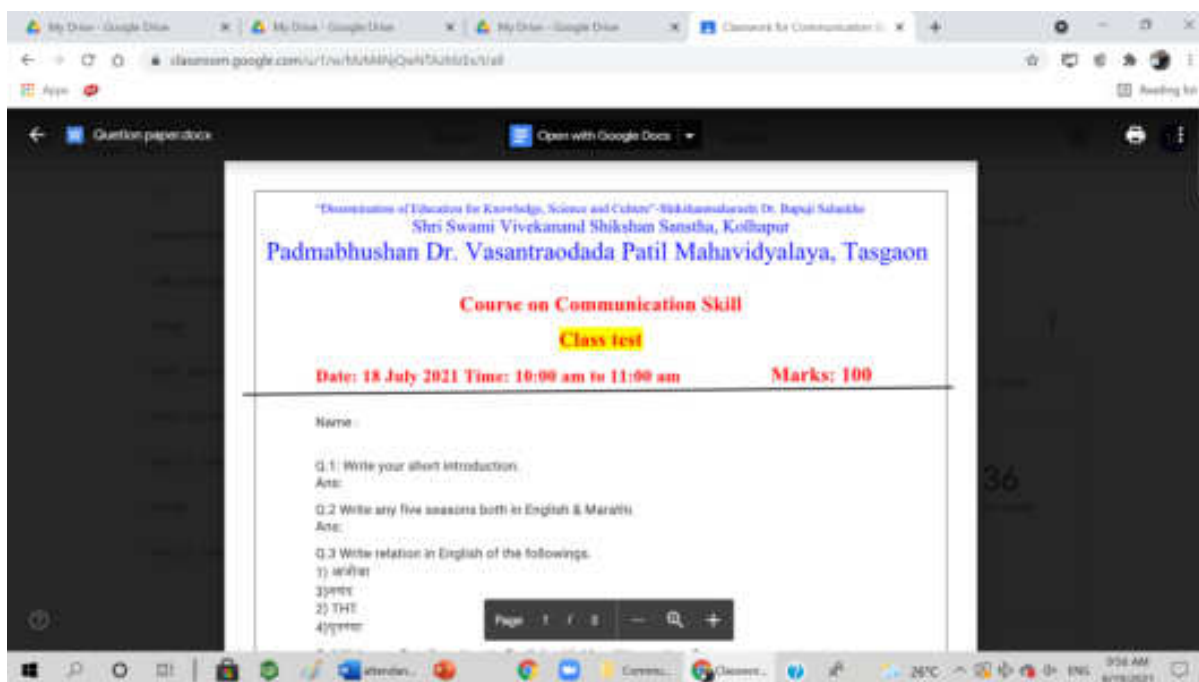
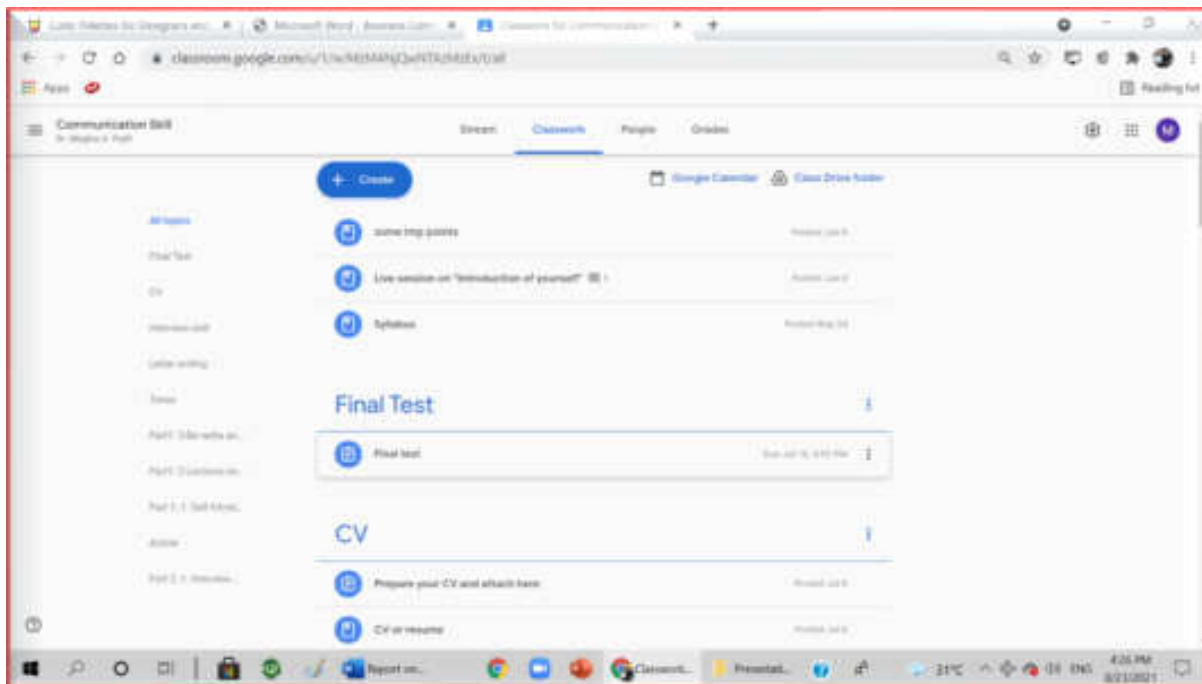


Use of Google Classroom

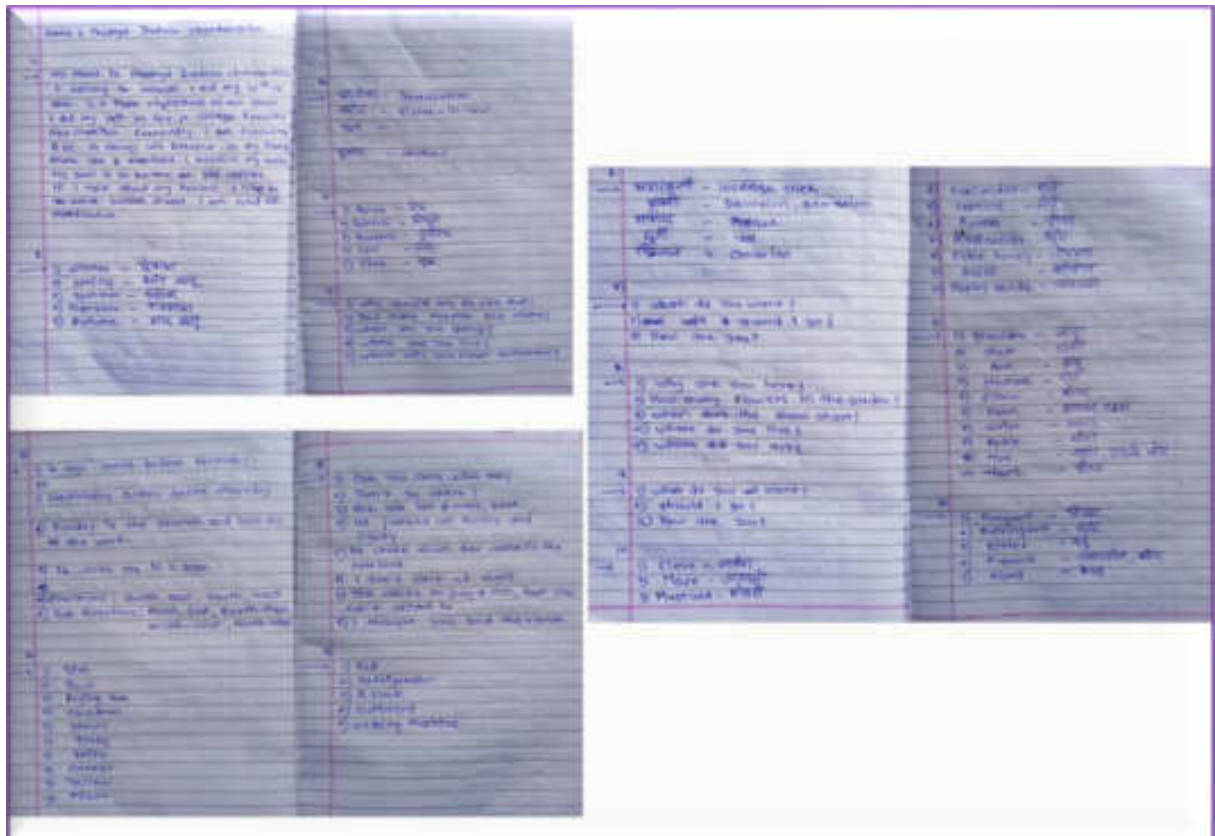
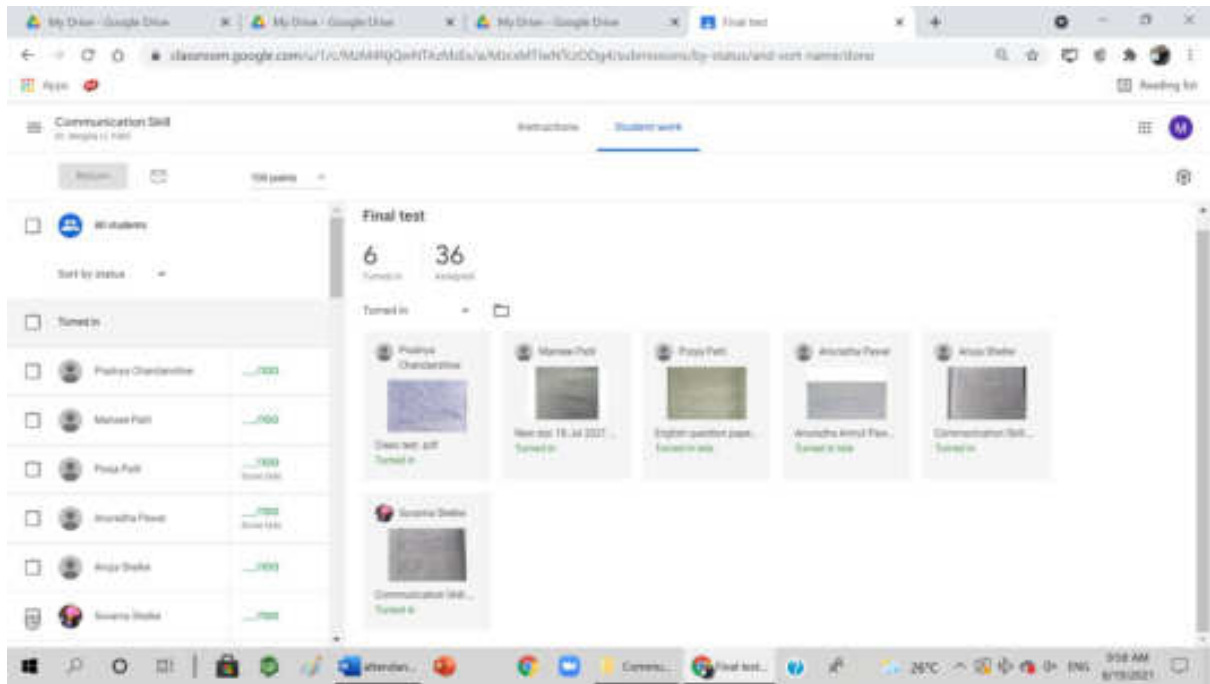








COURSE ON COMMUNICATION SKILLS



At the end of course online test was conducted to analyse their dept of knowledge, we have checked their confidence and communication skill on zoom meeting.

We have provided certificate to students those who have passed all tests with min. 40 % and completed course regularly and sincerely.

Feedback of course from students

It's no secret that feedback is an important component of effective learning. Feedback improves learner confidence, motivation to learn and ultimately, a learner's attainment. Effective feedback, both positive and negative, is very helpful. Feedback is valuable information that will be used to make important decisions.

Name	Department	How did you feel the course/teacher rate your overall in that aspects of the session were that useful or valuable and how will you implement them in your life?
Mare Honeka Pogol	Statistics	Excellent
Ran Isahbathi Inghamane	Chemistry	Very Good
Marese jansurung pati	Bsc 1	Very Good
Dhyaneshwar Akhota High Science	Very Good	Excellent
Dhyaneshwar Akhota High Science	Excellent	Good
Dhyaneshwar Akhota High Science	Excellent	Good
Pall marase jansurung	Bsc 1	Very Good
Pall marase h	Bsc 1	Very Good
Marese jansurung pati	Bsc 1	Very Good
Pabun Ananya Jemman	B.Sc. 1 year	Excellent
Shree Anup Anil	Bsc 2	Very Good
Shree Anup Anil	Bsc 2	Very Good
Pranava Dadas Chaudhari P. Y. B. Sc.	Excellent	Excellent
Suvarna Shikhar Shinde	Bsc 1	Very Good
Sakun Dadas Choudhary	Bsc 1	Good
Radhika Tarang Wode	Science Bsc 1	Excellent
Wika Sanjay Patil	Science	Excellent
Radhika Tarang Wode	Science Bsc 1	Excellent
Tarun Dhanraj Aj	Statistics	Excellent
Falgun Manojkumar Dattaraj	Statistics	Very Good
Saemrath Sanjay Kote	Organic chemistry	Very Good

Sample Certificate



1/25/2021

**Report on Subjective
course conducted on
MOOC Course on
ELECTROCHEMISTRY
conducted on
MOOC on**

Submitted by,

Dr. Megha U. Patil
ASSISTANT PROFESSOR IN CHEMISTRY

“Dissemination of Education through Knowledge, Science and Culture.”

—Shikshanmaharshi Dr. Bapuji Salunkhe.

