

Shri Swami Vivekanand Shikshan Sanstha Kolhapur  
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon

**PG Department of Chemistry**  
**Notice**

Date : 25/07/2019


*Certificate Course in Analytical  
Techniques*

B. Sc. III Chemistry students are hereby informed that we are going to conduct  
**Certificate Course in Laboratory Management and Analytical Techniques.**

✓ This course includes

- Fundamentals in Laboratory Management
- Fundamentals in Analytical Chemistry

✓ The students can register their names upto 29<sup>th</sup> July 2019.

  
Prof. Dr. S. S. Patil  
Head of the Department  
Dept. of Chemistry  
P.D.V.P. College, Tasgaon

"Dissemination of Education for Knowledge, Science and Culture" - Shikshanmaharshi Dr. Bapuji Salunkhe  
Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's  
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon  
**Post Graduate Department of Chemistry**

A REPORT SUBMITTED  
TO  
**Internal Quality Assurance Cell**

Certificate Course in  
**Analytical Techniques**

**2019-20**

“ ज्ञान, विज्ञान आणि सुसंस्कार यांसाठी शिक्षण प्रसार ”

॥ शिक्षणमहर्षी डॉ. वापूजी साळुंखे

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur

**PADMABHUSHAN DR. VASANTRAODADA PATIL  
MAHAVIDYALAYA, TASGAON**

(AFFILIATED TO SHIVAJI UNIVERSITY, KOLHAPUR)

**Department of Chemistry**

**Certificate Course**

**in**

**Analytical Techniques**

**To be implemented from**

**June 2019**

Swami Vivekanand Shikshan Sanshtha, Kolhapur  
**Padmabhushan Dr. Vasantodada Patil**  
Mahavidhyalaya, Tasgaon  
**PG DEPARTMENT OF CHEMISTRY**

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Date: 10 July 2019

To,  
The Principal,  
P D V P College, Tasgaon  
Tal: Tasgaon, Dist: Sangli - 416 312.

Sub: - Regarding Permission to commencement of Certificate Course in Analytical  
Techniques

Respected Sir,

With reference to above subject, our department is going to conduct **Certificate Course in Analytical Techniques** for B. Sc. students in this academic year. So, I request you to give us permission to conduct above said course.

Thanking you

Yours faithfully

(Dr. S. S. Patil)

Head  
Dept. of Chemistry  
P.D.V.P. College, Tasgaon

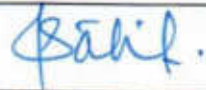







**DEPARTMENT OF CHEMISTRY**  
**CERTIFICATE COURSE IN ANALYTICAL TECHNIQUES**  
**Academic year 2019-20**


DT. 15 July 2019

**Notice of Meeting**

The meeting of the members of BOS in Certificate Course in Analytical Techniques is conveyed on 22<sup>th</sup> July, 2019 for the preparation of syllabus.

Name	Position	Signature
Prof. (Dr) Suresh S. Patil	Chairman	
Dr. S. D. Jadhav	Member	
Dr. A. N. Ambhore	Member	
Mr. S. A. Wadkar	Member	
Mr. V. M. Desai	Member	

  
**Dr. S. S. Patil**  
**Head**  
**Dept. of Chemistry**  
**P. R. College, Tasgaon**

  
**Dr. M. S. Hujare**  
**Principal**  
**Padmabhushan Dr. Vasantodada Patil**  
**Mahavidyalaya, Tasgaon. (Sangli)**

Swami Vivekanand Shikshan Sanshtha, Kolhapur

Padmabhushan Dr. Vasanthaodada Patil

Mahavidhyalaya, Tasgaon

**PG DEPARTMENT OF CHEMISTRY**

**CERTIFICATE COURSE IN ANALYTICAL TECHNIQUES**

**Academic year 2019-20**


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**NOTICE**

15 JULY, 2019

All the students of B. Sc. III (Chemistry) are hereby informed that, our department is going to conduct Certificate course in **Analytical Techniques** from 1<sup>st</sup> August, 2019. Students should take admissions upto 25 July 2019.

  
**Dr. S. D. Jadhav**  
Course Coordinator

  
**Dr. S. S. Patil**  
Head  
Dept. of Chemistry  
P.D.V.P. College, Tasgaon



## **Aims and Objectives**

In recent times, analytical chemistry has attracted interest in Pharma and Biotech Industry owing to strengthen the stringent regulatory guidelines. In view of this, the present course has been effectively designed to provide skilled manpower to Industry and Academia. The course will cover the aspects such as Introduction to Analytical Techniques, Basic Principles of Chromatography, Spectrometry and Spectroscopy, Analytical Method Development and Validation, Quality Assurance and Quality Control aspects relevant to Industry, Introduction to regulatory aspects related to analytical techniques, Basic operation of instruments such as HPLC, FT-IR, GC-MS, LC-MS and ICP etc., Practical training on analytical method development and validation as per regulatory guidelines, Visits to the nearby industry besides the interactions with industry people, Evaluation of the trained personnel through internal assessment

## **Learning Outcomes**

### **Undergraduate students upon graduation with a Bachelor degree in Sciences:**

1. Students will have a firm foundation in the fundamentals and application of recent scientific theories.
2. Students are able to use modern instrumentation and classical techniques, to design experiments, and to properly record the results of their experiment.
3. Students will be skilled in critical thinking, analytical reasoning and problems solving.
4. Students will be able to identify and solve chemical problems
5. Students will be able to explore new areas of research.
6. Students will able to use modern library searching and retrieval methods to obtain information about an issue relating to chemical techniques.
7. Students will have to knows the regulations for safe handling and use of chemicals
8. Students will be able to communicate the results of their work to society.
9. Students will have to understand the ethical, historic, philosophical, and environmental dimensions of problems and issues facing chemists.
10. Students will be able to explain in proper manner, why chemistry is an integral activity for addressing social, economic, and environmental problems.

11. Students will have to find gainful employment in industry or government, be accepted at graduate or professional schools, or find employment in school systems as instructors or administrators.
12. Students will appreciate the central role of chemistry in society and use knowledge as a basis for ethical behavior in issues facing chemists including an understanding of safe handling of chemicals, environmental issues.

1. Duration of course: One Semester
2. Eligibility to course: Students studying in B.Sc.
3. Admission: On the basis of merit
4. Maximum No. of students: 25

**Evaluation System:**

All the students will be continuously evaluated by,

a. Attendance	10M
b. Assignments	10 M
c. Class tests	20 M
d. Seminar	10 M
d. One final examination (Two papers)	100 M

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**150 M**

**Practical Course**

a. Final examination	
i. Instrumentation	20 M
ii Non-nstrumentation	20 M
iii. Oral	05 M
iv. Laboratory Journal	05 M

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**50 M**



### Nature of Theory question paper

1. Q-1 is of short answer type (one or two sentences) and compulsory, containing 10 sub-questions. (10 marks)
2. From Q-2 to Q-6 (to be divided into sub questions A, B, C &/or D) **any four** questions to be solved (40 marks)

**Grades:** A grade= above 75, B grade = above 60, C grade = above 50

**Certification:** A certificate will be issued on successful completion of the course.

## Paper- I [Total periods: 15]

### Fundamentals in Analytical Chemistry and Separation Techniques

#### Unit I:

4 L

Safety in the laboratory - General safety considerations, Fire, Explosions, Poisoning – solid, liquid, gas, Cuts, Burns, Eye accidents, Electrical safety.

#### Unit II

4 L

Basic Techniques – Preparing substances for analysis, Weighing the sample, Dissolving the sample, Decomposing organic compounds, Precipitation, Filtration, Filter papers, Crucibles with permanent porous plates, Washing precipitates, Drying and igniting precipitates.

#### Unit III

4 L

- a) Sampling of solids, liquids and gases : Definition, types of samples, sampling plan, quality of sample, sub-sampling, sample registration and storage, acceptance sampling, etc.
- b) Instrumental methods of analysis, their classification and advantages of instrumental methods, the limitations, sensitivity and detection limits, precision and accuracy, calibration of glassware.
- c) Preparation of indicators, reagents and standard solutions.

#### Unit IV

3 L

Applications of following techniques of analysis:

pH metery, colorimetry, Spectrofluorimetry, Analysis of Biological samples, Redoximetric methods, Amperometry.