Shri. Swami Vivekanand Shikshan Sanstha's

**Padmabhushan Dr. Vasantodada Patil Mahavidyalaya,
Tasgaon, Sangli-416312, Maharashtra
(Affiliated to Shivaji University Kolhapur)**



**DEPARTMENT OF
CHEMISTRY**



Report on Subjective course conducted on MOOC Course on **ELECTROCHEMISTRY** conducted on **MOOC on MyeClass Portal**

Submitted to

INTERNAL QUALITY ASSURANCE CELL

Organising Department Department of Chemistry

Duration

23rd October to 30th November 2020

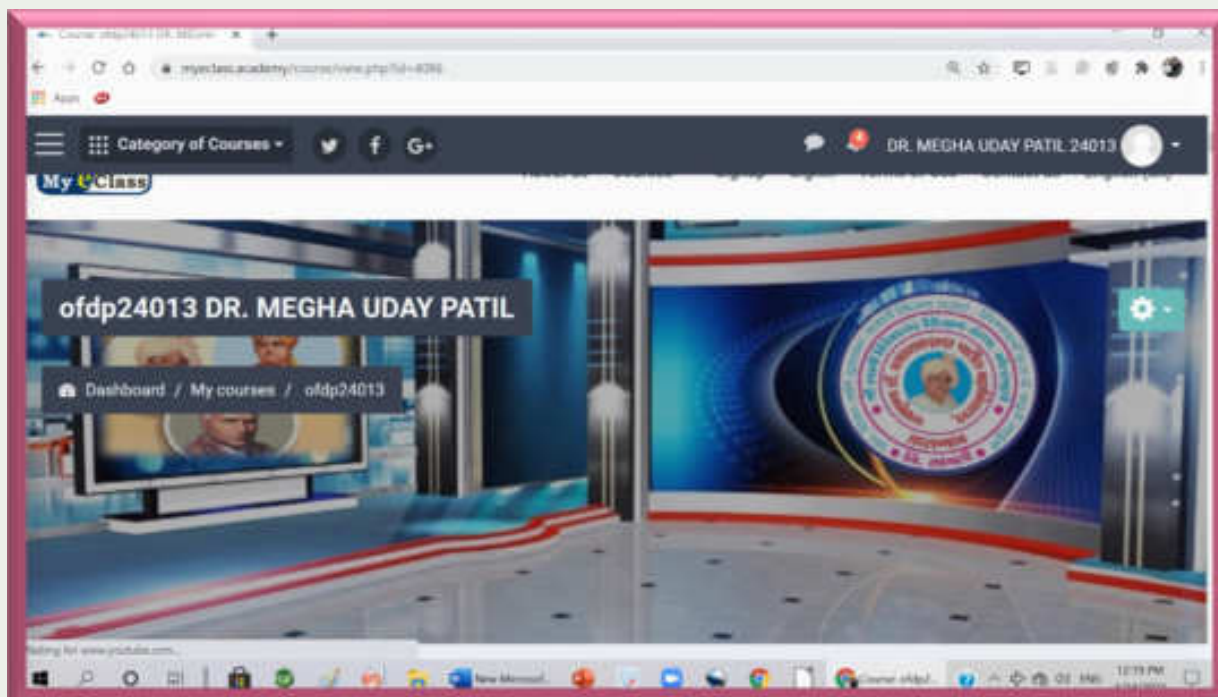
Platform

MyeClass Portal

Participants

- **Total: 130**
- **Male: 48**
- **Female: 82**

About Course:



We have been keen interested to conduct online lecture on Massive Open Online Courses (MOOC) to engage our students in learning process. We are thrilled that GAD-TLC provide us MyeClass portal to conduct such courses on this portal. Which help us to share Study material in terms of Module, Notes, Assessment and Assignment.

MOOC has become an effective solution to conduct online lectures at a low cost.

We conduct this course by providing Study material in terms of self-prepared modules and notes regarding

module. Also we have taken daily assessment and assignments on module of each point in syllabus.

We have covered topic Electrochemistry for Physical Chemistry student of B.Sc. III- Sem- V. In last session we have taken Feedbacks from student. Finally, we conduct Test on complete topic and we issued certificate to those students who quality final test with minimum 70% marks.

List of participants:

First name	Surname	Email address	Phone	Mobile phone	Institution
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			
Dr. Vikramad	Dr. Vikramad	vikramad@univ.ac.in			

REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC

courseid_4086_participants.pdf - Adobe Acrobat Reader DC

Home Tools iio-ia-puz-15-03... courseid_4086_part... Try Acrobat Pro DC Sign In

First name	Surname	Email address	Phone	Mobile phone	Institution
Pradjito	Paul	pradjito2019@gmail.com			
Alexandra	Paul	alexandra21199@gmail.com			
Baniy	Nurwahid	baniy2019@gmail.com			
Rafaja	Paul	rafaja2019@gmail.com			
Melita	Reza	melita2019@gmail.com			
Susanto	Mari	susanto2019@gmail.com			
Arif	Paul	arif2019@gmail.com			
Priscilla	Melita	priscilla2019@gmail.com			
Nella	Sarrah	nella2019@gmail.com			
Uyuni	Jadler	uyuni2019@gmail.com			
Prati	Jadler	prati2019@gmail.com			
Sari	Kherka	sari2019@gmail.com			
Magdalena	Maria	magdalena2019@gmail.com			
Alexandra	David	alexandra2019@gmail.com			
Yoni	Gerard	yoniphon2019@gmail.com			
Magdalena	Jadler	magdalena2019@gmail.com			
Anggi	Mari	anggi2019@gmail.com			
Maryi	Chares	maryi2019@gmail.com			
Ramadhan	Paul	ramadhan2019@gmail.com			

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First name	Surname	Email address	Phone	Mobile phone	Institution
Tejawan	Agus	tejawan2019@gmail.com			
Seputi	Mari	seputi2019@gmail.com			
Aji	Mari	aji2019@gmail.com			
Andriyana	Jadler	andriyana2019@gmail.com			
Karya	Shah	karya2019@gmail.com			
Pradana	Paul	pradana2019@gmail.com			
Ayaka	Shah	ayaka2019@gmail.com			
Ersal	Mari	ersal2019@gmail.com			
Melita	Mari	melita2019@gmail.com			
Ani	Shah	ani2019@gmail.com			
Ersal	Paul	ersal2019@gmail.com			
Prati	Jadler	prati2019@gmail.com			
Rizka	Mari	rizka2019@gmail.com			
Ani	Rahma	ani2019@gmail.com			
Suryani	Chares	suryani2019@gmail.com			
Seputi	Jadler	seputi2019@gmail.com			

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REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC

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First name	Surname	Email address	Phone	Mobile phone	Institution
Parvati	Mal	parvati171@gmail.com			
Saptha	Man	sapthaman121@gmail.com			
Arjun	Jaffer	jafferarjun74@gmail.com			
Vijay	Patil	vijay1269@gmail.com			
Madhuri	Jaffer	Madhuri13@gmail.com			
Ashika	Shank	Ashikashank171@gmail.com			
Sanja	Shree	sanjashree21@gmail.com			
Sanika	Channa	Sanikaanna212@gmail.com			
Shakal	Patil	shakalpatil@gmail.com			
Bonika	Mal	bonika171@gmail.com			
Ajay	Zachary	ajayzachary199@gmail.com			
Surbat	Madhur	surbatmadhur2@gmail.com			
Krunal	Mal	krunal119@gmail.com			
Kalyani	Shank	kalyanishank147@gmail.com			
Manikata	Wagh	manikatawagh212@gmail.com			
Pratiksha	Shendge	pratikshashendge@gmail.com			
Madhur	Man	manmadhur131@gmail.com			

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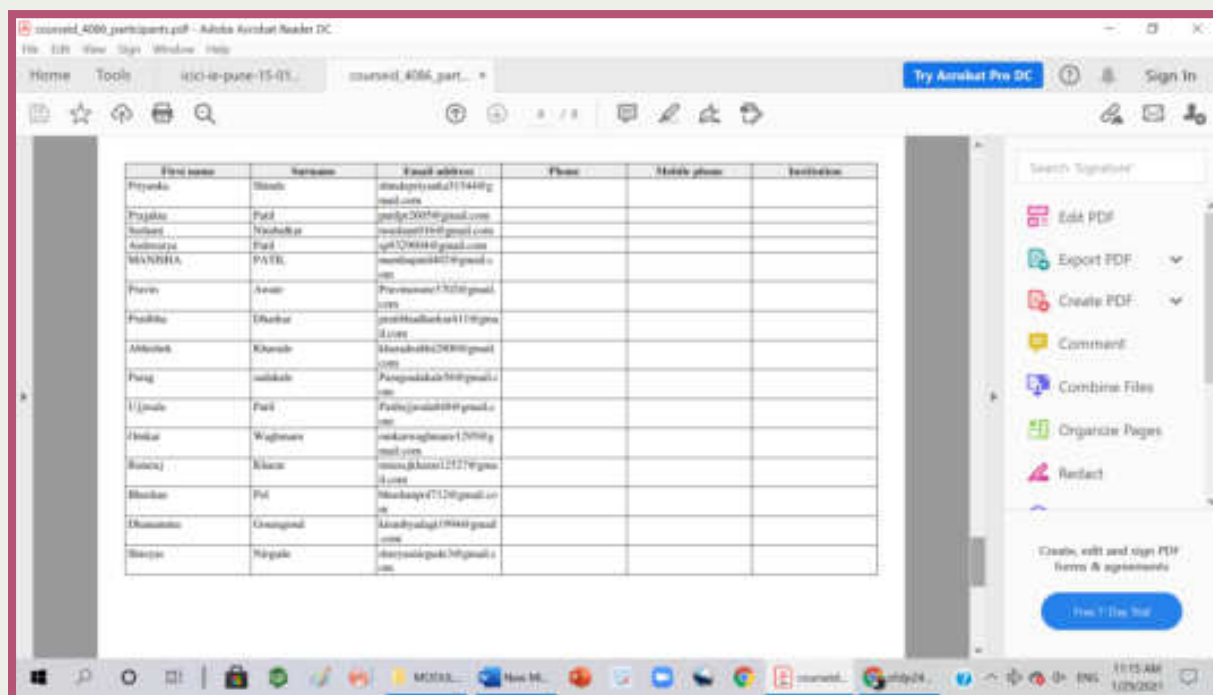
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First name	Surname	Email address	Phone	Mobile phone	Institution
Anika	Shendge	anikashendge212@gmail.com			
Rishabh	Jaffer	rishabhjaffer23@gmail.com			
Saurabh	Patil	saurabhpatil234@gmail.com			
Rupali	Patil	rupalipatil24@gmail.com			
Ashika	Shank	ashikashank171@gmail.com			
Rishi	Shank	rishishank121@gmail.com			
Anshu	Ashu	anshuanshu197@gmail.com			
Shree	Shivale	shreesivale111@gmail.com			
Shubham	Rishabh	shubhamrishabh131@gmail.com			
Shweta	Patil	shwetapatil19@gmail.com			
Vijay	Patil	vijaypatil242@gmail.com			
Surbat	Surbat	surbatsurbat199@gmail.com			
Falguni	Kumbhar	falgunikumbhar171@gmail.com			
Vandana	Patil	vandana1212@gmail.com			
Vandana	Patil	vandana123@gmail.com			
Ami	Patil	ami123@gmail.com			
Falguni	Patil	falgunipatil234@gmail.com			
Shreyas	Man	shreyas121@gmail.com			
Prachi	Channa	prachichanna171@gmail.com			

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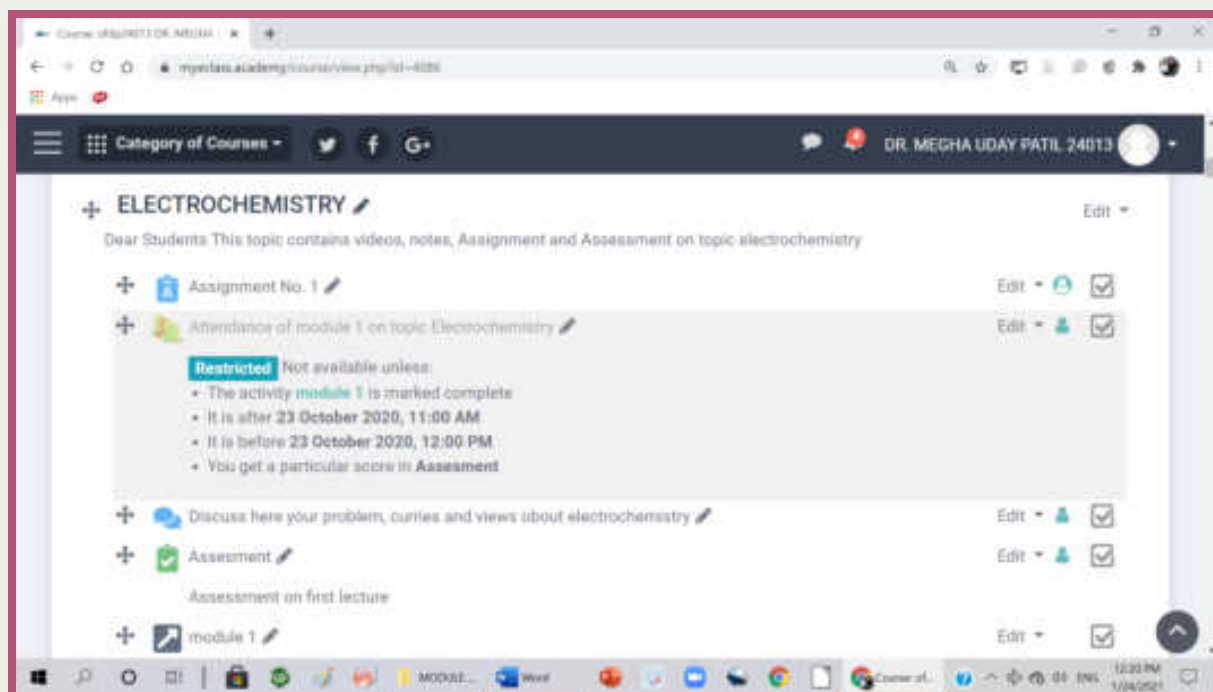
REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC



First name	Surname	Email address	Phone	Mobile phone	Institution
Prayada	Shank	shankprayada11144@gmail.com			
Prayada	Patil	prayada2004@gmail.com			
Sanjay	Nayak	sanjay1111@gmail.com			
Anshu	Patil	anshu1111@gmail.com			
MANISHA	PATEL	manishapatil111@gmail.com			
Harin	Arav	harin1111@gmail.com			
Pratik	Dhoke	pratikdhoke111@gmail.com			
Ashish	Kharab	ashishkharab111@gmail.com			
Pang	Patil	pangpatil111@gmail.com			
Ujjwal	Patil	ujjwalpatil111@gmail.com			
Harsh	Waghmare	harshwaghmare111@gmail.com			
Ravi	Arora	raviarora1111@gmail.com			
Shantanu	Patil	shantanupatil111@gmail.com			
Dhruv	Chavhan	dhruvchavhan111@gmail.com			
Sanjay	Patil	sanjaypatil111@gmail.com			

Insights of Course:

Day 1:



Category of Courses

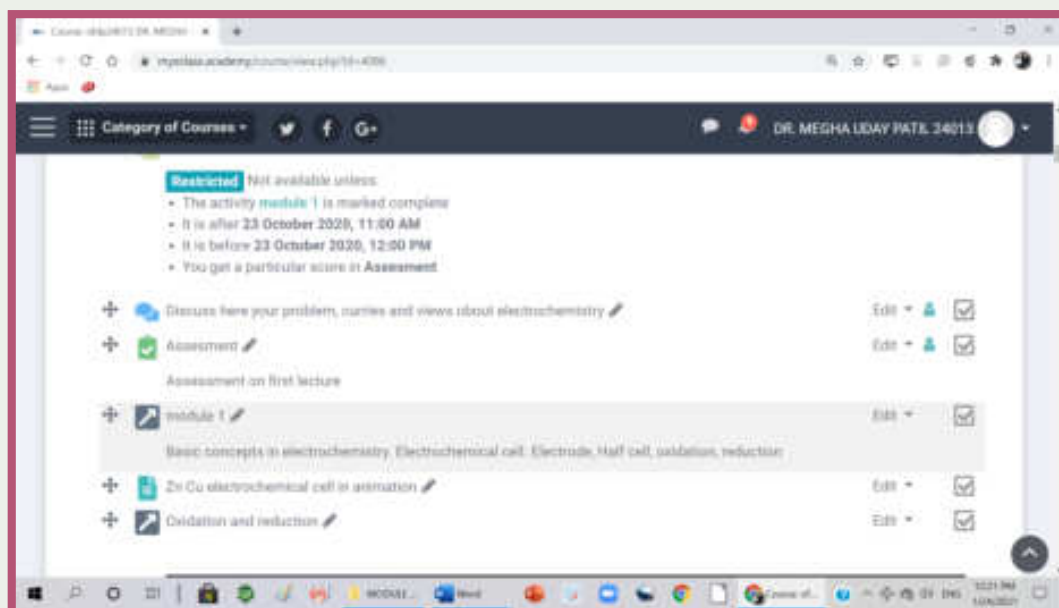
DR. MEGHA UDAY PATIL 24013

ELECTROCHEMISTRY

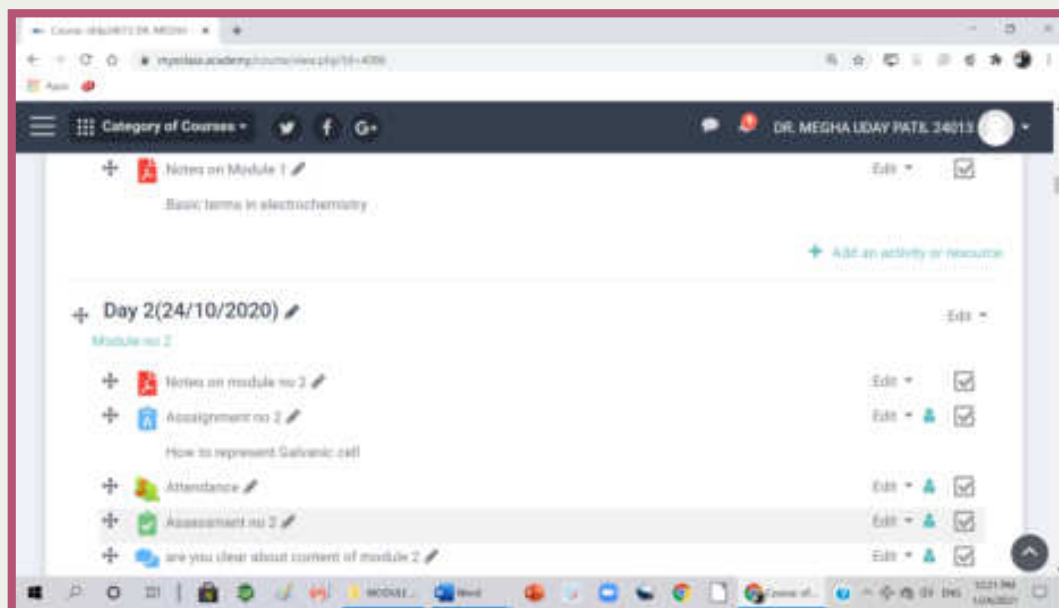
Dear Students This topic contains videos, notes, Assignment and Assessment on topic electrochemistry

- Assignment No. 1
- Attendance of module 1 on topic Electrochemistry
- Restricted** Not available unless:
 - The activity **module 1** is marked complete
 - It is after 23 October 2020, 11:00 AM
 - It is before 23 October 2020, 12:00 PM
 - You get a particular score in **Assessment**
- Discuss here your problem, queries and views about electrochemistry
- Assessment
- Assessment on first lecture
- module 1

REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC



Day 2:

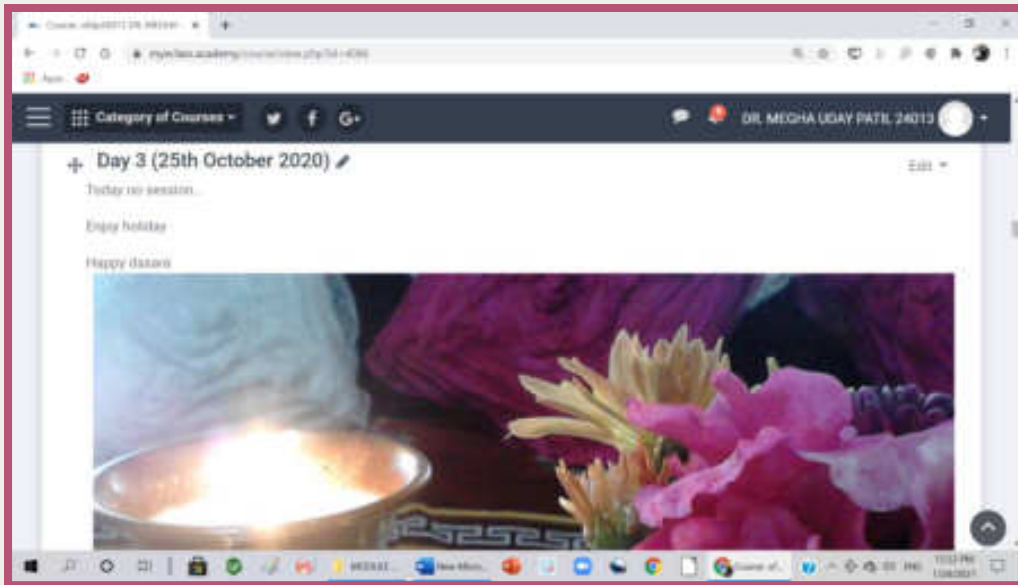


REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC

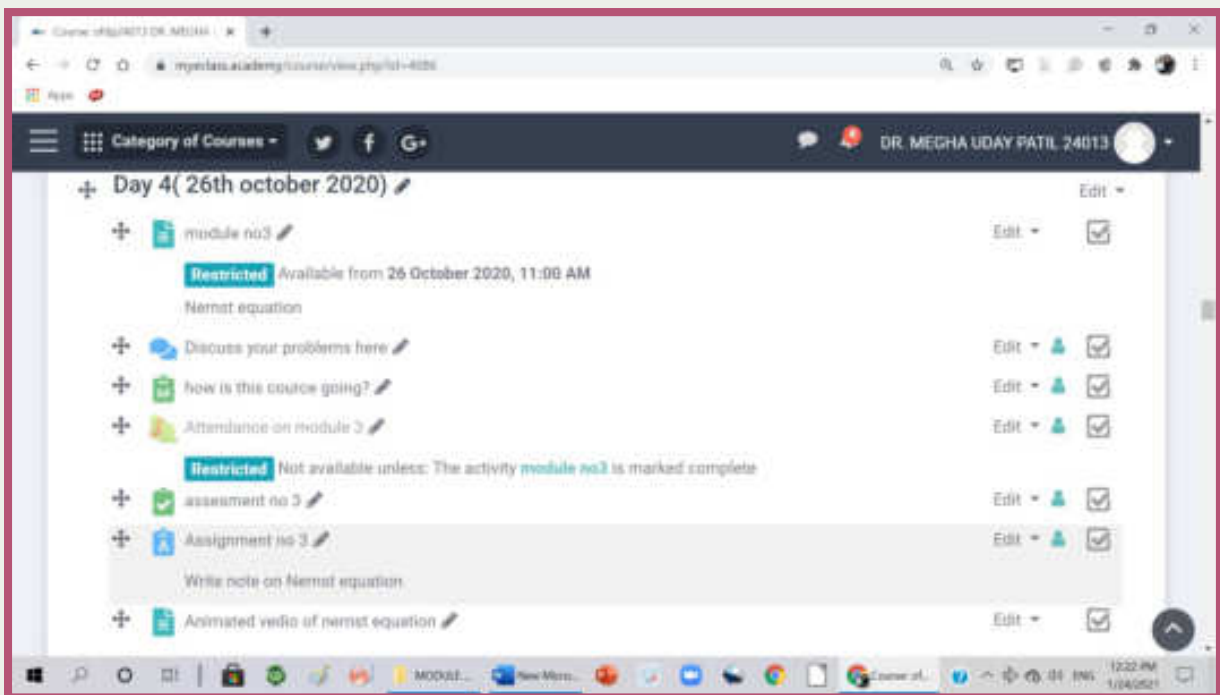
The screenshot shows a web browser window displaying a MOOC page. The page title is "Types of an electrochemical cell, salt bridge, representation of galvanic cell". The video player shows a diagram of a galvanic cell. The cell consists of two half-cells connected by a salt bridge. The left half-cell contains a Zn anode in a ZnSO₄ solution. The right half-cell contains a Cu cathode in a CuSO₄ solution. A voltmeter is connected between the two electrodes, showing a reading of 2.10 V. The salt bridge is labeled "salt bridge (NaCl)". The diagram also shows the flow of electrons from the Zn anode to the Cu cathode through an external circuit.

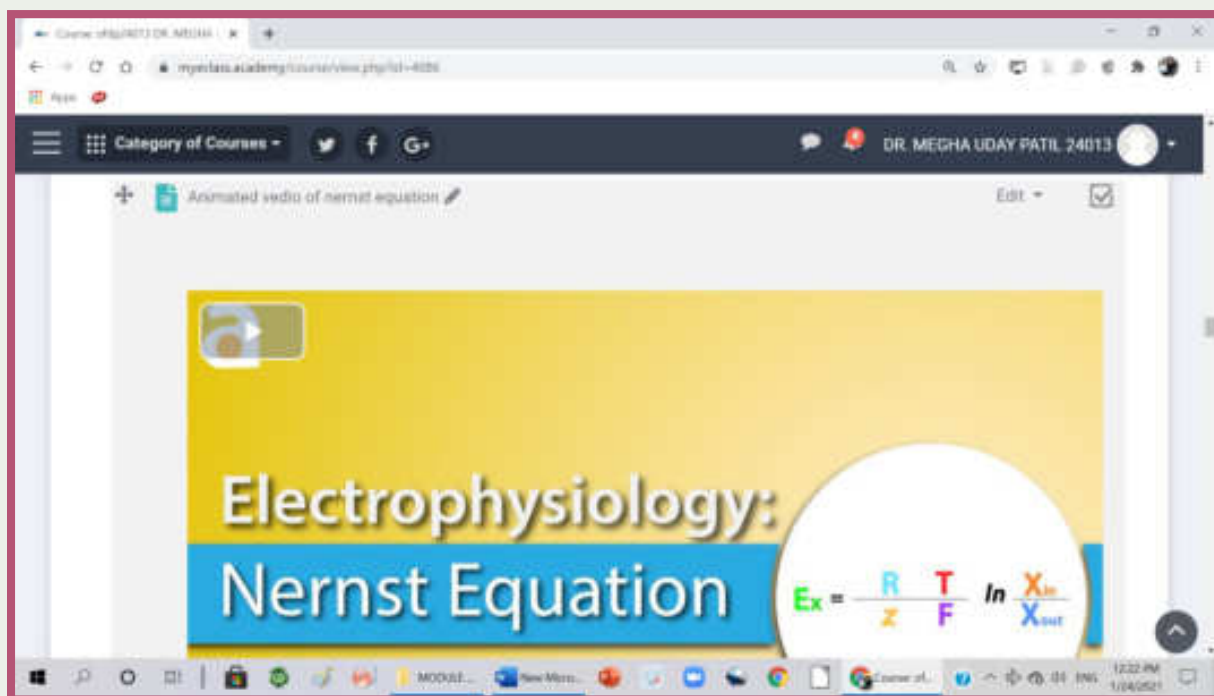
The screenshot shows a web browser window displaying a MOOC page. The page title is "types of electrochemical cell". The video player shows a blackboard with handwritten text. The text reads: "There are two types of electrochemical cells". Below this, there is a list of two types: "1. ELECTROLYTIC CELLS" and "2. GALVANIC CELLS".