Chromatographic methods: Liquid chromatography, Partition, Ion exchange, Paper, Thin layer, Column, Gel chromatography, GC, GC-MS, HPLC, HPLC-MS & HPTLC.

**Reference books:**

- 1) Analytical Chemistry ( J.W ) G. D. Christian
- 2) Introduction to Chromatography: Bobbit
- 3) Instrumental Methods of Analysis (CBS) H. H. Willard; L.L Merit; J. A. Dean & F. A. Settle
- 4) Instrumental Methods of Analysis: Chatwal and Anand
- 5) Instrumental Methods of Inorganic Analysis (ELBS): A.I. Vpge;
- 6) Chemical Instrumentation, A Systematic approach; H. S. Strobel
- 7) Physical Chemistry ; P. W. Atkins.
- 8) Principles of Instrumental Analysis- D. Skoog and D. West
- 9) Treatise on Analytical Chemistry; Vol. I to VII I, .M. Kolthoff.
- 10) Computer' Fundamentals: P. K. Sinha.
- 11) Programming in BASICS: Balaguruswamy.
- 12) Computer Programming made simple: I Maynard.
- 13) Vogel's textbook of Quantitative Chemical Analysis – Sivasankar.
- 14) Laboratory techniques in organic chemistry – Aggarwal.



## Paper- II [Total periods: 15]

### Instrumental Methods in Analytical Chemistry

#### Unit I

4 L

Introduction to Spectroscopic techniques in chemical analysis- UV, IR, NMR and MS spectroscopic analysis

#### Unit II

4 L

Introduction to AAS, Flame emission, fluorescence, XRD, Nephelometry, and related techniques

#### Unit III

4 L

Electro analytical techniques A)Polarography, Cyclic voltametry:

Theory, Apparatus; derivative, polarography, modified polarographic techniques, and their application in qualitative and quantitative analysis.

#### Unit IV

3 L

Conductometry: Defination of conductometric titration, measurement of conductance by Wheatstone Bridge, direct reading Wheatstone Bridge, general procedure of conductometric titration, types of conductometric titration, cell constant, types of cell constant.

#### References Books:

1. Introduction to Instrumental Analysis; R. D. Braun,
2. Instrumental Methods of Analysis; Willard, Merritt, Dean and Settle
3. Standard Methods of Chemical Analysis Vol.3, Part A & B F. J. Welcher:
4. Instrumental Methods of Analysis, 4<sup>th</sup> & 5<sup>th</sup> editions; G. W. Ewing,.
5. Instrumental Methods of Analysis; Chatwal and Anand.
6. Electroanalytical Chemistry, Ed. H. W. Nurnberg.
7. A Textbookk of Electrochemistry, Kortum and Bockris,
8. Principles of Electrochemistry; D. A. Maclines,



9. Analytical Chemistry – G. D. Christain
10. Introduction to Chromatography, Bobbit.
11. Instrumental Methods of Inorganic Analysis (ELBS): A.I. Vogel
12. Chemical Instrumentation: A Systematic Approach, H.A. Strobel
13. The Principles of ion-selective electrodes membrane transport, W.E. Morf.
14. Physical Chemistry, P. W. Atkins.
15. Principles of Instrumental Analysis, D. Skoog & D. West
16. Principles of Physical Chemistry, Puri and Sharma.

## Practical Course

### **Part – I: Non-Instrumentation (Any ten)**

1. Estimation of nickel from chocolate sample
2. Fertilizer analysis
3. To determine the percentage of nitrogen in fertilizer by Kjeldhal method.
4. Study of adsorption of acetic acid on charcoal.
5. Determination of isoniazide from pharmaceutical tablet.
6. Determination of iron from iron tablet samples.
7. To determine the acid value of given oil.
8. Determination of calcium from given drug sample.
9. Determination of hardness, alkalinity and salinity of water sample.
10. Thin layer chromatography (TLC).
11. Column chromatography.
12. Soil analysis: Environmental samples BOD, COD, DO etc.
13. Determination of caffeine from tea powder.

### **Part – II: Instrumentation (Any five)**

#### **Potentiometry:**

1. Determination of binary mixture of weak and strong acid.
2. To determine the iron potentiometrically by titrating with potassium dichromate

#### **Colorimetry:**

1. To verify the Beer-Lamberts Law and determine the concentration of given organic dye solution colorimetrically.
2. Verify Beer-Lamberts Law by Colorimetric method.

**Conductometry:**

1. Determination of mixture of acids and relative strength of weak acids.
2. Determination of solubility of lead sulphate.

**Spectrophotometer:**

1. To verify Beer-Lambert's Law for potassium permanganate solution and hence to determine the molar extinction coefficient and unknown concentration of given sample Spectrophotometrically

**Refractometry:**

1. Determination of concentration of sugar in unknown sample.
2. Determination of composition of mixture of liquids

**Polarimetry:**

1. Kinetics of inversion of cane sugar in presence of strong acid.

**pH- metry:**

1. Determination of dissociation constant of dibasic acid.
2. Determination of pH of different soil samples by using pH metry

**Reference books:**

- 1) Text book of Quantitative Inorganic Analysis by A. I. Vogel.
- 2) A Text book of Practical Organic Chemistry by A. I. Vogel



















Certificate course in Analytical Techniques (Batch I) (P)  
2019-20

Sr. No.	Roll No.	Name of the student	20/09/19 01:30	26/09/19 01:30	27/09/19 01:30	03/10/19 01:30	04/10/19 01:30
1	6041	Patil Shweta Balasa	Patil Shweta Patil	Patil Shweta Patil	Patil Shweta Patil	Patil Shweta Patil	Patil Shweta Patil
2	6044	Patil Supriya Vasant	Patil Supriya Patil	Patil Supriya Patil	Patil Supriya Patil	Patil Supriya Patil	Patil Supriya Patil
3	6203	Patil Rutuja Rajendra	Patil Rutuja Patil	Patil Rutuja Patil	Patil Rutuja Patil	Patil Rutuja Patil	Patil Rutuja Patil
4	6005	Peshmuth. Parva. S.	Peshmuth. Parva. S.	Peshmuth. Parva. S.	Peshmuth. Parva. S.	Peshmuth. Parva. S.	Peshmuth. Parva. S.
5	6048	Patil. Sneha Nilas.	Patil. Sneha Patil	Patil. Sneha Patil	Patil. Sneha Patil	Patil. Sneha Patil	Patil. Sneha Patil
6	6042	Patil Shweta (Patil)	Patil Shweta Patil	Patil Shweta Patil	Patil Shweta Patil	Patil Shweta Patil	Patil Shweta Patil
7	6019	Mahajan Bhagyashree. K.	Mahajan Bhagyashree. K.	Mahajan Bhagyashree. K.	Mahajan Bhagyashree. K.	Mahajan Bhagyashree. K.	Mahajan Bhagyashree. K.
8	6210	Patil Yashashri D. P.	Patil Yashashri Patil	Patil Yashashri Patil	Patil Yashashri Patil	Patil Yashashri Patil	Patil Yashashri Patil