Day 3:

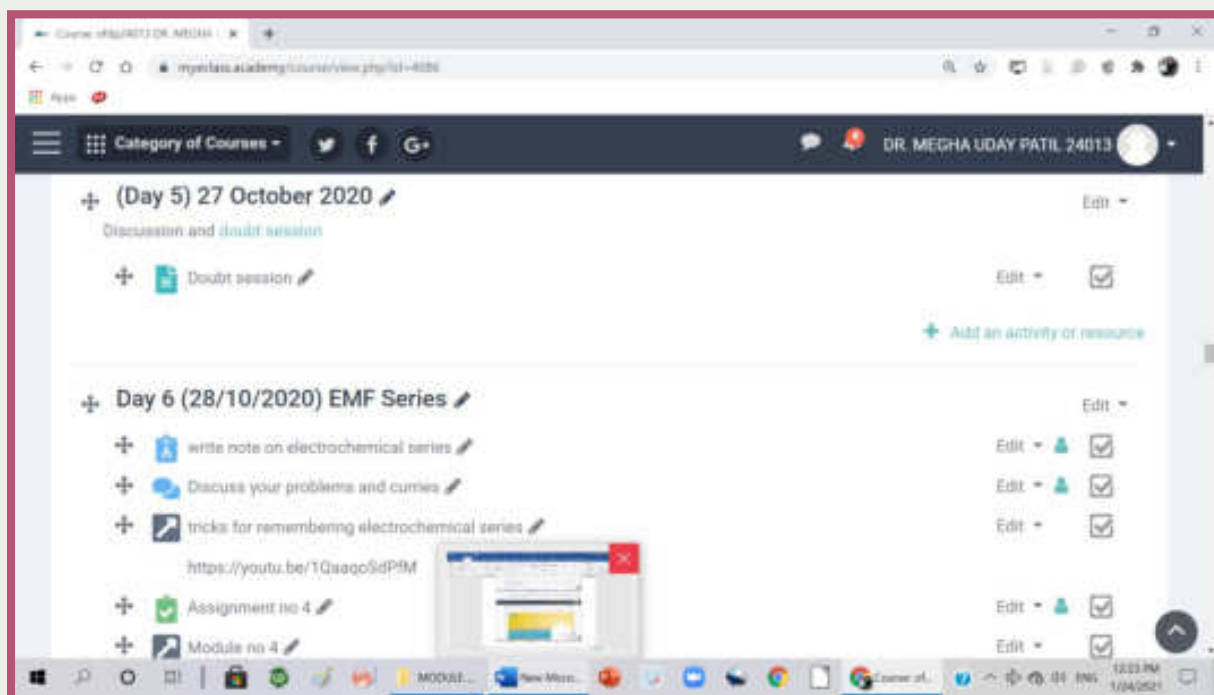


Day 4:





Day 5:



Day 6:

The screenshot shows a web browser window displaying a MyeClass course page. The URL is <https://myclass.academy/course/view.php?id=4026>. The user is logged in as DR. MEGHA UDAY PATIL 24013. The course content for Day 6 (28/10/2020) EMF Series includes:

- write note on electrochemical series
- Discuss your problems and queries
- tricks for remembering electrochemical series (with a YouTube link: <https://youtu.be/1Qaago5tPIM>)
- Assignment no 4
- Module no 4

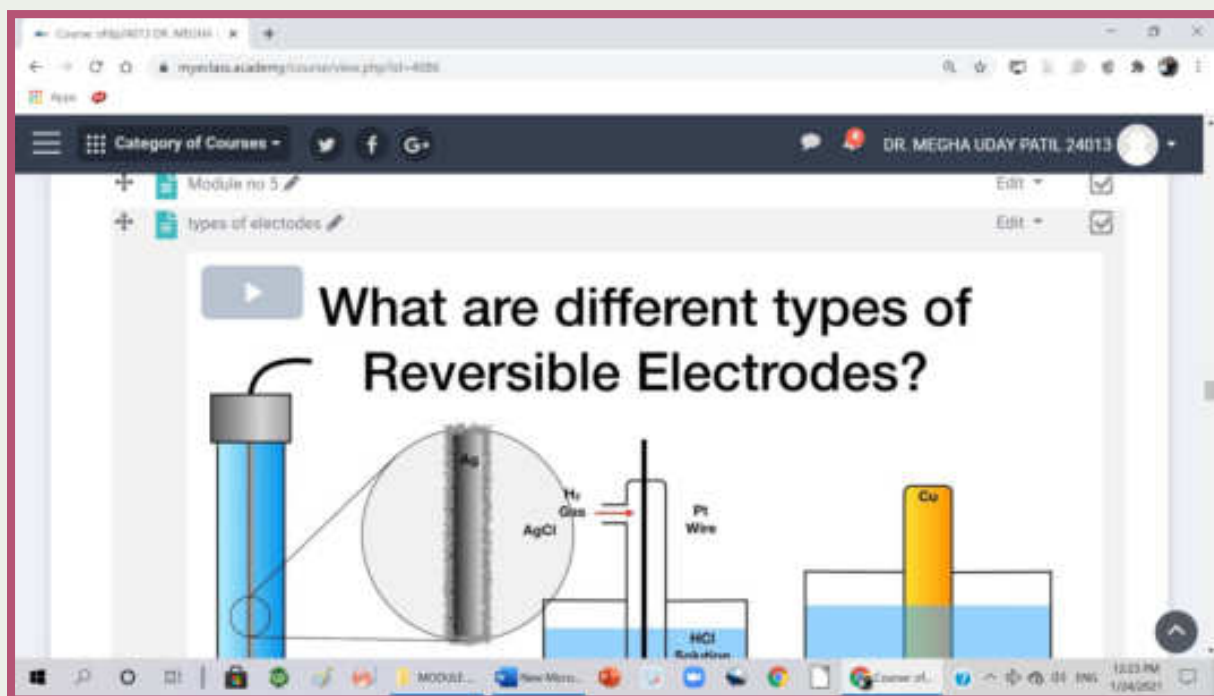
Each activity has an 'Edit' button and a checkmark icon. The page also features a 'Category of Courses' dropdown menu and social media icons for Twitter, Facebook, and Google+. A 'Add an activity or resource' button is visible at the top right and bottom right of the content area.

Day 7:

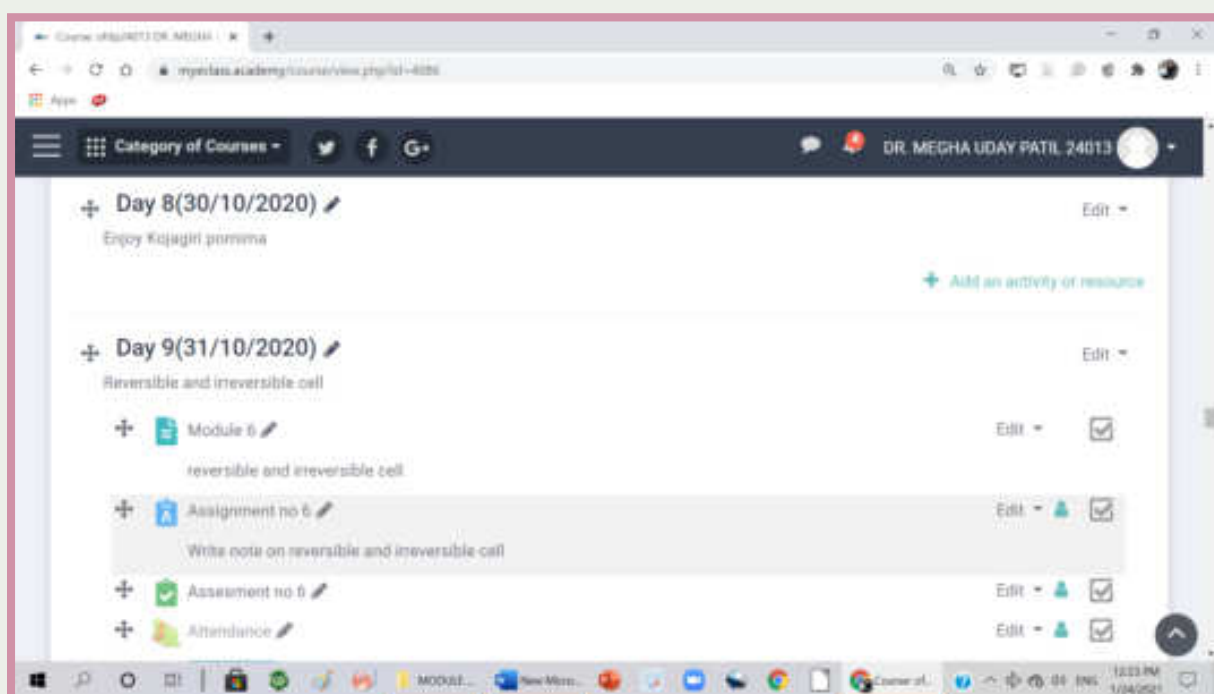
The screenshot shows a web browser window displaying a MyeClass course page. The URL is <https://myclass.academy/course/view.php?id=4026>. The user is logged in as DR. MEGHA UDAY PATIL 24013. The course content for Day 7 (29/10/2020) types of electrode includes:

- Attendance (Restricted: Not available unless:
 - The activity **Module no 5** is marked complete
 - The activity **Assesment no 5** is marked complete
 - You get a particular score in **Assesment no 5**On module no 5(types of electrode))
- Assignment no 5 (Write note on types of electrode in detail)
- Assesment no 5
- are you understand electromotive force topic on this platform of MyeClass
- Module no 5

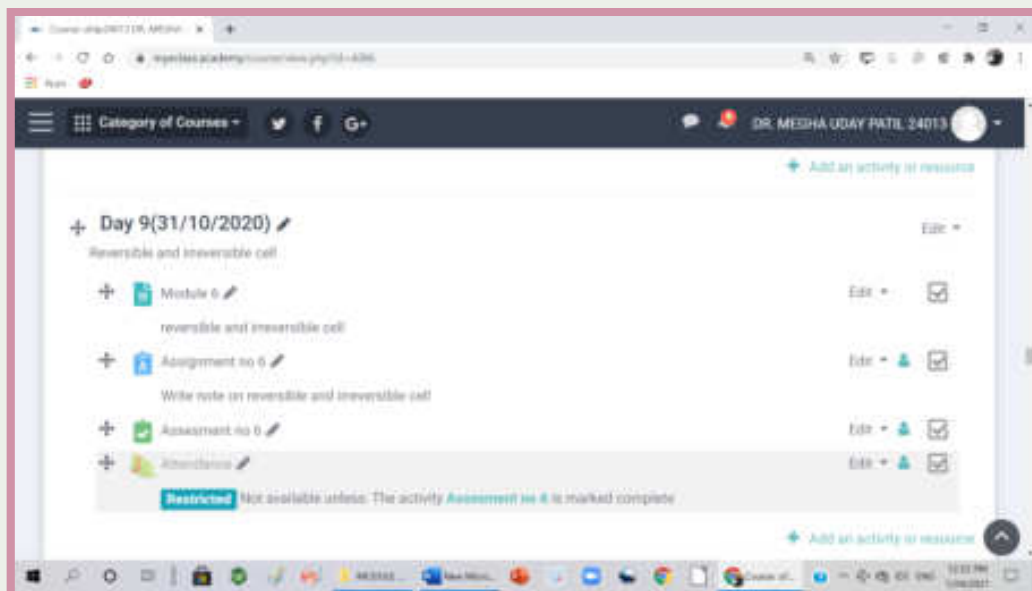
Each activity has an 'Edit' button and a checkmark icon. The page also features a 'Category of Courses' dropdown menu and social media icons for Twitter, Facebook, and Google+. A 'Add an activity or resource' button is visible at the top right and bottom right of the content area.



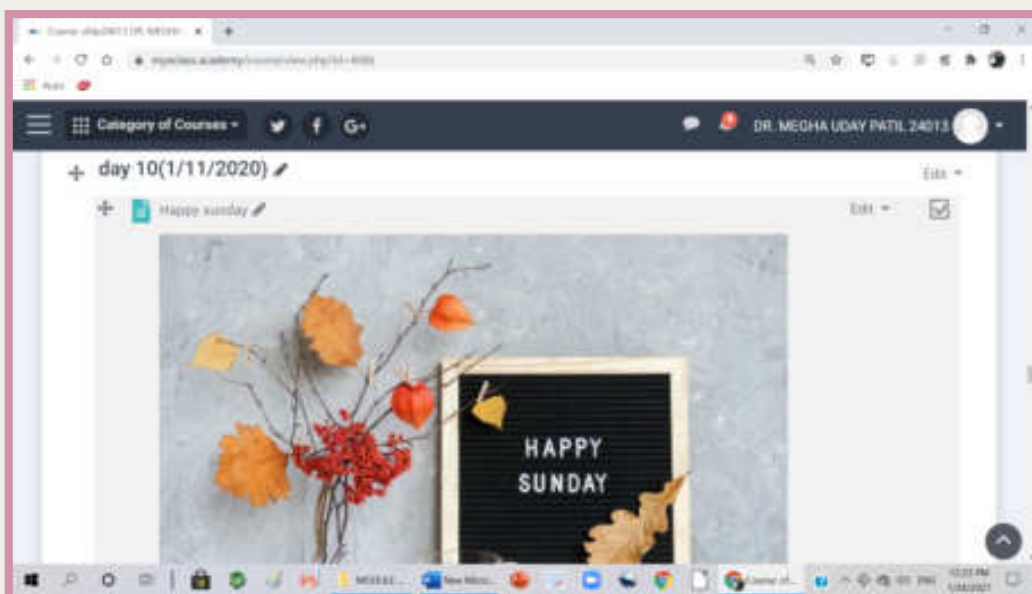
Day 8:



Day 9:



Day 10:



Day 11:

Category of Courses - DR. MEGHA UDAY PATIL 24013

Day 11(03/11/2020)

Chemical cell without transference

Module no 7

CHEMICAL CELL WITHOUT TRANSFERENCE

$\text{Pt} | \text{H}_2 (\text{g}, p) \quad | \quad \text{Ag} | \text{AgCl}$

$Q_a = \frac{a_{\text{Ag}} \cdot a_{\text{HCl}}}{P_{\text{H}_2}^{1/2} \cdot a_{\text{AgCl}}}$