9	6209	Patil Vaibhavi Sachin	Patil Vaibhavi Patil	Patil Vaibhavi Patil	Patil Vaibhavi Patil	Patil Vaibhavi Patil	Patil Vaibhavi Patil
10	6057	Tejashree Nandev. Sanjay	Tejashree Nandev. Sanjay	Tejashree Nandev. Sanjay	Tejashree Nandev. Sanjay	Tejashree Nandev. Sanjay	Tejashree Nandev. Sanjay
11	6051	Pilgade Mahin. S.	Pilgade Mahin. S.	Pilgade Mahin. S.	Pilgade Mahin. S.	Pilgade Mahin. S.	Pilgade Mahin. S.
12	6055	Salunkhe Anurata. B.	Salunkhe Anurata. B.	Salunkhe Anurata. B.	Salunkhe Anurata. B.	Salunkhe Anurata. B.	Salunkhe Anurata. B.
13	6034	Patil Bhagyashree Hanuman	Patil Bhagyashree Patil	Patil Bhagyashree Patil	Patil Bhagyashree Patil	Patil Bhagyashree Patil	Patil Bhagyashree Patil
14	6179	Patil Abhishek S.	Patil Abhishek Patil	Patil Abhishek Patil	Patil Abhishek Patil	Patil Abhishek Patil	Patil Abhishek Patil
15	6175	Nalawade Shubham B.	Nalawade Shubham Nalawade	Nalawade Shubham Nalawade	Nalawade Shubham Nalawade	Nalawade Shubham Nalawade	Nalawade Shubham Nalawade
16	6151	Jadhav Rahul Pandurang	Jadhav Rahul Jadhav	Jadhav Rahul Jadhav	Jadhav Rahul Jadhav	Jadhav Rahul Jadhav	Jadhav Rahul Jadhav
17	6194	Disale Pallavi. J.	Disale Pallavi Disale	Disale Pallavi Disale	Disale Pallavi Disale	Disale Pallavi Disale	Disale Pallavi Disale
18	6012	Karade Sonali. Dadasa	Karade Sonali Karade	Karade Sonali Karade	Karade Sonali Karade	Karade Sonali Karade	Karade Sonali Karade
19	6048	Patil Vaishali Vitthal	Patil Vaishali Patil	Patil Vaishali Patil	Patil Vaishali Patil	Patil Vaishali Patil	Patil Vaishali Patil
20	6066	Shinde Rushikesh Bhanu	Shinde Rushikesh Shinde	Shinde Rushikesh Shinde	Shinde Rushikesh Shinde	Shinde Rushikesh Shinde	Shinde Rushikesh Shinde
21	6052	Rajmane Vitaya Danda	Rajmane Vitaya Rajmane	Rajmane Vitaya Rajmane	Rajmane Vitaya Rajmane	Rajmane Vitaya Rajmane	Rajmane Vitaya Rajmane
22	6218	Shinde Aishwarya. M.	Shinde Aishwarya Shinde	Shinde Aishwarya Shinde	Shinde Aishwarya Shinde	Shinde Aishwarya Shinde	Shinde Aishwarya Shinde
23	6056	Salunkhe Samiksha Prakash	Salunkhe Samiksha Salunkhe	Salunkhe Samiksha Salunkhe	Salunkhe Samiksha Salunkhe	Salunkhe Samiksha Salunkhe	Salunkhe Samiksha Salunkhe
24	6069	Pranita Tasgaonkar	Pranita Tasgaonkar Pranita	Pranita Tasgaonkar Pranita	Pranita Tasgaonkar Pranita	Pranita Tasgaonkar Pranita	Pranita Tasgaonkar Pranita
25	6017	Lathande P. J.	Lathande P. J. Lathande	Lathande P. J. Lathande	Lathande P. J. Lathande	Lathande P. J. Lathande	Lathande P. J. Lathande



