$a_{\text{Ag}} = a_{\text{AgCl}} = 1$

Category of Courses - DR. MEGHA UDAY PATIL 24013

Assignment no 7

make note on chemical cell without transference

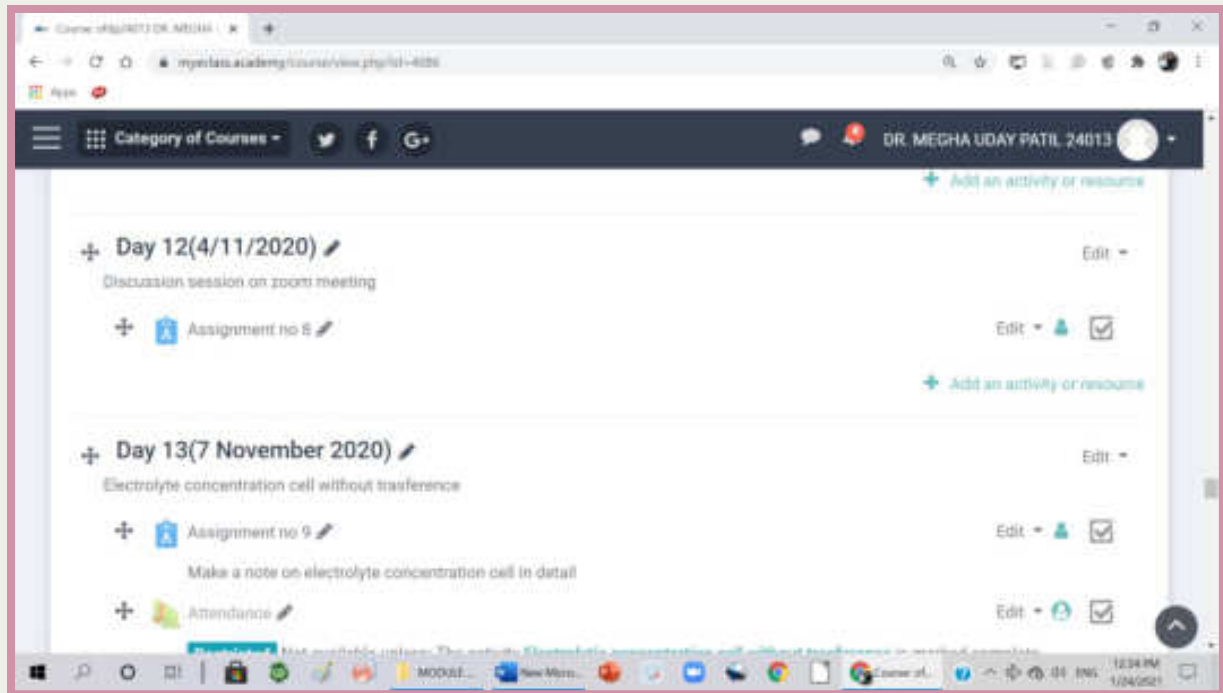
Attendance

Restricted Not available unless: The activity Module no 7 is marked complete

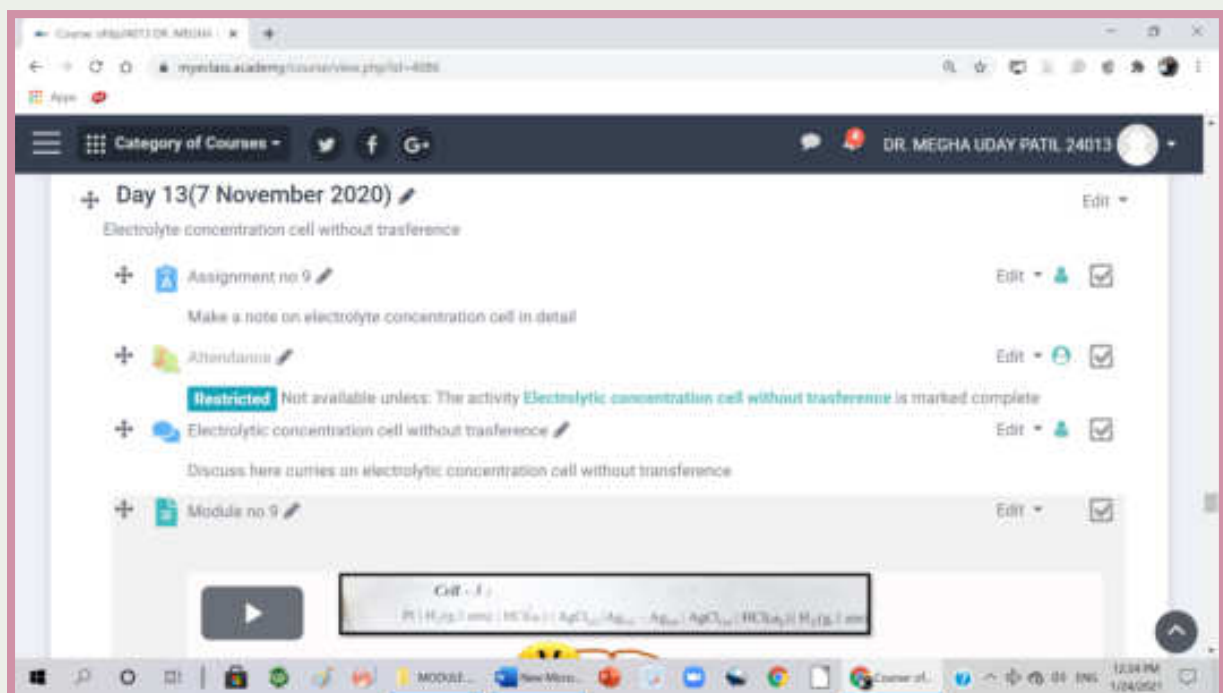
notes on chemical cell without transference

$Q_a = \frac{a_{\text{HCl}}}{P_{\text{H}_2}^{1/2}}$

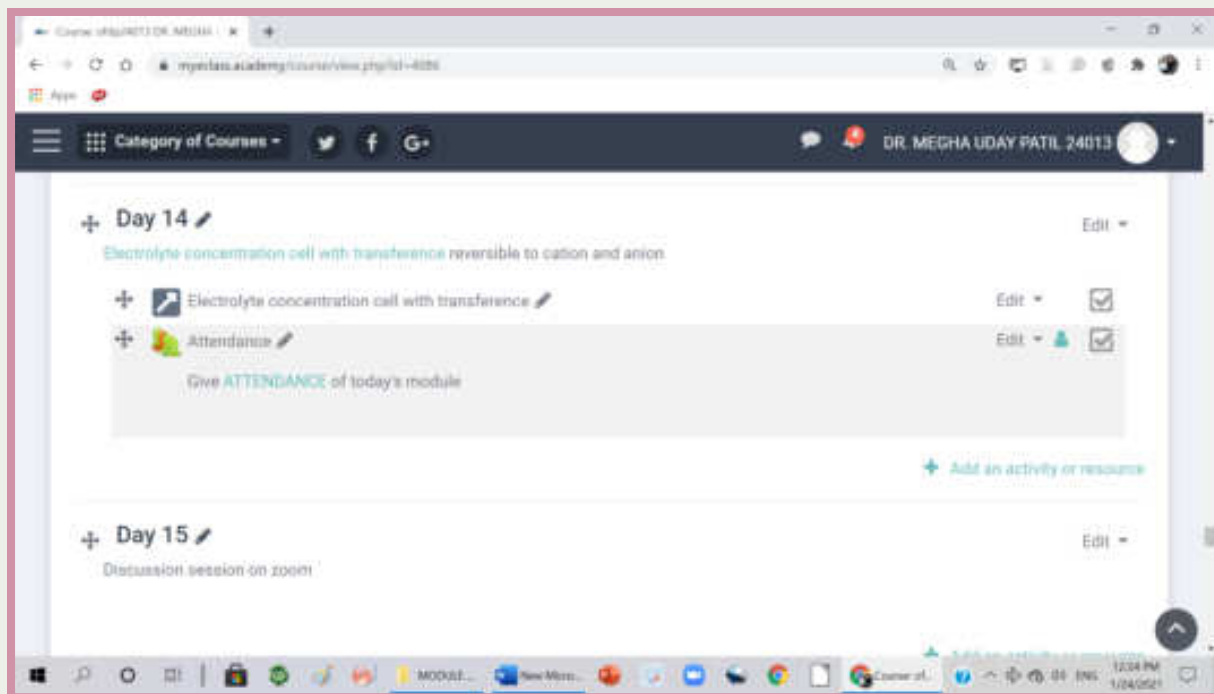
Day 12:



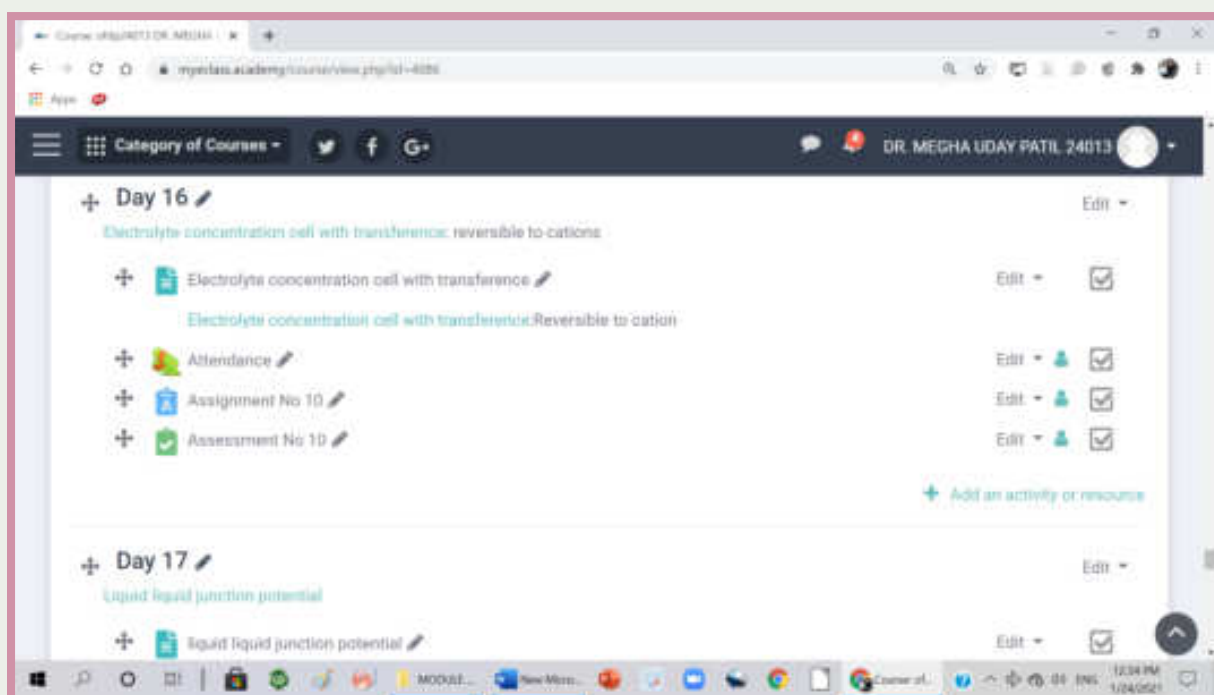
Day 13:



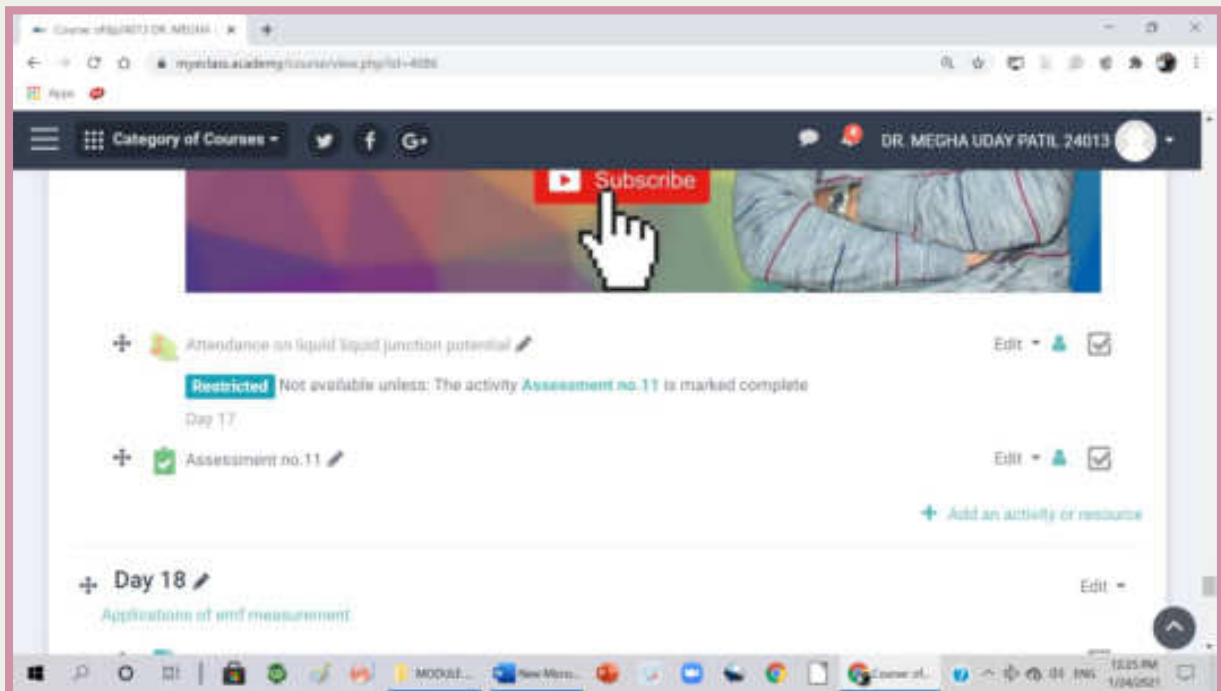
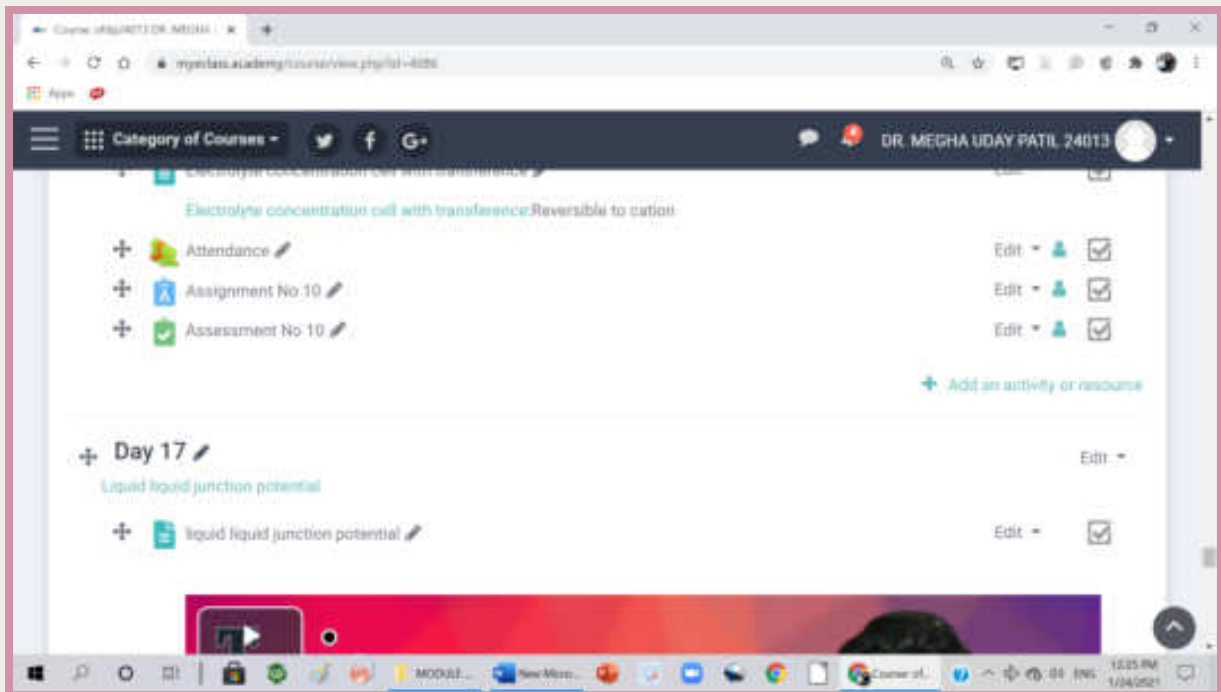
Day 14 & 15:



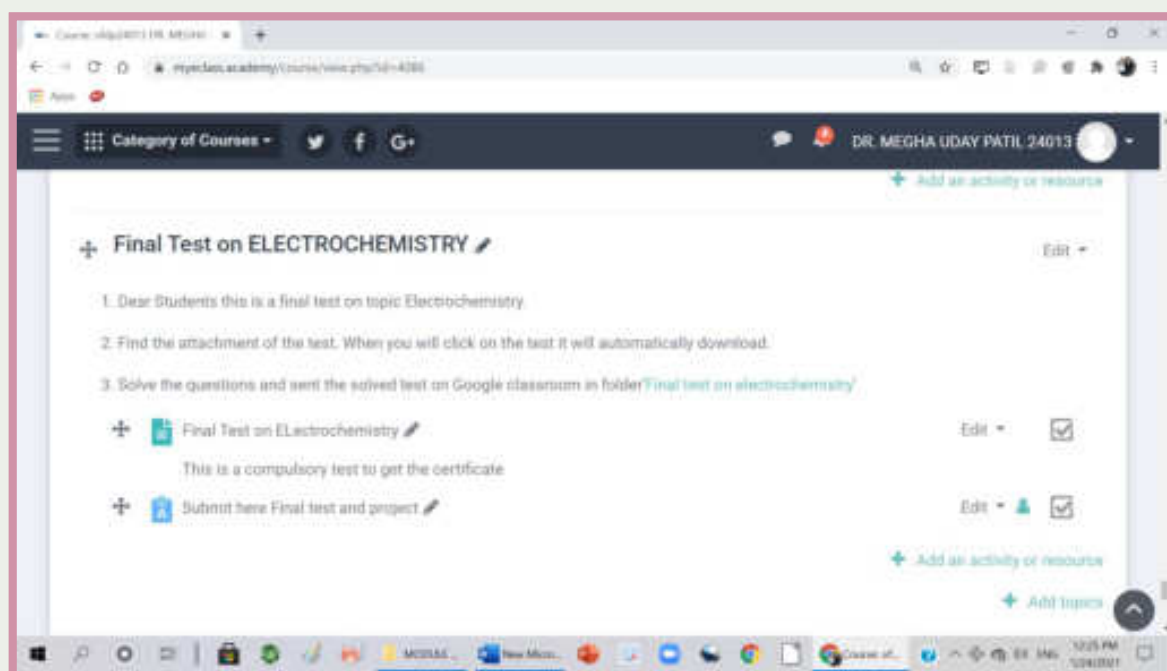
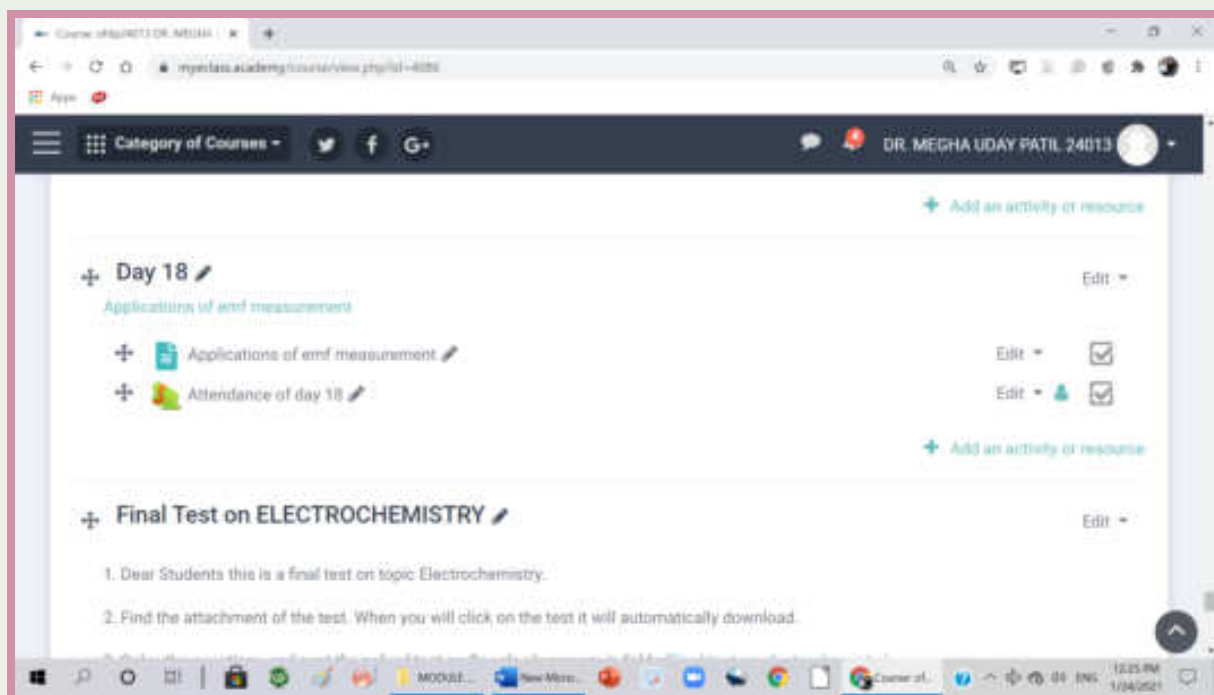
Day 16:



Day 17:

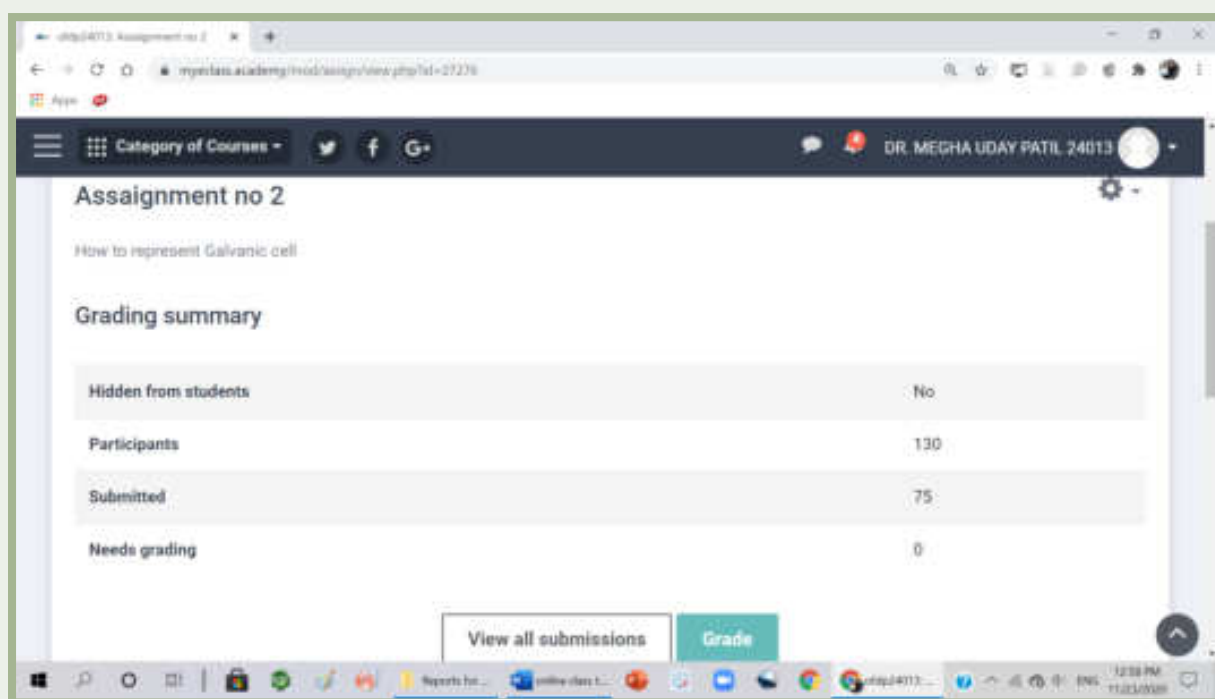


Day 18:

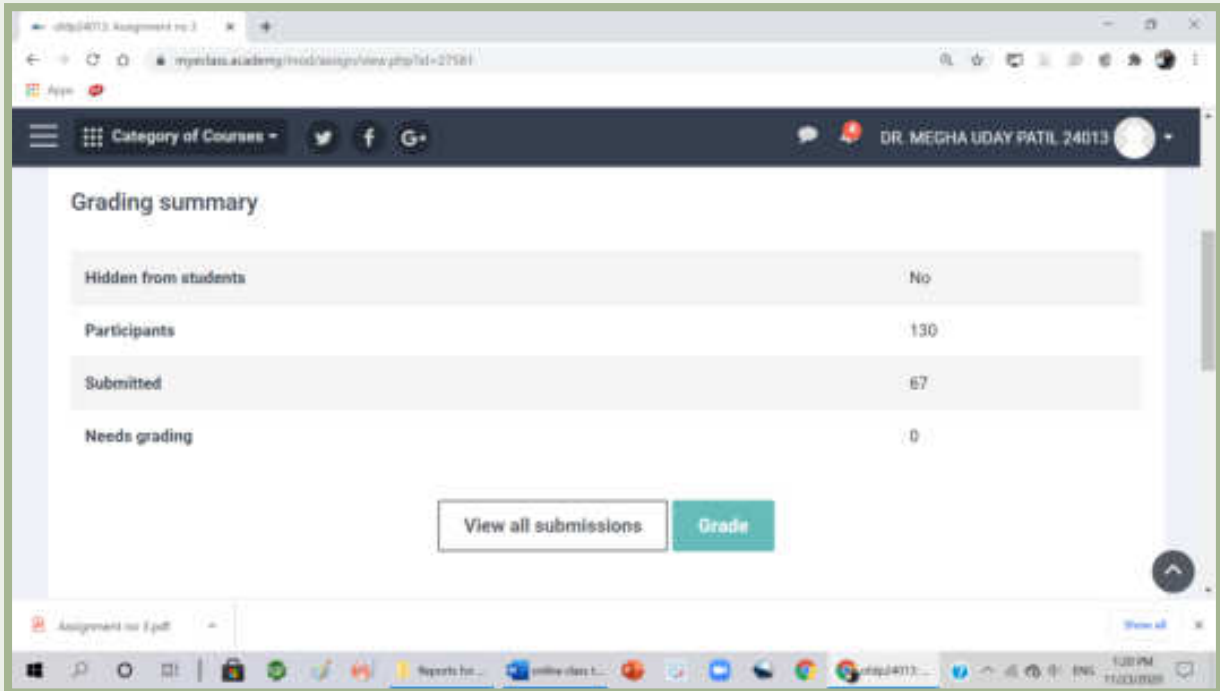


CLASS TEST AND ASSIGNMENTS TAKEN DURING COURSE:

Here we attached some screenshots of Checked Assignments conducted during course



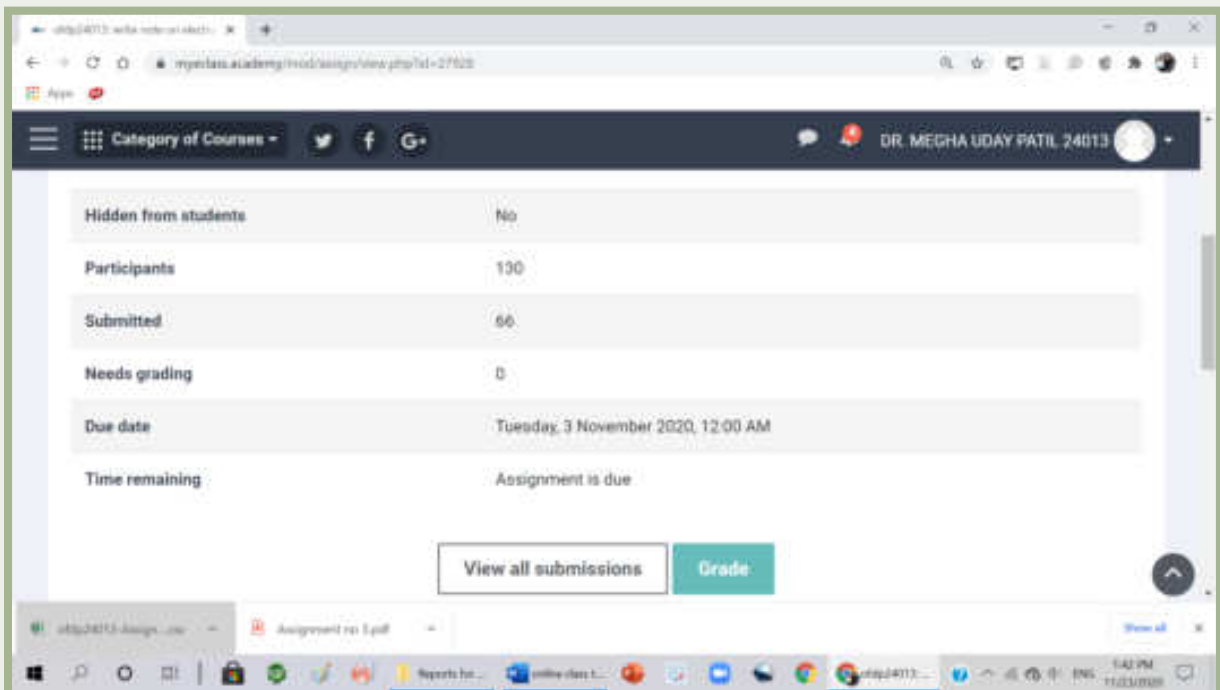
REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC



A screenshot of a web browser displaying a MOOC assignment page. The page title is "Assignment no. 2" and the URL is "myclass.academy/mod/assign/view.php?id=27528". The user is identified as "DR. MEGHA UDAY PATIL 24013". The page shows a "Grading summary" section with the following data:

Category	Value
Hidden from students	No
Participants	130
Submitted	67
Needs grading	0

At the bottom of the summary, there are two buttons: "View all submissions" and "Grade".