Certificate course in Analytical Techniques (Batch II) (P)  
 2019-20

Sr. No.	Roll No.	Name of the student	09/01/20 01:30	10/01/20 01:30	23/01/20 01:30	24/01/20 01:30	30/01/20 01:30	31/01/20 01:30
1	6174	Nalawade Rutuja Ravasa	RV	RV	RV	RV	RV	
2	6212	Pawar Pooja Panderang	Pawar	Pawar	Pawar	Pawar	Pawar	
3	6026	Patil Manisha Sunil	Ab	Patil	Patil	Patil	Patil	
4	6221	Shinde Amrita Sambhaji	Shinde	Shinde	Shinde	Shinde	Shinde	
5	6016	Kumbhare Kausar P.	Pawar	Kumbhare	Kumbhare	Kumbhare	Kumbhare	
6	6217	Salunkhe Pratiksha Shripad	Ab	Salunkhe	Salunkhe	Salunkhe	Ab	
7	6001	Amgutsagar Pranali Babasaheb	Amgutsagar	Pranali	Pranali	Pranali	Pranali	
8	6213	Pawar Snehal Prakash	Pawar	Pawar	Pawar	Pawar	Pawar	
9	6220	Shinde Shital Shivaji	Shinde	Ab	Shinde	Shinde	Shinde	
10	6193	Patil Minakal Prakash	Patil	Patil	Patil	Ab	Patil	
11	6222	Juyavarkar K. R.	RV	RV	RV	RV	RV	
12	6194	Patil Nilesh	Patil	Patil	Patil	Patil	Patil	
13	6197	Patil Prasad P.	Patil	Patil	Patil	Patil	Patil	
14	6152	Jadhav Ranpreet	Jadhav	Jadhav	Jadhav	Jadhav	Jadhav	
15	6004	Chavan Vaibhav E.	Chavan	Chavan	Chavan	Chavan	Chavan	
16	6204	Patil Saurav Vilas	Su. Patil	Su. Patil	Su. Patil	Su. Patil	Su. Patil	
17	6157	Fadnis Sachin M.	Fadnis	Fadnis	Fadnis	Fadnis	Fadnis	
18	6042	Patil Vaibhav V.	Ab	Patil	Patil	Patil	Patil	
19	6018	Madane Vikas S.	Madane	Madane	Madane	Madane	Madane	
20	6023	Motale Rahul B.	Motale	Motale	Motale	Motale	Motale	
21	6205	Shubham Patil	Shubham	Shubham	Shubham	Shubham	Shubham	
22	6211	Pawar Akash P.	Pawar	Pawar	Pawar	Pawar	Pawar	
23	6049	Pawar Gauri R.	Pawar	Pawar	Pawar	Pawar	Pawar	
24	6142	Jadhav Amit. A.	Jadhav	Jadhav	Jadhav	Jadhav	Jadhav	
25	6154	Jadhav Sushant Sunil	Jadhav	Jadhav	Jadhav	Jadhav	Jadhav	



Swami Vivekanand Shikshan Sanshtha, Kolhapur

Padmabhushan Dr. Vasantodada Patil

Mahavidhyalaya, Tasgaon

**PG DEPARTMENT OF CHEMISTRY**

**CERTIFICATE COURSE IN ANALYTICAL TECHNIQUES**

**Academic year 2019-20**


Date : 05 March 2020

### Notice

All the students are hereby informed that the examination in Certificate Course in **Analytical Techniques** will be held as per following schedule. Students should present on time.

Date	Time	Paper No.
11 March 2020	11.00 am to 12.30 pm	I
12 March 2020	11.00 am to 12.30 pm	II
13 March 2020	11.00 am to 5.00 pm	Practical

  
Dr. S. D. Jadhav  
Course Coordinator

  
Dr. S. S. Patil  
Head  
Dept. of Chemistry  
P.D.V.P. College, Tasgaon

## Department of Chemistry

### Certificate Course in Analytical Techniques 2019-20

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#### Paper I

#### Fundamentals in Analytical Chemistry and Separation Techniques

Marks - 50

Time :

Date :

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**Q1. Answer the following question in one sentence.**

10M

- i) What is the basic principle of gravimetric analysis?
- ii) What is Accuracy?
- iii) What is the basic units of weights?
- iv) Define normal solution.
- v) What is precipitation?
- vi) When you heat the substance in the test tube ,where should it point
- vii) What is the principle of pH?
- viii) Why digestion important in gravimetric analysis?
- ix) What is the equivalent weight?
- x) What is the significant figure?

**Q2. Attempt any four.**

40M

- i) Discuss the different steps involved in gravimetric analysis .Explain yhe mechanism of precipitation.
- ii) What are the the filters in colorimetry? Explain its types.
- iii) Discuss the classification and advantages of instrumental methods.
- iv) What is digestion? How is it useful in the purification of precipitation?
- v) Explain ,how particle size is amatter of impotance in sampling of solid. Describe methods to reduce the sampling size.



"Dissemination of Education for Knowledge, Science and Culture" -Shikshanmaharashi Dr. Bapuji Salunkhe  
Shri Swami Vivekanand Shikshan Sanstha Kolhapur  
Padmabhushan Dr. Vasantodada Patil Mahavidyalaya, Tasgaon  
**Department of Chemistry**  
**Certificate Course in Analytical Techniques 2019-20**

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**Paper II**  
**Instrumental Methods in Analytical Chemistry**

**Date and Time:**

**50Mark**

Q1. Answer the following question in one sentence.

10 Mark

- 1) What is Nephelometry test.
- 2) Define the term conductivity .
- 3) Write down the formula of cell constant .
- 4) What is the principle of flame emission spectroscopy.
- 5) What is mean by finger print region.
- 6) What is chemical shift.
- 7) Define the term electromagnetic radiation.
- 8) State Hooke's law.
- 9) What is bathochromic shift .
- 10) What is coupling constant.

Q2. Attempt any four question.

40 Mark

- 1) What is selection rule? Explain IR active & inactive transitions the suitable example.
- 2) What are the fragment ions? How are they formed.
- 3) What is conductivity cell? Why the electrodes of conductivity cell are blackened with platinum black.
- 4) What is spectroscopy? Name the spectroscopic techniques that use radiowave ultraviolet visible & IR radiations.
- 5) What is XRD & how it works.



Dissedimentation of education through Knowledge, Science and Wisdom-Shikshanmaharshi Dr. Bapuji Salunkhe  
Swami Vivekanand Shikshan Sanshtha, Kolhapur

**Padmabhushan Dr. Vasandraodada Patil Mahavidhyalaya, Tasgaon**  
**PG DEPARTMENT OF CHEMISTRY**

**Certificate Course in Analytical Techniques 2019-20**  
**Mark Sheet**

**Batch A (01 August 2019 to 31 October 2019)**

Sr. No.	Name of Candidate	Attendance (10)	Assignment (10)	Class tests (20)	Seminar (10)	Paper I (50)	Paper II (50)	Practical (50)	Total (200)
1	Ms. Patil Sweta Balaso	9	10	16	7	46	45	47	180
2	Ms. Patil Supriya Vasant	9	10	15	9	40	41	46	170
3	Ms. Patil Rutuja Rajendra	9	10	16	7	42	40	46	170
4	Ms. Deshmukh Purva Surendra	9	9	17	6	45	42	45	173
5	Ms. Patil Sneha Vilas	9	10	19	7	44	40	42	171
6	Ms. Patil Sweta Popat	9	10	19	9	44	49	45	183
7	Ms. Mahajan Bhagyashree K.	9	9	15	7	46	42	45	173
8	Ms. Patil Yashshree Dilip	9	9	16	7	42	40	44	166
9	Ms. Patil Vaibhavi Sachin	9	10	12	6	40	44	42	163
10	Mr. Pirjade Mahin Shafik	9	9	14	9	40	42	45	168
11	Ms. Salunkhe Amruta Bajarang	9	9	15	9	41	44	42	169
12	Ms. Sankpal Tejshree Namdev	9	9	17	6	40	42	45	168
13	Ms. Patil Bhagyashree Hanamant	9	10	16	7	44	40	42	152
14	Ms. Disale Pallavi J.	9	10	12	7	40	42	41	161



15	Ms. Karade Sonali Dadaso	9	10	15	7	44	40	42	167
16	Mr. Patil Abhishek Shashikant	8	9	14	7	40	41	41	160
17	Mr. Nalawade Shubham B.	9	10	12	9	44	42	40	166
18	Mr. Jadhav Rahul P.	9	9	11	7	40	41	41	158
19	Ms. Patil Vaishali Uttam	9	9	14	7	40	42	44	165
20	Mr. Rushikesh Bhanu	9	10	16	8	42	42	45	172
21	Ms. Rajmane Vijaya Sanjay	8	10	14	7	44	40	42	165
22	Ms. Shinde Aishwarya M.	9	10	15	7	40	42	44	167
23	Ms. Salunkhe Sarika Prakash	8	10	16	9	41	40	42	168
24	Ms. Tasgaonkar Pranita Dayanand	9	10	14	7	40	40	45	165
25	Ms. Lokhande P. J.	9	10	15	7	41	42	44	168

*Jadhav*

**Dr. S. D. Jadhav**  
Course Coordinator

*S. S. Patil*

**Dr. S. S. Patil**  
Head  
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Dissedimentation of education through Knowledge, Science and Wisdom-Shikshanmaharshi Dr. Bapuji Salunkhe  
Swami Vivekanand Shikshan Sanshtha, Kolhapur