A screenshot of a web browser displaying a MOOC assignment page, similar to the one above. The page title is "Assignment no. 2" and the URL is "myclass.academy/mod/assign/view.php?id=27528". The user is identified as "DR. MEGHA UDAY PATIL 24013". The page shows a "Grading summary" section with the following data:

Category	Value
Hidden from students	No
Participants	130
Submitted	66
Needs grading	0
Due date	Tuesday, 3 November 2020, 12:00 AM
Time remaining	Assignment is due

At the bottom of the summary, there are two buttons: "View all submissions" and "Grade".

REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC

The screenshot shows a web browser window displaying a MOOC assignment page. The URL is <https://myclass.aademyg.com/assignment/view.php?id=27966>. The page title is "Assignment no 5". The assignment instruction is "Write note on types of electrode in detail". The "Separate groups" section is set to "All participants". The "Grading summary" table shows the following data:

Category	Count
Hidden from students	No
Participants	130
Submitted	57
Needs grading	0

The page also features a navigation menu, social media icons, and a user profile for "DR. MEGHA UDAY PATIL 24013". The system clock shows 2:40 PM on 11/23/2020.

The screenshot shows a web browser window displaying a MOOC assignment page. The URL is <https://myclass.aademyg.com/assignment/view.php?id=28041>. The page title is "Assignment no 6". The assignment instruction is "Write note on reversible and irreversible cell". The "Grading summary" table shows the following data:

Category	Count
Hidden from students	No
Participants	130
Submitted	54
Needs grading	0

At the bottom of the grading summary, there are two buttons: "View all submissions" and "Grade". The page also features a navigation menu, social media icons, and a user profile for "DR. MEGHA UDAY PATIL 24013". The system clock shows 3:07 PM on 11/23/2020.

REPORT ON SUBJECTIVE COURSE CONDUCTED ON MOOC

make note on chemical cell without transference

Grading summary

Hidden from students	No
Participants	130
Submitted	42
Needs grading	0

View all submissions Grade

CERTIFICATES ISSUED TO STUDENT:



Report Submitted by,

Dr. Megha U. Patil
Assistant Professor in Chemistry



“Dissemination of Education through Knowledge, Science and Culture”

-Shikshanmaharshi Dr. Babuji Salunkhe.

Shri. Swami vivekanand shikshan sanstha's

**Padmabhushan Dr. Vasantodada Patil Mahavidyaya,
Tasgaon, Sangli-416312, Maharashtra**

Department of English and IQAC

ONE DAY WORKSHOP ON

SOFT SKILLS AND PERSONALITY DEVELOPMENT

In Collaboration with Lead College Scheme, Shivaji University, Kolhapur

6th February 2019

2018-19

Activity	ONE DAY WORKSHOP ON SOFT SKILLS AND PERSONALITY DEVELOPMENT Jointly Organized by Lead College Scheme, Shivaji University, Kolhapur
Day and Date	Wednesday 6th February 2019
Chief Guest	Prof. A. N. Patil Dr. P. M. Patil
Participants	116, M: 34, F: 82

On the special occasion of Birth Centenary Year of Shikshanmahrshi Dr. Bapuji Salunkhe, the Promoter, Founder of our institute, The Department of English was organized One Day Workshop on Soft Skills and Personality Development on Wednesday 06th February 2019 under the Lead College Scheme.



Prof. A. N. Patil, The Former Head, Dept of English, P D V P College, Tasgaon was the Chief Guest of inaugural function. In his inaugural speech, he emphasized on the importance of Soft Skills and Personality Development.



Dr. P. M. Patil, the Resource Person of the second session discussed descriptively on the use of soft skills and how they can be applied for personality development.




Prof. P. V. Patil, the Resource Person of the third session stressed on how to development personality and also he explained various examples of the persons working in the various fields of the society.

In the fourth session valedictory function, Prof. K. S. Patil, Coordinator of Lead College Scheme P D V P College, Tasgaon spoke on various important activities done by under lead college scheme.

Principal, Dr. R. R. Kumbhar was the chairperson of the workshop. His presence and thought not only motivated but also enlightened for the students and teachers. Dr. T. K. Badame, Dr. B. T. Kanse, Dr. Maya Patil, Prof R B Mankar,

Prof. S. R. Ghogare, Prof. D. Y. Sakhare, Prof. A. S. Bagal , Dr. A. S. Wagh Dr. H. D. Nadaf were presented for the workshop. The workshop was organized successfully by Prof. A. S. Pachore, Prof. P. R. Khade, Prof. A. R. Patil, and Prof. N. R. Waghmare. The students of cluster colleges were actively participated in the workshop under the lead college scheme.

Dr. D. B. Thorbole
Coordinator
One Day Workshop
Department of English


(Dr. R. R. Kumbhar)

Principal

Padmabhushan Dr. Vasantodada Patil
Mahavidyalaya, Tasgaon, (Sangli)

06/02/2019 One day workshop on
Soft Skills & Personality Development

Sr.No.	Name of the students	class	Mobile NO.
1	Mrs. Manisha Dharmendra ^{Throat}	M.A.II	7219343636
2	Mrs. Mujawar Asma Shabbir	M.A.II	8956206005
3	Mrs. Awale Mayuri Dilip	M.A.II	8689500467
4	Mrs. Salunkhe Dipali Prakash	M.A.II	9823780814
5	Mrs. Dabhade Snehal Pandharinath	M.A.II	7387803926
6	Mrs. Jadhav Aarathi Mohan	M.A.I	9503896680
7	Mrs. Mandale Pranali Prakash	M.A.I	9373020098
8	Mr. Koligudde Pujarajendra	M.A.I	9370126831
9	Mrs. Shivasharan Tejswini Gurubas	B.A.III	9075163255
10	Mrs. Kumbhar Pragati Jalindar	B.A.III	9309960077
11	Mrs. Chavan Smita Devdas	B.A.III	7219631437
12	Mrs. Jadhav Kshitiya Dattatraya	B.A.III	9800573353
13	Mrs. Salunkhe Punam Jotiram	B.A.II	9665688399
14	Mrs. Todkar Ankita Hanmant	B.A.II	9975699813
15	Mrs. Pol Rutuja Sunil	B.A.II	8805856711
16	Mrs. Nikam Aishwarya Ramesh	B.A.II	9665223400
17	Patil Pragati Abasaheb	B.A.II	7558666577
18	Mrs. Nikam Snehal Suresh	B.A.II	7387495572
19	Gele Vaijayanti Sanjay	B.A.II	7556912532
20	Vajarinkar snehal. R.	B.A.II	8483828038
21	Mrs. Patil Dhanashri Bhagavat	B.A.II	9518308829
22	Mr. Koligudde Ashwin Kumar	B.A.II	9075085145
23	Mr. Sathe Sushant Sudam	B.A.II	7066319660
24	Desai Akshay Ramesh	B.A.II	9764618579
25	Mrs. Jadhav Pooja Jayram	B.A.II	7248996341
26	Mr. Patil Manasi Nivas	B.A.I	9637259955

Sr.No.	Name of the students	class	Mobile No.
27	Patil Yivek Suryakant	B.A.I	8053918615
28	Korde Vidyarani Ramchandra	B.A.I	8888338374.
29	Mane Aditya Ravindra	B.A.I	8380939659
30	Kiwale Megha Ekanath	B.A.I	7028274414
31	Aiwale Mayuri Shivanand	B.A.I	7066825037
32	Chavan Prantl Baburao	B.A.I.	9096526733
33.	Patil Bhagyashri Govind	B.A.I	7517424552.
34.	Landage Sachin Sambhaji	B.com.III	9172584508
35	Kapse Mahesh Vijay	B.A.com.III	9309995410
36	Patil Praphulla Namdev	B.com.III	9766400413
37.	Patil Shweta Subhash	B.com.I	9970350919
38	Chavan Ranjana Devdas	B.com.I	8459505039
39	Chavan Monika Rajendra	B.com.I	8459505039
40	Mane Sonali Madhukar	B.com.I	9503350309
41	Sagare Arati Sudesh	B.com.I	8329060715
42	Mithare Rutuja Kalgonda	B.com.I	7972399007.
43.	Chavan Usha Sanjay	B.com.I	7841921707
44	Chavan Sarika Shahaji	B.Sc.I	7420959716
45	Shendage Rupali Balaso	B.Sc.I	7972625816
46	Shendage Pratiksha Ashok	B.Sc.I	8956481561
47	Giri Kavita Sanjay	B.Sc.I	9503047566
48	Shinde Kalyani Sanjay	B.Sc.I	9637274199
49	Kamble Shivani Sanjay	B.Sc.I	8856918648
50	Desai Amruta Madhukar	B.Sc.I	8208642982
51	Salunkhe Snehal Dinesh	B.Sc.I	9960933580
52	Patil Manisha Sanjit	B.Sc.I	8956810448

sr.No.	Name of the students	class	Mobile No.
53	Patole Sayali Ramchandra	B.Sc.I	9764291680
54	Jadhav Swapnali Subhash	B.Sc.I	9970033354
55	Mali Pronal Balaso	B.Sc.I	9172754073
56	Bile Adarshrao Sptaram	B.Sc.I	7057313661
57.	Mali Dhananjay Rajkumar	B.Sc.I	8600574981
58	Shendge Rushikesh Ravindra	B.Sc.I	9561868697
59	shinde Ashish Ashokrao	B.Sc.I	8600441628
60	Patil Prathmesh Manik	B.Sc.I	9373750265
61	Savvashe Snehal Arun	B.Sc.I	9595587882
62	Patil Sonali Balaso	B.Sc.I	8624910933
63	Zende Gouri Subhash	B.Sc.I	7499374647
64	Rajmane Gayatri Vipshwas	B.Sc.I	9325805499
65	Patil Nikita Nagesh	B.Sc.I	9665362025
66	Sabale Swapnali Rajendra	B.Sc.I	9579813394
67	Patil Amruta Anil	B.Sc.I	9730340890
68	Zarekar Snehal Mahadev	B.Sc.I	9767517945
69	chavan Tejashree Shreekant	B.Sc.I	9421310176
70	Graikwad Prajakta Nagesh	B.Sc.I	8007723554
71	Mane Mohini Sandip	B.Sc.I	8551827421
72	Mane Nilam Sampat	B.Sc.I	7756959068
73	Mulla Misba Jamirahmad	B.Sc.I	7385999947.
74.	Jadhav Pritee Ravindra	B.Sc.I	9325147802
75	Nadaf Tarannum Suraj	B.Sc.I	7385623886
76	Jamdade Rutuja Vijay	B.Sc.I	9096374483
77	Patil Asavari Shivaji	B.Sc.I	9767573880
78	Watharkar Aditi Suresh	B.Sc.I	7028583442.

Sr.No.	Name of the students	class	Mobile
79	Malkar Neha Bharat	B.Sc.I	8600918891
80	Patil Ankita Shankar	B.Sc.I	7028369650
81.	Godave Sanket Sidhaling	B.Sc.I	8805129220
82.	Gole Sourabh Dinakar	B.Sc.I	7620150821
83.	Gaikwad Indrajit Rajkamal	B.Sc.I	7559338766
84	Desai Prathmesh.	B.Sc.I	
85	Jadhav Amruto Jagannath	M.A-I	9503591504
86)	Jadhav Sambuddhi Dattatray	M.A-I	9860679353
87	Yogita Yuvraj Yuvraj Patil	BA III	9545589066
88	Komal Tanaji Mulik	BA III	9665377632
89	Pooja Mahadev mali	BA III	8208353671
90	Peachita Prakash. Patil	BA III	7719048142
91	Poonam Vinayak Pore	BA III	9284886851
92	Peetik Sanjay Chauhan	BA III	8830404508
93	Sahil Santosh Sadanand	B.com I	7448172142
94.	Jadhav Npkhil Prabhakar	B.com I	7888126530
95	Jadhav Harshali Vijaysinh	M.A-I	7350202304
96	Sutar Saniya Manuti	B.com I	7378939564
97	Mayuri Sunil mali	BA III	7499273092
98	Jadhav Snehal Suresh	BA III	8788540081
99	Surywansh Satyajit vijay	BA III	9637821140
100	Devkule Prasad Devk Dinkar	B.A. I	8605841368
101	Kumbhar Yrushi Bhat	B.Sc I	9545420954
102	Sakhare Pallavi Mahadev	B.Sc I	9096710955
103	Jadhav Yaishali Hanmani	BA I	9923284817
104	Mulani Nishad Rafik	BA I	8999603311

Sr.No.	Name of the students	Class	Mobile No.
105	Buwa Pooja Dattatray	M.A.II	9637183617
106	Sudae Nilam Suresh	B.com	9766291687
107	Salankhe Manoj Shashikant	B.A.II	7219088850
108	Ronkhambe Subhas Baban	M.A. (B.Ed)	8999940128
109	Shinde Ajinnya Vasant	B.Sc.I	9975207910
110	Pandit Shubham Bhagvat	B.A.III	8275496493
111	Pawar Rohan Bhageerat	B.A.III	9673236423
112	Adsul Ruturaj Balaso	B.A.III	8446295121
113	Kale Ardhut Ashokrao	B.A.II	7709833827
114	Patil Harshali Ashok	B.A.III	9975881704
115	Awale Karishma Suresh	B.A.III	8530399594
116	Shinde vijay vishram	B.A.III	8308212342

Male - 34
Female - 82
Total 116

Other College, (Vidyaniketan)

Sr.No.	Name of the students	class	Mobile No.
1)	Vaishnavi Satish Upadhye	B.com III	7620067595
2)	Sunita Sanjay Patil	B.com III	9730075659
3)	Pallavi Prabhakar Sutar	B.com III	9579246449
4)	Jadhav Snehal Narayan	BA II	9923138280
5)	Chavan Komal Bajarang	BA II	- "
6)	Patil Arati Jagnath	BA II	- "
7)	Jadhav Rutuja Jalindar	BA II	- "
8)	Patil Pratiksha Prakash	BA III	917767830575
	Name of the teacher		
①	Patil Aparna Suryakant	Teacher	



“Dissemination of Education for Knowledge, Science and Culture”
-Shikshanmaharashi Dr. Bapuji Salunkhe

Swami Vivekanand Shikshan Sanstha Kolhapur

**Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,
Tasgaon,
Sangli-416412, (MS) India**

Report on

VASANT AVISHKAR

Research Convention

30th December 2019



2019-20

“Dissemination of Education for Knowledge, Science and Culture”
-Shikshanmaharashi Dr. BapujiSalunkhe

Swami Vivekanand Shikshan Sanstha Kolhapur

**Padmabhushan Dr. Vasantraodada Patil Mahavidyalaya,
Tasgaon, Sangli-416412, (MS) India**

Organize

VASANT AVISHKAR

➤ Event	: Research computation
➤ Date	: 30th December 2019
➤ Organizer	: PDVP College Tasgaon
➤ venue	: College Campus

PARTICIPANT

Participant	Male	Female	Total
Students	31	47	78
Faculty	12	05	17
			95

Innovative research is a search for new business and strategic techniques and methods. They develop and optimize well-known methodologies, enabling the implementation of new and better solutions. Innovative research focuses on creating new ideas, analyzing problems, diagnosing them and identifying their causes.

Accordingly, our college initiating the organization of “Vasant Avishkar” every year to provide a platform to the students from the various groups and extending the helping hands to understand the research attitude and acquiring the scientific knowledge thus transforming

for the cause of development. This will also educate students and teachers to understand their responsibility towards societal development. In this year also, our college was organized the “Vasant Avishkar” research computation to initiate the scientific attitude in students and to express their research ideas.

In this year “Vasant Avishkar” research poster computation was organized on 30th December 2019 in college campus. For the inaugural function we invite as a chief guest to **Dr. P. M. Patil Sir**, A.S.C. College, Palus. Also we invite **Dr. Sunil Kamble** SMDBS College, Miraj and **Dr. Sajjan Kamble** ACS College, Palus as an examiner for the same computation. Principal **Dr. Milind Hujare** as a precedent of the program guided the student about the research and informs him how such activities auspice there features. Chief Guest of program Dr. P. M. Patil In his speech said that such event gives the right opportunity to the students at the ground level that leads to forming such a young researchers.

INAUGURAL FUNCTION



Principal Dr. Milind Hujare sir delivered speech on Inaugural function



Chief Guest Dr. P. M. Patil sir delivered speech on inaugural function



Introductory speech by coordinator Dr. Ajay Ambhore



Felicitation of Chief Guest Dr. P. M. Patil



Felicitation of Guest Dr. Sunil Kamble



Felicitation of Guest Dr. Sunil Kamble



Inauguration of poster presentation



Invited guest with college staff

Introductory speech was delivered by the program coordinator **Dr. Ajay Ambhore**. Total 78 students are participated and present their research poster in this computation. All these research projects are evaluated by the examiners.



Chief Guest and Principal observing the project in the computation



Examiners examine the research projects



Examiners examine the research projects



Vote of thanks by Dr. Amol Sonawale

After the completion of evaluation process all the examiners handover the result towards our Principal and it was declared to students on the same day. From the result 12 students are selected to participate in district level “Avishkar” organized by Shivaji University, Kolhapur at Balvant College, Vita.

Photo Gallery





REGISTRATION

"Department of Education for Knowledge, Science & Culture"
 Shri Swami Vivekananda Shiksha Sanstha Kalyana
 Padmabhanu Dr. Vasantaramada Patil Mahavidyalaya, Tasgaon, Dist. Sangli
Yashant Arishkar 2019-20 Date: 06/12/2019
 1. CATEGORY: HUMANITIES, LANGUAGE, FINE ARTS, EDUCATION, ETC

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Mark No.	Sign
1.1	MANOJ YADAV S.	B.A. II	Geography	मानवसंसाधन विकास का भू-विकास	557416104	[Signature]
1.2	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416105	[Signature]
1.3	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416106	[Signature]
1.4	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416107	[Signature]
1.5	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416108	[Signature]
1.6	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416109	[Signature]
1.7	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416110	[Signature]

"Department of Education for Knowledge, Science & Culture"
 Shri Swami Vivekananda Shiksha Sanstha Kalyana
 Padmabhanu Dr. Vasantaramada Patil Mahavidyalaya, Tasgaon, Dist. Sangli
Yashant Arishkar 2019-20 Date: 06/12/2019
 1. CATEGORY: HUMANITIES, LANGUAGE, FINE ARTS, EDUCATION, ETC

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Mark No.	Sign
1.1	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416111	[Signature]
1.2	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416112	[Signature]
1.3	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416113	[Signature]
1.4	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416114	[Signature]
1.5	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416115	[Signature]
1.6	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416116	[Signature]
1.7	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416117	[Signature]
1.8	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416118	[Signature]
1.9	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416119	[Signature]
1.10	ADARSH KADAM S.	B.A. II	English	मानवसंसाधन विकास का भू-विकास	557416120	[Signature]

"Department of Education for Knowledge, Science & Culture"
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 Padmabhanu Dr. Vasantaramada Patil Mahavidyalaya, Tasgaon, Dist. Sangli
Yashant Arishkar 2019-20 Date: 06/12/2019
 2. CATEGORY: COMMERCE, MANAGEMENT, LAW, ETC

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Mark No.	Sign
2.01	ADARSH KADAM S.	B.Com	Commerce	मानवसंसाधन विकास का भू-विकास	557416121	[Signature]
2.02	ADARSH KADAM S.	B.Com	Commerce	मानवसंसाधन विकास का भू-विकास	557416122	[Signature]
2.03	ADARSH KADAM S.	B.Com	Commerce	मानवसंसाधन विकास का भू-विकास	557416123	[Signature]
2.04	ADARSH KADAM S.	B.Com	Commerce	मानवसंसाधन विकास का भू-विकास	557416124	[Signature]

"Department of Education for Knowledge, Science & Culture"
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Yashant Arishkar 2019-20 Date: 06/12/2019
 3. CATEGORY: PHYSICENCE

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Mark No.	Sign
3.01	ADARSH KADAM S.	B.Sc	Physics	मानवसंसाधन विकास का भू-विकास	557416125	[Signature]
3.02	ADARSH KADAM S.	B.Sc	Physics	मानवसंसाधन विकास का भू-विकास	557416126	[Signature]
3.03	ADARSH KADAM S.	B.Sc	Physics	मानवसंसाधन विकास का भू-विकास	557416127	[Signature]
3.04	ADARSH KADAM S.	B.Sc	Physics	मानवसंसाधन विकास का भू-विकास	557416128	[Signature]
3.05	ADARSH KADAM S.	B.Sc	Physics	मानवसंसाधन विकास का भू-विकास	557416129	[Signature]
3.06	ADARSH KADAM S.	B.Sc	Physics	मानवसंसाधन विकास का भू-विकास	557416130	[Signature]

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Bikramnagar, Dr. Jagdish Chandra
Bai Swami Vivekananda Mahavidyalaya College

Pattadakshin Dr. Vasantnagar Palli Mahavidyalaya, Tasgaon, Dist. Durg
Vasant Avishkar 2019-20 Date: 20/12/2019

3. CATEGORY: PURE SCIENCE

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Marked No.	Rank
1	Pradeep, Dushyant	B.Sc.	Chemistry	Application of the use of green catalyst for synthesis of bioactive molecule	80/100	1st
2	Yashraj, Pradyumn	B.Sc.	Chemistry	Study of the synthesis of bioactive molecule	75/100	2nd
3	Smiti, Vikramjit	B.Sc.	Chemistry	Study of the synthesis of bioactive molecule	70/100	3rd
4	Aravind, Pradyumn	B.Sc.	Chemistry	Study of the synthesis of bioactive molecule	65/100	4th
5	Garvit, Nikhil	B.Sc.	Chemistry	Comparative study of college students in educational way	60/100	5th
6	Neha, Pooja	B.Sc.	Chemistry	Comparative study of college students in educational way	55/100	6th
7	Kaushik, Vijay	B.Sc.	Chemistry	Comparative study of college students in educational way	50/100	7th
8	Shruti, Manoj	B.Sc.	Chemistry	Comparative study of college students in educational way	45/100	8th

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Vasant Avishkar 2019-20 Date: 20/12/2019

4. CATEGORY: AGRICULTURE & ANIMAL HUSBANDRY

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Marked No.	Rank
1	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	85/100	1st
2	Chaitanya, N. N.	B.Sc.	Botany	Thin Film Solar Cell	80/100	2nd
3	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	75/100	3rd
4	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	70/100	4th
5	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	65/100	5th
6	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	60/100	6th
7	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	55/100	7th
8	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	50/100	8th

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Vasant Avishkar 2019-20 Date: 20/12/2019

4. CATEGORY: AGRICULTURE & ANIMAL HUSBANDRY

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Marked No.	Rank
1	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	85/100	1st
2	Chaitanya, N. N.	B.Sc.	Botany	Thin Film Solar Cell	80/100	2nd
3	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	75/100	3rd
4	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	70/100	4th
5	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	65/100	5th
6	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	60/100	6th
7	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	55/100	7th
8	Harshita, N. N.	B.Sc.	Botany	Thin Film Solar Cell	50/100	8th

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Pattadakshin Dr. Vasantnagar Palli Mahavidyalaya, Tasgaon, Dist. Durg
Vasant Avishkar 2019-20 Date: 20/12/2019

5. CATEGORY: ENGINEERING & TECHNOLOGY

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Marked No.	Rank
1	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	85/100	1st
2	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	80/100	2nd
3	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	75/100	3rd
4	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	70/100	4th
5	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	65/100	5th
6	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	60/100	6th
7	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	55/100	7th
8	Smiti, Karan	B.Sc.	Physics	Thin Film Solar Cell	50/100	8th

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Vasant Avishkar 2019-20 Date: 20/12/2019

6. CATEGORY: MEDICINE & PHARMACY

Sr. No.	Name of Student	Class	Name of Department	Title of Project	Marked No.	Rank
1	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	85/100	1st
2	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	80/100	2nd
3	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	75/100	3rd
4	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	70/100	4th
5	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	65/100	5th
6	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	60/100	6th
7	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	55/100	7th
8	Smiti, Karan	B.Sc.	Botany	Thin Film Solar Cell	50/100	8th



वसन्त अविष्कार 2019-20
Bikramnagar, Dr. Jagdish Chandra
Bai Swami Vivekananda Mahavidyalaya, Tasgaon, Dist. Durg
Vasant Avishkar - 2019-2020

Category I : Humanities, language, fine arts, education etc

Poster No.	Name of the Student	Class	Department	Title of the Project	Remark
1.01	Pallavi Bhagya	B.A.II	Economics	...	First
1.02	Vandana Suresh	B.Com.II	Geography	...	Second
1.03	Lakshmi Anil	B.A.II	Political Science	...	Third

Category II : Commerce Management and law, etc

Poster No.	Name of the Student	Class	Department	Title of the Project	Remark
2.01	Pallavi Bhagya	B.Com.II	Commerce	...	First
2.02	Hare Ashwini	B.Com.II	Commerce	...	Second
2.03	Hare Shanta	B.Com.II	Commerce	...	Third

Category III : Pure Science

Poster No.	Name of the Student	Class	Department	Title of the Project	Remark
3.01	Harshita N. N.	B.Sc.III	Chemistry	Application of Alloys	First
3.02	Jayashree Suresh	B.Sc.III	Zoology	Effect of radiation	Second
3.03	Sakshita Suresh	B.Sc.III	Botany	Model of cell	Third

Category IV : Agriculture & Animal Husbandary

Poster No.	Name of the Student	Class	Department	Title of the Project	Remark
4.01	Yashraj Pradyumn	B.Sc.III	Zoology	LTPP Model	First
4.02	Garvit Nikhil	B.A.II	Economics	...	Second
4.03	Harshita N. N.	B.A.II	Geography	Green House	Third

Category V : Engineering and Technology

Poster No.	Name of the Student	Class	Department	Title of the Project	Remark
5.01	Smiti Karan	B.Sc.III	Physics	Thin Film Solar Cell	First
5.02	Pallavi Bhagya	B.Sc.III	Physics	Diode Rectifier	Second
5.03	Sakshita Suresh	B.Sc.III	Physics	Thin Film Solar Cell	Third

Yatani Akshar 2019-20
(Date - 30.11.2019)

➤ Introduction	Dr. Ajay N. Ambhore Program Coordinator
➤ President	Prin. Dr. Milind S. Hujare
➤ Chief Guest	Dr. P. M. Patil Art, Commerce and Science College, Bale, Dist. Sangli
➤ Guest & Examinee	Dr. Sujata Kamble Art, Commerce and Science College, Bale, Dist. Sangli
➤ Guest & Examinee	Dr. Snehal Kamble MADRIS College, Malegaon, Dist. Sangli
➤ Vice of Thesis	Dr. A. S. Wagh Member, Research Committee
➤ Anchoring	Dr. A. G. Sonawale Member, Research Committee

Yatani Akshar 2019-20
(Date - 30.11.2019)

➤ Registration	8:30 AM to 9:30 AM
➤ Paper Arrangement	10:30 AM to 11:30 AM
➤ Inaugural Function	12:30 AM to 1:30 AM
➤ Paper Examination	2:30 AM to 04:00 PM
➤ Lunch	1:00 PM to 1:45 PM
➤ Result	4:00 PM

Yatani Akshar 2019-20
(Date - 30.11.2019)

Program Committee

➤ Registration Committee	➤ Dr. A. G. Sonawale ➤ Dr. B. J. Kadam ➤ Dr. R. B. Joshi
➤ Inaugural Function Committee	➤ Dr. V. V. Joshi ➤ Dr. S. C. Kadam
➤ Result Analysis Committee	➤ Dr. A. S. Wagh ➤ Dr. A. S. Wagh
➤ Headery Committee	➤ Prin. Dr. Milind S. Hujare ➤ Prof. V. S. Patil ➤ Dr. V. V. Joshi ➤ Prof. Dr. H. A. Kulkarni ➤ Prof. Dr. S. S. Patil


Research Committee Chairman: 
Dr. Ajay N. Ambhore

Principal: 
Dr. Milind S. Hujare



Ajay N. Ambhore

**Dr. Ajay N. Ambhore
(Coordinator)**


Prin. Dr. Milind S. Hujare
Principal
Padmabhushan Dr. Vasantraodada Patil
Mahavidyalaya, Tasgaon (Sangli).