## Padmabhushan Dr. Vasantraodada Patil Mahavidhyalaya, Tasgaon PG DEPARTMENT OF CHEMISTRY

### Certificate Course in Analytical Techniques 2019-20 Mark Sheet

Batch B (01 December 2019 to 29 February 2020)

Sr. No.	Name of Candidate	Attendance (10)	Assignment (10)	Class tests (20)	Seminar (10)	Paper I (50)	Paper II (50)	Practical (50)	Total (200)
1	Ms. Nalawade Rutuja Ravso	9	9	14	7	40	41	44	164
2	Ms. Pawar Pooja Pandurang	9	10	12	9	42	41	42	165
3	Ms. Patil Manisha Sanjay	9	9	14	7	43	44	40	166
4	Ms. Shinde Smita Sambhaji	9	10	15	7	41	42	45	169
5	Ms. Kumbhar Karuna Popat	9	10	16	7	44	40	42	168
6	Ms. Salunkhe Prajкта Shamkumar	8	10	17	8	40	41	44	168
7	Ms. Amrutsagar Pranali B.	9	10	15	7	42	46	42	171
8	Ms. Pawar Snehal Prakash	9	10	17	7	41	40	45	169
9	Ms. Shinde Shital Shivaji	9	9	19	9	40	42	42	170
10	Ms. Patil Minal Popat	9	9	15	7	42	42	41	165
11	Mr. Suryawanshi Kunal K.	8	9	14	9	44	40	42	166
12	Mr. Patil Nilesh T.	9	10	16	7	40	41	41	164
13	Mr. Patil Prasad Pandurang	9	10	15	7	44	40	40	165



14	Mr. Jadhav Ranjeet R.	8	10	14	7	40	44	42	165
15	Mr. Chavan Vaibhav R.	9	10	15	7	40	41	42	165
16	Mr. Patil Sourav Vilas	8	10	16	8	41	42	42	167
17	Mr. Kadam Satish M.	9	9	12	7	39	40	41	<del>157</del>
18	Mr. Patil Vaibhav Vijay	9	9	14	7	40	41	41	161
19	Mr. Madane Vikas S.	8	10	15	9	41	42	42	167
20	Mr. Mokashi Rahul B.	9	9	16	7	44	40	40	165
21	Mr. Patil Shubham S.	9	9	12	7	35	39	41	152
22	Mr. Pawar Akash P.	8	9	13	7	40	39	40	156
23	Ms. Pawar Gouri R.	9	10	14	8	40	37	42	160
24	Mr. Jadhav Sushant Sharad	9	10	16	9	41	42	42	169
25	Mr. Jadhav Amit Anil	8	9	15	7	40	41	41	161

*Jadhav*

**Dr. S. D. Jadhav**  
Course Coordinator

*Patil*

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Padmabhushan Dr. Vasandraodada Patil Mahavidyalaya,  
Tasgaon, Sangli, (MS), India-416312



## Department of Chemistry

### Certificate Course in Analytical Techniques

This is to certify that Ms. Mahajan Bhagyashree K. has  
completed successfully 'Certificate Course in Analytical Techniques' conducted  
by Department of Chemistry, Padmabhushan Dr. Vasandraodada Patil  
Mahavidyalaya, Tasgaon in academic year 2019-20.

  
**Dr. Swati D. Jadhav**

Organizing Secretary

  
**Prof. (Dr.) Suresh S. Patil**

Head of the Department

  
**Dr. Milind S. Hujare**

Principal





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This is to certify that Ms. Disale Pallavi J. has  
completed successfully 'Certificate Course in Analytical Techniques' conducted  
by Department of Chemistry, Padmabhushan Dr. Vasandraodada Patil  
Mahavidyalaya, Tasgaon in academic year 2019-20.

**Dr. Swati D. Jadhav**  
Organizing Secretary

**Prof. (Dr.) Suresh S. Patil**  
Head of the Department

**Dr. Milind S. Hujare**  
Principal





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## Department of Chemistry

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Mahavidyalaya, Tasgaon in academic year 2019-20.

  
Dr. Swati D. Jadhav

Organizing Secretary

  
Prof. (Dr.) Suresh S. Patil

Head of the Department

  
Dr. Milind S. Hujare

Principal